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Evaluating and Categorizing the Reliability of Distribution Coefficient Values in the Sorption Database(2)

Yoshihiko SAITO, Michael OCHS* Susanne KUNZE*, Akira KITAMURA Yukio TACHI and Mikazu YUI

Radionuclide Migration Research Group Geological Isolation Research and Development Directorate

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〒319-1195 茨城県那珂郡東海村白方白根2番地4 日本原子力研究開発機構 研究技術情報部 研究技術情報課 電話 029-282-6387, Fax 029-282-5920

*〒319-1195 茨城県那珂郡東海村白方白根2番地4 日本原子力研究開発機構内

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Yoshihiko SAITO^{**}, Michael OCHS^{*}, Susanne KUNZE^{*}, Akira KITAMURA, Yukio TACHI and Mikazu YUI

> Geological Isolation Research Unit Geological Isolation Research and Development Directorate Japan Atomic Energy Agency Tokai-mura, Naka-gun, Ibaraki-ken

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Japan Atomic Energy Agency (JAEA) has been developing a number of databases that formed an important basis for the H12 performance assessment (PA) of high-level radioactive waste disposal in Japan. These databases include extensive compilations of sorption K_d data (JNC-SDB). JAEA has been and is continuing to improve and update the JNC-SDB in view of potential future data needs. This on-going program has focused on testing the usefulness of the existing databases for possible applications to K_d -setting and keeping the databases up-to-date and assuring the desired quality level.

The JNC-SDB includes more than 24,000 K_d data which have a great variety of K_d and related experimental information from many different literatures. Accordingly, the quality assuring (QA) and classifying criteria for K_d has been developed in order to evaluate the reliability of each K_d value. The reliability of K_d values for bentonite system has been already evaluated.

In this report, the QA/classification of selected entries in the JNC-SDB, especially of K_d values for mudstone systems to use in the K_d -derivation exercise for Horonobe rocks, was done following the approach defined in our previous report. As a result, the reliability of 1,056 K_d values was evaluated and classified with numerical rating. This classification scheme is expected to make it possible to obtain quick overview of the available data from the SDB, and to have suitable access to the respective data for K_d -setting in PA.

Keywords: Sorption Database, Distribution Coefficient, Kd, Classification, Reliability, HLW Disposal

^{*}Collaborating Engineer

^{*} BMG ENGINEERING LTD, Switzerland

収着データベースに登録された収着分配係数の信頼度評価 (2)

日本原子力研究開発機構

地層処分研究開発部門

地層処分基盤研究開発ユニット

齋藤 好彦^{**}, Michael OCHS^{*}, Susanne KUNZE^{*}, 北村 暁, 舘 幸男, 油井 三和

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日本原子力研究開発機構(JAEA)では、高レベル放射性廃棄物地層処分研究開発の第2次取り まとめにおいて、その重要な基礎と位置づけられる幾つかのデータベースを整備してきた。これ らのデータベースの一つが収着分配係数 K_dに関するデータベース(JNC-SDB)である。JAEAで は、将来の性能評価におけるニーズへの対応を念頭に、JNC-SDBの改良・更新を継続的に実施し てきている。この開発計画の一環として、現在、実際の地質環境に対する K_d設定におけるデータ ベース適用の有効性検討、データベースに含まれる K_dデータの信頼度評価を進めている。

JNC-SDB に登録されている K_dは 24,000 データを超え,膨大な文献データから収集された様々 な条件での幅広い K_dデータと関連する実験条件等の情報が含まれている。このため,各々の K_d データの信頼性を評価するため,K_dの品質確認とクラス分けの手法の開発を進めてきた。また, この手法に基づき,ベントナイト系の K_dデータを対象に信頼性評価を行った結果を既に報告した。

本報告書においては、 JNC-SDB に含まれる K_d データのうち,特に幌延岩石を対象にした K_d 設定検討への活用を念頭に、泥岩系の K_d データを主な対象として、前報告書にて設定した手法に 従って信頼性評価を行った。その結果、1,056 の K_d データに対して、新たな信頼度情報が付与さ れた。この信頼度評価手法は、収着データベースから利用可能な関連データ集を速やかに抽出し、 K_d データ設定の際に参照すべきデータを適切に選定する上で、有効な手法となると考えられる。

核燃料サイクル工学研究所(駐在):〒319-1194 茨城県那珂郡東海村村松 4-33 ※技術開発協力員

^{*} BMG ENGINEERING LTD, スイス連邦共和国

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1. Introduction

The Japan Atomic Energy Agency (JAEA) has been developing a number of databases that formed an important basis for the H12 performance assessment (PA) of high-level nuclear waste disposal in Japan (JNC, 2000). These databases include extensive compilations of sorption (JNC-SDB) (Shibutani et al., 1999). JAEA has been and is continuing to improve and update its PA-related databases, in particular the databases on sorption of radionuclides in view of potential future data needs (Suyama et al., 2004, Saito et al., 2007).

As part of this on-going development program, JAEA has focused on

- testing the usefulness of the existing databases to K_d-setting for possible PA-related applications;
- keeping the databases up-to-date and assuring the desired quality level.

As part of this on-going development program, a project to classify the entries in the JNC-SDB according to defined QA-criteria has been started. The JNC-SDB includes more than 24,000 K_d data which have a great variety of K_d and related experimental information from many different literatures. Accordingly, the quality assuring (QA) and classifying criteria for K_d has been developed in order to evaluate the reliability of each K_d value (Saito et al., 2005). The reliability of K_d values for bentonite system has been already evaluated (Ochs et al., 2007).

In this report, the QA/classification of selected entries in the JNC-SDB, especially of K_d values for mudstone systems to use in the K_d -derivation exercise for Horonobe rocks (Ochs et al., 2008), was done following the approach defined in our previous report. The K_d values for the sorption of Th, Np, Se, Cs, Pu and Am on mudstone, bentonito and related minerals.

In the chapter 2, the QA criteria and classification scheme defined in our previous report (Ochs et al., 2007) are described. And in the chapter 3, classifications of selected entries for mudstone and related minerals in the JNC-SDB are described by each item. In the appendix, the details of the evaluation and classifications for the selected entries of the SDB are listed.

2. Reliability of JNC-SDB: Classification Guideline

2.1 Introduction, description of main criteria

The reliability of K_d values in the JNC-SDB can be assessed using the following three main criteria. The three main criteria are listed in the expected sequence of application during a classification of entries in the JNC-SDB. Criteria Ia and Ib are related to documentation and data entry, whereas the technical and scientific quality of an entry is addressed by criteria II and III.

I Completeness of documentation and type of K_d information:

- a) It needs to be verified that the documentation of each entry is detailed enough to allow further examination according to the main criteria II-III. At this point, only the completeness of the documentation is examined; the appropriateness of the reported data and approaches is evaluated under criterion II below.
- b) This point takes also into account that the reliability of data input to the JNC-SDB will be substantially high if K_d values are directly available in table format in comparison to literature that reports e.g. %-adsorbed values in a graph. The latter way of reporting requires the operator to i) manually read values off a graph and ii) to calculate K_d from the %-adsorbed and Solid/water ratio (s/w) ratio values given, which significantly increases the likelihood of an operator error during data input.

II Quality of reported data

This is the most important issue from a technical and scientific point of view. This criterion encompasses an evaluation of the appropriateness of the experimental system to produce reliable K_d data. The methods used (or lacking) for determining experimental uncertainty are also examined for each literature source. Further, it is considered whether the data represent single-point measurements or are part of e.g. an isotherm, which would provide additional support for their reliability.

III Consistency of data:

While the previous two main criteria address the reliability of each K_d entry in the JNC-SDB, criterion No. III requires an examination of the level of support that other K_d values in similar systems can lend to the entry under consideration. Any disagreement with data from related systems will have to be evaluated as well. It could be argued that this kind of data examination may be left to the user of the JNC-SDB. However, the classification of data entries in the JNC-SDB in terms of reliability adds an aspect of quality that is above that for a pure compilation, and users may expect that the listed K_d

values passed some kind of check for internal consistency.

Internal consistency means that data from different sources should not be in obvious disagreement. An example would be the dependency on pH of K_d values for a certain radionuclide, which should be approximately similar in all studies. Similarly, if many studies indicate e.g. stronger sorption of U(IV) than of Th(IV), for any study that indicates the opposite an appropriate explanation should be given. If no good reason can be found, such deviations make a study less reliable. These types of considerations will only be possible for sufficiently well researched elements.

2.2 Description of checkpoints within each main criterion

2.2.1 General

Each entry in the JNC-SDB (each K_d value identified in the JNC-SDB by a unique ID) should be evaluated and classified individually. Because many studies report K_d values under different experimental conditions, it is not sufficient to evaluate all data based on a given reference globally. Depending on conditions, different entries related to a given study may receive a different rating.

2.2.2 Criterion I: Completeness of documentation and type of K_d information

The checkpoints under I-a are used for a screening prior to a further classification. Failure to satisfy these checkpoints will not to be used (unreliable).

- **I-a.1** Are all mandatory fields completed? Here it is only verified that all fields have been completed by the operator; an entry "not reported" is counted, therefore. The following entries are considered mandatory:
 - element
 - solid phase
 - solution composition
 - atmosphere
 - pH (or other information that allows to derive pH, e.g. portlandite equilibrium)
 - pe/redox condition (only in case of redox-sensitive systems)
 - method of pe control (only in case of redox sensitive systems and imposed reducing conditions)
 - initial radionuclide (RN) concentration (except for RN that are not solubility controlled)

- method for phase separation
- type of experiment, if different from batch
- → In case of missing entries, the corresponding K_d is excluded from further evaluation and classified as unreliable (until remedied by operator). If all fields are completed, proceed to Ia.2.
- **I-a.2** Is all mandatory information provided? Here it is evaluated whether critical information is provided or lacking completely. The quality of the information provided is evaluated under criterion II. In addition to the information listed under Ia.1, further mandatory information includes:
 - units
 - → In case of missing mandatory information, the corresponding K_d is excluded from further evaluation and classified as unreliable. If all fields are completed, proceed to Ib.
- I-b Does the type of K_d information provided require manipulation by the operator?
 → The following levels are distinguished:

class 1:	table with K _d values given		
class 2:	table with % sorbed given		
	table with residual concentration given		
class 3:	linear graph K _d		
class 4:	linear graph % sorbed		
	linear graph residual concentration		
class 5:	logarithmic graph K _d		
class 6:	logarithmic graph % sorbed		
	logarithmic graph residual concentration		

2.2.3 Criterion II: Technical and scientific quality of reported data

It is generally assumed that the entries presently contained in the JNC-SDB correspond to a minimum quality standard; i.e. are assumed to be basically reliable. The different checkpoints regarding experimental quality are designed to distinguish different levels of reliability. However, if

in case of critical checkpoints even the requirements leading to the lowest rating are not met, the respective entry should be classified as unreliable (indicated for each checkpoint).

II-a Solid phase (substrate)

It is evaluated whether the solid phase has been sufficiently characterized. This is equally important for properly designing experiments, as well as for using the measured K_d values. In general, three types of key information are required:

- Information about major mineral composition.
- Information about accessory minerals or impurities.
- Information about surface characteristics: Minimum is a measure of sorption capacity per mass of sorbent, such as CEC or a different measure of site density per mass.

However, the amount of information required to sufficiently characterize a given solid phase also depends on the complexity of the substrate:

- It needs to be known whether a substrate consists of a single pure mineral phase, or whether it contains impurities or additional minerals. In general, some measure of site density per mass (e.g. CEC) needs to be known to properly design experiments, in particular with respect to achieving reasonable surface loading.
- 2. In case of simple substrates (pure minerals), no further information is necessary.
- 3. In case of complex substrates (i.e., where significant impurities are present, or where a substrate is composed of several minerals), and in particular in case of natural samples, detailed information on composition has to be provided in addition.
- 4. In cases where sample treatment (such as crushing or sieving) had been performed, the respective information on particle size also needs to be provided (see II-f). Where any chemical treatments (e.g. acid washing to remove calcite; but also change of redox conditions in case of redox-sensitive substrates, see II-c) had been applied, the applied method and resulting mineralogy should be given as well.
- 5. In case of many commercially available substrates (e.g., MX-80 or Kunigel-V1 bentonite; standard clay minerals from the Clay Minerals Society, such as SWy-1; Min-U-Sil SiO₂, etc.) detailed solid phase information is widely known and can be retrieved from a large number of publications. Therefore, characterization of such solids is not required for each entry in the JNC-SDB; i.e., level A or B can be reached even if such information is not reported. Note that this holds only when such solids have been used as received. Where washing procedures etc. have been applied, the procedures and resulting changes still need to be documented.

- \rightarrow Three levels of reliability:
- A) Major and minor mineralogy as well as surface characteristics are known.
 For example: The substrate is a single, well-defined mineral; or comprehensively characterized complex mineral assemblage. Either no sample treatment has been carried out, or it is described in detail and the result are documented.
- B) Major mineralogy as well as surface characteristics are known.
 For example: The substrate is a single mineral that may contain impurities (such as a non-purified clay mineral) or a complex mineral assemblage where additional impurities could be present. Sample treatment may have led to minor changes in mineralogy.
- C/D) Information on both major mineralogy or surface characteristics is lacking. For example: There is no information on CEC (or another measure of sorption capacity); or the substrate is a natural clay sample where it is not clear whether it is smectite, kaolinite, or illite; or a non-characterized soil or crushed rock. Sample treatment may have led to major changes in mineralogy that are not documented.

II-b Adjustment and control of pH

One of the most important solution parameters controlling radionuclide(RN) sorption is pH. It needs to be known to interpret K_d values, but also for proper experimentation: The pH needs to be known to evaluate the solubility limits of radionuclides and some major ions, as well as the stability of certain mineral phases (in particular carbonates). Further, pH has to be approximately constant during a sorption experiment in order to reach equilibrium of sorption reactions. There are two basically different approaches in sorption experiments with regard to pH control:

- The pH is not controlled, but allowed to reach an equilibrium value according to the experimental conditions and is then measured at the end of the experiment. In this case, it is important that the pH has been verified after experimentation, in order to know its equilibrium value.
- 2. The pH is controlled during the experiment by acid-base addition and/or buffers. Where it is desired to determine K_d values as a function of pH, this cannot be avoided. In this case, it needs to be shown (or known from the literature) that the added acids, bases, or buffers do not interfere with RN reactions at the surface (which obviously influence sorption) or with RN reactions in solution (which influence sorption through changing the RN speciation). Therefore, use of a

non-inert pH buffer at unspecified concentration levels leads to a classification as unreliable.

- \rightarrow Four levels of reliability:
- A) To achieve rating A it is sufficient, but required, that the pH is verified at the end of the experiment. This is based on the assumption that equilibrium or at least a stable state of near-equilibrium conditions has been achieved (see also II-a, II-d, and II-j). In such systems, a determination of the experimental end pH will represent an adequate measure of the actual equilibrium pH. Second, rating A is given where the successful use of inert buffers has been demonstrated (e.g. by measuring K_d in the presence and absence of buffers at some pH, or by showing through speciation calculations that the buffer does not influence RN behavior). In some cases, level A may also apply if a non-inert buffer is part of the experimental setup (see the example of K_d determination as a function of carbonate concentration under point C).
- B) The final pH is reported, but only a pH range (within 1 pH unit) is given instead of a discrete pH value (the same assumptions regarding equilibrium can be made as for level A above). Rating B also applies in cases where only the initial pH is provided, but the experimental system is well buffered (for example, because a inert buffer is used, or because of the presence of a natural buffer system, such as carbonate).
- C) Only the initial pH is provided, no attempt is made to control final pH. All cases where non-inert pH-buffers are being added. Note that this refers to the addition of an additional complexing ligand, such as acetate, for the control of pH. On the other hand, if a sorption experiment is carried out where K_d is measured as a function of carbonate concentration and this is simultaneously used to control pH, level A applies (given that the effect of carbonate on K_d is documented).
- D) Only a range (within 1 pH unit) of initial pH is provided, no information on final pH is given.
- → If a lower quality than required for level D is evident, the respective entry is excluded from further evaluation as unreliable. If a non-inert buffer (e.g. acetate or carbonate) is used at unspecified concentration levels, the respective entry is excluded from further evaluation as unreliable.

II-c Redox conditions

Here it needs to be differentiated between systems that are not redox-sensitive and

systems that are. Within the redox-sensitive systems, it needs to be further taken into account whether only the sorbing RN is redox-sensitive or whether other components of the system (such as solid phase or groundwater components) are redox-sensitive as well. In this sense, checkpoint II-c deals with the redox control of the sorbing RN, not with redox control of an overall redox-sensitive system. If the experimental system comprises a range of redox-sensitive dissolved (e.g. organics) and solid (e.g. Fe- and Mn-phases) components, imposing redox conditions different from the original level may influence many redox-equilibria simultaneously. In such a case it can be very difficult to ascertain equilibrium or to know which solid phases are present. Such effects on solution and solid phase chemistry are addressed by checkpoints II-a and II-d. It also needs to be pointed out in this context that "imposed redox condition" does not necessarily refer only to imposing reducing conditions by adding a reducing agent, it also includes imposing oxidizing conditions by e.g. transferring a reduced natural sediment to the laboratory and exposing it to O_2 (as a matter of fact, the latter may be the more common problem).

Given the focus of this checkpoint on redox control of sorbing radionuclides explained above, two different requirements on data quality can be distinguished. Levels of reliability reflect the degree to which these two requirements are met:

- 1. Reliability regarding control and confirmation of the redox status of the sorbing RN.
- 2. Reliability regarding the absence of unwanted side effects, such as changes in RN speciation induced by the addition of a reducing agent.
- \rightarrow Two levels of reliability:
- A/B) Level A/B applies to entries in the JNC-SDB where it is demonstrated that both of the above requirements are met: This includes the following cases:
 - Systems which are not redox-sensitive in terms of sorption and where no reducing agents needed to be added (i.e., where the sorbing RN can take on only one oxidation state in aqueous solutions).
 - Redox-sensitive systems that have been pre-equilibrated with and are being kept at ambient conditions.
 - Experiments where reducing conditions are imposed on redox-sensitive RN (in otherwise stable systems) and where similar results are obtained using several reducing agents.
- C/D) Level C/D applies to entries in the JNC-SDB where meeting the above requirements may not have been demonstrated, but can be assumed with high certainty. This includes the following cases:

- Reducing conditions imposed on redox-sensitive RN (in otherwise stable systems) using one reducing agent that can be estimated (e.g. from experience or from the literature) to be effective and to be sufficiently inert with respect to influencing RN behavior.
- In cases where complexing reducing agents have been used, level C/D still can be achieved if the influence of the reducing agent on RN speciation has been estimated.

• All cases where redox conditions may be less well defined than for level A/B, but where it can be assumed that no significant artifacts regarding RN behavior are introduced and where the oxidation state of RN has been measured independently (in some cases, this may include low-O₂ conditions with a subsequent confirmation of RN oxidation state). Evaluating the reliability of such measurements is likely to require an expert decision by the operator.

- → If a lower quality than required for level C/D is evident, the respective entry is excluded from further evaluation as unreliable. For example, cases where it has been attempted to achieve reducing conditions only by minimizing the level of O₂ (e.g., by performing experiments in a N₂ atmosphere) generally should be labeled "unreliable" (except where the oxidation state of a RN somehow has been confirmed, see description of level C/D). Also, if a strongly complexing reducing agent (such as many organic acids) is used at unspecified concentration levels, the respective entry is excluded from further evaluation as unreliable.
- **II-d** Final solution composition

Note that solution composition includes dissolved carbonate concentration, which may be controlled through, or expressed as pCO₂. Added pH-buffers or reducing agents are also included, and are addressed in checkpoints II-b and II-c.

- \rightarrow Two levels of reliability:
- A/B) The final solution composition is known (either from direct measurements or from the initial experimental setup and speciation calculations) and corresponds to equilibrium or is otherwise well constrained. All major components are included in the analysis. Relevant minor components (e.g. traces of carbonate or of other complexing ligands) may only be estimated. Some minor components may be unknown. In case of natural water samples, solutions are (or can be) shown to be charge balanced (within 5 %). The information on final solution composition can be obtained from i) analyses of the actual sorption samples or from ii) using pre-equilibrated solutions that had been analyzed prior to the actual sorption experiments.

- C/D) The critical major solution components are known, or can be estimated approximately. There may be unknown minor components and/or less critical major components. In case of natural water samples, solutions are approximately charge balanced (within 10 %).
- \rightarrow If a lower quality than required for level C/D is evident, the respective entry is excluded from further evaluation as unreliable.

II-e Temperature

Here, it is evaluated whether temperature is specified and kept constant.

- \rightarrow Two levels of reliability:
- A/B) Temperature is approximately specified (e.g. room temperature) and constant, or varied in a controlled fashion.
- C/D) Temperature is not specified at all (i.e., it is not clear whether the experiments had been performed at room temperature or not).
- II-f Solid/water ratio (S/W) and grain size

It is evaluated whether enough solid had been added to avoid a significant influence by the vessel walls (see II-m), and to ensure sample reproducibility and representativeness in case of complex substrates, especially in case of large grain sizes: It is estimated that in cases where less than ca. 100 mg of solid (this value depends on grain size) has been added to each experimental vessel, sample reproducibility and representativeness becomes difficult to achieve in case of complex or crushed samples.

- \rightarrow Two levels of reliability:
- A/B) Enough solid had been added to each vessel to assume that
 - a) [surface area sorbent] » [surface area vessel], i.e. that at least 5 m² of sorbent surface had been added to each vessel, and to assume that
 - b) samples are reproducible and representative.

What is enough substrate clearly depends on specific surface area and homogeneity. Fulfilling the above two requirements is typically not a problem in case of relatively homogeneous sorbents with a high specific surface are (such as clay minerals or bentonite), where "enough" may mean at least ca. 100 mg. On the other hand, "enough" may mean at least one to several grams in case of rocks (depending on specific surface area, grain size and complexity of the sample).

C/D) Any other than the above.

II-g Sorption value

It is evaluated whether an appropriate experimental design had been employed to avoid sorption values near 0% or 100%, which can lead to higher experimental uncertainty. This problem can be addressed by choosing an appropriate S/W ratio (see II-f) or/and an appropriate initial concentration of RN ([RN]) (see II-h). However, the choice of [RN] is more restricted by solubility and analytical detection limits.

A) The sorption value is in the range of 5% - 95% sorbed.

- B) The sorption value is inside the range of 2% 98% sorbed.
- C/D) Any other than the above.

II-h Initial RN concentration ([RN])

This parameter is used to evaluate the likelihood of a possible supersaturation of RN-phases:

- \rightarrow Three levels of reliability:
- A) RN is not solubility limited, or initial [RN] was clearly (at least a factor of 5) below the solubility limit. Note that factor 5 does not take into account uncertainties in RN solubility; i.e., if the solubility of a given RN cannot be estimated with more certainty than e.g. 10^{-6} to 10^{-8} M, then initial [RN] has to be $\leq 2 \times 10^{-9}$ M for rating A to apply.
- B) Initial [RN] was clearly below the solubility limit, but maybe less than a factor of 5 (see above).
- C/D) [RN] was very small, and in all likelihood below their maximum solubility, but the solubility limit cannot be established clearly due to missing information (solution composition) or lacking thermodynamic data.
- \rightarrow Note that the solubility limit can be defined on either thermodynamic calculations or on experimental data obtained under the relevant conditions.
- → If initial RN concentration had been clearly above the respective solubility limit, the respective entry is excluded from further evaluation as unreliable.

II-i Phase separation

Here, the appropriateness of phase separation is evaluated: Note that in cases where colloids or other artifacts are important, different phase separation methods will not lead to the same results. Identical or very similar results with different efficient methods are probably the best direct proof of absence of important colloid effects; hence such studies

are rated A. Rating B would be given for methods that can be presumed to remove colloids, but where no direct proof as in A is given.

- \rightarrow Three levels of reliability:
- A) Identical (very similar) results are obtained with different methods of phase separation, where at least one method needs to be efficient in terms of colloids removal (ultrafiltration or high-speed centrifugation). Accordingly, the best comparison would be between two efficient methods, such as ultrafiltration and high-speed centrifugation.

Note that such a comparison of phase separation methods is not required for each individual K_d value: For example: If the absence of artifacts has been demonstrated for some representative samples of a study by comparing an efficient and a standard method of phase separation, the rating A may be given to all datapoints of this study, even if they correspond to the standard method only.

- B) Only one, but efficient method (high-speed centrifugation, ultrafiltration) is used, and there is no evidence for artifacts such as colloid effects or significant sorption to the filter.
- C/D) Only one general method (normal centrifugation, membrane filtration with nominal pore sizes of $0.01-0.45 \ \mu m$) is used, and there is no evidence for artifacts such as colloid effects or significant sorption to the filter.
- → If no phase separation is used, or in case of obvious evidence for artifacts (colloid effect, adsorption on filter) the respective entry is excluded from further evaluation as unreliable.
- II-j Reaction time
 - \rightarrow Two levels of reliability:
 - A/B) Identical (similar) results are obtained with different reaction times, or some other demonstration of near-equilibrium is provided (e.g. separate kinetic experiments).
 - C/D) Only one, but reasonably long reaction time is used. What is "reasonably long" is highly dependent on the experimental system: In general, the time needed to reach equilibrium will increase with the complexity of the sorbing substrate and the strength of sorption. Sorption of Sr onto a pure clay mineral through ion exchange can be assumed to be complete within a day; sorption of a trivalent actinide onto a complex substrate may need several days to weeks for completion. In the absence of kinetic information, operator expert decisions will be required to assess this point. If possible, reaction times reported for similar

systems included in the JNC-SDB could be used to evaluate what is reasonably long. Further, even for the most simple systems a reaction time of 1 day is considered as minimum requirement.

- → If the requirement for level C/D is not met (i.e., if the reaction time cannot be assumed to be reasonably long), the respective entry is excluded from further evaluation as unreliable.
- **II-k** Agitation method
 - \rightarrow Two levels of reliability:
 - A/B) Appropriate agitation is required in all cases, except where enough kinetic information is provided to show that equilibrium has been reached. Shaking is the preferred method, as use of stir bars can lead to abrasion of samples. In case of simple and well crystallized substrates (such as Al-oxide) or of substrates with very small grain size that are easily suspended, stir bars can also be accepted.
 - C/D) Any other than the above.

II-I RN loading

Ideal are values as a function of RN loading (i.e., K_d values that form part of an isotherm), otherwise low loading is preferred. RN loading (e.g. in moles RN/kg substrate) refers to the amount of RN adsorbed in relation to the amount of different surface sites available. It is known from classical isotherms (e.g. Langmuir) that a linear sorption can only be assumed if sufficient unoccupied sites are present. In case of simple substrates (including some bentonites), the linear portion of an isotherm extends to fairly high RN loading. There are other cases where K_d depends significantly on RN loading over many orders of RN concentration.

- \rightarrow Three levels of reliability:
- At least one isotherm has been determined (for a constant solution composition and S/W), and at least some experiments have been carried out using trace level RN concentration (i.e., at least some data are included within a linear sorption region).
- B) No isotherm is available, but at least a limited variation of initial [RN] or S/W has been carried out, and some experiments have been carried out using trace level RN concentration (i.e., some data are included within a linear sorption

region).

C/D) No variation as in A or B has been carried out.

II-m Reaction vessels

High-density polyethylene (HDPE) or Teflon are preferred over normal PE, which is preferred over glass, which may lead to sorption of radionuclides by the vessel walls. Especially at high or very low pH, glass dissolution and release of dissolved or colloidal silica may also occur. On the other hand, glass is more gas-tight (especially than PE); if that is of experimental relevance. Corrections for sorption on vessel walls should not be necessary if blank tests show that it can be neglected.

Correction for sorption on vessel walls may be needed to estimate K_d values correctly in some cases, but only in cases where a) sorption on the vessel is much stronger than on the solid sorbent, or b) if the vessel offers a significant surface area in comparison to the sorbent (see II-f). If that is not the case, the sorption on the added solid will be much greater than on the vessel in a system where both solid and vessel are present. It is further an erroneous assumption that sorption on the vessel will be same in i) the absence of the solid (no competition for RN by solid) as ii) in the presence of the solid (strong competition for RN by solid). The sorption on the walls is typically much smaller in ii) than in i). Therefore, the overall mistake is often bigger if sorption on the vessel wall is accounted for than if it is neglected.

If effects of vessel walls are corrected for, it has to be done by extracting any RN sorbed to vessel walls after experimentation (e.g. by acid washing) and establishing a complete mass balance.

- \rightarrow Three levels of reliability:
- A) An appropriate vessel has been used (taking into account sorption as well as tightness with respect to CO₂ or O₂, where required), and corrections for sorption on vessel wall have been performed or no sorption on vessel wall has been observed by blank tests. If effects of vessel walls are corrected for, it has to be done by extracting any RN sorbed to vessel walls after experimentation (e.g. by acid washing) and establishing a complete mass balance. If the sorption on vessel wall has been determined as significantly lower (at least two orders of magnitude in terms of K_d) than the actual K_d value and thus corrections for sorption on vessel wall have not been performed, such a case would also correspond to level A
- B) An appropriate vessel has been used, and corrections for sorption on vessel walls have not been performed.

C/D) The vessel used may have been not appropriate (this is often the case with glass, see above), or corrections for sorption on vessel wall have been performed based on a blank test only (i.e., without verifying that sorption on vessel walls is relevant in the presence of a solid added, thus possibly leading to overcorrection).

II-n Uncertainty estimates

In general, uncertainties based on repeated experiments (i.e., actual observations of K_d) are preferred over uncertainties based on error propagation, as the latter is an estimate based on a type of extrapolation. Thus, the difference between levels of reliability is mainly based on the amount of actual information gained by repetitions: For level A, the entire experiment is repeated; for level B, only sampling and analysis are repeated; for C, no repetitions are carried out.

Values that are based on repetitive experiments are preferred over single experimental data points. Note, however, that this checkpoint refers to single-point K_d values and may be overruled by data being part of e.g. pH-edge, isotherm, kinetic experiment, etc., which may provide independent evidence of good reproducibility or systematic errors (see checkpoint II-o).

- \rightarrow Four levels of reliability:
- A) Uncertainties in K_d are derived based on entire, replicated sorption experiments (i.e., at least duplicate experiments).
- B) Uncertainties in K_d are derived based on single sorption experiments that are sampled and analyzed repeatedly. This may be supplemented by error propagation.
- C) Uncertainties in K_d are based on error propagation of estimated analytical and/or procedural uncertainties.
- D) No error estimate is given, no repeated sampling is done.

II-o Parameter variation

Studies with a systematic variation of key parameters are much more valuable and reliable than single K_d measurements. In this context, key parameters are those that influence sorption (for example, chemical parameters such as RN concentration, pH, pCO₂, but also temperature, S/W, or grain size in case of crushed substrates), but not parameters that only help to determine the experimental framework (such as vessel type or reaction time). In particular, variation of key parameters allows improved detection of

experimental problems and systematic errors. Especially the latter are not detected by repeating experiments under identical conditions. In the application of this checkpoint, care has to be taken to take into account the characteristics of the particular system studied. For example, more parameter variation may be required to show clear trends in a complicated system in comparison to a simpler one. On the other hand, the pH and carbonate concentration in experiments with calcite are quite constrained by the solid itself, and only limited variations are possible.

- \rightarrow Four levels of reliability:
- A) Both RN surface loading (isotherm) as well as a chemical parameter, such as pH or pCO₂ (edge), or e.g. [Na] in case of ion exchange, are varied systematically.
- B) Either RN concentration (isotherm) and/or chemical parameters, such as pH or pCO₂ (edge), or e.g. [Na] in case of ion exchange (i.e., at least two parameters in total), are varied. These variations are less systematic than in A, but still allow to observe trends.
- C) As B, but only one parameter in total is varied.
- D) No parameter variation is done.

2.2.4 Criterion III: Consistency of data

Here it will be evaluated whether data from a particular study can be supported by other studies. Comparisons should only be made with studies that are at least as (or more) reliable than the study under investigation, based on criteria I and II. In many cases, only approximate consistencies or inconsistencies may be apparent, because of different conditions used in the different studies.

- → Therefore, the evaluation of criterion III will only be reported in the form of a comment. Any such comments will be included both in a classification report as well as in the corresponding rating summary sheets.
- → If the K_d values under investigation are *clearly* inconsistent with the majority of related reliable studies, and if the reason for this observation cannot be explained, they may also be labeled unreliable based on criterion III. As this requires an expert decision by the operator, the underlying reasoning needs to be clearly documented.

2.3 Overall classification

The above criteria are applied to an overall classification system as follows:

- The three criteria I-III are evaluated separately, the respective results are reported separately as well.
- Criterion I: The checkpoints under I-a are used in a yes/no screening fashion, entries not fulfilling I-a are labeled as unreliable and are not evaluated further. Criterion I-b is then used to assign classes 1-6 for documentation.
- Criterion II: a) The datasets that pass Criterion I are again classified according to a 6-level system, where classes 1-6 represent the highest and lowest levels of reliability. To ensure a minimum quality level, certain checkpoints are regarded as critical (marked with * in Table 2.3.1) If the quality of the data does not correspond to the respective minimum requirements, the entries are not to be used and are classified as unreliable.
 - b) To facilitate transparent averaging of all checkpoints, the following numerical system is suggested: A=3, B=2, C=1, D=0 (A/B=3 and C/D=0 in some cases).
 - c) Initially, checkpoints II-b, II-c, II-d, and II-h are evaluated (indicated in bold letters below). If an entry is rated unreliable for any of these checkpoints, it is excluded from further evaluation.
 - d) Weighting of individual checkpoints at this level is done according to the factors given in Table 2.3.1 below.

e) The total sum of points obtained for Criterion II is then used to indicate the level of reliability. With the present system, the maximum number of points would be 183, leading to an overall classification as follows (<u>Table</u> 2.3.2):

checkpoint	description	weighting factor
II-a	solid phase (substrate)	A-C/D $\times 2$
II-b	pH	A-D $\times 8$
[I-c	redox conditions	A/B-C/D × 8
II-d	final solution composition	A/B-C/D × 8
I-e	temperature	A/B-C/D $\times 1$
II-f	S/W, grain size	A/B-C/D $\times 2$
II-g	sorption value	A-C/D $\times 2$
II-h	initial RN concentration	A-/CD $\times 8$
I-i	phase separation	A-C/D $\times 8$
I-j	reaction time	A/B-C/D × 2
II-k	agitation	A/B-C/D $\times 1$
II-l	RN loading	A-C/D $\times 2$
I-m	reaction vessel	A-C/D $\times 1$
II-n	uncertainty estimates	A-D $\times 2$
II-o	parameter variation	A-D $\times 8$

Table 2.3.1Weighting of individual checkpoints under Criterion II.

* indicates critical checkpoints with minimum requirements;

bold letters indicate the checkpoints to be evaluated initially

points	rating	
183-151	class 1	
150-121	class 2	
120-91	class 3	
90-61	class 4	
60-31	class 5	
30-0	class 6	

Table 2.3.2Overall classes of reliability for Criterion II

- Criterion III: Criterion III is used to qualitatively assess consistency with other studies. In case of clear inconsistencies, an entry may be labeled as unreliable.
- Overall, the following classification system is used, with Criterion II as the main basis for assessing the reliability of entries in the JNC-SDB

Criterion	classification
I-a	accept/reject
I-b	6 classes of K _d information
II	6 classes of data quality and reliability
III	qualitative level of consistency with other studies

Table 2.3.3The classification system

3 Classification of selected entries for mudstone and related minerals in the JNC-SDB

This section presents the classification results for selected entries in the JNC-SDB covering the datasets for the sorption of Th, Np, Se, Se, Pu and Am on mudstone and related minerals that were mainly used for the K_d-setting exercise discussed in Ochs et al.(2008).

- An overview of the results as well as the corresponding numerical rating is given in the Appendix.
- For transparency and ease of presentation, all results of Criteria I and II are presented in tabular form, using the format of the following table throughout. The results pertaining to Criterion III are discussed subsequently and are illustrated in the form of plots of K_d vs. a relevant master variable (typically pH), where applicable.
- According to the established classification guideline, criterion I and checkpoints II-b, II-c, II-d, II-h were evaluated first. Classification and final numerical rating were only completed when an entry was evaluated as reliable based on these checkpoints. Otherwise, entries were labeled "unreliable" and were excluded from further evaluation. For most reliable entries, criterion III was evaluated as well.

3.1 Criteria I and II

1

Data table Element/#: REF: original reference					
JNC-SDB version # – DATA: element/solid sorbent, datapoint #					
GUIDELINE:	Revision # (date)				
Checkpoint ¹	Rating				
"SDB"/"REF"					
I-a.1	Are all mandatory fields completed?	Yes/No			
I-a.2	Is all mandatory information provided?	Yes/No			
I-b	Type of K _d information	class 1-6			
II-a	Solid phase (substrate)	A-C/D			
II-b	Adjustment and control of pH	A-D			
II-c	Redox conditions.	A/B-C/D			
II-d	Final solution composition	A/B-C/D			
II-e	Temperature	A/B-C/D			
II-f	Solid/water ratio (S/W) and grain size	A/B-C/D			
II-g	Sorption value	A-C/D			
II-h	Initial RN concentration [RN]	A-C/D			
II-i	Phase separation	A-C/D			
II-j	Reaction time	A/B-C/D			
II-k	Agitation method	A/B-C/D			
II-l	RN loading	A-C/D			
II-m	Reaction vessels	A-C/D			
II-n	Uncertainty estimates	A-D			
II-o	Parameter variation	A-D			

Application of the checkpoint to the information given in the JNC-SDB is indicated with "SDB". "REF" indicates the additional application to the original reference indicated in each table.

3.1.1. Thorium

Data t	able Th	/1: REF: Lieser et al. (1990)	
JNC-S	DB vers	ion 2 – DATA: Th/Mudstone, # 56105-56107	
GUID	ELINE:	Revision 4b (May 19, 2005)	
Checkpoint Evaluation			Rating
I-a.1	-a.1 SDB All mandatory fields are completed.		Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	It is indicated that the distribution between size fractions is used for	
		calculation of K _d values.	
	REF	It is a linear graph with indication of relative concentrations c/c_0 of Th in	class 4
		percent in the effluent groundwater ($c_0 =$ total initial conc. of Th in the	
		groundwater).	
II-a	SDB	It is indicated that Elster glacial sediment (specific surface area: 0.003) and	
		Weichsel glacial sediment (specific surface area: 0.01) has been used.	
		No mineral composition is available, only the chemical compositions of the	
	REF	sediments are listed in a table.	C/D
II-b SDB Initial and final pH-values are given.		А	
II-c SDB It is indicated that all groundwaters and sediments are kept under 99% argon			
and 1% carbon dioxide.			
		Th is not redox sensitive.	A/B
II-d	SDB	It is indicated that solutions are low and medium salinity groundwaters.	
	REF	The final solution compositions after equilibration with the corresponding	
		sediments are listed in a table. The concentrations are given for unfiltered,	A/B
		0.45 µm filtered and ultrafiltered groundwater solutions.	
II-e	SDB	Temperature is not specified.	C/D
II-f	SDB	A solution/solid ratio of 2.5 mL/g is indicated. 100 g solid are added to 250	
		mL solution.	A/B
II-g	SDB	Based on the information given in the SDB, and on a solution/solid ratio of 2.5	
		mL/g, sorption values between 97.4-99.5 % sorbed can be calculated.	
		• Datapoints 56105, 56107	C/D
		• Datapoint 56106	В
II-h	SDB	The initial Th concentrations are reported to be 1.0×10^{-11} M in all three	
		groundwaters:	
	REF	Based on thermodynamic calculations with PHREEQC Interactive 2.12.5	А
		(database: JNC-TDB_011213c2 and NAGRA-PSI) for the groundwaters G1,	

		G8 and G9 the indicated i	nitial [Th] is below the calc	ulated solubility limits:	
		Calculated solubility limits for ThO_2 in mol/kg			
		JNC-TDB_011213c2 NAGRA-PSI			
		G1 a) unfiltered 2.44×10^{-7} 2.57×10^{-7}			
		G8 a) unfiltered	1.87×10^{-6}	1.97×10^{-6}	
		G9 a) unfiltered	1.11×10^{-6}	1.20×10^{-6}	
		The calculated solubility l (1999).	imits agree with the solubil	ity data given in Rai et al.	
II-i	REF	Samples are analyzed w	ithout filtration, after filtra	ation through membrane	
		filters (Millipore) of 0.45 μ m pore size and after ultrafiltration through filter			
		sets (Millipore Immersible CX-30) of 0.002 μ m pore size. Sorption values			А
		from the different method	s are similar.		
II-j	SDB	It is indicated that a conta	ct time of 14 days had been	C/D	
II-k	REF	Sufficient mixing had bee	ing had been achieved, considering the relatively long reaction		A/B
		time. For mixing the closed reaction vessels are slowly turned over head (1			
		revolution / 3 min.).			
II-l	SDB	Only one datapoint is available	ilable for each groundwater.		C/D
II-m	REF	No description of reaction vessels is available. However, a sufficient amount			В
		of sediment (400 g/L) is used in the experiments and therefore no correction			
		for vessel walls is necessa	ry in case of all experiment	s.	
II-n	SDB	No repeated sampling is d	one and no uncertainties are	e reported.	D
II-o	SDB	Only one datapoint is available	ilable for each groundwater.		D

Data t	Data table Th/2: REF: Legoux et al. (1992)				
JNC-S	DB vers	ion 2 – DATA: Th/Mudstone, # 56055			
GUIDI	ELINE:	Revision 4b (May 19, 2005)			
Check	point	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided.	Yes		
I-b	SDB	It is indicated that tables with K_d are given.	class 1		
II-a	SDB	The solid phase is indicated as soil A. The mineralogical composition and	А		
		specific surface of the sample are given in a table.			
II-b	SDB	Initial and final pH is indicated.	А		
II-c	SDB	It is indicated that no redox conditions are reported.			

			1
	REF	No reducing agents are added to the experiment and no inert gas atmosphere is	A/B
		reported. Therefore it is assumed that the experiment is conducted under	
		ambient conditions.	
		Thorium is not redox sensitive.	
II-d	REF	The final solution composition is reported.	A/B
II-e	SDB	It is indicated that no temperature is reported.	C/D
II-f	SDB	A L/S ratio of 10:1 (10 mL/1 g) is indicated.	
	REF	A specific surface area (BET) of 8.6 m^2/g is reported. The grain sizes of the	A/B
		sample is between 150 and 500 μ m.	
II-g	SDB	Based on the information given in the SDB, a sorption value of	C/D
		99.99 %-sorbed can be calculated.	
II-h	SDB	An initial Th concentration of 4×10^{-11} M is indicated.	
	REF	Based on the solubility data given in Rai et al. (1999), the experiments are	А
		considered to be undersaturated with respect to $ThO_2(s)$ by a factor of 5 or	
		more.	
II-i	SDB	Centrifugation and ultrafiltration with 0.025 and 0.002 µm Millipore filters is	А
		reported. Centrifugation is done at 4000 rpm prior to filtration.	
II-j	SDB	It is indicated that contact time is 2 days.	C/D
II-k	REF	Samples are agitated continuously.	A/B
II-l	SDB	No variation of initial [Th] or S/W has been carried out.	C/D
II-m	REF	The experiments are carried out in polypropylene vessels. It is reported that	C/D
		sorption to the vessel wall has been corrected for. No information about the	
		correction procedure is available. It is assumed that the correction is based on	
		the data obtained of a blank experiment.	
II-n	REF	The reported error represents the standard deviation of the average of at least	А
		four experimental repetitions.	
II-o	SDB	No parameter variation is indicated.	D

Data table Th/3: REF: Östhols (1995)					
JNC-S	JNC-SDB version 2 – DATA: Th/Other minerals, # 57172-57220				
GUIDI	ELINE:	Revision 4b (May 19, 2005)			
Checkpoint		Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided.	Yes		

I-b	SDB	It is indicated that tables with % sorption are given.	class 2
II-a	SDB	It is indicated that SiO_2 (Aerosil OX200) had been used.	А
II-b	SDB	Final pH-values are given.	А
II-c	REF	It is indicated that during all titrations water-saturated N ₂ is bubbled through	
		the solutions.	A/B
		Th is not redox sensitive.	
II-d	SDB	It is indicated that solutions consisted of 0.1 M and 1.0 M NaClO ₄ .	A/B
		Due to the simplicity of the given solutions, the final solution composition can	
		be estimated.	
II-e	SDB	It is indicated that a temperature of 25 ± 0.6 °C had been used.	A/B
II-f	SDB	The BET surface area is indicated to be 169.3 m^2/g .	
		Rating is done based on the calculated surface area of the added sorbent	A/B
		indicated. In all cases it is $> 5 \text{ m}^2$.	
II-g	SDB	Rating is done based on %-sorbed values calculated from the information	
		given in the JNC-SDB:	
		 Datapoints 57199, 57205, 57208, 57209, 57213, 57219 (<1.7%, > 98.88%) 	C/D
		• All other datapoints (between 5.09%- 93.4%)	А
II-h	SDB	The following initial Th concentrations are indicated:	
		• 0.1 M NaClO ₄ : initial [Th] = 3.45×10^{-6} M	
		• 1.0 M NaClO ₄ : initial [Th] = 1.00×10^{-6} M	
		$3.18 \times 10^{-6} \mathrm{M}$	
		$6.33 \times 10^{-6} \text{ M}$ $3.00 \times 10^{-5} \text{ M}$	
		$3.16 \times 10^{-6} \text{ M}$	
		Based on the solubility data given in Rai et al. (1999), initial [Th] in the	
		experiments with 0.1 M and 1.0 M NaClO ₄ at the indicated acidic pH values	А
		are far below the solubility limit given by $ThO_2(s)$.	
II-i	SDB	It is indicated that all samples had been centrifuged for 20 min. at 19000 rpm.	В
II-j	SDB	It is indicated that a contact time of at least 24 hours had been used.	C/D
II-k	REF	For mixing, the samples are shaken for at least 24 hours.	A/B
II-l	SDB	[Th] and the S/W have been varied.	В
II-m	REF	Experiments are performed in glass vessels. A blank titration is made using an	
_	-	appropriate [Th]. At a pH of 3.5 the losses to vessel walls are around 20%.	
		Information about a possibly performed correction is not available.	
			C/D

II-n	REF	It is indicated that forward (alkaline direction) and reverse titration had been	
		performed. The corresponding uncertainties of % adsorbed Th are given as	В
		error bars in a graph.	
II-0	SDB	Distribution ratios are determined for six Th concentrations; pH is also varied	А
		(sorption edge).	

Data t	able Th	/4: REF: Bradbury & Baeyens (2003)*		
JNC-S	DB vers	ion 2 – DATA: Th/Na-SWy1 montmorillonite, # 67039-67054		
GUIDI	ELINE:	Revision 4b (May 19, 2005)		
* REMARK: The literature was classified assuming the same experimental conditions as described in				
		Lauber et al. (2000) and Baeyens and Bradbury (1997)		
Check	Checkpoint Evaluation		Rating	
I-a.1	REF	All mandatory fields are completed.	Yes	
I-a.2	REF	All mandatory information is provided.	Yes	
I-b	REF	A sorption edge graph with logarithmic K _d values is given.	class 5	
II-a	REF	Purified Na-SWy1montmorillonite with determined mineralogical	А	
		composition and characteristics is used for the experiments.		
II-b	REF	Final pH-values are measured.	А	
II-c	REF	Experiments are carried out in an inert atmosphere glove box ($CO_2 < 10^{-5.5}$	A/B	
		bar). Th is not redox sensitive.		
II-d	REF	A 0.1 M NaClO ₄ -solution is used.	A/B	
II-e	REF	It is not specified at which temperature experiments had been performed. It is	C/D	
		assumed that experiments had been made at ambient temperature.		
II-f	REF	Enough solid had been added to each vessel (0.42 g/L).	A/B	
II-g	REF	Rating is done based on %-sorbed values calculated from sixteen K _d -values		
		(pH-range 1-12) out of the sorption edge graph. Sorption values are in the		
		range of 78% and 99.5%.		
		• Datapoints: 67039, 67040, 67043	А	
		• Datapoints: 67041, 67042, 67044, 67046	В	
		• Datapoints: 67045, 67047-67054	C/D	
II-h	REF	The initial Th concentration is reported to be $< 10^{-9}$ M. According to the data	В	
		of Rai et al. (1999), in a pH range of 1-8.5, the reported Th concentration is		
		below the solubility limit.		
		Datapoints above pH 8.5 are close or above the solubility limit and are	unreliable	

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		considered as unreliable.	
II-i	REF	It is reported that samples had been centrifuged for 60 min. at 95,000g max.	В
II-j	REF	Only one reaction time is reported.	C/D
II-k	REF	For mixing the samples, the tubes are shaken end-over-end for 7 days.	A/B
II-l	REF	No variation in [Th] or the S/W is indicated. A sorption isotherm on MX-80	C/D
		bentonite is reported. However these data are not considered for the	
		classification of Th on Swy1 montmorillonite.	
II-m	REF	Experiments are performed in polyethylene vessels. A maximum uncertainty	А
		in the log K_d values of ~0.05 log units due to wall sorption is indicated.	
II-n	REF	Estimates of the maximum error in each operation in similar batch sorption	С
		tests to those performed in Lauber et al. (2000) have been done.	
II-o	REF	Experiments are performed with constant [Th] at various pH values.	В

Data	Data table Th/5: REF: Meier et al. (1994)					
JNC-S	JNC-SDB version 2 – DATA: Th /Mudstone, # 62232 – 62236					
GUID	ELINE:	Revision 4b (May 19, 2005)				
Check	Checkpoint Evaluation					
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
		on initial and final pH, final solution composition, phase separation, initial and				
		final Eh and atmosphere/ redox condition are given.	unreliable			
	REF	No further information is available. Thorium is not redox sensitive.				

Data ta	Data table Th/6: REF: Ishii et al. (2001)					
JNC-SE	JNC-SDB version 2 – DATA: Th /Mudstone, # 65701 – 65704					
GUIDE	ELINE: I	Revision 4b (May 19, 2005)				
Checkp	Checkpoint Evaluation Rating					
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b S	SDB	Tables with K_d are given.	class 1			
II-a S	SDB	The solid phase is reported as loam(I) and loam(II). Physico- chemical	А			
		properties and specific surface of loams are given in a table.				
II-b S	SDB	Initial pH values are reported. Final pH values are not reported.	С			

II-c	SDB	It is reported that experiments are done under atmosphere and controlled	A/B
		atmosphere. Also, Th is not a redox-sensitive element.	
II-d	SDB	The final chemical properties of spring and synthesized rain water are	A/B
		reported.	
II-e	SDB	A temperature of 15°C is reported.	A/B
II-f	SDB	Solution/solid ratios are reported for 50mL/1g or 100mL/1g.	A/B
	REF	A specific surface area (BET) of 99.5 or 43.9 m^2/g are reported.	
II-g	SDB	Based on the information given in the SDB, and on a solution/solid ratio of	C/D
		100ml/g, sorption values between 98.7%- 99.9% sorbed can be calculated.	
II-h	SDB	The initial Th concentrations are reported to be 8.0×10^{-8} M, 4.3×10^{-7} M,	
		8.6×10^{-7} M.	
	REF	Based on thermodynamic calculations with CHESS (Chemical Equilibrium	
		with Species and Surface, database: Yamaguchi JAERI-Data/Code 2000-031,	В
		2000), the indicated initial [Th] is below the calculated solubility limits (but it	
		is not clear by how much, due to the uncertainty in initial concentration as	
		well as in solubility data for Th).	
II-i	SDB	The membrane filtration with pore size of $0.45 \mu m$ is used.	C/D
II-j	SDB	It is indicated that contact time is 14 days. This reaction time is reasonably	
		long.	C/D
	REF	The kinetics of sorption is not tested.	
II-k	REF	The batch sorption experiments are continuously performed by shaking	A/B
		(50rpm). And handshakes are performed once a day.	
II-l	REF	[Th] and the W/S have been varied.	В
II-m	REF	The experiments are carried out in polypropylene or Teflon. The sorption on	В
		vessel wall has been determined as lower than 5% than the actual value and	
		thus corrections for sorption on vessel wall have not been performed.	
II-n	SDB	Uncertainties are reported (n=3).	
		-)Datapoints: 65701, 65702, 65704	А
		Uncertainties are not reported.	
		-)Datapoint: 65703	D
II-0	SDB	The initial pH and initial Th concentration are varied.	В

3.1.2 Neptunium

JNC-S	DB vers	ion 2 – DATA: Np/Mudstone, # 56095			
GUIDELINE: Revision 4b (May 19, 2005)					
Check	point	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided.	Yes		
I-b	SDB	It is indicated that K _d values are tabulated.	class 1		
II-b	SDB	Both initial and final pH values are indicated.	А		
II-c	SDB	Reducing conditions and initial and final Eh values are indicated. As			
		additional information it is indicated that Np(V) is reduced to Np(IV) during			
		groundwater circulation (column experiment).			
	REF	Original redox potentials of the groundwaters are reported to be between	C/D		
		-100 mV and + 200 mV. The original concentration of molecular oxygen is			
		indicated to be < 0.06 mmol/L. All experiments under anaerobic conditions			
		are performed in inert gas boxes filled with a mixture of 99% Ar and 1%			
		CO_2 . The oxygen content in these inert gas boxes is reported to be between			
		0.02 and 0.05%. Eh and pH are continuously monitored during the			
		experiments. No reducing agent is reported. Final amount of Np(IV) is			
		determined by extraction with TTA in xylene.			
II-d	SDB	Natural groundwater is indicated as input solution; final solution			
		compositions are provided.			
	REF	K_d values of seven different natural groundwaters are reported. All			
		groundwaters are characterized.	A/B		
II-h	SDB	The initial Np concentration is indicated to be 4.0×10^{-9} M in groundwater G7.			
		Based on calculations with PHREEQC-2.12.5 (databases:			
		JNC-TDB_011213c2 and NAGRA-PSI) for the indicated groundwater, the			
		initial [Np(IV)] is above the calculated solubility limits. In addition, the			
		calculated data indicate that the main species is Np(V) given the reported Eh			
		of 80 mV.			
	REF	Apparent sorption due to the formation of sparingly soluble species of	unreliable		
		Np(IV) that are precipitated and held back by the sediments is reported. The			
		formation of colloids and precipitate lead to erroneous K _d values.			

Data	Data table Np/2: REF: Barney and Anderson (1979)				
JNC-S	SDB vers	ion 2 – DATA: Np /Mudstone, # 44274 – 44279			
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	Checkpoint Evaluation				
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on final solution composition is given.			
	REF	No further information is available. The initial composition of groundwater is			
		not reported. The final solution composition can not be estimated.	unreliable		

Data 1	Data table Np/3: REF: Barney and Brown (1979)					
JNC-S	JNC-SDB version 2 – DATA: Np /Mudstone, # 44340 – 44342					
GUID	ELINE:	Revision 4b (May 19, 2005)				
Check	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
		on final solution composition, Eh (and pH) at beginning or end of the				
		experiments, experimental redox/atmosphere conditions and the initial RN				
		concentration are given.				
	REF	No further information is available. The initial composition of groundwater is				
		not reported. The final solution composition can not be estimated.	unreliable			

Data	Data table Np/4: REF: Berry et al. (1990a)				
JNC-S	SDB vers	ion 2 – DATA: Np /Mudstone, # 48144 – 48172			
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	Checkpoint Evaluation Ratin				
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on Eh at beginning or end of the experiments and contact time are given.			
	REF	No further information is available.	unreliable		

Data	Data table Np/5: REF: Higgo et al.(1987)				
JNC-S	SDB vers	ion 2 – DATA: Np /Mudstone, # 53249 – 53260			
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	Checkpoint Evaluation				
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on final solution composition, Eh at beginning or end of the experiments and			
		contact time are given.			
	REF	No further information is available.	unreliable		

Data	Data table Np/6: REF: Kim et al. (1994)				
JNC-S	SDB vers	ion 2 – DATA: Np /Mudstone, # 53968			
GUID	ELINE:	Revision 4b (May 19, 2005)			
Checkpoint		Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on separation and final solution composition are given.			
	REF	No further information is available.	unreliable		

Data	table Np	/7: REF:Legoux et al.(1992)					
JNC-S	JNC-SDB version 2 – DATA: Np /Mudstone, # 56023 – 56026						
GUID	DELINE:	Revision 4b (May 19, 2005)					
Chec	kpoint	Evaluation	Rating				
I-a.1	SDB	All mandatory fields are completed.	Yes				
I-a.2	SDB	All mandatory information is provided.	Yes				
I-b	SDB	Tables with K_d are given.	class 1				
II-a	SDB	Four soils with known mineral composition are indicated. CEC value and	А				
		specific surface area are reported.					
II-b	SDB	Initial and final pH values are indicated.	А				
	REF	No use of any pH-buffer is reported.					
II-c	SDB	It is indicated that no redox conditions are reported. There is no information					
		about Eh. Experiments are carried out with Np (V).					
	REF	No reducing agents are added to the experiment and no inert gas atmosphere is	C/D				
		reported. Therefore it is assumed that the experiment is conducted under					

II-d REF The final solution composition is reported. A/B II-e SDB A temperature is not reported. C/D II-f SDB Solution/solid ratio is 10mL/g. A/B REF The used mass of 1g soil (A, B, C, D) has a surface area of about 6.6 to 14.7m ² . A/B II-g SDB The sorption value is between 43.18% and 85.61% (calculated with W/S and K _d values). A II-h SDB Initial [Np] is 5.0×10 ⁻¹² M. A REF Based on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limiti (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np). B II-i SDB The centrifugation is performed at 4000rpm followed by ultrafiltration through as well as in solubility data for Np). A/B II-i SDB A reaction time of 2 days is indicated. C/D II-i SDB A reaction time of 2 days is indicated. C/D II-i REF Agitation is carried out in reaction time. A/B II-i REF No isotherms, no variation of the Np concentration is indicated. C/D II-i REF No isotherms, no variation of the N				
II-e SDB A temperature is not reported. C/D II-f SDB Solution/solid ratio is 10mL/g. A/B REF The used mass of 1g soil (A, B, C, D) has a surface area of about 6.6 to 14.7m ² . A/B II-g SDB The sorption value is between 43.18% and 85.61% (calculated with W/S and K _d values). A II-h SDB Initial [Np] is 5.0×10 ⁻¹² M. A Based on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np). B II-i SDB The centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated. C/D II-i SDB A reaction time of 2 days is indicated. C/D II-i REF No isotherms, no variation of the Np concentration is indicated. C/D II-i REF The experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment. A II-m REF The reported error represents the standard deviation of the average of at least four e			ambient conditions.	
II-f SDB Solution/solid ratio is 10mL/g. A/B REF The used mass of 1g soil (A, B, C, D) has a surface area of about 6.6 to 14.7m ² . A/B II-g SDB The sorption value is between 43.18% and 85.61% (calculated with W/S and K _d values). A II-h SDB Initial [Np] is 5.0×10 ⁻¹² M. A REF Based on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np). B II-i SDB The centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated. C/D II-i SDB A reaction time of 2 days is indicated. C/D II-i REF Agitation is carried out in reaction time. A/B II-i REF No isotherms, no variation of the Np concentration is indicated. C/D II-i REF The experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment. A II-m REF The reported error represents the standard deviation of	II-d	REF	The final solution composition is reported.	A/B
REFThe used mass of 1g soil (A, B, C, D) has a surface area of about 6.6 to 14.7m².II-gSDBThe sorption value is between 43.18% and 85.61% (calculated with W/S and Kq values).AII-hSDBInitial [Np] is 5.0×10^{-12} M.AREFBased on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np).BII-iSDBThe centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated.C/DII-iSDBA reaction time of 2 days is indicated.C/DII-iREFAgitation is carried out in reaction time.A/BII-iREFNo isotherms, no variation of the Np concentration is indicated.C/DII-iREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-inREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A	II-e	SDB	A temperature is not reported.	C/D
II-gSDBThe sorption value is between 43.18% and 85.61% (calculated with W/S and K _d values).AII-hSDBInitial [Np] is 5.0×10 ⁻¹² M.AREFBased on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np).BII-iSDBThe centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002μm Millipore filters is indicated.C/DII-iSDBA reaction time of 2 days is indicated.C/DII-iREFAgitation is carried out in reaction time.A/BII-iREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A	II-f	SDB	Solution/solid ratio is 10mL/g.	A/B
II-g SDB The sorption value is between 43.18% and 85.61% (calculated with W/S and K _d values). A II-h SDB Initial [Np] is 5.0×10^{-12} M. Based on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np). B II-i SDB The centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated. A II-i SDB A reaction time of 2 days is indicated. C/D II-k REF Agitation is carried out in reaction time. A/B II-n REF The experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment. A II-n REF The reported error represents the standard deviation of the average of at least four experimental repetitions. A		REF	The used mass of 1g soil (A, B, C, D) has a surface area of about 6.6 to	
II-h SDB Initial [Np] is 5.0×10 ⁻¹² M. REF Based on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np). Based on 0.002 μm Millipore filters is indicated. II-i SDB The centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002 μm Millipore filters is indicated. C/D II-i SDB A reaction time of 2 days is indicated. C/D II-i REF Agitation is carried out in reaction time. A/B II-i REF No isotherms, no variation of the Np concentration is indicated. C/D II-m REF The experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment. A II-n REF The reported error represents the standard deviation of the average of at least four experimental repetitions. A			14.7m ² .	
II-hSDBInitial [Np] is 5.0×10 ⁻¹² M.REFBased on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np).BII-iSDBThe centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated.AII-jSDBA reaction time of 2 days is indicated.C/DII-kREFAgitation is carried out in reaction time.A/BII-1REFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A	II-g	SDB	The sorption value is between 43.18% and 85.61% (calculated with W/S and	А
REFBased on speciation calculations (for 25° C) with PHREEQC using the thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np).BII-iSDBThe centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated.AII-jSDBA reaction time of 2 days is indicated.C/DII-kREFAgitation is carried out in reaction time.A/BII-1REFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A			K _d values).	
Hermodynamic data in JNC-TDB the initial [Np] are below the solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np).BII-iSDBThe centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated.AII-jSDBA reaction time of 2 days is indicated.C/DII-kREFAgitation is carried out in reaction time.A/BII-1REFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A	II-h	SDB	Initial [Np] is 5.0×10^{-12} M.	
 (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Np). II-i SDB The centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated. II-j SDB A reaction time of 2 days is indicated. II-k REF Agitation is carried out in reaction time. II-n REF No isotherms, no variation of the Np concentration is indicated. II-n REF The experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment. II-n REF The reported error represents the standard deviation of the average of at least four experimental repetitions. 		REF	Based on speciation calculations (for 25° C) with PHREEQC using the	
II-ias well as in solubility data for Np).AII-iSDBThe centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002µm Millipore filters is indicated.AII-jSDBA reaction time of 2 days is indicated.C/DII-kREFAgitation is carried out in reaction time.A/BII-1REFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A			thermodynamic data in JNC-TDB the initial [Np] are below the solubility limit	
II-iSDBThe centrifugation is performed at 4000rpm followed by ultrafiltration through 0.025 and 0.002μm Millipore filters is indicated.AII-jSDBA reaction time of 2 days is indicated.C/DII-kREFAgitation is carried out in reaction time.A/BII-1REFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A			(but it is not clear by how much, due to the uncertainty in initial concentration	В
II-jSDBA reaction time of 2 days is indicated.C/DII-jSDBA reaction time of 2 days is indicated.C/DII-kREFAgitation is carried out in reaction time.A/BII-1REFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A			as well as in solubility data for Np).	
II-jSDBA reaction time of 2 days is indicated.C/DII-kREFAgitation is carried out in reaction time.A/BII-lREFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A	II-i	SDB	The centrifugation is performed at 4000rpm followed by ultrafiltration through	А
II-kREFAgitation is carried out in reaction time.A/BII-lREFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.AII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A			0.025 and 0.002µm Millipore filters is indicated.	
II-IREFNo isotherms, no variation of the Np concentration is indicated.C/DII-mREFThe experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment.C/DII-nREFThe reported error represents the standard deviation of the average of at least four experimental repetitions.A	II-j	SDB	A reaction time of 2 days is indicated.	C/D
II-m REF The experiments are carried out in polypropylene vessels. It is reported sorption to the vessel wall has been corrected for. No information about the correction procedure is available. It is assumed that the correction is based on the data obtained of a blank experiment. C/D II-n REF The reported error represents the standard deviation of the average of at least four experimental repetitions. A	II-k	REF	Agitation is carried out in reaction time.	A/B
II-n REF The reported error represents the standard deviation of the average of at least four experimental repetitions. A	II-l	REF	No isotherms, no variation of the Np concentration is indicated.	C/D
II-n REF The reported error represents the standard deviation of the average of at least four experimental repetitions. A	II-m	REF	The experiments are carried out in polypropylene vessels. It is reported	C/D
II-n REF The reported error represents the standard deviation of the average of at least four experimental repetitions. A			sorption to the vessel wall has been corrected for. No information about the	
II-n REF The reported error represents the standard deviation of the average of at least four experimental repetitions.			correction procedure is available. It is assumed that the correction is based on	
four experimental repetitions.			the data obtained of a blank experiment.	
	II-n	REF	The reported error represents the standard deviation of the average of at least	A
II-0 REF No parameter variation is indicated. D			four experimental repetitions.	
	II-o	REF	No parameter variation is indicated.	D

Data	Data table Np/8: REF: Mackenzie et al.(1983)				
JNC-S	SDB vers	ion 2 – DATA: Np /Mudstone, # 56147 – 56150			
GUID	ELINE:	Revision 4b (May 19, 2005)			
Checkpoint Evaluation		Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		about final solution composition is given.			
	REF	No further information is available.	unreliable		

Data	table Np	/9: REF: Sakamoto et al. (1990)	
JNC-	SDB vers	ion 2 – DATA: Np /Mudstone, # 58697 –58703	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	It is indicated that K _d values are scanned from a graphical presentation.	
	REF	The K_d –axis is in logarithmic form.	class 5
II-a	SDB	The solid phase is TAKADATE loam which is hydrated halloysite 40% and	
		quartz 60%. It is reported that CEC is measured between pH 2 and 10 by $\mathrm{NH_4^+}$	
		ion saturation (Wada and Okada, 1977). Specific surface area is not reported.	
		The loam contains a small amount of colored minerals such as mica, olivine	В
	REF	and feldspar.	
II-b	SDB	Ranges (1 pH-unit) of final pH values are given.	В
II-c	SDB	There is no information about Eh. The experiment is conducted under	
		atmospheric conditions. Experiments are carried out with Np(V).	A/B
	REF	No reducing agents are added to the experiment and no inert gas atmosphere is	
		reported.	
II-d	REF	The initial composition is indicated.	
		The final solution composition is not reported.	C/D
II-e	SDB	Temperatures of 30°C are reported.	A/B
II-f	SDB	Solution/solid ratios are 100mL/g.	C/D
	REF	Solid masses of 0.05g are added to a liquid volume of 5 mL.	
II-g	SDB	Based on the information given in the JNC-SDB, all sorption values are	А
		calculated to lie in the range 18.37- 87.50% sorbed.	
II-h	SDB	An initial [Np] of 1.0×10^{-5} M is indicated.	
	REF	Based on thermodynamic calculations with PHREEQC (database:	
		JNC-TDB_011213c2), the experiments are considered to be below the	
		solubility limit (but it is not clear by how much, due to the uncertainty in	В
		initial concentration as well as in solubility data for Np).	
II-i	SDB	The phase separation is carried out by centrifugation of 3000g for 20 min.	C/D
II-j	SDB	A reaction time of 7 days is indicated.	C/D
II-k	REF	Samples are shaken for 7 days at 30°C. This reaction time is reasonably long.	A/B
II-l	REF	No variation is indicated for either W/S or [Np].	C/D
II-m	REF	The loam is added to the solution in polyethylene container. The experiments	А
		without loam are also carried out to examine whether neptunium adsorbs to	

		the container wall.	
II-n	REF	Error of K_d is not reported, no replicated are performed.	D
II-o	REF	Only pH is varied.	С

Data	table Np	/10: REF: Sakamoto (1994)	
JNC-	SDB vers	ion 2 – DATA: Np /Mudstone, # 58718 – 58731	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	It is indicated that K_d values are scanned from a graphical presentation.	
	REF	A linear graph % sorbed K_d are provided.	class 4
II-a	SDB	It is reported that CEC is measured 10-30 meq/100g. Specific surface area is	
		reported to 130m ² /g.	
	REF	The solid phase is TAKADATE loam which is hydrated halloysite 40% and	В
		quartz 60%. The loam contains a small amount of colored minerals such as	
		mica, olivine and feldspar.	
II-b	SDB	Ranges (0.5 or 1 pH-unit) of final pH values are given.	В
II-c	SDB	It is indicated that no redox conditions are reported. There is no information	
		about Eh. Experiments are carried out with Np(V).	
	REF	No reducing agents are added to the experiment and no inert gas atmosphere is	
		reported. Therefore it is assumed that the experiment is conducted under	A/B
		ambient conditions.	
II-d	REF	The initial composition is indicated in Sakamoto et al. (1990).	
		The final solution composition is not reported.	C/D
II-e	SDB	Temperatures of 30°C are reported.	A/B
II-f	SDB	Solution/solid ratios are 100mL/g.	C/D
	REF	Solid masses of 0.05g are added to a liquid volume of 5 mL.	
II-g	SDB	Based on the information given in the SDB, all sorption values are calculated	А
		to lie in the range 15.97-85.07% sorbed.	
II-h	SDB	An initial [Np] of 6.0×10^{-6} M is indicated.	
	REF	Based on thermodynamic calculations with PHREEQC (database:	
		JNC-TDB_011213c2), the experiments are considered to be below the	
		solubility limit (but it is not clear by how much, due to the uncertainty in	В
		initial concentration as well as in solubility data for Np).	

II-i	SDB	The phase separation is carried out by centrifugation of 3000g for 20 min.	C/D
II-j	SDB	A reaction time of 7 days is indicated. This reaction time is reasonably long.	C/D
II-k	REF	Samples are shaken for 7 days at 30°C.	A/B
II-l	REF	No variation is indicated for either W/S or [Np].	C/D
II-m	REF	The loam is added to the solution in polyethylene container. The experiments	А
		without loam are also carried out to examine whether neptunium adsorbs to	
		the container wall.	
II-n	REF	Error of K _d is not reported, no replicate is performed.	D
II-o	REF	Only pH is varied.	С

Data	table Np	/11: REF: Tachi et al. (1999a)				
JNC-	SDB vers	tion 2 – DATA: Np /Mudstone, # 59864 – 59893				
GUII	GUIDELINE: Revision 4b (May 19, 2005)					
Chec	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b	SDB	K _d is given in tabular form.	class 1			
II-a	SDB	CEC and specific surface area are not reported.				
	REF	The sold phase is mudstone which is product at TOMAI in Miyagi prefecture.				
		Information about mineral composition and chemical composition are given.	C/D			
II-b	SDB	Initial as well as final pH are given.	А			
	REF	HCl or NaOH are used to adjust pH.				
II-c	SDB	Initial and final Eh values are indicated. It is further indicated that 0.01M				
		$Na_2S_2O_4$ is used as reducing agent. Experiments are performed under Ar	C/D			
		(O ₂ :<1ppm). The oxidation state of Np is not indicated. The initial oxidation				
		state of Np is not given.				
II-d	REF	The major initial solution composition is reported (Na ^{$+$} , SiO ₂ , Fe, SO ₄ ²⁻).	C/D			
II-e	SDB	Temperatures of 25°C are reported.	A/B			
II-f	SDB	Solution/solid ratios are 50mL/g.	A/B			
	REF	Solid masses of 2g are added to a liquid volume of 100mL.				
II-g	SDB	The %-sorbed can be calculated with K_d and W/S ratio:				
		-) Datapoints 59864-59866, 59879-59881: between 89.1% and 91.5% is	А			
		sorbed -) Datapoints 59867-59869, 59883-59884: between 97.5% and 97.9% is sorbed	В			

		-) Datapoints 59870-59878, 59882, 59885-59893: between 98.0% and 99.8% is sorbed	C/D
II-h	SDB	An initial [Np] of 2.0×10^{-9} M is indicated.	
	REF	Based on thermodynamic calculations with PHREEQC (database:	
		JNC-TDB_011213c2), the experiments are considered to be below the	
		solubility limit (but it is not clear by how much, due to the uncertainty in	В
		initial concentration as well as in solubility data for Np).	
II-i	SDB	The phase separation is carried out by filtration with 10000 MWCO	В
		membranes.	
II-j	SDB	A reaction time of 1, 7, 14, 21 and 35 days are indicated. Similar results are	
		obtained with 14, 21 and 35 days.	
		-) Datapoints: 59864-59869, 59879-59887	C/D
		-) Datapoints: 59870-59878, 59888-59893	A/B
II-k	REF	Samples are shaken for 1, 7, 14, 21 and 35 days at 30°C.	A/B
II-l	REF	No variation is indicated for either W/S or [Np].	C/D
II-m	REF	The mudstone is added to the solution in Teflon container. The experiments	А
		without mudstone are also carried out to examine whether neptunium adsorbs	
		to the container wall.	
II-n	REF	Error of K _d is not reported. 3 replicates are performed.	D
II-o	REF	It is indicated that contact time and Eh are varied.	В

Data	table Np	/12: REF: Tanaka and Muraoka (1999)				
JNC-S	JNC-SDB version 2 – DATA: Np /Mudstone, # 62312 - 62314					
GUID	GUIDELINE: Revision 4b (May 19, 2005)					
Check	Checkpoint Evaluation		Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
II-h	SDB	An initial [Np] of 1.62×10^{-3} M is indicated.				
	REF	Based on thermodynamic calculations with PHREEQC (database:				
		JNC-TDB_011213c2), the experiments are considered to be above the				
		solubility limit.	unreliable			

Data	table Np	/13: REF: Tanaka and Muraoka (1998)	
JNC-	SDB vers	ion 2 – DATA: Np /Mudstone, # 62353 - 62355	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	Tables with K_d are given.	class 1
II-a	SDB	CEC and chemical composition are reported about the ando soil, yellowish	
		soil and reddish soil.	
	REF	Humic substance and humic acid are reported. Surface area are not reported.	В
II-b	SDB	Initial pH values and final pH values (as range< 1 pH unit) are reported.	В
II-c	SDB	It is reported that experiments are done under air; i.e., under oxidizing conditions.	A/B
II-d	SDB	Physico-chemical properties of the equilibrated solution are reported.	A/B
II-e	SDB	A temperature of 25°C is reported.	A/B
II-f	SDB	Solution/solid ratio is 20mL/g.	A/B
	REF	The used mass of 0.5g soil is added to liquid volumes (de-ionized water) of	
		10mL. It is assumed that enough solid had been added to each vessel.	
II-g	SDB	The sorption value is between 14.53% and 67.74% (calculated with W/S and	А
		K _d values).	
II-h	SDB	An initial [Np] of 1.62×10^{-4} M is indicated.	
	REF	Based on thermodynamic calculations with PHREEQC (database:	
		JNC-TDB_011213c2), the experiments are considered to be below the	
		solubility limit (but it is not clear by how much, due to the uncertainty in	
		initial concentration as well as in solubility data for Np).	В
II-i	SDB	The 5000MWCO filters are used for phase separation.	А
II-j	SDB	A reaction time of 7 days is indicated.	
	REF	Identical (similar) results are obtained with different reaction times (>48	A/B
		hours).	
II-k	REF	The sample is shaking for 7 days.	A/B
II-l	REF	No isotherms, no variation of the Np concentration is indicated.	C/D
II-m	REF	The reaction vessel is made from teflon. Corrections for sorption on vessel	В
		walls are not mentions.	
II-n	SDB	Error of K_d is not reported, no replicated are performed.	D
II-o	REF	Only pH (end) is varied.	С

Data	table Np	/14: REF: Tanaka et al. (1999)				
JNC-S	JNC-SDB version 2 – DATA: Np /Mudstone, # 62805 - 62809					
GUID	GUIDELINE: Revision 4b (May 19, 2005)					
Check	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
		on phase separation is given.				
	REF	No further information is available.	unreliable			

Data	table Np	/15: REF: El-Naggar et al. (2000)			
JNC-S	SDB vers	ion 2 – DATA: Np /Mudstone, # 63027 - 63046			
GUID	GUIDELINE: Revision 4b (May 19, 2005)				
Check	spoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on Eh, contact time, initial Np concentration and final solution composition			
		are given.			
	REF	No further information is available. The Eh is not a sensitive parameter for Cs,			
		but the phase separation method and final solution composition are key points			
		which are missing in the reference.	unreliable		

Data	table Np	/16: REF: Kitamura et al. (2001)			
JNC-	JNC-SDB version 2 – DATA: Np /Bentonite, # 65073 – 65107				
GUIDELINE: Revision 4b (May 19, 2005)					
Chec	kpoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided.	Yes		
I-b	SDB	Tables with K_d are given.	class 1		
II-a	SDB	The solid phase is reported as a purified Na-Smectite. The bentonite sample of	А		
		Kunipia-F(Kunimine Industries Co.), which is commercially purified from			
		Kunigel-V, is used as a starting material for purification of smectite.			
II-b	SDB	Final pH values are reported.	А		
II-c	SDB	Experiments are performed under N ₂ . The oxidation state of Np is indicated as			
		+IV or +V. Final Eh values are provided, as well as the reducing agent used			

		$(0.01 \text{M Na}_2\text{S}_2\text{O}_4)$. Reagent of NaHCO ₃ is also added for ranging a total	
		carbonate concentration from 0.09M to 1.0M.	C/D
	REF	O ₂ levels are reportedly <1 ppm. A TTA-extraction method is applied to the	
		determination of the oxidation state of neptunium.	
II-d	SDB	A NaCl and NaHCO ₃ are reported as input solution; final solution	
		compositions are reported in the additional file "solution composition".	A/B
II-e	SDB	A temperature of $22 \pm 3^{\circ}$ C is reported.	A/B
II-f	SDB	A solution/solid ratio of 100 mL/g is reported.	
		0.2 g solid are added to 20 mL solution.	
		From Shibutani et al.(1999), the BET (outer) surface areas are 50 - 58 m^2/g for	A/B
		a purified Na-Smectite.	
II-g	SDB	Based on the information given in the JNC-SDB, and on a solution/solid ratio	
		that may range from 100 mL/g, sorption values between 1.3 - 72.9 %	
		sorbed can be calculated.	
		• Datapoints 65073-65088, 65090, 65092- 65103, 65105-65107	А
		• Datapoints 65089, 65091	В
		• Datapoint 65104	C/D
II-h	SDB	Initial Np concentration is 1.0×10^{-6} M.	
	REF	Based on calculations (Kitamura and Tomura, 2003) using the thermodynamic	
		data (in JNC-TDB and OECD/NEA-TDB) and the data given in Yamaguchi et	
		al.(1991) as a function of pH and dissolved carbonate concentration, it is	
		estimated that initial Np concentration is considered to be below the solubility	
		limit (but it is not clear by how much, due to the uncertainty in initial	В
		concentration as well as in solubility data for Np).	
II-i	SDB	10000MWCO filters are used for phase separation. This method is considered	В
		to be efficient for removal of colloids.	
II-j	SDB	It is reported that contact time is 16 or 42 days for different datapoints. Similar	A/B
		results are obtained with different reaction times.	
II-k	REF	The Samples are gently shaken for 16 or 42 days (during the entire	A/B
		experiment) to attain sorption equilibrium.	
II-l	SDB	All experiments are presumably conducted at the same S/W and initial RN	C/D
		concentration.	
II-m	REF	The experiments are carried out in polypropylene tubes. And corrections for	В
		sorption on vessel walls have not been performed.	
II-n	SDB	No information is given with regard to errors. Error estimates are not reported.	D
	REF	It is not clear, whether repeated sampling is done.	
		•	•

0.00M to 1.0M) are varied	II-o	SDB	Reaction time (16 days or 42days) and total carbonate concentration (from	В
			0.09M to 1.0M) are varied.	

Data	table Np	/17: REF: Kitamura and Tomura (2003)	
JNC-	SDB vers	ion 2 – DATA: Np /Bentonite, # 64869 – 64974	
GUIE	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	Tables with K _d are given.	class 1
II-a	SDB	The solid phase is reported as a purified Na-Smectite. The bentonite sample of	А
		Kunipia-F(Kunimine Industries Co.), which is commercially purified from	
		Kunigel-V, is used as a starting material for purification of smectite.	
II-b	SDB	Initial pH ranges and final pH values are reported.	А
II-c	SDB	Experiments are performed under N_2 . The oxidation state of Np is reported as	
		+IV or +V. Final Eh values are provided, as well as the reducing agent used	
		$(0.01M Na_2S_2O_4)$ or nothing. Reagent of NaHCO ₃ is also added for ranging a	
		total carbonate concentration from 0.04M to 1.0M.	
	REF	O ₂ levels are reportedly <1 ppm. A TTA-extraction method is applied to the	
		determination of the oxidation state of neptunium.	
		-) Datapoints 64955 –64974	A/B
		-) Datapoints 64869 –64954	C/D
II-d	SDB	NaCl and NaHCO ₃ are reported as input solution; main solution compositions	
		are indicated in the additional file "solution composition".	C/D
II-e	SDB	A temperature of 25°C is reported.	A/B
II-f	SDB	A solution/solid ratio of 100, 1000 and 20 mL/g are reported.	
	REF	0.02g, 0.2g and 1 g solid are added to 20 mL solution. The 0.02g solid is not	
		satisfactory for reproducibility and representativeness.	
		From Shibutani et al.(1999), the BET (outer) surface areas are $50-58 \text{ m}^2/\text{g}$ for	
		a purified Na-Smectite.	
		-) Datapoints 64869–64904, 64930–64974	A/B
		-) Datapoints 64905–64929	C/D
II-g	SDB	Solution/solid ratios are reported; 20, 100 and 1000 mL/g, sorption values	
		between $0.2 - 97.1$ % sorbed can be calculated.	

		• Datapoints: 64869, 64925, 64963	В
		• Datapoints: 64870–64924, 64926–64962, 64964–64972, 64974	А
		• Datapoints: 64973	C/D
II-h	SDB	Initial Np concentration is 1.3×10^{-6} (in the presence of air), 1.3×10^{-7} and	
		6.3×10^{-8} M. Based on calculations (Kitamura and Tomura, 2003) using the	
		thermodynamic data (in JNC-TDB and OECD/NEA-TDB) and the data given	
		from Neck et al.(1992) in the presence of air, it is estimated that initial Np is at	
		least five-fold below the solubility in the samples with pH \leq ca. 10(except the	
		case in the presence of air).	
		-) Datapoints: 64869–64954	А
		-) Datapoints: 64955–64974	В
II-i	SDB	10000MWCO filters are used for phase separation. And membrane filtrations	А
		with pore size of $0.45 \mu m$ and $3000 MWCO$ filters are partly used in order to	
		compare with samples. There is the direct proof about absence of important	
		colloid effects.	
II-j	SDB	It is indicated that contact time is 14, 15, 28, 29, 30, 92 or 156 days for	A/B
		different datapoints. Similar results are broadly obtained with different	
		reaction times.	
II-k	REF	Samples are shaken by orbital shaker at 70 rpm for 14, 15, 28, 29, 30, 92 or	A/B
		156 days (during the entire experiment).	
II-l	SDB	A limited variation of initial Np concentration has been used, including some	В
		experiments at trace concentrations.	
II-m	REF	The experiments are carried out in polypropylene tubes. And corrections for	В
		sorption on vessel walls have not been performed.	
II-n	SDB	Uncertainties in K_d are derived based on duplicate experiments. Uncertainties	
		in K_d of the case in the presence of air are derived based on experiments.	
		-) Datapoints: 64869–64954	А
		-) Datapoints: 64955–64974	В
II-o	SDB	Solid/water ratio, reaction time and total carbonate concentration are varied.	В

3.1.3 Selenium

Data t	able Se/	1: REF: Ticknor et al. (1988)				
JNC-S	DB vers	ion 2 – DATA: Se/Bentonite, # 60751-60759				
GUIDI	GUIDELINE: Revision 4b (May 19, 2005)					
Check	point	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed	Yes			
I-a.2	SDB	Separation of the samples is not reported.				
	REF	It is reported, that aliquots of the aqueous phases are taken for the radiometric	Yes			
		analysis.				
		Phase separation is performed by high-speed centrifugation (oral				
		communication with T. T. Vandergraaf).				
I-b	SDB	A table with log K _d values is provided.	class 1			
II-a	SDB	As solid phase, Kaolinite is indicated and a detailed chemical composition of				
		the mineral is available. Surface characteristics (CEC and specific surface	А			
		area) are reported.				
	REF	It is reported that the Kaolinite is used as received from the supplier (KGa-1				
		from Source Clays Repository of the Clay Minerals Society, Columbia,				
		Missouri).				
II-b	SDB	Initial as well as final pH values are reported.	В			
	REF	The solution pH is determined with litmus paper. No pH correction with				
		buffer is performed after hydrazine addition.				
II-c	SDB	Redox conditions are reported for each experiment. Experiments are				
		performed under aerobic conditions, under nitrogen and under reducing				
		conditions (addition of hydrazine under nitrogen atmosphere). All				
		experiments are conducted in pre-equilibrated systems. Initial and final Eh				
		values are given in the SDB.				
		• # 60753-60755 (aerobic conditions, Se(IV))	A/B			
		• # 60756-60758 (nitrogen atmosphere, Se(IV))	A/B			
		• # 60751, 60752 and 60759 (Se(-II), hydrazine, nitrogen atmosphere).				
		Only one reducing agent is used.	C/D			
II-d	SDB	As water type, 1%, 10% and 100% SCSSS (Standard Canadian Shield saline				
		solution, synthetic brine) is indicated. Under reducing conditions, the				
		solutions are 0.08 M in hydrazine.				
	REF	Final solution compositions are not given, but it is supposed that no essential	A/B			

		changes take place with 10% and 100% SCSSS.	
		Chlorite and muscovite are reported to be of high purity and final solution	C/D
		compositions with the reported S/W and 1% SCSSS can be estimated	
		approximately.	
II-e	SDB	A temperature of 22°C is indicated.	A/B
II-f	SDB	A liquid/solid ratio of 25 mL/g is indicated.	
	REF	It is reported that standard-scale experiments used 10 mL of solution and 0.4	A/B
		g of solid.	
II-g	SDB	The sorption value (calculated from K_d and S/W ratios) ranges from 25% to 68%.	А
II-h	SDB	Initial Se concentrations between 4.43×10^{-11} and 8.84×10^{-11} M are given.	А
		Based on speciation calculations (for 25°C) with PHREEQC 2.12.5 using the	
		thermodynamic data in JNC-TDB_011213c2 the initial Se concentrations are	
		below the solubility limit.	
II-i	REF	It is indicated that no specific separation is reported.	
		High-speed centrifugation of the samples is used for phase separation (oral	В
		communication with T. T. Vandergraaf).	
II-j	SDB	A reaction time of 14 days is indicated.	C/D
	REF	No further (kinetic) information is provided.	
II-k	REF	The samples are intermittently agitated.	A/B
II-l	SDB	No variation in S/W or initial Se concentration is indicated.	C/D
II-m	REF	The experiments are carried out in polypropylene centrifuge tubes. Sorption	C/D
		on vessel walls is not tested and no corrections had been done.	
II-n	REF	No error information is available.	D
II-o	SDB	Only the SCSSS salinity is varied.	С

Data table Se/2: REF: Ticknor et al. (1988)						
JNC-SDB version 2 – DATA: Se/Other Minerals, # 60706-60714 (Se/Chlorite) and # 60760-60768						
(Se/Mu	(Se/Muscovite)					
GUIDI	GUIDELINE: Revision 4b (May 19, 2005)					
Checkpoint		Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed	Yes			
I-a.2	SDB	Separation of the samples is not reported.				
	REF	It is reported, that aliquots of the aqueous phases are taken for the				
		radiometric analysis. Phase separation is performed by high-speed				

		centrifugation (oral communication with T. T. Vandergraaf).	Yes
I-b	SDB	A table with log K_d values is provided.	class 1
II-a	SDB	As solid phases, chlorite and muscovite are indicated and a detailed mineral	А
		composition is listed. Surface characteristics (CEC values and specific	
		surface areas) are reported.	
	REF	It is indicated that chlorite and muscovite are obtained as bulk mineral	
		samples from Wards Natural Science Limited. The samples are further	
		treated. It is indicated that chlorite and muscovite are of high purity.	
II-b	SDB	Initial as well as final pH values are indicated.	В
	REF	The solution pH is determined with litmus paper. No pH correction with	
		buffer is performed after hydrazine addition.	
II-c	SDB	Redox conditions are reported for each experiment. Experiments are	
		performed under aerobic conditions, under nitrogen and under reducing	
		conditions with addition of hydrazine under nitrogen atmosphere. All	
		experiments are conducted in pre-equilibrated systems. Initial and final Eh	
		values are given in the JNC-SDB.	
		 # 60708, 60709, 60712, 60761, 60764, 60768 (aerobic conditions, Se(IV)) 	A/B
		 # 60707, 60710, 60714, 60760, 60765, 60767 (nitrogen atmosphere, Se(IV)) 	A/B
		• # 60706, 60711, 60713, 60762, 60763, 60766 (hydrazine, nitrogen atmosphere, Se(-II), only one reducing agent is indicated)	C/D
II-d	SDB	As water type, 1%, 10% and 100% SCSSS (Standard Canadian Shield saline	
		solution, synthetic brine) is indicated. Under reducing conditions, the	
		solutions are 0.08 M in hydrazine.	
		Final solution compositions are not given, but it is supposed that no essential	
		changes take place with 10% and 100% SCSSS.	A/B
		Chlorite and muscovite are reported to be of high purity and final solution	
		compositions with the reported S/W and 1% SCSSS can be estimated	
		approximately.	C/D
II-e	SDB	A temperature of 22°C is indicated.	A/B
II-f	SDB	A liquid/solid ratio of 25 mL/g is indicated.	
	REF	It is reported that standard-scale experiments used 10 mL of solution and 0.4	A/B
		g of solid.	
II-g	SDB	The sorption value (calculated from K_d and S/W ratios) ranges between 8% and 75%.	А

II-h	SDB	Initial Se concentrations between 4.43×10^{-11} and 1.21×10^{-10} are given. Based	
		on speciation calculations (for 25°C) with PHREEQC 2.12.5 using the	
		thermodynamic data in JNC-TDB_011213c2 the initial Se concentrations are	
		below the solubility limit.	А
II-i	SDB	It is indicated that no specific separation is reported.	
	REF	High-speed centrifugation of the samples is used for phase separation (oral	
		communication with T. T. Vandergraaf).	В
II-j	SDB	A reaction time of 14 days is indicated.	C/D
	REF	No further (kinetic) information is provided.	
II-k	REF	The samples are intermittently agitated.	A/B
II-l	SDB	No variation in S/W or initial Se concentration is indicated.	C/D
II-m	REF	The experiments are carried out in polypropylene centrifuge tubes. Sorption	C/D
		on vessel walls is not tested and no corrections had been done.	
II-n	REF	No error information is available.	D
II-o	SDB	Only the SCSSS salinity is varied.	С

Data t	able Se/	3: REF: Ticknor et al. (1988)				
JNC-S	INC-SDB version 2 – DATA: Se/Other Minerals, Illite-Muscovite, #67055-67063					
GUIDI	ELINE:	Revision 4b (May 19, 2005)				
Check	point	Evaluation	Rating			
I-a.1	SDB	-	-			
I-a.2	REF	It is reported, that aliquots of the aqueous phases are taken for the	Yes			
		radiometric analysis.				
		Phase separation is performed by high-speed centrifugation (oral				
		communication with T. T. Vandergraaf).				
I-b	REF	A table with log K_d values is provided.	class 1			
II-a	REF	As solid phase illite-muscovite obtained from a drill core (Cigar Lake,	В			
		Saskatchewan) is reported. A detailed mineral composition of the mineral				
		sample and its specific surface area are available. The sample is crushed but				
		not purified.				
II-b	REF	Initial as well as final pH values are reported. The solution pH is determined	В			
		with litmus paper. No pH correction with buffer is performed after hydrazine				
		addition.				
II-c	REF	Redox conditions are reported for each experiment. Experiments are				
		performed under aerobic conditions, under nitrogen and under reducing				

		conditions with addition of hydrazine under nitrogen atmosphere. All	
		experiments are conducted in pre-equilibrated systems. Initial and final Eh	
		values are given.	4 /D
		• aerobic conditions, Se(IV): # 67055, 67056, 67061	A/B
		• nitrogen atmosphere, Se(IV): # 67057, 67058, 67062	A/B
		 hydrazine, nitrogen atmosphere (only one reducing agent has been used): # 67059, 67060, 67063 	C/D
II-d	REF	As water type, 1%, 10% and 100% SCSSS (Standard Canadian Shield saline	
		solution, synthetic brine) is reported. Under reducing conditions, the	
		solutions are 0.08 M in hydrazine.	
		Final solution compositions are not given, but it is supposed that no essential	
		changes take place with 10% and 100% SCSSS. : #67055-67060	A/B
		Chlorite and muscovite are reported to be of high purity and final solution	
		compositions with the reported S/W and 1% SCSSS can be estimated	
		approximately. : #67061-67063	C/D
II-e	REF	A temperature of 22°C is indicated.	A/B
II-f	REF	It is reported that standard-scale experiments used 10 mL of solution and 0.4	A/B
		g of solid. This corresponds to a liquid/solid ratio of 25 mL/g.	
II-g	REF	Sorption values (calculated from K_d and S/W ratios) range from 19% to 58%.	А
II-h	REF	Initial Se concentrations between 5.65×10^{-11} and 8.56×10^{-11} M are given.	А
		Based on speciation calculations (for 25°C) with PHREEQC 2.12.5 using the	
		thermodynamic data in JNC-TDB_011213c2 the initial Se concentrations are	
		below the solubility limit.	
II-i	REF	High-speed centrifugation of the samples is used for phase separation (oral	В
		communication with T. T. Vandergraaf).	
II-j	REF	A reaction time of 14 days is indicated. This reaction time is reasonably long.	C/D
		No further (kinetic) information is provided.	
II-k	REF	The samples are intermittently agitated.	A/B
II-l	REF	No variation in S/W or initial Se concentration is indicated.	C/D
II-m	REF	The experiments are carried out in polypropylene centrifuge tubes. Sorption	C/D
		on vessel walls is not tested and no corrections had been done.	
II-n	REF	No error information is available.	D
II-o	SDB	Only the SCSSS salinity is varied.	С

Data	Data table Se/4: REF: Fujikawa and Fukui (1997b)					
JNC-S	JNC-SDB version 2 – DATA: Se /Mudstone, # 51979 - 52002					
GUID	GUIDELINE: Revision 4b (May 19, 2005)					
Checkpoint		Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
		on Eh, separation method and final solution composition are given.				
	REF	No further information is available.	unreliable			

Data	table Se/	5: REF: Legoux et al. (1992)				
JNC-	JNC-SDB version 2 – DATA: Se/Mudstone, # 56039 - 56042					
GUII	GUIDELINE: Revision 4b (May 19, 2005)					
Chec	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b	SDB	Tables with K_d are given.	class 1			
II-a	SDB	The solid phase is indicated as soil (A, B, C and D).	А			
		Mineralogical composition, specific surfaces and cationic exchange capacity				
		(CEC) are reported.				
II-b	SDB	Initial pH values and final pH values are reported.	А			
II-c	SDB	It is indicated that no redox conditions are reported. There is no information				
		about Eh. Redox of Se is reported VI.				
	REF	Se is in solution in the forms of $SeO_4^{2^2}$. No reducing agents are added to the	C/D			
		experiment and no inert gas atmosphere is reported. Therefore it is assumed				
		that the experiment is conducted under ambient conditions.				
II-d	REF	The final solution composition is reported.	A/B			
II-e	SDB	A temperature is not reported.	C/D			
II-f	SDB	It is indicated that a W/S ratio of 10 mL/g had been used. The specific solid				
		mass (1g) and solution volume (10ml) are reported.	A/B			
	REF	The soil mass of 1g has a surface area of $6.6-14.7 \text{m}^2$.				
II-g	SDB	The sorption value is between 19.4% and 93.0% (calculated with W/S and $K_{\rm d}$	А			
		values).				
II-h	SDB	The value of initial concentration is 2.0×10^{-7} M.				
		Based on thermodynamic calculations with PHREEQC (database:				
		JNC-TDB_011213c2), the experiments are considered to be below the				

solubility limit (but it is not clear by how much, due to the uncertainty in initial concentration as well as in solubility data for Se). -) Datapoints: 56039, 56041, 56042BAnd it is estimated that the initial Se concentration may have been at, or slightly above, the solubility limit. -) Datapoints: 56040unreliableII-iSDBUltrafiltration through 0.002μm and 0.025μm and membranes, as well as centrifugation at 4000rpm is reported.AII-jSDBIt is indicated that contact time is 2 days. This reaction time is reasonably long.C/D
 -) Datapoints: 56039, 56041, 56042 And it is estimated that the initial Se concentration may have been at, or slightly above, the solubility limit. -) Datapoints: 56040 II-i SDB Ultrafiltration through 0.002µm and 0.025µm and membranes, as well as centrifugation at 4000rpm is reported. II-j SDB It is indicated that contact time is 2 days. This reaction time is reasonably C/D
And it is estimated that the initial Se concentration may have been at, or slightly above, the solubility limit. -) Datapoints: 56040unreliableII-iSDBUltrafiltration through 0.002µm and 0.025µm and membranes, as well as centrifugation at 4000rpm is reported.AII-jSDBIt is indicated that contact time is 2 days. This reaction time is reasonablyC/D
slightly above, the solubility limit. unreliable -) Datapoints: 56040 unreliable II-i SDB Ultrafiltration through 0.002µm and 0.025µm and membranes, as well as centrifugation at 4000rpm is reported. A II-j SDB It is indicated that contact time is 2 days. This reaction time is reasonably C/D
II-iSDBUltrafiltration through 0.002μm and 0.025μm and membranes, as well as centrifugation at 4000rpm is reported.AII-jSDBIt is indicated that contact time is 2 days. This reaction time is reasonablyC/D
II-i SDB Ultrafiltration through 0.002μm and 0.025μm and membranes, as well as centrifugation at 4000rpm is reported. A II-j SDB It is indicated that contact time is 2 days. This reaction time is reasonably C/D
centrifugation at 4000rpm is reported. II-j SDB It is indicated that contact time is 2 days. This reaction time is reasonably C/D
II-j SDB It is indicated that contact time is 2 days. This reaction time is reasonably C/D
long.
II-k REF Samples are agitated continuously. A/B
II-ISDBNo variation of initial [Se] or S/W has been carried out.C/D
II-m REF The experiments are carried out in polypropylene vessels. It is reported C/D
sorption to the vessel wall has been corrected for. No information about the
correction procedure is available. It is assumed that the correction is based on
the data obtained of a blank experiment.
II-n SDB The reported error represents the standard deviation of the average of at least A
four experimental repetitions.
II-o SDB No parameter variation is indicated. D

3.1.4 Cesium

Data	Data table Cs/1: REF: Barney and Anderson (1979)					
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 44256 – 44261					
GUID	GUIDELINE: Revision 4b (May 19, 2005)					
Check	Checkpoint Evaluation Rating					
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
		on "Adjustment and control of pH", final solution composition and				
		atmosphere/ redox condition are given.				
	REF	No further information is available.	unreliable			

Data	Data table Cs/2: REF: Barney and Brown (1979)					
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 44331 – 44333					
GUID	GUIDELINE: Revision 4b (May 19, 2005)					
Checkpoint Evaluation		Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
		on final solution composition is given.				
	REF	No further information is available.	unreliable			

Data	Data table Cs/3: REF: Daniels (1981)					
JNC-	JNC-SDB version 2 – DATA: Cs /Mudstone, # 50063 – 50078					
GUII	DELINE:	Revision 4b (May 19, 2005)				
Chec	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b	SDB	Tables with K_d are given.	class 1			
II-a	SDB	The solid phase is reported as Eleana argillite.	C/D			
	REF	Additional Eleana argillite is crushed in the controlled-atmosphere boxes.				
		Physico- chemical properties and specific surface are not reported.				
II-b	SDB	Initial pH value and final pH value are reported.				
		-) Datapoints: 50063, 50064, 50066, 50071, 50073, 50074	А			

		Initial pH value is reported.	
		-) Datapoints: 50065, 50067-50070, 50072, 50075-50078	С
II-c	SDB	It is reported that experiments are done under atmosphere and controlled	A/B
		atmosphere. Also, Cs is not a redox-sensitive element.	
II-d	SDB	The final composition of the solution is not reported.	
	REF	Composition of water obtained in the field (groundwaters) is described in	
		Erdal et al.(1979b): Los Alamos Scientific Laboratory report LA-7455-MS.	C/D
II-e	SDB	A temperature of 32°C or room temperature is reported.	A/B
II-f	SDB	Solution/solid ratios are not reported.	C/D
	REF	The $<75\mu$ m and $75-500\mu$ m fractions are used for the controlled- atmosphere	
		studies. Several different particle-size fractions are used under atmospheric	
		conditions.	
II-g	SDB	The sorption value cannot be calculated, because the Solution/solid ratios are	C/D
		lacking.	
II-h	SDB	The initial Cs concentration varied between 1.95×10^{-10} mol/L and 2.42×10^{-3}	
		mol/L. Some initial [Cs] is not reported.	А
	REF	However, Cs is not solubility limited.	
II-i	SDB	The membrane filtration with pore size of 0.45µm is used.	
		-) Datapoints: 50065, 50067-50070, 50072, 50075-50078	C/D
		No information is given about the separation method for some experiments.	
		-) Datapoints: 50063, 50064, 50066, 50071, 50073, 50074	unreliable
II-j	SDB	It is indicated that contact time is 20.63 days, 21 days, 41.63 days or	
		91.63days for different datapoints. This reaction time is reasonably long.	
	REF	The kinetics of sorption is not tested. The K_d values on argillite increased	C/D
		slightly with time.	
II-k	REF	The batch sorption experiments are performed by shaking. It is indicated in	A/B
		Erdal et al.(1979b): Los Alamos Scientific Laboratory report LA-7455-MS.	
II-l	REF	A limited variation of initial Cs concentration has been used, including some	В
		experiments at trace concentrations.	
II-m	REF	The experiments are carried out in polyethylene centrifuge tubes. It is	В
		indicated in Erdal et al.(1979b): Los Alamos Scientific Laboratory report	
		LA-7455-MS.	
II-n	SDB	Uncertainties are not reported.	D
II-o	SDB	The initial pH, reaction time, initial Cs concentration and temperature are	В
		varied.	

Data	Data table Cs/4: REF: Duursma (1973)					
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 50310 – 50341					
GUID	ELINE:	Revision 4b (May 19, 2005)				
Check	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
	REF	on phase separation is given. No further information is available.	unreliable			

Data	table Cs	/5: REF: Erdal et al. (1977)	
JNC-	SDB vers	ion 2 – DATA: Cs /Mudstone, # 50648 – 50662	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided (the initial Cs concentration is not	Yes
		provided: see later comment at checkpoint II-h).	
I-b	SDB	Tables with K_d are given.	class 1
II-a	SDB	Alluviums are reported as mineral phase (Hoffman et al., 1977:	В
		LA-6877-MS). CEC (10 to 17 meq/100 g) and surface area (6.11 to 20.2 m^2/g)	
		are provided in table.	
II-b	SDB	Initial pH value and final pH range (from 8.3 to 8.7) are reported.	А
II-c	SDB	It is reported that experiments are done under oxidizing conditions. Moreover,	A/B
		Cs is not a redox-sensitive element.	
II-d	REF	The initial composition of the groundwater is reported.	
		The final solution composition is not reported, but can be estimated based on	C/D
		mineral and initial solution composition.	
II-e	SDB	A temperature of 20°C or 70°C is reported.	A/B
II-f	SDB	Solution/solid ratio is 20mL/g.	A/B
	REF	The used mass of 1g alluvium has a surface area of about 6 to $20m^2$.	
II-g	SDB	The sorption value is between 99.26% and 99.81% (calculated with W/S and	C/D
		K _d values).	
II-h	SDB	Initial [Cs] is not reported. However, Cs is not solubility limited.	А
II-i	SDB	The centrifugation is performed at 16000rpm for 1 hour.	
	REF	The supernatant is then centrifuged twice prior to measuring the Cs activity.	В
II-j	SDB	A reaction time of between 2.8 days and 41.8 days is indicated.	

	REF	The kinetics of sorption is tested. No difference between 2.8 days and 41.8	A/B
		days.	
II-k	REF	Stirring is used as agitation method in reaction time.	C/D
II-l	REF	No isotherms, no variation of the Cs concentration is indicated.	C/D
II-m	REF	The reaction vessel is a polyethylene tube. No corrections for sorption on	В
		vessel walls are reported.	
II-n	REF	No replicates and no error estimates are reported.	D
II-o	REF	No parameter variation.	D

Data	Data table Cs/6: REF: Erdal et al. (1979b)				
JNC-	JNC-SDB version 2 – DATA: Cs /Mudstone, # 50837 – 50858				
GUII	DELINE:	Revision 4b (May 19, 2005)			
Chec	kpoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided.	Yes		
I-b	SDB	Tables with K _d are given.	class 1		
II-a	SDB	Four different samples of the Eleana argillite are obtained from hole at the			
		Nevada Test Site. The samples had all been wrapped in aluminum foil and			
		waxed when they are originally taken. A petrographic description of the			
		fractions is given in Table. CEC (8 meq/100 g to 14 meq/100 g) and Surface			
		area (5.22 m ² /g to 12.5 m ² /g) are provided in table.	В		
II-b	SDB	Initial pH value and final pH are reported.			
		-) Datapoints: 50839, 50842, 50843, 50846-50848			
		Final pH is reported.			
		-) Datapoints: 50837, 50838, 50840, 50841, 50844, 50845, 50849-50858			
		After contact tine of 56-60 days, it can be assumed to be enough for reaching			
		equilibrium. However, pH is not a critical parameter for Cs.	А		
II-c	SDB	It is reported that experiments are done under aerobic conditions. Moreover,	A/B		
		Cs is not a redox-sensitive element.			
II-d	REF	The initial composition of the groundwater is reported.			
		The final solution composition is not reported, but can be estimated based on	C/D		
		mineral and initial solution composition.			
II-e	SDB	A temperature of 22°C or 70°C is reported.	A/B		
II-f	SDB	Solution/solid ratio is 20mL/g.			
	REF	The used mass of 1g argillite has a surface area of about 5.22 to $12.5m^2$.	A/B		

II-g	SDB	The %-sorbed can be calculated with K _d values and W/S ratio:	
		-) Datapoint 50839: 97.45%	В
		-) Datapoints 50837, 50838, 50840-50858: between 98.01% and 99.5%	C/D
II-h	SDB	Initial [Cs] is reported 1.4×10 ⁻⁹ M. Cs is not solubility limited.	А
II-i	REF	The centrifugation is performed at 16000rpm for one hour.	В
II-j	SDB	A reaction time of between 56 days and 60 days is indicated. This reaction	
		time is reasonably long.	C/D
II-k	REF	The solid sample is dispersed with vigorous shaking, and the mixture is	A/B
		agitated gently for given time. The shaking rates are 200 oscillations per	
		mimute for the ambient temperature studies, and 80 oscillations per minute for	
		the 70° C samples.	
II-l	REF	No isotherms, no variation of the Cs concentration is indicated.	C/D
II-m	REF	The reaction vessel is a polyethylene tube. It is reported that cesium did not	А
		adsorb on the container walls.	
II-n	REF	One replicate and no error estimates are reported.	D
II-o	REF	Only temperature is varied.	С

Data	table Cs/	/7: REF: Erdal et al. (1979d)			
JNC-	JNC-SDB version 2 – DATA: Cs /Mudstone, # 50932 – 50934				
GUII	DELINE:	Revision 4b (May 19, 2005)			
Chec	kpoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided (the initial Cs concentration is not	Yes		
		provided: see later comment at point II-h).			
I-b	SDB	Tables with K_d are given.	class 1		
II-a	REF	Alluviums and Argillite are reported as mineral phase (Hoffman et al., 1977:			
		LA-6877-MS). CEC and Surface area about alluviums are provided in table			
		(Erdal et al. ,1977).			
		-)Datapoints: 50932, 50933	C/D		
		-)Datapoint: 50934	В		
II-b	SDB	Neither initial nor final pH values are reported.			
	REF	There is no further information about pH conditions. However, pH is not a			
		critical parameter for Cs.	D		
II-c	SDB	It is reported that experiments are done under atmosphere. Moreover, Cs is not			
	REF	a redox-sensitive element.	A/B		

			1
II-d	SDB	The initial composition of the groundwater is not reported.	
	REF	Composition of water obtained in the field (groundwaters) is described in	C/D
		Erdal et al. (1979b).	
II-e	SDB	A temperature of 22°C or 70°C is reported.	A/B
II-f	SDB	Solution/solid ratio is 20mL/g.	A/B
	REF	The used mass of 1g alluvium has a surface area of about $6m^2$ to $20m^2$.	
II-g	SDB	The sorption value is between 98.96% and 99.72% (calculated with W/S and	C/D
		K _d values).	
II-h	SDB	Initial [Cs] is not reported.	
	REF	No further information. However, Cs is not solubility limited.	А
II-i	SDB	The centrifugation is performed at 16000rpm for one hour.	В
II-j	SDB	A reaction time of 7 days, 14 days, 28 days or 56 days is indicated.	
	REF	These reaction times are reasonably long.	C/D
II-k	REF	The sample is dispersed with vigorous shaking, and mixture is agitated gently	
		for a given time. The shaking rates are 200 oscillations per minute for the	
		ambient temperature studies, and 80 oscillations per minute for the 70° C	
		samples.	A/B
II-l	REF	No isotherms, no variation of the Cs concentration is indicated.	C/D
II-m	REF	The reaction vessel is a polyethylene tube. It is reported that cesium did not	А
		adsorb on the container walls.	
II-n	REF	No replicates and no error estimates are reported.	D
II-o	REF	Temperature and reaction time are varied.	В

Data	Data table Cs/8: REF: Erdal (1980)					
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 51200 – 51225					
GUID	ELINE:	Revision 4b (May 19, 2005)				
Checkpoint		Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
	REF	on final solution composition is given. No further information is available.	unreliable			

Data	Data table Cs/9: REF: Fujikawa and Fukui (1997a)					
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 51763 – 51786					
GUID	ELINE:	Revision 4b (May 19, 2005)				
Check	Checkpoint Evaluation Rat					
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
		on Eh, separation method and final solution composition are given.				
	REF	No further information is available. The Eh is not a sensitive parameter for Cs,				
		but the phase separation method and final solution composition are key points				
		which are missing in the reference.	unreliable			

Data	Data table Cs/10: REF: Higgo et al.(1987)					
JNC-	JNC-SDB version 2 – DATA: Cs /Mudstone, # 53224 – 53248					
GUII	DELINE:	Revision 4b (May 19, 2005)				
Chec	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b	SDB	Tables with K _d are given.	class 1			
II-a	SDB	Major and minor mineralogy is not known. Specific surface area and CEC are	C/D			
		reported. It is not clear whether impurities are determined.				
II-b	SDB	Initial pH values are indicated.	С			
	REF	No use of any pH-buffer is reported.				
II-c	SDB	Experiments are carried out at aerobic condition. Cs is not redox sensitive.	A/B			
II-d	SDB	Seawater is indicated.				
	REF	The critical major components are approximately Known. Minor components	C/D			
		may be unknown. Charge balance cannot be calculated, but seawater can be				
		expected to be well poised.				
II-e	SDB	A temperature of 4°C is reported.	A/B			
II-f	SDB	Solution/solid ratios are not reported.	C/D			
II-g	SDB	The sorption value cannot be calculated, because the Solution/solid ratios are	C/D			
		lacking.				
II-h	SDB	Initial [Cs] ranged from 3.4×10^{-9} M to 1.0×10^{-3} M. Cs is not solubility limited.	А			
II-i	SDB	The centrifugation (7000rpm / 90 min) and followed by filtration through 0.1	C/D			
		and 0.22µm Millipore filters is indicated.				

II-j	SDB	A reaction time is 45days. This reaction time is reasonably long.	C/D
II-k	REF	Agitation is carried out by shaken at 120 strokes/min.	A/B
II-l	REF	The variation of initial Cs concentration is indicated.	В
II-m	REF	The quality of reaction vessel is not reported.	C/D
II-n	SDB	No replicates are reported. But error estimates are reported.	С
	REF	Uncertainties are based on error propagation, not on replicate measurements.	
II-o	REF	Initial Cs concentration is varied.	С

Data	table Cs/	/11: REF: Konishi et al. (1988)			
JNC-	JNC-SDB version 2 – DATA: Cs / Mudstone, # 54009 – 54012				
GUII	DELINE:	Revision 4b (May 19, 2005)			
Chec	kpoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided.	Yes		
I-b	SDB	A table with K _d is provided.	class 1		
II-a	REF	The loam is reported as mineral's content. Takadate loam (CEC: 10-30	В		
		meq/100g and specific surface area $130 \text{cm}^2/\text{g}$) and Hachinohe loam (30-135			
		meq/100g and $60 \text{cm}^2/\text{g}$) are provided.			
II-b	SDB	Final pH value is reported (after a contact time of 7 days, which can be	А		
		assumed to be enough for reaching equilibrium).			
II-c	SDB	Experiments are done under oxidizing conditions.	A/B		
		Moreover, Cs is not a redox-sensitive element.			
II-d	SDB	The initial composition for datapoints 54009 and 54012 is reported (deionized			
		water). The final solution composition can be estimated based on mineral and			
		initial solution composition.			
	REF	The initial composition for datapoints 54010 and 54011 is not reported			
		(synthetic groundwater). The composition of the synthetic water could be	C/D		
		reported in Konishi et al.(1986). The final solution composition can be			
		estimated based on mineral and initial solution composition.			
II-e	SDB	A room temperature is reported.	A/B		
II-f	SDB	A W/S ratio is not reported.	C/D		
	REF	No further information about W/S ratio.			
II-g	SDB	The sorption value cannot be evaluated, because of the lacking W/S ratio.	C/D		
II-h	SDB	The final Cs concentrations are between 5×10^{-7} - 1×10^{-3} mol/L. The initial			

		concentration is higher than the given equilibrium aqueous concentration, but	
		cannot be estimated because no information is provided about the amount of	А
		loam used in the experiment.	
		Since Cs is not solubility limited, the datapoints are classed as A.	
II-i	SDB	Centrifugation at 39000g is reported as separation method.	В
II-j	SDB	Contact time is 7 days. This reaction time is reasonably long.	C/D
II-k	REF	There is no information about the agitation method. This information may be	C/D
		reported in Konishi et al.(1986).	
II-l	REF	An isotherm has been measured (Experiment T-2 and H-2, Datapoints 54010,	А
		54011). The lower points of the isotherm are in the linear sorption range.	
II-m	REF	The material of the vessel is not reported.	C/D
II-n	SDB	No replicates and no error estimates are reported.	D
II-o	SDB	The pH value is varied (from 6.1 to 8.3).	С

Data	table Cs	/12: REF: Legoux et al. (1992)				
JNC-	JNC-SDB version 2 – DATA: Cs/Mudstone, # 55999 – 56002					
GUII	DELINE:	Revision 4b (May 19, 2005)				
Chec	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b	SDB	Tables with K_d are given.	class 1			
II-a	SDB	The solid phase is indicated as soil (A, B, C and D).	А			
		Mineralogical composition, specific surface area and cationic exchange				
		capacity (CEC) are reported.				
II-b	SDB	Initial pH values and final pH values are reported.	А			
	REF	Variation of the equilibrium pH versus the initial pH of the solution is showed				
		by the figure. The equilibrium pH of each of the samples is studied as a				
		function of the initial pH of the reconstituted water.				
II-c	SDB	There is no information about Eh and redox conditions.				
	REF	Based on the information in the paper, experiments are done under air; i.e,	A/B			
		under oxidizing conditions. Moreover, Cs is not a redox sensitive element.				
II-d	SDB	The final solution composition is reported.	A/B			
II-e	SDB	A temperature is not reported.	C/D			

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II-f	SDB	It is indicated that a W/S ratio of 10 mL/g had been used. The specific solid	
		mass (1g) and solution volume (10mL) are reported.	
	REF	The soil mass of 1g has a surface area of $6.6-14.7 \text{m}^2$.	A/B
II-g	SDB	The sorption value is between 99.7% and 99.8% (calculated with W/S and $K_{\rm d}$	C/D
		values).	
II-h	SDB	2.0×10^{-7} M is indicated as initial Cs concentration. Cs is not solubility limited.	А
II-i	SDB	Centrifugation and ultrafiltration through $0.002\mu m$ and $0.025\mu m$ Millipore	
		filters is reported. Centrifugation is done at 4000rpm prior to filtration.	А
II-j	SDB	Contact time is 2 days. This reaction time is reasonably long.	C/D
II-k	REF	Samples are agitated continuously.	A/B
II-l	SDB	No variation of W/S ratio and the initial Cs concentration is reported.	C/D
II-m	REF	The experiments are carried out in polypropylene vessels. It is reported that	C/D
		sorption to the vessel wall has been corrected for. No information about the	
		correction procedure is available. It is assumed that the correction is based on	
		the data obtained of a blank experiment.	
II-n	SDB	The reported error represents the standard deviation of the average of at least	А
	REF	four experimental repetitions.	
II-o	SDB	No parameter variation is indicated.	D

Data	Data table Cs/13: REF: Mackenzie et al. (1983)					
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 56123 – 56142					
GUID	ELINE:	Revision 4b (May 19, 2005)				
Check	Checkpoint Evaluation Rat					
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No			
		on initial and final solution composition are given.				
	REF	Solid phase; LLM-marine sediment; LLL-lacustrine sediment are not reported				
		about elemental analysis. No further information is available.	unreliable			

Data	table Cs/	/14: REF: Maclean et al. (1978)	
JNC-	SDB vers	ion 2 – DATA: Cs /Mudstone, # 56183 – 56184	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	Tables with K _d are given.	class 1
II-a	SDB	Conasauga shale is reported as mineral phase. CEC and surface area are not	C/D
		reported.	
II-b	SDB	Initial pH value is reported. Final pH value is not reported.	C
	REF	There is no further information about pH conditions. However, pH is not a	
		critical parameter for Cs.	
II-c	SDB	There is no information about Eh.	
	REF	It is reported that experiments are done under aerobic condition; i,e., under	A/B
		oxidizing conditions. Moreover, Cs is not a redox sensitive element.	
II-d	REF	The initial composition of the groundwater is reported.	
		The final solution composition is not reported, but can be estimated based on	C/D
		mineral and initial solution composition.	
II-e	SDB	A temperature of 25°C is reported.	A/B
II-f	SDB	A solution/solid ratio is 25mL/g. The used mass of 1g shale is considered to	A/B
		have enough surface area.	
II-g	SDB	The sorption values are 99.6 % and 99.8% (calculated with W/S and $K_{\rm d}$	C/D
		values).	
II-h	SDB	Initial [Cs] is reported 2.8×10^{-7} M.	
	REF	Cs is not solubility limited.	А
II-i	SDB	Filtration through 0.10µm membranes is indicated	C/D
II-j	SDB	A reaction time of 7 days is indicated. This reaction time is reasonably long.	C/D
II-k	REF	Agitation of the 130 oscillations/min is used as method in reaction time.	A/B
II-l	REF	No isotherms, no variation of the initial [Cs] is indicated.	C/D
II-m	REF	The reaction vessel is a polycarbonate centrifuge tube. No corrections for	В
		sorption on vessel walls are reported.	
II-n	REF	Three replicates and error estimates are reported.	А
II-o	REF	No parameter variation.	D

Data	table Cs/	/15: REF: Mucciardi et al. (1979)	
JNC-	SDB vers	ion 2 – DATA: Cs /Mudstone, # 57578 – 57589	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	Table with K_d values is provided.	class 1
II-a	SDB	Argillaceous shale is reported as mineral phase. CEC and surface area are	В
		reported.	
II-b	SDB	Initial pH value and final pH value are reported.	А
II-c	SDB	It is reported that experiments are done under air; i,e., under oxidizing	A/B
		conditions. Moreover, Cs is not redox sensitive.	
II-d	SDB	Only the initial composition is indicated (synthetic groundwater).	
	REF	It is assumed that the final solution composition can be estimated based on	C/D
		mineral and initial solution composition.	
II-e	SDB	Temperature of 20°C or 23.5°C is reported.	A/B
II-f	SDB	Fraction size of shale is reported 300µm-850µm. Solution/solid ratio is	A/B
		15mL/g. The used mass of 0.5g shale is considered to have enough surface	
		area ($\sim 5m^2$).	
II-g	REF	The %-sorbed can be calculated with K_d and W/S ratio:	
		-) Datapoint 57578: 81.0%	А
		-) Datapoints 57579, 57580: 97.1% and 97.7%	В
		-) Datapoints 57581-57589: between 98.0% and 99.8% is sorbed	C/D
II-h	SDB	Initial [Cs] is not reported. No information about initial [Cs] is reported.	
	REF	However, Cs is not solubility limited.	А
II-i	SDB	Centrifugation (7000rpm for 20 minutes) is indicated.	C/D
II-j	SDB	A reaction time of 3~ 30 days is indicated. This reaction time is reasonably	C/D
		long.	
II-k	REF	Agitation is carried out by shaking.	A/B
II-l	REF	No isotherms, no variation of W/S ratio and no initial [Cs] have been performed.	C/D
II-m	REF	The reaction vessel is a polycarbonate centrifuge tube.	В
II-n	REF	Three replicates and error estimates in K _d are reported.	А
II-o	REF	Two temperatures are used (20 and 23.5°C), two pH values (8.2 and 7) and	В
		four groundwaters; Eh (end) is varied.	

Data	Data table Cs/16: REF: Yamamoto et al. (1989)				
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 61762 – 61767				
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	kpoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on Eh, separation method, reaction time and final solution composition are			
		given.			
	REF	No further information is available. The Eh is not a sensitive parameter for Cs,			
	but the phase separation method, contact time and final solution composition				
		are key points which are missing in the reference.	unreliable		

Data	Data table Cs/17: REF: Rajec et al. (1998)				
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 62331 – 62350				
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	Checkpoint Evaluation Rating				
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on initial and final solution composition are given. Only one reaction time (2			
		hours) is reported.			
	REF	No further information is available. The final solution composition is a key	unreliable		
		point which is missing in the reference. The kinetics of sorption is not tested.			

Data	Data table Cs/18: REF: Mollah and Ullah (1998)					
JNC-	JNC-SDB version 2 – DATA: Cs /Mudstone, # 62474 – 62485					
GUIE	DELINE:	Revision 4b (May 19, 2005)				
Chec	Checkpoint Evaluation Rating					
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b	SDB	Table with K_d values is provided.	class 1			
II-a	SDB	Physical and chemical characteristics of the soils at the atomic energy research	В			
	REF	establishment campus in Bangladesh are reported. CEC is reported. Surface				
		area is not reported.				

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II-b	SDB	Initial pH value and final pH value are indicated.	А
II-c	SDB	It is reported that experiments are done under air; i,e., under oxidizing	A/B
		conditions. Moreover, Cs is not redox sensitive.	
II-d	SDB	Only initial characteristics are indicated (groundwater).	
	REF	It is assumed that the final solution composition can be estimated based on	C/D
		characteristics of the soils and initial solution composition.	
II-e	SDB	A temperature of 25°C is reported.	A/B
II-f	SDB	Solution/solid ratio is 50mL/g. The used mass of 1g soil is considered to have	A/B
		enough surface area.	
II-g	REF	The %-sorbed can be calculated with $K_d$ and W/S ratio: between 96.2% and	
		97.7% is sorbed.	В
II-h	SDB	Initial [Cs] is reported $1.68 \times 10^{-10}$ M.	
	REF	Cs is not solubility limited.	А
II-i	SDB	Centrifugation (5000rpm for 1 hour) is indicated.	C/D
II-j	SDB	A reaction time of 2 days is indicated. This reaction time is reasonably long.	C/D
II-k	REF	Agitation is carried out by shaking for 2 days.	A/B
II-l	REF	No isotherms, no variation of W/S ratio and no variation of initial [Cs] have	C/D
		been performed.	
II-m	REF	The reaction vessel is a polypropylene centrifuge tube. Blank experiments	А
		showed that no sorption of Cs on the walls of the tubes.	
II-n	REF	Three replicates estimates in $K_d$ are reported. Error is not reported.	D
II-o	REF	No isotherm is recorded; no parameter variation.	D

Data table Cs/19: REF: Kamel and Ibrahim (1998)							
JNC-	JNC-SDB version 2 – DATA: Cs /Mudstone, # 62721 – 62736						
GUIL	GUIDELINE: Revision 4b (May 19, 2005)						
Checkpoint		Evaluation	Rating				
I-a.1	SDB	All mandatory fields are completed.	Yes				
I-a.2	SDB	All mandatory information is provided.	Yes				
I-b	SDB	Table with $K_d$ values is provided.	class 1				
II-a	SDB	CEC is reported. Surface area is not reported.					
	REF	Mineralogical and chemical characteristics of compacted Egyptian soils are					
		given in previous reference (Ibrahim and Kamel, 1998).	C/D				
II-b	SDB	Initial pH value and final pH value are indicated.					

		- ) Datapoints 62731-62736	А
		Initial pH value is indicated. The sorption of Cs is not strongly pH dependent.	
		-) Datapoints 62721-62730	В
II-c	SDB	It is reported that experiments are done under air; i,e., under oxidizing	A/B
		conditions. Moreover, Cs is not redox sensitive.	
II-d	SDB	Only initial physical and chemical characteristics are indicated (natural	
		Ismailia canal water).	
	REF	It is assumed that the final solution composition can be estimated based on	C/D
		characteristics of the soils and initial canal water solution composition.	
II-e	SDB	A temperature of 30°C is reported.	A/B
II-f	SDB	Solution/solid ratio is 30mL/g. The used mass of 0.5g soil is considered to	
		have enough surface area.	A/B
	REF	The soil samples are crushed and grinded.	
II-g	REF	The %-sorbed can be calculated with $K_d$ and W/S ratio:	
		-) Datapoints 62721-62736: between 12.2% and 87.8% is sorbed	А
II-h	SDB	Initial [Cs] is reported from $1.0 \times 10^{-4}$ M to $1.0 \times 10^{-2}$ M.	
	REF	Cs is not solubility limited.	А
II-i	SDB	Centrifugation and filtered through 0.45µm is indicated;	
		- ) Datapoints 62721-62730	C/D
		Centrifugation is indicated;	
		- ) Datapoints 62731-62736	C/D
II-j	SDB	A reaction time of 7 days is indicated. This reaction time is reasonably long.	C/D
II-k	REF	Agitation is carried out by shaking for 7 days.	A/B
II-l	REF	Variation of initial [Cs] has been performed.	В
II-m	REF	The reaction vessel is a polyethylene bottle.	В
II-n	REF	Replicates estimates in $K_d$ are not reported. Error is not reported.	D
II-o	REF	The pH and [Cs] are varied.	В

Data table Cs/20: REF: Strezov et al. (2000)						
JNC-SDB version 2 – DATA: Cs /Mudstone, # 62772 – 62786						
GUIDELINE: Revision 4b (May 19, 2005)						
Checkpoint		Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			

I-b	SDB	It is indicated that values are extracted from a graph (linear).	class 3
II-a	SDB	Mineral composition and chemical composition about the soil are reported.	C/D
	REF	CEC and Surface area are not reported.	
II-b	SDB	Initial pH values and final pH values are indicated.	
		- ) Datapoints 62777-62786	А
		Initial pH values are indicated.	
		- ) Datapoints 62772-62776	С
II-c	SDB	It is reported that experiments are done under air; i,e., under oxidizing	A/B
		conditions. Moreover, Cs is not redox sensitive.	
II-d	SDB	The compositions of the groundwater before equilibration with the solid phase	
		and after equilibration (the final solution composition) are given in table.	A/B
II-e	SDB	A temperature of 20°C is reported.	A/B
II-f	SDB	Solution/solid ratio is 30mL/g. The used mass of 0.3g soil is considered to	A/B
		have enough surface area.	
	REF	The geological formation is crushed and sieved into the fractions from $80\mu m$	
		to 400µm.	
II-g	SDB	Based on the information given in the SDB, the %-sorbed values are between	А
		33.3% and 92.3%.	
II-h	SDB	Initial [Cs] is reported $1.0 \times 10^{-6}$ M. Cs is not solubility limited.	А
II-i	SDB	Centrifugation at 6000rpm is indicated.	C/D
	REF	No further information.	
II-j	SDB	A reaction time of 0.2 day, 0.3 day and 0.4 day is indicated, which is not	
		reached a steady-state.	
		- ) Datapoints 62772-62774	unreliable
		A reaction times of 6 days, 30 days and 48 days are indicated, which is	
		reached a steady-state. Sorption kinetics of Cs on geological formation is	
		showed in figure.	
		- ) Datapoints 62775-62786	A/B
II-k	REF	The samples are shaken for 1 month with groundwater for equilibration. The	A/B
		kinetics of Cs sorption is measured for a soil.	
II-l	REF	The influence of Cs concentration on sorption on the geological media is	В
		investigated (sorption isoterm of Cs on geological formation).	
II-m	REF	The measurements are performed in 50 mL polyethylene vials. Corrections for	В
		sorption on vials have not been performed.	
II-n	REF	Each experiment is tripled. Error is not reported.	D
II-o	REF	An isotherm is reporded. The pH is varied.	В

Data	table Cs/	/21: REF: Ashida et al. (1999)	
JNC-	SDB vers	ion 2 – DATA: Cs /Mudstone, # 62892, 62896	
GUIE	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	It is indicated that values are extracted from a graph (log).	class 5
II-a	REF	Mineral composition list about the mudstone (Miyagi prefecture Tomegun) is	C/D
		reported. Surface area is reported. CEC is not reported.	
II-b	SDB	Final pH value is indicated.	А
II-c	SDB	It is reported that experiments are done under air; i,e., under oxidizing	A/B
		conditions. Moreover, Cs is not redox sensitive.	
II-d	SDB	0.01M-NaCl and 0.1M-NaCl are reported.	
	REF	It is assumed that the final solution composition can be estimated based on	C/D
		mineral composition of the mudstone and NaCl water.	
II-e	SDB	A temperature of 22°C is reported.	A/B
II-f	SDB	Solution/solid ratio is 100mL/g. The used mass of mudstone is not reported.	C/D
II-g	SDB	Based on the information given in the JNC-SDB, the %-sorbed values are	А
		between 39.4% and 44.4%.	
II-h	SDB	Initial [Cs] is reported $1.9 \times 10^{-9}$ M.	
	REF	Cs is not solubility limited.	А
II-i	SDB	Centrifugation at 10000rpm is indicated.	В
	REF	No further information.	
II-j	SDB	A reaction time of 30 days is indicated, which is considered for enough to	A/B
		reached a steady-state.	
II-k	REF	The samples are shaken for 1 month with NaCl water for equilibration.	A/B
II-l	REF	Neither [Cs] nor the W/S has been varied.	C/D
II-m	REF	The measurements are performed in polyethylene vessels. Corrections for	В
		sorption on vessels have not been performed.	
II-n	REF	Each experiment is duplicated. Error is not reported.	D
II-o	REF	No isotherm is recorded; no parameter variation.	D

Data	Data table Cs/22: REF: Inoue and Morisawa (1975)				
JNC-S	JNC-SDB version 2 – DATA: Cs /Mudstone, # 63346				
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	Checkpoint Evaluation Rating				
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on Eh, separation method and reaction time are given.			
	REF	No further information is available. The Eh is not a sensitive parameter for Cs,			
		but the phase separation method and contact time are key points which are			
		missing in the reference.	unreliable		

Data	table Cs/	/23: REF: Silva et al. (1979)			
JNC-	SDB vers	ion 2 – DATA: Cs /Bentonite, # 65509 – 65658			
GUII	GUIDELINE: Revision 4b (May 19, 2005)				
Chec	kpoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided.	Yes		
I-b	SDB	Tables with $K_d$ are given.	class 1		
II-a	SDB	The solid phase is reported as a montmorillonite #27(Belle Fourche clay). The			
		chemical compositions of solid substrates are reported in the table. Major and			
		minor mineralogy as well as surface characteristics are reported.	А		
II-b	SDB	Only initial pH values are reported.			
	REF	Just prior to the sorption measurements, the pH of the solutions are checked			
		and readjusted if necessary.	В		
II-c	SDB	Experiments are performed under an argon filled, inert atmosphere box. Since			
		the effect of dissolved $CO_2$ , i.e., $HCO_3^-$ and $CO_3^{2-}$ ions, it is decided to exclude			
		$CO_2$ from the system. Cs is not a redox-sensitive element.	A/B		
II-d	SDB	A NaCl (0.002M, 0.01M, 0.1M and 1M) is reported as input solution. Initial			
		chemical compositions of synthetic groundwater are reported in the additional			
		file "solution composition" and had an ionic strength of 0.003M.	C/D		
II-e	SDB	A temperature of $26 \pm 2^{\circ}$ C is reported.	A/B		
II-f	SDB	A solution/solid ratio of 100 mL/g is reported.			
	REF	1 g solid is added to 100 mL solution. The average particle size is 1.14 $\mu m$			
		with 90% of the particle sizes in the range $0.5 - 2.0 \mu m$ . The BET (outer)			
		surface areas are $18.1 \text{m}^2/\text{g}$ for a Belle Fourche Clay.	A/B		

II-g	SDB	On a solution/solid ratio that may from 100 mL/g, sorption values between 5.3	
0		- 98.3% sorbed can be calculated.	
		<ul> <li>Datapoints: 65510-65512, 65515-65517, 65520-65522, 65525-65527, 65530-65532, 65535-65537, 65540-65542, 65545-65547, 65550-65552, 65555-65557, 65560-65562, 65565-65567, 65570-65572, 65575-65577, 65580-65582, 65584-65587, 65590-65592, 65595-65597, 65600-65602, 65605-65607, 65609-65612, 65614-65617, 65620-65622, 65625-65627, 65630-65632, 65634-65647, 65650-65652, 65654-65657</li> </ul>	А
		<ul> <li>Datapoints: 65509, 65513, 65514, 65518, 65519, 65523, 65524, 65528, 65529, 65533, 65534, 65538, 65539, 65543, 65544, 65549, 65553, 65554, 65558, 65559, 65563, 65564, 65568, 65569, 65573, 65574, 65578, 65579, 65583, 65588, 65589, 65593, 65594, 65598, 65599, 65603, 65604, 65608, 65613, 65618, 65619, 65623, 65624, 65628, 65629, 65633, 65648, 65649, 65653, 65658</li> </ul>	В
		Datapoints: 65548	C/D
II-h	SDB	Initial Cs concentrations are $0.93 \times 10^{-3}$ , $1.00 \times 10^{-4}$ , $1.00 \times 10^{-5}$ , $1.01 \times 10^{-7}$ and	
		$1.04 \times 10^{-9}$ M. Cs is not solubility limited.	А
II-i	SDB	Centrifuge for 3 minutes at 12000 rpm and membrane filtration with pore size	
		of 0.2µm are used for phase separation.	В
II-j	SDB	It is reported that contact time is 2 days.	A/B
	REF	The sorption of Cs as a function of time is investigated in separate	
		experiments and no detectable increase in sorption is measured after 15	
		minutes.	
II-k	REF	The solutions are agitated for $15 - 20$ minutes. But the method is not specified.	C/D
II-l	SDB	A limited variation of initial Cs concentration has been used, including some	А
		experiments at trace concentrations.	
	REF	Cesium sorption isotherms using the sodium form of the Belle Fourche clay is	
		obtained as a function of Cs concentration, pH and supporting electrolyte	
		concentration in batch-type experiments.	
II-m	REF	The experiments are carried out in polyethylene bottles. Corrections for	В
		sorption on vessel walls have not been performed.	
II-n	SDB	Uncertainties in $K_d$ are given for each $K_d$ value.	
	REF	The assigned errors result from a consideration of errors associated with	
		counting and sampling. Duplicates are run on only about 10% of the samples	В
		and agreements are within the estimated error. Cesium sorption isotherms are	
		obtained.	

II-o	SDB	Within the set of reliable experiments, Cs concentration (isotherm), pH and	А
		supporting electrolyte concentration are varied. The initial Cs concentrations	
		ranged from 10 ⁻³ M to 10 ⁻⁹ M; pH values are 5, 6, 7, 8, 9 and 10; electrolyte	
		solutions are (0.002M, 0.01M, 0.1M and 1M) NaCl and the simulated	
		groundwater.	

# 3.1.5 Plutonium

JNC-	SDB vers	ion 2 – DATA: Pu /Bentonite, # 65922 – 65927	
		Revision 4b (May 19, 2005)	
	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	Tables with K _d are given.	class 1
II-a	SDB	The solid phase is indicated as Kunigel-V1.	А
	REF	The surface area is 6.80 m ² /g. The concentration of sites $2.88 \times 10^{-2}$ mol dm ⁻³ .	
II-b	SDB	Final pH values are reported.	А
	REF	No use of any pH-buffer is reported. The pH value is monitored every 7 days,	
		the pH adjusted back to the appropriate value.	
II-c	SDB	Experiments are performed under N ₂ . The oxidation state of Pu is reported as	
		+IV. Final Eh values are provided, as well as the reducing agent used	C/D
		$(Na_2S_2O_4)$ at an initial concentration of $2.5 \times 10^{-3}$ M.	
	REF	O ₂ levels are reportedly <1 ppm. All solutions are de-oxygenated by bubbling	
		an argon (96%) /hydrogen (4%) mixture through them before posting into the	
		glovebox. The Eh value is monitored every 7 days, and when it increased	
		towards -300mV more sodium hydrosulphite is added.	
II-d	SDB	A NaCl, NaHCO ₃ is reported as input solution; main solution compositions are	
		indicated in the additional file "solution composition".	A/B
	REF	Samples are filtered using a 10000MWCO filter prior to analysis.	
II-e	SDB	A temperature of 60°C is reported.	A/B
II-f	SDB	It is indicated that a W/S ratio of 20 mL/g had been used, but that the specific	
		solid mass and solution volume are not reported.	
		Assuming that the solution volume is $\geq 20$ mL would mean a corresponding	
		solid mass of at least 1 g. This is accepted as reasonable.	A/B
II-g	SDB	Based on the information provided in the JNC-SDB, all $K_d$ values can be	
		calculated to correspond to %-sorbed values >99%.	C/D
II-h	SDB	$4.3 \times 10^{-11}$ M is indicated as initial Pu concentration.	
	REF	Based on the experimental data shown for Th and Pu(III) in Rai et al.(1999), it	
		is estimated that the initial Pu concentration is below the solubility limit, but	
		possibly by less than a factor of 5.	В
		The solutions are pre-filtered ( $45\mu m$ ) before being equilibrated with the solid.	

			1
		Presumably, a possible Pu-precipitate would have been removed in this way.	
II-i	SDB	Filtration through 10000MWCO and $45\mu m$ membranes, as well as	
		centrifugation at 1100g for 2.5 h is reported.	
	REF	Three methods of phase separation are used: 1) centrifugation at 1100 g for 2.5	
		h, 2) centrifugation at 1100 g for 2.5 h followed by filtration through 0.45 $\mu$ m	
		membranes, and 3) centrifugation at 1100 g for 2.5 h followed by filtration	
		through 0.45 $\mu$ m membranes and then by filtration through 10000 MWCO	
		filters. It appears that the resulting $K_d$ values are slightly dependent on the	
		method of phase separation: surprisingly, the highest $K_d$ is generally obtained	
		with centrifugation alone, whereas filtration with 10000 MWCO membranes	
		leads to the lowest values. However, the respective variation is of about the	
		same magnitude as observed for different W/S. Therefore, it may be	А
		considered to be within the overall experimental uncertainties.	
II-j	SDB	A reaction time of 122 days (four months) is reported. This reaction time is	
		reasonably long.	C/D
II-k	REF	The experiments are gently agitated on a shaker table.	A/B
II-l	SDB	No variation of W/S ratio or no variation of the initial Pu concentration is	
		reported.	C/D
II-m	REF	The experiments are carried out in polypropylene centrifuge tubes. And	
		corrections for sorption on vessel walls have been performed.	А
II-n	SDB	It is indicated that each experiment is done in duplicate.	
	REF	Error estimates are given for each replicate, based on analytical uncertainties.	А
II-o	SDB	No relevant parameter variation is indicated.	D

# 3.1.6 Americium

Data	Data table Am/1: REF: Barney and Anderson (1979)				
JNC-S	JNC-SDB version 2 – DATA: Am /Mudstone, # 44238 – 44243				
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	Checkpoint Evaluation				
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on initial and final pH, final solution composition and atmosphere/ redox			
		condition are given.	unreliable		
	REF	No further information is available. Americium is not redox sensitive			

Data	Data table Am/2: REF: Barney and Brown (1979)				
JNC-S	JNC-SDB version 2 – DATA: Am /Mudstone, # 44322 – 44324				
GUID	ELINE:	Revision 4b (May 19, 2005)			
Chec	Checkpoint ¹ Evaluation				
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on initial and final pH, initial and final Eh, final solution composition, initial			
		Am concentration and atmosphere/ redox condition are given.			
	REF	No further information is available. Americium is not redox sensitive	unreliable		

Data t	able Am/	3: REF: Baston et al. (1995b)				
JNC-	JNC-SDB version 2 – DATA: Am /Mudstone, # 46664 – 46667					
GUII	DELINE:	Revision 4b (May 19, 2005)				
Chec	Checkpoint ¹ Evaluation					
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b	SDB	Tables with $K_d$ are given.	class 1			
II-a	SDB	The solid phase is reported as cover soil at Rokkasho-Mura. Physico- chemical	C/D			
	REF	properties and specific surface of cover soil are not given.				
II-b	SDB	Initial and final pH values are reported.	А			
II-c	SDB	It is reported that experiments are done in nitrogen-atmosphere glove-boxes.	A/B			
		Also, Am is not a redox-sensitive element.				

II-d	SDB	The compositions of solutions (synthetic groundwater) used in experiments	
		are reported.	A/B
II-e	SDB	A temperature is not reported.	C/D
II-f	SDB	Solution/solid ratios are reported for 50mL/g.	C/D
II-g	SDB	Based on the information given in the SDB, and on a solution/solid ratio of	
		50ml/g, sorption values between 99.77%- 99.88% sorbed can be calculated.	C/D
II-h	SDB	The initial Am concentrations are reported to be $4.\times 10^{-11}$ M to $8.\times 10^{-11}$ M.	
	REF	The exact initial Am concentration is not reported. Based on data from Rai et	
		al. (1999), the solubility for Am at the highest pH of 8 is about $1.\times 10^{-7}$ M.	
		Therefore, the initial concentration is clearly below the solubility limit (the	А
		solution composition will not strongly influence the solubility of Am).	
II-i	SDB	The membrane filtration with pore size of $0.45 \mu m$ is used.	
		-) Datapoints: 46664, 46666	C/D
		Filtration with a 30000MWCO filter is reported as separation method.	
		-) Datapoints: 46665, 46667	В
II-j	SDB	It is indicated that contact time is not reported exactly.	
	REF	It is reported that one tube is analysed regularly to monitor the approach to a	C/D
		steady state concentration of radionuclide. The reaction time considered to be	
		reasonably long.	
II-k	REF	The batch sorption experiments are gently agitated continuously. But the	C/D
		agitation method is not specified.	
II-l	REF	A limited variation of [Am] has been used, including some experiments at	В
		trace concentrations.	
II-m	REF	The experiments are carried out in polypropylene centrifuge tubes. The	А
		sorption on vessel wall has been determined as lower than 0.5% than the	
		actual value and thus corrections for sorption on vessel wall have not been	
		performed.	
II-n	SDB	Uncertainties are reported (n=3).	А
II-o	SDB	The phase separation and final pH are varied.	В

Data	table An	n/4: REF: Daniels et al.(1981)	
JNC-	SDB vers	ion 2 – DATA: Am /Mudstone, # 50002 – 50025	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	Tables with $K_d$ are given.	class 1
II-a	SDB	The solid phase is reported as Eleana argillite.	C/D
	REF	Additional Eleana argillite is crushed in the controlled-atmosphere boxes.	
		Physico- chemical properties and specific surface are not given.	
II-b	SDB	Initial pH value and final pH value are reported.	А
II-c	SDB	It is reported that experiments are done under atmosphere and controlled	A/B
		atmosphere. Also, Am is not a redox-sensitive element.	
II-d	SDB	The final composition of the solution is not reported.	
	REF	Composition of water obtained in the field (groundwaters) is described in	
		Erdal et al.(1979b): Los Alamos Scientific Laboratory report LA-7455-MS.	C/D
II-e	SDB	A temperature of 32°C or 22°C is reported.	A/B
II-f	SDB	Solution/solid ratios are not reported.	C/D
II-g	SDB	The sorption value cannot be calculated, because the Solution/solid ratios are	C/D
		lacking.	
II-h	SDB	The initial Am concentrations are $1.4 \times 10^{-7}$ mol/L and $4.6 \times 10^{-8}$ mol/L.	
		Based on data from Rai et al. (1999), the solubility for Am at the highest pH of	
	REF	8 is about $1 \times 10^{-7}$ M. Therefore, the initial concentration of $4.6 \times 10^{-8}$ mol/L is	
		below the solubility limit (the solution composition will not strongly influence	
		the solubility of Am). And the initial concentration of $1.4 \times 10^{-7}$ mol/L is to	
		close to the saturation concentration. Therefore following rating is given:	
		-) Datapoints: 50004, 50005, 50007, 50008, 50011, 50012, 50015,	В
		50017, 50020-50022, 50024	
		-) Datapoints: 50002, 50003, 50006, 50009, 50010, 50013, 50014, 50016, 50018, 50019, 50023, 50025	unreliable
II-i	SDB	Centrifugation and filtration are indicated. But there is no information about	
		the concrete method.	C/D
II-j	SDB	It is indicated that contact time is 20.7 days, 21.9 days, 41.6 days, 42.8 days,	
		83.7days or 83.8days for different datapoints. These reaction times are	
		reasonably long.	C/D
	REF	The kinetics of sorption is not tested.	

II-k	REF	The batch sorption experiments are performed by shaking. It is indicated in	A/B
		Erdal et al.(1979b): Los Alamos Scientific Laboratory report LA-7455-MS.	
II-l	REF	No isotherm is performed.	C/D
II-m	REF	The experiments are carried out in polyethylene centrifuge tubes. It is	В
		indicated in Erdal et al.(1979b): Los Alamos Scientific Laboratory report	
		LA-7455-MS.	
II-n	SDB	Uncertainties are not reported.	D
II-o	SDB	The initial pH, reaction time and temperature are varied.	В

Data	Data table Am/5: REF: Erdal (1980)				
JNC-S	JNC-SDB version 2 – DATA: Am /Mudstone, # 50977 – 51062				
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	spoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on initial and final pH (# $50977 - 51000$ ), initial and final Eh, final solution			
		composition and atmosphere/ redox condition are given.			
	REF	No further information is available. Americium is not redox sensitive.	unreliable		

Data table	Data table Am/6: REF: Higgo et al.(1987)				
JNC-SDB v	ersion 2 – DATA: Am /Mudstone, # 53185 – 53223				
GUIDELIN	E: Revision 4b (May 19, 2005)				
Checkpoin	Evaluation	Rating			
I-a.1 SDF	All mandatory fields are completed.	Yes			
I-a.2 SDF	All mandatory information is provided.	Yes			
I-b SDE	Tables with $K_d$ are given.	class 1			
II-a SDB	Major and minor mineralogy is not known. Specific surface area and CEC are	C/D			
	reported. It is not clear whether impurities are determined.				
II-b SDB	Initial pH values are indicated.	С			
REF	No use of any pH-buffer is reported.				
II-c SDB	Experiments are carried out at aerobic condition. Americium is not redox	A/B			
	sensitive.				
II-d SDB	Seawater is indicated.				

REF       The critical major components are approximately Known. Minor components may be unknown. Charge balance cannot be calculated, but seawater can be expected to be well poised.       C/D         II-e       SDB       A temperature of 4°C is reported.       A/B         II-f       SDB       Solution/solid ratios ranged from 1.69mL/g to 25000mL/g.       C/D         REF       Liquid volumes are reported 50mL.       C/D         II-g       SDB       The sorption value is between 98.64% and 99.99% (calculated with W/S and K4 values).       C/D         II-h       SDB       Initial [Am] ranged from 6.24×10 ⁻¹⁰ M to 1.19×10 ⁻⁹ M. Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below the respective solubility limit, but maybe by a factor <5.       B         II-i       SDB       The centrifugation (7000rpm / 90 min) and followed by filtration through 0.1 and 0.22µm Millipore filters is indicated.       C/D         II-j       SDB       A reaction time ranged from 23 days to 74days. This reaction time is C/D reasonably long.       C/D         II-k       REF       No substantial variation of initial Am concentration is indicated.       C/D         II-n       SDB       No replicates are reported. But error estimates are reported.       C/D         II-k       REF       No replicates are neported. But error estimates are reported.       C/D         II-n       SDB       No replicates are rep				
expected to be well poised.C/DII-eSDBA temperature of 4°C is reported.A/BII-fSDBSolution/solid ratios ranged from 1.69mL/g to 25000mL/g. Liquid volumes are reported 50mL.C/DII-gSDBThe sorption value is between 98.64% and 99.99% (calculated with W/S and $K_d$ values).C/DII-hSDBInitial [Am] ranged from $6.24 \times 10^{-10}$ M to $1.19 \times 10^{-9}$ M. Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below the respective solubility limit, but maybe by a factor <5.BII-iSDBThe centrifugation (7000rpm / 90 min) and followed by filtration through 0.1 and $0.22 \mu$ m Millipore filters is indicated.C/DII-iSDBA reaction time ranged from 23 days to 74days. This reaction time is reasonably long.C/DII-aREFAgitation is carried out by shaken at 120 strokes/min.A/BII-aREFNo substantial variation of initial Am concentration is indicated.C/DII-mREFThe quality of reaction vessel is not reported.C/DII-nSDBNo replicates are reported. But error estimates are reported.C/D		REF	The critical major components are approximately Known. Minor components	
II-eSDBA temperature of 4°C is reported.A/BII-fSDBSolution/solid ratios ranged from 1.69mL/g to 25000mL/g. Liquid volumes are reported 50mL.C/DII-gSDBThe sorption value is between 98.64% and 99.99% (calculated with W/S and $K_d$ values).C/DII-hSDBInitial [Am] ranged from $6.24 \times 10^{-10}$ M to $1.19 \times 10^{-9}$ M. Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below the respective solubility limit, but maybe by a factor <5.BII-iSDBThe centrifugation (7000rpm / 90 min) and followed by filtration through 0.1 and 0.22µm Millipore filters is indicated.C/DII-iSDBA reaction time ranged from 23 days to 74days. This reaction time is reasonably long.C/DII-iREFAgitation is carried out by shaken at 120 strokes/min.A/BII-iREFNo substantial variation of initial Am concentration is indicated.C/DII-mREFThe quality of reaction vessel is not reported.C/DII-nSDBNo replicates are reported. But error estimates are reported.C/D			may be unknown. Charge balance cannot be calculated, but seawater can be	
II-f       SDB       Solution/solid ratios ranged from 1.69mL/g to 25000mL/g.       C/D         REF       Liquid volumes are reported 50mL.       C/D         II-g       SDB       The sorption value is between 98.64% and 99.99% (calculated with W/S and K _d values).       C/D         II-h       SDB       Initial [Am] ranged from 6.24×10 ⁻¹⁰ M to 1.19×10 ⁻⁹ M. Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below the respective B       B         solubility limit, but maybe by a factor <5.       II       In ecentrifugation (7000rpm / 90 min) and followed by filtration through 0.1 and 0.22μm Millipore filters is indicated.       C/D         II-j       SDB       A reaction time ranged from 23 days to 74days. This reaction time is reasonably long.       C/D         II-k       REF       Agitation is carried out by shaken at 120 strokes/min.       A/B         II-1       REF       No substantial variation of initial Am concentration is indicated.       C/D         II-m       REF       The quality of reaction vessel is not reported.       C/D         II-n       SDB       No replicates are reported. But error estimates are reported.       C/D			expected to be well poised.	C/D
REF       Liquid volumes are reported 50mL.         II-g       SDB       The sorption value is between 98.64% and 99.99% (calculated with W/S and K _d values).       C/D         II-h       SDB       Initial [Am] ranged from 6.24×10 ⁻¹⁰ M to 1.19×10 ⁻⁹ M. Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below the respective B solubility limit, but maybe by a factor <5.	II-e	SDB	A temperature of 4°C is reported.	A/B
II-gSDBThe sorption value is between 98.64% and 99.99% (calculated with W/S and $K_d$ values).C/DII-hSDBInitial [Am] ranged from $6.24 \times 10^{-10}$ M to $1.19 \times 10^{-9}$ M. Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below the respective solubility limit, but maybe by a factor <5.	II-f	SDB	Solution/solid ratios ranged from 1.69mL/g to 25000mL/g.	C/D
Kd values).Kd values).II-hSDBInitial [Am] ranged from $6.24 \times 10^{-10}$ M to $1.19 \times 10^{-9}$ M. Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below the respective solubility limit, but maybe by a factor <5.		REF	Liquid volumes are reported 50mL.	
II-hSDBInitial [Am] ranged from 6.24×10 ⁻¹⁰ M to 1.19×10 ⁻⁹ M. Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below the respective solubility limit, but maybe by a factor <5.	II-g	SDB	The sorption value is between 98.64% and 99.99% (calculated with W/S and	C/D
Rai et al. (1999), it is estimated that initial [Am] is below the respective solubility limit, but maybe by a factor <5.       B         II-i       SDB       The centrifugation (7000rpm / 90 min) and followed by filtration through 0.1 and 0.22 μm Millipore filters is indicated.       C/D         II-j       SDB       A reaction time ranged from 23 days to 74 days. This reaction time is reasonably long.       C/D         II-k       REF       Agitation is carried out by shaken at 120 strokes/min.       A/B         II-I       REF       No substantial variation of initial Am concentration is indicated.       C/D         II-m       REF       The quality of reaction vessel is not reported.       C/D         II-n       SDB       No replicates are reported. But error estimates are reported.       C			K _d values).	
solubility limit, but maybe by a factor <5.	II-h	SDB	Initial [Am] ranged from $6.24 \times 10^{-10}$ M to $1.19 \times 10^{-9}$ M. Based on the data in	
II-iSDBThe centrifugation (7000rpm / 90 min) and followed by filtration through 0.1 and 0.22μm Millipore filters is indicated.C/DII-jSDBA reaction time ranged from 23 days to 74days. This reaction time is reasonably long.C/DII-kREFAgitation is carried out by shaken at 120 strokes/min.A/BII-lREFNo substantial variation of initial Am concentration is indicated.C/DII-mREFThe quality of reaction vessel is not reported.C/DII-nSDB REFNo replicates are reported. But error estimates are reported.C/DREFUncertainties are based on error propagation, not on replicate measurements.C			Rai et al. (1999), it is estimated that initial [Am] is below the respective	В
and 0.22μm Millipore filters is indicated.         II-j       SDB         A reaction time ranged from 23 days to 74days. This reaction time is C/D reasonably long.         II-k       REF         Agitation is carried out by shaken at 120 strokes/min.         II-l       REF         No substantial variation of initial Am concentration is indicated.       C/D         II-m       REF         The quality of reaction vessel is not reported.       C/D         II-n       SDB         No replicates are reported. But error estimates are reported.       C         REF       Uncertainties are based on error propagation, not on replicate measurements.			solubility limit, but maybe by a factor <5.	
II-jSDBA reaction time ranged from 23 days to 74days. This reaction time is reasonably long.C/DII-kREFAgitation is carried out by shaken at 120 strokes/min.A/BII-1REFNo substantial variation of initial Am concentration is indicated.C/DII-mREFThe quality of reaction vessel is not reported.C/DII-nSDBNo replicates are reported. But error estimates are reported.CREFUncertainties are based on error propagation, not on replicate measurements.C	II-i	SDB	The centrifugation (7000rpm / 90 min) and followed by filtration through 0.1	C/D
II-kREFAgitation is carried out by shaken at 120 strokes/min.A/BII-lREFNo substantial variation of initial Am concentration is indicated.C/DII-mREFThe quality of reaction vessel is not reported.C/DII-nSDBNo replicates are reported. But error estimates are reported.CREFUncertainties are based on error propagation, not on replicate measurements.C			and 0.22µm Millipore filters is indicated.	
II-kREFAgitation is carried out by shaken at 120 strokes/min.A/BII-IREFNo substantial variation of initial Am concentration is indicated.C/DII-mREFThe quality of reaction vessel is not reported.C/DII-nSDBNo replicates are reported. But error estimates are reported.CREFUncertainties are based on error propagation, not on replicate measurements.C	II-j	SDB	A reaction time ranged from 23 days to 74days. This reaction time is	C/D
II-IREFNo substantial variation of initial Am concentration is indicated.C/DII-mREFThe quality of reaction vessel is not reported.C/DII-nSDBNo replicates are reported. But error estimates are reported.CREFUncertainties are based on error propagation, not on replicate measurements.C			reasonably long.	
II-mREFThe quality of reaction vessel is not reported.C/DII-nSDBNo replicates are reported. But error estimates are reported.CREFUncertainties are based on error propagation, not on replicate measurements.C	II-k	REF	Agitation is carried out by shaken at 120 strokes/min.	A/B
II-n       SDB REF       No replicates are reported. But error estimates are reported. Uncertainties are based on error propagation, not on replicate measurements.       C	II-l	REF	No substantial variation of initial Am concentration is indicated.	C/D
REF     Uncertainties are based on error propagation, not on replicate measurements.	II-m	REF	The quality of reaction vessel is not reported.	C/D
	II-n	SDB	No replicates are reported. But error estimates are reported.	С
II-o REF Solution/solid ratio and reaction time are varied.		REF	Uncertainties are based on error propagation, not on replicate measurements.	
	II-o	REF	Solution/solid ratio and reaction time are varied.	В

Data	Data table Am/7: REF: Kim et al.(1994)				
JNC-S	SDB vers	ion 2 – DATA: Am /Mudstone, # 53965 – 53966			
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	cpoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		about phase separation is given.			
	REF	No further information is available.	unreliable		

Data	Data table Am/8: REF: Konishi et al. (1988)				
JNC-	SDB vers	tion 2 – DATA: Am / Mudstone, # 54003 – 54004			
GUIE	DELINE:	Revision 4b (May 19, 2005)			
Chec	kpoint	Evaluation	Rating		
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	All mandatory information is provided.	Yes		
I-b	SDB	A table with $K_d$ is provided.	class 1		
II-a	REF	The loam is reported as mineral's content. Takadate loam (CEC: 10~ 30			
		meq/100g and specific surface area130cm ² /g) and Hachinohe loam (30 $\sim$ 135			
		$meq/100g$ and $60cm^2/g$ ) are provided (Konishi et al.1986).	В		
II-b	SDB	Final pH value is reported (after a contact time of 7 days, which can be			
		assumed to be enough for reaching equilibrium).	А		
II-c	SDB	Experiments are done under oxidizing conditions.			
	REF	Moreover, Am is not a redox-sensitive element.	A/B		
II-d	SDB	The initial composition is reported (deionized water).			
	REF	The final solution composition can be estimated based on mineral and initial			
		solution composition.	C/D		
II-e	SDB	A room temperature is reported.	A/B		
II-f	SDB	A W/S ratio is not reported.			
	REF	No further information about W/S ratio.	C/D		
II-g	SDB	The sorption value cannot be evaluated, because of the lacking W/S ratio.	C/D		
II-h	SDB	The final Am concentrations are between $2 \times 10^{-9}$ - $9 \times 10^{-9}$ mol/L.			
	REF	Based on the data in Rai et al. (1999), it is estimated that initial [Am] is below			
		the respective solubility limit, but maybe by a factor $<5$ .	В		
II-i	SDB	Centrifugation at 39000g is reported as separation method.	В		
II-j	SDB	Contact time is 7 days. This reaction time is reasonably long.	C/D		
II-k	REF	There is no information about the agitation method.	C/D		
II-l	REF	No isotherms and no variation of W/S ratio have been performed.	C/D		
II-m	REF	The material of the vessel is not reported.	C/D		
II-n	SDB	The replicate is reported to one. No error estimates are reported.	D		
II-o	SDB	Two pH values are varied (6.0 to 6.5).	С		

Data	table An	n/9: REF:Legoux et al.(1992)	
JNC-	SDB vers	ion 2 – DATA: Am /Mudstone, # 55991 – 55994	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	Tables with K _d are given.	class 1
II-a	SDB	Four soils with known mineral composition are indicated. CEC value and	А
		specific surface area are reported.	
II-b	SDB	Initial and final pH values are indicated.	А
	REF	No use of any pH-buffer is reported.	
II-c	SDB	It is indicated that no redox conditions are reported. There is no information	
		about Eh. Experiments are carried out with Am (III).	
	REF	No reducing agents are added to the experiment and no inert gas atmosphere is	
		reported. Therefore it is assumed that the experiment is conducted under	
		ambient conditions. Moreover, Am is not redox-sensitive element.	A/B
II-d	REF	The final solution composition is reported.	A/B
II-e	SDB	A temperature is not reported.	C/D
II-f	SDB	Solution/solid ratio is 10mL/g.The used mass of 1g soil (A, B, C, D) has a	A/B
		surface area of about 6.6 to 14.7m ² .	
II-g	SDB	The sorption value is between 99.93% and 99.94% (calculated with W/S and	C/D
		K _d values).	
II-h	SDB	Initial [Am] is $1.3 \times 10^{-7}$ M.	
	REF	Based on thermodynamic calculations with PHREEQC (database:	
		JNC-TDB_011213c2), the initial [Am] are below the solubility limit (but it is	
		not clear by how much, due to the uncertainty in initial concentration as well	
		as in solubility data for Am).	В
II-i	SDB	The centrifugation is performed at 4000rpm followed by ultrafiltration through	А
		0.025 and 0.002µm Millipore filters is indicated.	
II-j	SDB	A reaction time of 2 days is indicated. This reaction time is reasonably long.	C/D
II-k	REF	Samples are agitated continuously.	A/B
II-l	REF	No isotherms, no variation of the Am concentration is indicated.	C/D
II-m	REF	The experiments are carried out in polypropylene vessels. It is reported	
		sorption to the vessel wall has been corrected for. No information about the	
		correction procedure is available. It is assumed that the correction is based on	
		the data obtained of a blank experiment.	C/D

II-n	REF	The reported error represents the standard deviation of the average of at least	А
		four experimental repetitions.	
II-o	REF	No parameter variation is indicated.	D

Data	Data table Am/10: REF: Tanaka et al. (1996)				
JNC-S	SDB vers	ion 2 – DATA: Am /Mudstone, # 60389 – 60394			
GUID	ELINE:	Revision 4b (May 19, 2005)			
Check	Checkpoint Evaluation Rating				
I-a.1	SDB	All mandatory fields are completed.	Yes		
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No		
		on initial and final Eh, final solution composition and atmosphere/ redox			
		condition are given.			
	REF	No further information is available. Americium is not redox sensitive.	unreliable		

table All	n/11: REF: Tanaka and Muraoka (1999)	
SDB vers	ion 2 – DATA: Am /Mudstone, # 62326-62328	
DELINE:	Revision 4b (May 19, 2005)	
kpoint	Evaluation	Rating
SDB	All mandatory fields are completed.	Yes
SDB	All mandatory information is provided.	Yes
SDB	It is indicated that K _d values are scanned from a graphical presentation.	
REF	The K _d –axis is in logarithmic form.	class 5
SDB	The sold phases are reddish soil, yellowish soil and ando soil. It is reported	
	CEC (reddish soil: 8.7 meq/100g, yellowish soil: 5.2meq/100g, ando soil: 17.2	
	meq/100g). Specific surface area is not reported. Physico-chemical properties	А
	of the sedimentary samples are reported.	
SDB	The initial and final pH values are given.	А
SDB	It is reported that experiments are done under air; i.e., under oxidizing	A/B
	conditions. Experiments are carried out with Am (III). Also, Am is not a	
	redox-sensitive element.	
REF	Physico-chemical properties of the equilibrated solution are reported.	A/B
SDB	Temperatures of 25°C by using a water jacket are reported.	A/B
SDB	Solution/solid ratios are 20mL/g.	
	DELINE: kpoint SDB SDB SDB SDB SDB SDB REF SDB	SDBAll mandatory fields are completed.SDBAll mandatory information is provided.SDBIt is indicated that K _d values are scanned from a graphical presentation.REFThe K _d –axis is in logarithmic form.SDBThe sold phases are reddish soil, yellowish soil and ando soil. It is reportedCEC (reddish soil: 8.7 meq/100g, yellowish soil: 5.2meq/100g, ando soil: 17.2meq/100g). Specific surface area is not reported. Physico-chemical propertiesof the sedimentary samples are reported.SDBSDBIt is reported that experiments are done under air; i.e., under oxidizing conditions. Experiments are carried out with Am (III). Also, Am is not a redox-sensitive element.REFPhysico-chemical properties of the equilibrated solution are reported.SDBTemperatures of 25°C by using a water jacket are reported.

r			1 1
	REF	Solid masses of 0.5g are added to a liquid volume of 10 mL. The used mass	A/B
		minerals may be considered enough surface area compared to that of the	
		vessel.	
II-g	SDB	Based on the information given in the JNC-SDB, all sorption values are	C/D
		calculated to >99.8% sorbed.	
II-h	SDB	An initial [Am] of $3.27 \times 10^{-7}$ M is indicated.	
	REF	Based on thermodynamic calculations with PHREEQC (database:	
		JNC-TDB_011213c2), the experiments are considered to be below the	
		solubility limit (but it is not clear by how much, due to the uncertainty in	В
		initial concentration as well as in solubility data for Am).	
II-i	SDB	The phase separation is carried out by Millipore filters of $0.45 \mu m$ pore	C/D
		diameter.	
II-j	SDB	A reaction time of 7 days is indicated. This reaction time is reasonably long.	C/D
II-k	REF	The sample is gently agitated on a shaker at 60 rpm for 7 days.	A/B
II-l	REF	No variation is indicated for either W/S or [Am].	C/D
II-m	REF	The material of the vessel is not reported.	C/D
II-n	REF	Error of $K_d$ is not reported. No replicated are performed.	D
II-o	REF	No parameter variation is indicated.	D

Data	Data table Am/12: REF: Tanaka and Muraoka (1998)					
JNC-	JNC-SDB version 2 – DATA: Am /Mudstone, # 62367 - 62369					
GUIE	DELINE:	Revision 4b (May 19, 2005)				
Chec	kpoint	Evaluation	Rating			
I-a.1	SDB	All mandatory fields are completed.	Yes			
I-a.2	SDB	All mandatory information is provided.	Yes			
I-b	SDB	Tables with K _d are given.	class 1			
II-a	SDB	CEC and chemical composition are reported about the ando soil, yellowish				
		soil and reddish soil.				
	REF	Humic substance and humic acid are reported. Surface area are not reported.	В			
II-b	SDB	Initial pH values and final pH values (as range< 1 pH unit) are reported.	В			
II-c	SDB	It is reported that experiments are done under air; i.e., under oxidizing	A/B			
		conditions. Also, Am is not a redox-sensitive element.				
II-d	SDB	Physico-chemical properties of the equilibrated solution are reported.	A/B			
II-e	SDB	A temperature of 25°C is reported.	A/B			

II-f	SDB	Solution/solid ratio is 20mL/g.	A /D
11-1			A/B
	REF	The used mass of 0.5g soil is added to liquid volumes (de-ionized water) of	
		10mL. It is assume that enough solid had been added to each vessel.	
II-g	SDB	The sorption values are >99.64 % (calculated with W/S and $K_d$ values).	C/D
II-h	SDB	An initial [Am] of $3.27 \times 10^{-7}$ M is indicated.	
	REF	Based on thermodynamic calculations with PHREEQC (database:	
		JNC-TDB_011213c2), the experiments are considered to be below the	
		solubility limit (but it is not clear by how much, due to the uncertainty in	
		initial concentration as well as in solubility data for Am).	В
II-i	SDB	The 5000MWCO filters are used for phase separation.	В
II-j	SDB	A reaction time of 7 days is indicated.	
	REF	Identical (similar) results are obtained with different reaction times (>48	A/B
		hours).	
II-k	REF	The sample is shaking for 7 days.	A/B
II-l	REF	No isotherms, no variation of the Am concentration is indicated.	C/D
II-m	REF	The reaction vessel is made from teflon. Corrections for sorption on vessel	В
		walls are not mentions.	
II-n	SDB	Error of $K_d$ is not reported. And no replicated are performed.	D
II-o	REF	No parameter variation is indicated.	D

Data	table Am	n/13: REF: Tanaka et al. (1999)	
JNC-S	SDB vers	ion 2 – DATA: Am /Mudstone, # 62810 - 62814	
GUID	ELINE:	Revision 4b (May 19, 2005)	
Check	spoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	Not all mandatory information is provided: it is indicated that no information	No
		on phase separation is given.	
	REF	No further information is available.	unreliable

Data	table An	n/14: REF: Baston et al., (1998)	
JNC-	SDB vers	ion 2 – DATA: Am /Bentonite, # 65940 – 65945	
GUII	DELINE:	Revision 4b (May 19, 2005)	
Chec	kpoint	Evaluation	Rating
I-a.1	SDB	All mandatory fields are completed.	Yes
I-a.2	SDB	All mandatory information is provided.	Yes
I-b	SDB	Tables with K _d are given.	class 1
II-a	SDB	The solid phase is indicated as Kunigel-V1.	А
	REF	The surface area is $6.80 \text{m}^2/\text{g}$ . The concentration of sites $2.88 \times 10^{-2} \text{ mol dm}^{-3}$ .	
II-b	SDB	Final pH values are reported.	А
	REF	No use of any pH-buffer is reported. The pH value is monitored every 7 days,	
		the pH adjusted back to the appropriate value.	
II-c	SDB	Experiments are performed under $N_2$ . The oxidation state of Am is reported as	
		III. Final Eh values are provided, as well as the reducing agent used $(Na_2S_2O_4)$	
		at an initial concentration of $2.5 \times 10^{-3}$ M.	
	REF	O ₂ levels are reportedly <1 ppm. All solutions are de-oxygenated by bubbling	
		an argon(96%) /hydrogen(4%) mixture through them before posting into the	
		glovebox. Also, Am is not a redox sensitive element.	A/B
II-d	SDB	A NaCl, NaHCO ₃ is reported as input solution; main solution compositions are	
		indicated in the additional file "solution composition".	A/B
	REF	Samples are filtered using a 10000MWCO filter prior to analysis.	
II-e	SDB	A temperature of 60°C is reported.	A/B
II-f	SDB	It is indicated that W/S ratio of 20 mL/g had been used, but that the specific	
		solid mass and solution volume are not reported.	A/B
		Assuming that the solution volume is $\geq 20$ mL would mean a corresponding	
		solid mass of at least 1 g. This is accepted as reasonable.	
II-g	SDB	Based on the information provided in the JNC-SDB, all $K_d$ values can be	C/D
		calculated to correspond to %-sorbed values >99%.	
II-h	SDB	$1.2 \times 10^{-10}$ M is indicated as initial Am concentration. Based on the data in Rai	
		et al.(1999), as well as on speciation calculations using the thermodynamic	В
		data in Guillaumont et al.(2003), it is estimated that the initial Am	
		concentration is below the respective solubility limit, but possibly by less than	
		a factor of 5.	
II-i	SDB	Filtration through 10000MWCO and 45µm membranes, as well as	А
	REF	centrifugation at 1100g for 2.5 h is reported.	
		Three methods of phase separation are used: 1) centrifugation at 1100 g for 2.5	

		h, 2) centrifugation at 1100 g for 2.5 h followed by filtration through 0.45 $\mu m$	
		membranes, and 3) centrifugation at 1100 g for 2.5 h followed by filtration	
		through 0.45 $\mu$ m membranes and then by filtration through 10000 MWCO	
		filters. It appears that the resulting $K_d$ values are slightly dependent on the	
		method of phase separation: surprisingly, the highest $K_d$ is generally obtained	
		with centrifugation alone, whereas filtration with 10000 MWCO membranes	
		leads to the lowest values. However, the respective variation is of about the	
		same magnitude as observed for different W/S. Therefore, it may be	
		considered to be within the overall experimental uncertainties.	
II-j	SDB	A reaction time of 122 days (four months) is reported. This reaction time is	
		reasonably long.	C/D
II-k	REF	The experiments are gently agitated on a shaker table.	A/B
II-l	SDB	No variation of W/S ratio or of the initial Am concentration is reported.	C/D
II-m	REF	The experiments are carried out in polypropylene centrifuge tubes. And	А
		corrections for sorption on vessel walls have been performed.	
II-n	SDB	It is indicated that each experiment is done in duplicate.	А
	REF	Error estimates are given for each replicate, based on analytical uncertainties.	
II-o	SDB	No relevant parameter variation is indicated.	D

#### 3.2 Criterion III

#### 3.2.1 Examination of data in the JNC-SDB: Thorium

The following entries are evaluated in this section; the respective data are shown in Figure 3.2.1:

Reference	Solid phase (group/solids)
Lieser et al. (1990)	mudstone / glacial sediment
Legoux et al. (1992)	mudstone / soil
Östhols (1995)	other minerals / SiO ₂
Bradbury & Baeyens (2003)	Bentonite / SWy-1 montmorillonite

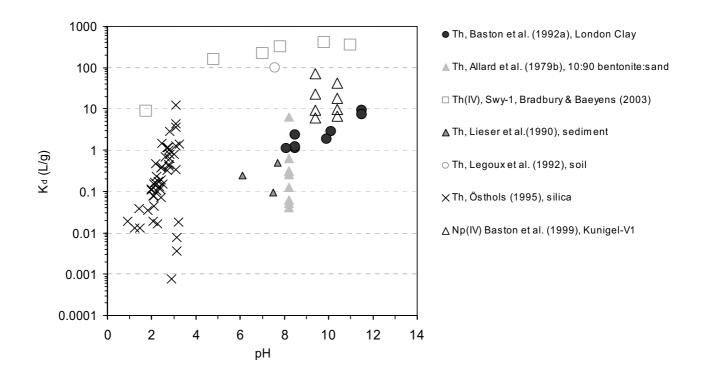


Figure 3.2.1 Overview of sorption data for Th on relevant substrates. The data by Baston et al. (1992a) refer to experiments with cement water in the absence of gluconate or degradation products, obtained after filtration through 30000 MWCO membranes. Relevant data for Np(IV) by Baston et al (1999) are also shown.

In comparison with the  $K_d$ -values obtained by Bradbury and Baeyens (2003) on pure montmorillonite (Swy-1), relatively low sorption of Th has been observed by both Allard et al. (1979) and Baston et al. (1992a), as well as Lieser et al. (1990). In all these cases, solution chemistry had been relatively complex, including presumably significant concentrations of carbonate, silicate, etc. Because the influence of such ligands on Th sorption is not known precisely, due to the lack of systematic sorption data, these data cannot be evaluated conclusively according to Criterion III, but they appear to be consistent among each other.

The data by Östhols appear to be consistent with the data by Bradbury and Baeyens (2003), considering that Th is expected to sorb on edge (i.e., oxide-type) surfaces and considering the hydrolysis behavior of Th at low-intermediate pH.

The datapoint by Legoux et al. (1992) appears to be consistent with the data by Bradbury and Baeyens (2003). On the other hand, the value measured by Legoux et al. (1992) is surprisingly high, considering the low CEC (3 meq/100 g) and the relatively complex solution chemistry in their experiments.

#### 4. Summary

The QA/classification of selected entries (1,056  $K_d$  values) in the JNC-SDB, especially of  $K_d$  values for mudstone systems to use in the  $K_d$ -derivation exercise for Horonobe rocks, was done following the classification guideline defined in our previous report.

Based on the results of the second application of classification guideline to  $K_d$  data for mudstone systems in the chapter 3, and the results of the application study of the JNC-SDB to  $K_d$ -setting in Horonobe rocks, some conclusions can be drawn;

- The classification guideline allows a suitable classification of the  $K_d$  values on the basis of the completeness of documented key information and the quality of the underlying experimental methods and conditions.
- The classification scheme made it possible to obtain quick overview of the available data, and to provide suitable access to the respective  $K_d$  values for the PA-related  $K_d$ -setting.
- Finaly, these approaches should be repeatedly tested through the application to various rocks and geochemical conditions.

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# Appendix

# Summary tables for $K_d$ classification

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	lusive, N.E. = not evaluated, m.i.	-																								
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			Rating 👻	I-a: yes/no, I-b: class 1-	6				Checkpoints 🛩 le	əvel: A-D (nı	umerical value	: 3-0) / unreliat	le Rating II	🖌 class 1-6	6 / unreli	iable			1 1							
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56105	Lieser et al. (1990)	yes		yes (can be used)	class 4	C/D 0	A 3	A/B 3 A/E		0 A/B	3 C/D 0			3 C/D	0 A/E		0 B	2 D	0 D	-	131	class 2	not conclusive		BMG January 2007	Revision 4b (May 19, 2005
56106 56107	Lieser et al. (1990) Lieser et al. (1990)	yes yes		yes (can be used) yes (can be used)	class 4 class 4	C/D 0 C/D 0	A 3 A 3			0 A/B 0 A/B	3 B 2 3 C/D 0				0 A/E	3 3 C/E 3 3 C/E		2 D 2 D	0 D		135 131	class 2 class 2	not conclusive not conclusive		BMG January 2007 BMG January 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
56055	Legoux et al. (1992)	Ves	Ves	ves (can be used)	class 1	A 3	A 3	A/B 3 A/E	3 C/D	0 A/B	3 C/D 0	A 3		3 C/D	0 0/6	3 3 0/		0 A	3 D	0 0	141	class 2	not conclusive	S Kupze/M Ochs	BMG November 20	6 Revision 4b (May 19, 2005
50055	Legoux et al. (1992)	yes	yes	yes (can be used)	CidSS 1	A J	A J	AB 5 AB	3 0/0	U AB	3 0/0 0	AJ		3 0/0	U A/E	5 3 C/L			3 0	, 0	141	CId55 Z	hot conclusive	3.Runze/w.ochs,	Bivig November 20	0 Revision 40 (May 19, 2000
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Other minerals								conditions com	position		value		separation	n time		Ioa	ading ves	sels estima	ates va	tota	al value	class				
57172	Östhols (1995)	yes	Vec	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B	3 A 3	Δ 3	в	2 C/D	0 A/E	2 3 B	2 C/D	0 B	2 A	A 3	168	class 1	consistent with independent data	S Kupze/M Ochs	BMG November 20	6 Revision 4b (May 19, 2005
57173	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B	3 A 3	A 3	B 2	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A	A 3	168	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57174 57175	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3 3 A 3			2 C/D 2 C/D	0 A/E		2 C/D 2 C/D		2 A	A 3 A 3	168 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57176	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3		A/B 3 A/E	3 A/B	3 A/B			B 2	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A	A 3	168	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57177 57178	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E	3 A/B	3 A/B 3 A/B	3 A 3 3 A 3	A 3 A 3	B 2		0 A/E 0 A/E	3 3 B	2 C/D 2 C/D	0 B	2 A 2 A	A 3	168 168	class 1 class 1	consistent with independent data consistent with independent data	S.Kunze/M.Ochs,	BMG November 20 BMG November 20	6 Revision 4b (May 19, 2005
57179 57180	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3 3 A 3			2 C/D 2 C/D	0 A/E		2 C/D 2 C/D		2 A		168 168	class 1 class 1	consistent with independent data		BMG November 20 BMG November 20	
57181	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B	3 A 3	A 3	B 2	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A 2 A	A 3	168	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57182 57183	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3 3 A 3	A 3		2 C/D 2 C/D	0 A/E		2 C/D 2 C/D		2 A	A 3 A 3	168 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57184	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 C/D	0 A 3	A 3	B	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A		162	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57185 57186	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3 3 A 3			2 C/D 2 C/D			2 C/D 2 C/D		2 A 2 A		168 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57187 57188	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 C/D 3 A/B	0 A 3 3 A 3				0 A/E		2 C/D 2 C/D		2 A		162 168	class 1 class 1	consistent with independent data		BMG November 20 BMG November 20	
57189	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B	3 A 3	A 3	B	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A	A 3	168	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57190 57191	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3			3 A/B 3 A/B	3 A 3 3 A 3			2 C/D 2 C/D			2 C/D 2 C/D		2 A 2 A		168 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57192	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B	3 A 3	A 3	B 2	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A	A 3	168	class 1	consistent with independent data		BMG November 20	6 Revision 4b (May 19, 2005
57193 57194	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3	A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3	A 3 A 3			0 A/E		2 C/D 2 C/D		2 A		168 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57195 57196	Östhols (1995) Östhols (1995)	yes ves		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 C/D 3 A/B	0 A 3 3 A 3			2 C/D 2 C/D	0 A/E		2 C/D 2 C/D				162 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57197	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B	3 A 3	A 3	B 2	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A	A 3	168	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57198 57199	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3 3 C/D 0				0 A/E		2 C/D 2 C/D		2 A 2 A		168 162	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57200 57201	Östhols (1995) Östhols (1995)	yes	yes	yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E	3 A/B	3 A/B 3 A/B	3 A 3 3 A 3		B	2 C/D	0 A/E	3 3 B	2 C/D 2 C/D	0 B	2 A	A 3	168 168	class 1 class 1	consistent with independent data		BMG November 20 BMG November 20	
57202	Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B 3 A/B	3 A 3		B 2	2 C/D	0 A/E		2 C/D 2 C/D		2 A		168	class 1	consistent with independent data consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57203 57204	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3 3 A 3	A 3			0 A/E		2 C/D 2 C/D		2 A		168 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57205	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B	3 C/D 0	A 3	B	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A	A 3	162	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57206 57207	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3 3 A 3			2 C/D 2 C/D	0 A/E		2 C/D 2 C/D		2 A 2 A		168 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57208 57209	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3	A 3	A/B 3 A/E A/B 3 A/E		3 C/D 3 A/B	0 C/D 0 3 C/D 0	A 3 A 3		2 C/D 2 C/D	0 A/E		2 C/D 2 C/D		2 A	A 3	156 162	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57210	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3		A/B 3 A/E	3 A/B	3 A/B	3 A 3	A 3	B 2	2 C/D	0 A/E	3 3 B	2 C/D	0 B		A 3	168	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57211 57212	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E A/B 3 A/E		3 A/B 3 A/B	3 A 3 3 A 3						2 C/D 2 C/D		2 A	A 3 A 3	168 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
57213	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E		3 A/B	3 C/D 0						2 C/D	0 B 0 B	2 A		162	class 1	consistent with independent data		BMG November 20	
57214 57215	Östhols (1995) Östhols (1995)	yes yes	yes	yes (can be used) yes (can be used)	class 2 class 2	A 3 A 3	A 3 A 3	A/B 3 A/E	3 A/B	3 A/B	3 A 3 3 A 3	A 3	B	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A	3	168 168	class 1 class 1	consistent with independent data consistent with independent data	S.Kunze/M.Ochs,	BMG November 20 BMG November 20	6 Revision 4b (May 19, 2005
57216 57217	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)				A/B 3 A/E A/B 3 A/E	3 A/B	3 A/B	3 A 3											class 1 class 1	consistent with independent data consistent with independent data			6 Revision 4b (May 19, 2005 6 Revision 4b (May 19, 2005
57218	Östhols (1995)	yes	yes	yes (can be used)	class 2	A 3	A 3	A/B 3 A/E	3 A/B	3 A/B	3 A 3	A 3	B	2 C/D	0 A/E	3 3 B	2 C/D	0 B	2 A	A 3	168	class 1	consistent with independent data	S.Kunze/M.Ochs,	BMG November 20	6 Revision 4b (May 19, 2005
57219 57220	Östhols (1995) Östhols (1995)	yes yes		yes (can be used) yes (can be used)	class 2 class 2			A/B 3 A/E A/B 3 A/E													156 168	class 1 class 1	consistent with independent data consistent with independent data		BMG November 20 BMG November 20	
67039	Bradbury & Baeyens (2003a)			yes (can be used)				A/B 3 A/E																	BMG January 2007	Revision 4b (May 19, 2005
67040	Bradbury & Baeyens (2003a)	yes	yes	yes (can be used)	class 5	A 3	A 3	A/B 3 A/E	3 C/D	0 A/B	3 A 3	B 2	B	2 C/D	0 A/E	3 3 C/E	A 0 C	3 C	1 B	3 2	146	class 2 class 2		S.Kunze/M.Ochs,	BMG January 2007	Revision 4b (May 19, 2005
67041 67042	Bradbury & Baeyens (2003a) Bradbury & Baeyens (2003a)	yes	yes	yes (can be used) yes (can be used)	class 5 class 5			A/B 3 A/E A/B 3 A/E													144 144	class 2 class 2			BMG January 2007 BMG January 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
67043	Bradbury & Baeyens (2003a)	yes	yes	yes (can be used)	class 5	A 3	A 3	A/B 3 A/E	3 C/D	0 A/B	3 A 3	B 2	B 2	2 C/D	0 A/E	3 3 C/E	A 0 C	3 C	1 B	3 2	146	class 2	and the set of the set	S.Kunze/M.Ochs,	BMG January 2007	Revision 4b (May 19, 2005
67044 67045	Bradbury & Baeyens (2003a) Bradbury & Baeyens (2003a)			yes (can be used) yes (can be used)	class 5 class 5	A 3	A 3	A/B 3 A/E A/B 3 A/E	3 C/D	0 A/B	3 C/D 0	B 2	B	2 C/D	0 A/E	3 3 C/E	A 0 C	3 C	1 B	3 2	144 140	class 2 class 2	used as reference data set for consistency-checks		BMG January 2007 BMG January 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
67046 67047	Bradbury & Baeyens (2003a)	yes	yes	yes (can be used)	class 5 class 5	A 3	A 3	A/B 3 A/E A/B 3 A/E	3 C/D	0 A/B	3 B 2	B 2	B 2	2 C/D	0 A/E	3 3 C/E	A 0 C	3 C	1 B	3 2	144	class 2 class 2		S.Kunze/M.Ochs,	BMG January 2007 BMG January 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
67048	Bradbury & Baeyens (2003a) Bradbury & Baeyens (2003a)	yes	yes	yes (can be used) yes (can be used)	class 5	A 3	A 3	A/B 3 A/E	3 C/D	0 A/B	3 C/D 0	B 2	B 2	2 C/D	0 A/E	3 3 C/E	A 0 C	3 C	1 B	3 2	140	class 2		S.Kunze/M.Ochs,	BMG January 2007	Revision 4b (May 19, 2005
67049 67050	Bradbury & Baeyens (2003a) Bradbury & Baeyens (2003a)			yes (can be used) yes (can be used)	class 5 class 5			A/B 3 A/E A/B 3 A/E													140 140	class 2 class 2			BMG January 2007 BMG January 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
67051	Bradbury & Baeyens (2003a)	yes	yes	yes (can be used)	class 5	A 3	A 3	A/B 3 A/E	3 C/D	0 A/B	3 C/D 0	B 2									140	class 2		S.Kunze/M.Ochs,	BMG January 2007	Revision 4b (May 19, 2005
67052 67053	Bradbury & Baeyens (2003a) Bradbury & Baeyens (2003a)	yes		yes (can be used) yes (can be used)	class 5 class 5			A/B 3 A/E A/B 3 A/E					+ +	+	_		+	+						S.Kunze/M.Ochs,	BMG January 2007 BMG January 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
67054	Bradbury & Baeyens (2003a)			yes (can be used)				A/B 3 A/E																	BMG January 2007	Revision 4b (May 19, 2005
62232	Meier et al. (1994)	yes		no (can not be used)																		unreliable		Y. Saito(JAE		Revision 4b (May 19, 2005
62233 62234	Meier et al. (1994) Meier et al. (1994)	yes yes		no (can not be used) no (can not be used)		+ $+$ $-$	+	+ $+$ $+$ $-$		-	+	+	+	+	-	+	+ $+$ $-$					unreliable		Y. Saito(JAE/ Y. Saito(JAE/		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
62235	Meier et al. (1994)	yes	no	no (can not be used)																	1	unreliable		Y. Saito(JAE/	A) October 2007	Revision 4b (May 19, 2005
62236	Meier et al. (1994)	yes	no	no (can not be used)			$\vdash$				+	+	+ +	+	_		+	+				unreliable		Y. Saito(JAE/	A) October 2007	Revision 4b (May 19, 2005
65701 65702	Ishii et al. (2001) Ishii et al. (2001)	yes		yes (can be used) yes (can be used)	class 1 class 1			A/B 3 A/E A/B 3 A/E													118 118	class 3 class 3		Y. Saito(JAE/ Y. Saito(JAE/		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
65703	Ishii et al. (2001)	yes yes	yes	yes (can be used)	class 1	A 3	C 1	A/B 3 A/E	3 A/B	3 A/B	3 C/D 0	B 2	C/D (	0 C/D	0 A/E	3 3 B	2 B	2 D	0 B	3 2	112	class 3		Y. Saito(JAE/	A) October 2007	Revision 4b (May 19, 2005
65704	Ishii et al. (2001)	yes	yes	yes (can be used)	class 1	A 3	C 1	A/B 3 A/E	3 A/B	3 A/B	3 C/D 0	B 2	C/D (	0 C/D	0 A/E	3 3 B	2 B	2 A	3 B	3 2	118	class 3		Y. Saito(JAE	A) October 2007	Revision 4b (May 19, 2005

N.C. = not con	lusive, N.E. = not evaluated, m.i. =	missing infe	ormation																							
Np				documentation and a: yes/no, I-b: class								C	II - Technical heckpoints V level: A-D (numeri	and scientific quality cal value: 3-0) / unrelia	of reported data	s 1-6 / unreliable								II - Consistency Operator	Date	Classification Guideline
Datapoint	Reference	I-a.1	I-a.2	Rating I-a	Rating I-b	II.a solid phase		I.b oH ree	II.c dox conditions	II.d solution composition	II.e temperature	II S/	f II.g	II.h inital [RN]	II.i phase separation	II.j	II.k agitation	II.I RN loading	II.m reaction vess	II.n els error estimates	II.o parameter variation		ing II	comment/rating		
							P				temperature							Refloading				total value	class			
56095	Lieser and Muhlenweg (1988)	yes	yes	yes (can be used)	class 1		A	3 C	/D 0	A/B 3				unreliable									unreliable	S.Kunze/M.Ochs, BMG	November 2006	Revision 4b (May 19, 2005)
44274	Barney and Anderson (1979)	yes	no	no (can not be used)																			unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
44276	Barney and Anderson (1979) Barney and Anderson (1979) Barney and Anderson (1979)	yes	no	no (can not be used)																			unreliable unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
44278	Barney and Anderson (1979) Barney and Anderson (1979) Barney and Anderson (1979)	yes	no	no (can not be used)																			unreliable		October 2007	
	Barney and Brown (1979)	yes	no	no (can not be used)																			unreliable	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
44341	Barney and Brown (1979) Barney and Brown (1979)	yes yes	no	no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Berry et al. (1990a)	yes	no	no (can not be used)															+				unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
48145 48146 48147	Berry et al. (1990a) Berry et al. (1990a) Berry et al. (1990a)	yes yes yes	no	no (can not be used) no (can not be used) no (can not be used)											+			· · · · · · · · · · · · · · · · · · ·	+				unreliable unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Berry et al. (1990a) Berry et al. (1990a)	yes yes	no	no (can not be used) no (can not be used) no (can not be used)														· · · · · · · · · · · · · · · · · · ·	+				unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
48151	Berry et al. (1990a) Berry et al. (1990a)	yes yes		no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
48153	Berry et al. (1990a) Berry et al. (1990a)	yes	no	no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
48154 48155 48156	Berry et al. (1990a) Berry et al. (1990a) Berry et al. (1990a)	yes yes	no	no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Berry et al. (1990a) Berry et al. (1990a) Berry et al. (1990a)	yes yes	no	no (can not be used) no (can not be used) no (can not be used)															+				unreliable unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
48159 48160	Berry et al. (1990a) Berry et al. (1990a)	yes yes	no no	no (can not be used) no (can not be used)																			unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
48161 48162	Berry et al. (1990a) Berry et al. (1990a)	yes yes	no no	no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
48163 48164 48165	Berry et al. (1990a) Berry et al. (1990a) Berry et al. (1990a)	yes yes	no	no (can not be used) no (can not be used)						1					1		1		+				unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 10, 2005)
48166	Berry et al. (1990a) Berry et al. (1990a) Berry et al. (1990a)	yes yes yes	no	no (can not be used) no (can not be used) no (can not be used)						+							1		++				unreliable unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
48168	Berry et al. (1990a) Berry et al. (1990a)	yes	no	no (can not be used) no (can not be used) no (can not be used)														· · · · · · · · · · · · · · · · · · ·	+				unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
48170 48171	Berry et al. (1990a) Berry et al. (1990a)	yes yes	no no	no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Berry et al. (1990a)			no (can not be used)																			unreliable	Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005)
53250	Higgo et al. (1987) Higgo et al. (1987) Higgo et al. (1987)	yes yes	no	no (can not be used) no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
53252	Higgo et al. (1987) Higgo et al. (1987) Higgo et al. (1987)	yes	no	no (can not be used) no (can not be used) no (can not be used)																			unreliable unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
53254	Higgo et al. (1987) Higgo et al. (1987)	yes	no	no (can not be used) no (can not be used)															+				unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
53256 53257	Higgo et al. (1987) Higgo et al. (1987)	yes	no no	no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
	Higgo et al. (1987) Higgo et al. (1987)	yes yes	no	no (can not be used) no (can not be used)																			unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Higgo et al. (1987) Kim et al. (1994)	yes		no (can not be used)																			unreliable	Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005)
	Legoux et al. (1992)			no (can not be used) yes (can be used)	class 1	A 3	A	3 C	/D 0	A/B 3	C/D 0	A/B	3 A 3	B 2	A 3	C/D 0	A/B 3	C/D 0	C/D	D A 3	D 0	115	class 3	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
56024	Legoux et al. (1992) Legoux et al. (1992)	yes	yes	yes (can be used) yes (can be used)	class 1 class 1	A 3	A	3 C	/D 0		C/D 0	A/B	3 A 3	B 2 B 2	A 3 A 3	C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0	C/D C/D	D A 3 D A 3		115 115	class 3 class 3	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
	Legoux et al. (1992)	yes	yes	yes (can be used)	class 1	A 3	A	3 C	:/D 0	A/B 3	C/D 0	A/B	3 A 3 3 A 3	B 2	A 3	C/D 0	A/B 3	C/D 0	C/D	D A 3	D 0	115	class 3		ļ	Revision 4b (May 19, 2005)
56148	Mackenzie et al. (1983) Mackenzie et al. (1983) Mackenzie et al. (1983)	yes	no	no (can not be used) no (can not be used) no (can not be used)																			unreliable unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Mackenzie et al. (1983)	yes yes		no (can not be used)																			unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
58697 58698	Sakamoto et al. (1990) Sakamoto et al. (1990)	yes yes	yes yes	yes (can be used) yes (can be used)	class 5 class 5	B 2 B 2	B		/B 3 /B 3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0 C/D 0	C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	C 1 C 1	83 83	class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
58700	Sakamoto et al. (1990) Sakamoto et al. (1990)	yes	yes	yes (can be used) yes (can be used)	class 5 class 5	B 2 B 2	B	2 A 2 A	/B 3 /B 3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0	C/D 0	A/B 3	C/D 0 C/D 0	A	3 D 0	C 1 C 1	83 83	class 4 class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
58701 58702	Sakamoto et al. (1990) Sakamoto et al. (1990) Sakamoto et al. (1990)	yes	yes	yes (can be used) yes (can be used)	class 5 class 5 class 5	B 2 B 2	B		/B 3 /B 3 /B 3	C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0 C/D 0	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	C/D 0 C/D 0	A A	3 D 0 3 D 0	C 1 C 1	83	class 4 class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Sakamoto (1994)			yes (can be used) yes (can be used)	class 5	B 2	B	1	/B 3	C/D 0	A/B 3	C/D	0 A 3	B 2	C/D 0	C/D 0	A/B 3	C/D 0	A	3 D 0	<u> </u>	83	class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
58719 58720	Sakamoto (1994) Sakamoto (1994)	yes yes	yes	yes (can be used) yes (can be used)	class 4 class 4	B 2 B 2	B	2 A 2 A	/B 3 /B 3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0 C/D 0	C/D 0 C/D 0	A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	C 1 C 1	83	class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
58722	Sakamoto (1994) Sakamoto (1994)	yes	yes yes	yes (can be used) yes (can be used)	class 4 class 4	B 2 B 2	B	2 A	/B 3 /B 3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0 C/D 0	C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	C 1 C 1	83 83	class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
58723 58724	Sakamoto (1994) Sakamoto (1994)	yes yes	yes	yes (can be used) yes (can be used)	class 4 class 4	B 2 B 2	B	2 A	/B 3 /B 3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0 C/D 0	C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	C 1 C 1	83	class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Sakamoto (1994) Sakamoto (1994) Sakamoto (1994)	yes	yes	yes (can be used) yes (can be used) yes (can be used)	class 4 class 4 class 4	B 2 B 2	B	2 A	/B 3 /B 3 /B 3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0 C/D 0	C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0		83	class 4 class 4 class 4		October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
58728 58729	Sakamoto (1994) Sakamoto (1994)	yes yes	yes yes	yes (can be used) yes (can be used)	class 4 class 4 class 4	B 2 B 2	B	2 A 2 A	/B 3 /B 3	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0 C/D 0	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0 C/D 0	A	3 D 0 3 D 0	C 1 C 1	83	class 4 class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Sakamoto (1994) Sakamoto (1994)	yes	yes	yes (can be used) yes (can be used)	class 4 class 4	B 2 B 2	B	2 A	/B 3 /B 3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0 A 3 0 A 3	B 2 B 2	C/D 0 C/D 0	C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	C 1 C 1	83 83	class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59864	Tachi et al. (1999a)	yes	yes	yes (can be used)	class 1	C/D 0	A		/D 0	C/D 0	A/B 3	A/B	3 A 3	B 2	B 2	C/D 0	A/B 3	C/D 0	A	3 D 0	B 2	93	class 3	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
59865 59866 59867	Tachi et al. (1999a) Tachi et al. (1999a) Tachi et al. (1999a)	yes	yes	yes (can be used) yes (can be used) yes (can be used)	class 1 class 1 class 1	C/D 0 C/D 0 C/D 0	A A	3 C	2/D 0 2/D 0 2/D 0	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	A/B A/B A/B	3 A 3 3 A 3	В 2 В 2	B 2	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	C/D 0 C/D 0 C/D 0	A A A	D         0           3         D         0           3         D         0	B 2 B 2	93 93 91	class 3 class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59868 59869	Tachi et al. (1999a) Tachi et al. (1999a) Tachi et al. (1999a)	yes yes yes	yes yes	yes (can be used) yes (can be used)	class 1 class 1 class 1	C/D 0 C/D 0 C/D 0	A	3 C	/D 0	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	A/B A/B	3 B 2 3 B 2	B 2 B 2	B 2 B 2	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	C/D 0 C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	91	class 3 class 3 class 3	Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59870 59871 59872	Tachi et al. (1999a) Tachi et al. (1999a)	yes yes	yes yes	yes (can be used) yes (can be used)	class 1 class 1	C/D 0 C/D 0	A	3 C 3 C	/D 0	C/D 0 C/D 0	A/B 3 A/B 3	A/B A/B	3 C/D 0 3 C/D 0	B 2 B 2	B 2 B 2	A/B 3 A/B 3	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	93 93	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59873	Tachi et al. (1999a) Tachi et al. (1999a)	yes yes	yes yes	yes (can be used) yes (can be used)	class 1 class 1	C/D 0 C/D 0	A	3 C	/D 0	C/D 0 C/D 0	A/B 3 A/B 3	A/B A/B	3 C/D 0 3 C/D 0	B 2 B 2	B 2 B 2	A/B 3 A/B 3	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	93	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59874 59875 59876	Tachi et al. (1999a) Tachi et al. (1999a) Tachi et al. (1999a)	yes yes	yes	yes (can be used) yes (can be used) yes (can be used)	class 1 class 1 class 1	C/D 0 C/D 0 C/D 0	A	3 C	I/D 0 I/D 0 I/D 0	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	A/B A/B A/B	3 C/D 0 3 C/D 0 3 C/D 0	В 2 В 2	B 2 B 2	A/B 3 A/B 3 A/B 3	A/B 3 A/B 3 A/B 3	C/D 0 C/D 0 C/D 0	A	3 D 0 3 D 0 3 D 0	B 2 B 2	93 93 93	class 3 class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Tachi et al. (1999a) Tachi et al. (1999a) Tachi et al. (1999a)	yes yes yes	yes	yes (can be used) yes (can be used) yes (can be used)	class 1 class 1 class 1	C/D 0 C/D 0 C/D 0	A	3 C	/D 0	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	A/B A/B A/B	3 C/D 0 3 C/D 0 3 C/D 0	B 2 B 2	B 2 B 2	A/B 3 A/B 3 A/B 3	A/B 3 A/B 3 A/B 3	C/D 0 C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	93	class 3 class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59879 59880	Tachi et al. (1999a) Tachi et al. (1999a)	yes yes	yes yes	yes (can be used) yes (can be used)	class 1 class 1	C/D 0 C/D 0	A	3 C 3 C	/D 0	C/D 0 C/D 0	A/B 3 A/B 3	A/B A/B	3 A 3 3 A 3	B 2 B 2	B 2 B 2	C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	93	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59881 59882	Tachi et al. (1999a) Tachi et al. (1999a)	yes yes	yes yes	yes (can be used) yes (can be used)	class 1 class 1	C/D 0 C/D 0	A	3 C 3 C	/D 0	C/D 0 C/D 0	A/B 3 A/B 3	A/B A/B	3 A 3 3 C/D 0	B 2 B 2	B 2 B 2	C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	93 87	class 3 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59883 59884 59885	Tachi et al. (1999a) Tachi et al. (1999a)	yes yes	yes yes	yes (can be used) yes (can be used)	class 1 class 1	C/D 0 C/D 0	A	3 C	/D 0	C/D 0 C/D 0	A/B 3 A/B 3	A/B A/B	3 B 2 3 B 2	B 2 B 2	B 2 B 2	C/D 0 C/D 0	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	91 91	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59886	Tachi et al. (1999a) Tachi et al. (1999a) Tachi et al. (1999a)	yes yes yes	yes	yes (can be used) yes (can be used) yes (can be used)	class 1 class 1 class 1	C/D 0 C/D 0 C/D 0	A	3 C	:/D 0 :/D 0 :/D 0	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	A/B A/B A/B	3 C/D 0 3 C/D 0 3 C/D 0	B 2 B 2	B 2 B 2	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	C/D 0 C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	87	class 4 class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59888 59889	Tachi et al. (1999a) Tachi et al. (1999a) Tachi et al. (1999a)	yes yes yes	yes	yes (can be used) yes (can be used) yes (can be used)	class 1 class 1 class 1	C/D 0 C/D 0 C/D 0	A	3 C	/D 0	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	A/B A/B	3 C/D 0 3 C/D 0 3 C/D 0	B 2 B 2	B 2 B 2	A/B 3 A/B 3	A/B 3 A/B 3 A/B 3	C/D 0 C/D 0 C/D 0	A A	3 D 0 3 D 0	B 2 B 2	93	class 4 class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59890 59891	Tachi et al. (1999a) Tachi et al. (1999a)	yes yes	yes yes	yes (can be used) yes (can be used)	class 1 class 1	C/D 0 C/D 0	A	3 C 3 C	/D 0	C/D 0 C/D 0	A/B 3 A/B 3	A/B A/B	3 C/D 0 3 C/D 0	B 2 B 2	B 2 B 2	A/B 3 A/B 3	A/B 3 A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	93 93	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
59892 59893	Tachi et al. (1999a) Tachi et al. (1999a)	yes	yes	yes (can be used) yes (can be used)	class 1 class 1	C/D 0 C/D 0	A	3 C	/D 0	C/D 0 C/D 0	A/B 3 A/B 3	A/B A/B	3 C/D 0	B 2 B 2	B 2 B 2	A/B 3 A/B 3	A/B 3	C/D 0 C/D 0	A	3 D 0 3 D 0	B 2 B 2	93 93	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
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 $-95 \sim 96 -$ 

62312 62313	Tanaka and Muraoka (1999) Tanaka and Muraoka (1999)	yes yes	no no (can not be used) no no (can not be used)																									unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
62314	Tanaka and Muraoka (1999)	yes																										unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
62353 62354	Tanaka and Muraoka (1998) Tanaka and Muraoka (1998)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	B 2 B 2	B	2	A/B A/B	3 A/E 3 A/E	B 3	A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	B 2 B 2	D	0 0	1	1 134 1 134	class 2 class 2	Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
62355	Tanaka and Muraoka (1998)	yes	yes yes (can be used)	class 1	B 2	В	2	A/B	3 A/E	B 3	A/B	3	A/B	3 A	3	В 2	2 B	2	A/B :	3 A/	/B 3 C/D	0	B 2	D	0 0	1	1 134	class 2	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
62805 62806	Tanaka et al. (1999) Tanaka et al. (1999)	yes ves	no no (can not be used) no no (can not be used)								_																	unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
62807	Tanaka et al. (1999) Tanaka et al. (1999)	yes	no no (can not be used) no no (can not be used)																									unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Tanaka et al. (1999)		no no (can not be used)																									unreliable			Revision 4b (May 19, 2005)
63027 63028	El-Naggar et al. (2000) El-Naggar et al. (2000)	yes	no no (can not be used) no no (can not be used)																							_		unreliable unreliable		October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63029 63030	El-Naggar et al. (2000) El-Naggar et al. (2000) El-Naggar et al. (2000)	yes yes	no no (can not be used) no no (can not be used) no no (can not be used)								1															_		unreliable	Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63031	El-Naggar et al. (2000)	yes yes	no no (can not be used)																									unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
63033	El-Naggar et al. (2000) El-Naggar et al. (2000)	yes yes																										unreliable	Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63034 63035	El-Naggar et al. (2000) El-Naggar et al. (2000)	yes yes	no no (can not be used)																									unreliable unreliable	Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63037	El-Naggar et al. (2000) El-Naggar et al. (2000)	yes yes	no no (can not be used)																									unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63038 63039	El-Naggar et al. (2000) El-Naggar et al. (2000)	yes	no no (can not be used) no no (can not be used)																									unreliable		October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63040 63041	El-Naggar et al. (2000) El-Naggar et al. (2000)	yes yes	no no (can not be used) no no (can not be used)																									unreliable unreliable		October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63042 63043	El-Naggar et al. (2000) El-Naggar et al. (2000)	yes yes	no no (can not be used) no no (can not be used)																									unreliable unreliable	Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63044 63045	El-Naggar et al. (2000) El-Naggar et al. (2000)	yes yes																										unreliable unreliable	Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
63046	El-Naggar et al. (2000)	yes	no no (can not be used)																									unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
Bentonite																										_					
65074	Kitamura et al. (2001) Kitamura et al. (2001)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E	B 3 B 3	A/B A/B	3	A/B A/B	3 A 3 A	3	в 2 В 2	2 B 2 B	2	A/B A/B	3 A/	/B 3 C/D /B 3 C/D /B 3 C/D	0	В 2 В 2	D	0 E	2	2 128 2 128 2 128	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65076	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E	B 3	A/B A/B	3	A/B A/B	3 A 3 A	3	в 2 В 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	В 2 В 2	D	0 E	2	2 128	class 2 class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Kitamura et al. (2001) Kitamura et al. (2001)		yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E	В 3	A/B A/B	3	A/B A/B	3 A 3 A	3	в 2 В 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	в 2 В 2	D	0 E	2	2 128 2 128		Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65080	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E	B 3	A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2	A/B S		/B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 128		Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65082	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E	B 3	A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2	A/B A/B		/B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 128	class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65084	Kitamura et al. (2001) Kitamura et al. (2001)		yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A A	3	C/D C/D		B 3	A/B A/B		A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2 2	A/B A/B		/B 3 C/D		B 2 B 2	D	0 E	2	2 128 2 128	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65086	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E		A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2 2	A/B A/B	3 A/	/B 3 C/D /B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 128 2 128	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65087 65088	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E		A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2 2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 128	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65089 65090	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E		A/B A/B	3	A/B A/B	3 B 3 A	2	B 2 B 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/		0	B 2 B 2	D	0 E	2	2 126 2 128	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65091 65092	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E		A/B A/B	3	A/B A/B	3 B 3 A	2	B 2 B 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	B 2 B 2	D	0 E	2	2 126 2 128	class 2 class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65093 65094	Kitamura et al. (2001) Kitamura et al. (2001)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E		A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 128		Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65095	Kitamura et al. (2001) Kitamura et al. (2001)		yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E	B 3 B 3	A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/	/B 3 C/D /B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 128		Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65097 65098	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E		A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/		0	B 2 B 2	D	0 E	2	2 128 2 128	class 2 class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65099 65100	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E	B 3	A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2 2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 128	class 2 class 2		July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65101 65102	Kitamura et al. (2001) Kitamura et al. (2001)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E		A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2 2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 128	class 2 class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65103 65104	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E	B 3	A/B A/B	3	A/B A/B	3 A 3 C/D	3	B 2 B 2	2 B 2 B	2	A/B A/B	3 A/ 3 A/	/B 3 C/D	0	B 2 B 2	D	0 E	2	2 128 2 122	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
65106	Kitamura et al. (2001) Kitamura et al. (2001)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 A/E 0 A/E	В 3	A/B A/B	3	A/B A/B	3 A 3 A	3	B 2 B 2	2 B 2 B	2	A/B A/B	3 A/		0	B 2 B 2	D	0 E	2	2 128 2 128	class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Kitamura et al. (2001)		yes yes (can be used)	class 1	A 3	Α	3	C/D		B 3	A/B	3	A/B	3 A	3	B 2	2 B	2	A/B :		/B 3 C/D	0	B 2	D	0 E	2		class 2	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
64869 64870	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	A/B A/B	3 B 3 A	2	A S	3 A 3 A	3	A/B A/B	3 A/ 3 A/	/B 3 B	2	B 2 B 2	A	3 E 3 E	2	2 128 2 130	class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64871 64872 64873	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	A 3 A 3	A	3	C/D C/D C/D	0 C/E 0 C/E 0 C/E	D 0	A/B A/B A/B	3	A/B A/B A/B	3 A 3 A	3	A 3	3 A 3 A	3	A/B A/B A/B	3 A/ 3 A/ 3 A/	/B 3 B	2	B 2 B 2	A	3 E	2	2 130 2 130 2 130	class 2 class 2 class 2	Y. Saito(JAEA)	July 2007 July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64874 64875	Kitamura and Tomura(2003)		yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	A 3	A	3	C/D	0 C/E	D 0	A/B	3	A/B A/B	3 A	3	A	3 A	3	A/B A/B	3 A/	/B 3 B	2	B 2	A	3 E	2	2 130	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64876	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1	A 3 A 3	A	3	C/D C/D C/D	0 C/E 0 C/E 0 C/E	D 0	A/B A/B	3	A/B A/B	3 A 3 A	3	A S	3 A 3 A	3	A/B A/B	3 A/ 3 A/ 3 A/	/B 3 B	2	B 2 B 2	A	3 E	2	2 130 2 130 2 130	class 2	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64878	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)		yes yes (can be used)	class 1 class 1 class 1		A		C/D	0 C/E	D 0	A/B		A/B A/B	3 A 3 A	3	A S	3 A 3 A	3	A/B A/B		/B 3 B	2		A	3 E	2	2 130	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64880 64881	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	A 3 A 3	A	3	C/D C/D C/D	0 C/F		A/B	3	A/B A/B	3 A 3 A	3	A	3 A 3 A	3	A/B A/B	3 A/	/B 3 B /B 3 B	2 2	B 2 B 2	A	3 E	2	2 130	class 2 class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64882	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3	A	3	C/D C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	A/B A/B A/B	3 A	3	A 3	3 A 3 A	3	A/B A/B	3 A/	/B 3 B /B 3 B	2	B 2 B 2	A	3 6	2		class 2	Y. Saito(JAEA)	July 2007 July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64884 64885	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	A 3	A	3	C/D C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	A/B A/B	3 A	3	A 3	3 A	3	A/B A/B	3 A/ 3 A/	/B 3 B	2	B 2	A	3	2	2 130 2 130 2 130	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64885 64886 64887	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1	A 3	A	3	C/D	0 C/E	D 0	A/B A/B	3	A/B	3 A	3	A 3	3 A 3 A	3	A/B :	3 A/	/B 3 B	2	B 2	A	3 E	2	2 130	class 2	Y. Salto(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64887 64888 64889	Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3	A	3	C/D C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	A/B A/B	3 A	3	A 3 A 3	3 A 3 A	3	A/B A/B	3 A/	/B 3 B /B 3 B	2	B 2	A	3 E	2		class 2 class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005)
64890	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)		yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	A 3	A	3	C/D	0 C/E	D O	A/B A/B		A/B A/B A/B	3 A 3 A	3	A A	3 A 3 A	3	A/B A/B	3 A/	/B 3 B	2	B 2	A	3 6	2	2 130		Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64891 64892 64893	Kitamura and Tomura(2003)	yes	yes yes (can be used)	class 1 class 1 class 1	A 3	A	3	C/D C/D C/D	0 C/L 0 C/E 0 C/E	D 0 D 0	A/B A/B	3	A/B A/B A/B	3 A	3	A A A	A A A	3	A/B	3 A/ 3 A/ 3 A/		2	B 2	A	3 6	2		class 2 class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64893 64894 64895	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1	A 3	A	3	C/D	0 C/E	D 0	A/B A/B	3	A/B A/B A/B	3 A 3 A	3	A S	3 A 3 A	3	A/B A/B A/B	3 A/		2	B 2	A	3 E	2	2 130	class 2		July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64895 64896 64897	Kitamura and Tomura(2003)	yes yes	yes yes (can be used)	class 1 class 1 class 1	A 3	A	3	C/D C/D	0 C/E 0 C/E 0 C/E	D 0	A/B A/B	3	A/B A/B A/B	3 A 3 A 3 A	3	A 3 A 3	3 A 3 A 3 A	3	A/B A/B	3 A/	/B 3 B	2	B 2	A	3 E	2	2 130 2 130 2 130	class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005)
64898	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1	A 3	A	3	C/D C/D	0 C/E	D 0	A/B	3	A/B	3 A 3 A	3	A S	3 A 3 A	3	A/B :	3 A/ 3 A/	/B 3 B	2	B 2 B 2	A	3 E	2	2 130	class 2 class 2	Y. Saito(JAEA)		Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64899 64900	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	A/B A/B	3 A 3 A	3	A 3	3 A 3 A	3	A/B C	3 A/		2	B 2	A	3 E	2	2 130	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64901 64902	Kitamura and Tomura(2003) Kitamura and Tomura(2003)		yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D		D 0	A/B A/B		A/B A/B	3 A 3 A	3	A 3	3 A 3 A	3	A/B	3 A/	/B 3 B /B 3 B	2	B 2	A	3 E	2		class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D		D 0 D 0	A/B A/B		A/B A/B	3 A	3	A 3	3 A 3 A	3	A/B A/B		/B 3 B /B 3 B	2	B 2	A	3 E	2	2 130	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64905 64906	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 C/E 0 C/E		A/B A/B	3	C/D C/D	0 A 0 A	3	A 3	3 A 3 A	3	A/B 3	3 A/ 3 A/		2	B 2	A	3 E	2	2 124 2 124		Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64907 64908	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	C/D C/D	0 A 0 A	3	A S	3 A 3 A	3	A/B A/B	3 A/		2	B 2	A	3 E	2	2 124 2 124	class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64909 64910	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	C/D C/D	0 A 0 A	3	A 3	3 A 3 A	3	A/B S		/B 3 B	2	B 2 B 2	A	3 E	2	2 124 2 124	class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64911 64912	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	C/D C/D	U A 0 A	3	A 3	3 A 3 A	3	A/B A/B	3 A/ 3 A/	/B 3 B	2	в 2 В 2	A	3 E	2		class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64913 64914	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A A	3	C/D C/D C/D	0 C/E 0 C/E		A/B A/B	3	C/D C/D	U A 0 A	3	A 3	3 A 3 A	3	A/B A/B	3 A/	/B 3 B /B 3 B	2	в 2 В 2	A	3 E	2		class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64915 64916 64917	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used)	class 1 class 1 class 1	A 3 A 3	A	3	C/D			A/B		C/D C/D	0 A 0 A	3	A 3	3 A 3 A	3	A/B A/B		/B 3 B	2	B 2	A	3 E	2	2 124	class 2 class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64917 64918 64919	Kitamura and Tomura(2003) Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	A 3	A	3	C/D C/D C/D	0 C/E 0 C/E 0 C/E	D 0	A/B A/B A/B	3	C/D C/D C/D	0 A	3	A S	3 A 3 A 3 A	3	A/B A/B	3 A/ 3 A/ 3 A/	/B 3 B	2	B 2	A	3 E	2	2 124		Y. Saito(JAEA)	July 2007 July 2007 July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64920	Kitamura and Tomura(2003)		yes yes (can be used) yes yes (can be used)	class 1	A 3 A 3	A	3	C/D	0 C/E	D 0	A/B	3	C/D	0 A	3	A 3	3 A 3 A	3	A/B :	3 A/	/B 3 B	2	B 2	A	3 E	2	2 124	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005)
64921 64922	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 C/E 0 C/E	D 0	A/B A/B	3	C/D C/D	0 A	3	A A	A A	3	A/B A/B		/B 3 B	2	B 2	A	3 E	2	2 124		Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64923 64924 64925	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3	A	3	C/D C/D	0 C/E 0 C/E 0 C/E	D 0	A/B A/B A/B	3	C/D C/D	0 A 0 A	3	A 3	3 A 3 A	3	A/B A/B	3 A/	/B 3 B /B 3 B /B 3 B	2	B 2 B 2	A	3 E	2	2 124	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
64926	Kitamura and Tomura(2003) Kitamura and Tomura(2003)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	A 3 A 3 A 3	A	3	C/D C/D				3	C/D C/D	0 B	3	A 3		3			/B 3 B /B 3 B /B 3 B	2	B 2	A	3 E	2		class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
04927	Kitamura and Tomura(2003)	yes	yes yes (can be used)	Class 1	A 3	A	1 3		U C/L		I A/B	1 3		U A	1. 3	n   3	ΑΑ	I	ND	, A/	JU J J B		PZ	A		12	2   124	UID35 2	Y. Saito(JAEA)	Joury 2007	1130VISIO11 40 (IVIAY 19, 2005)

 $-97 \sim 98 -$ 

$$97 \sim 98$$

64000	(2002)	(and he used)	alars 4		2 4	1 2	C/D	0	0/0	0 A/B	2	C/D	0		3 4	2		2	A/B	2 4/0	2			0	2	D 0	124	-l 0	X 0-it-(1454) I.t. 2007	Devision (h (May 40, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	· <u> </u>	C/D	0 0	0/0			5/D	0 A	<u> </u>	3 A	<u></u>	A			3 A/B		<u>2</u>		2 A		- <u>P</u> - 2	124	class 2	Y. Saito(JAEA) July 2007 Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	· · · · ·	C/D C/D	0 0	C/D	0 A/B		J/D		<u> </u>					A/B	3 A/B	3	<u></u>		2. A		- B . 2	124	class 2		
	tamura and Tomura(2003)	yes yes yes (can be used)		A 3	3 A	3						A/B	3 A		3 A	3	A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2		class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B		A/B	3 A		3 A	3	A	3	A/B	3 A/B	3	B 2	В	Z A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B		A/B	3 A	\	3 A	3	A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B		A/B	3 A	1	3 A	3	A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B		A/B	3 A	1	3 A		A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B		4/B	3 A	\	3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	amura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B		A/B	3 A	۱ L	3 A	3	A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64937 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B	3	A/B	3 A	۱ I.	3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64938 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 0	C/D	0 A/B	3	A/B	3 A	A	3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64939 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 0	C/D	0 A/B	3	A/B	3 A	1	3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64940 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 (	C/D	0 A/B	3	A/B	3 A	1	3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64941 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 0	C/D	0 A/B	3	A/B	3 A	1	3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64942 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B	3	A/B	3 A	1	3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64943 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 0	C/D	0 A/B	3	A/B	3 A	1	3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64944 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 0	C/D	0 A/B	3	A/B	3 A		3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64945 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 0	C/D	0 A/B	3	A/B	3 A	1	3 A	3	A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64946 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 0	C/D	0 A/B	3	A/B	3 A		3 A	3	A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64947 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D	0 0	C/D	0 A/B	3	A/B	3 A		3 A	3	A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	C/D		C/D	0 A/B		A/B	3 A		3 A	3	A	3	A/B	3 A/B	3	B 2	В	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	amura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 4	3	C/D		C/D	0 A/B		A/B	3		3 A	3	Δ	3	A/B	3 A/B	3	B 2	B	2 A	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 4	3	C/D		C/D	0 A/B		A/B	3 4		3 A	3	A	3	A/B	3 A/B	3	B 2	B	2 4	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	ves ves ves (can be used)	class 1	A 3	3 4	- 3	C/D		C/D	0 A/B		A/B	3 4		3 A		A	3	A/B	3 A/B	3	B 2	B	2 4	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	amura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3		C/D		C/D	0 A/B		A/B	3	<u> </u>	3 A		A		A/B	3 A/B	3	B 2		2 4	3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 4		C/D		C/D	0 A/B	3	A/B	3	<u> </u>	3 4	3	A		A/B	3 A/B	3	B 2		2 4	- 3	B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)		A 3	3 4	· · · · ·	C/D		C/D	0 A/B		A/B	3		3 A		A	3	A/B	3 A/B	3	B 2		2 A		B 2	130	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 4	3	A/B		C/D	0 A/B		A/B	3 4		3 R	2	A	3	A/B	3 A/B	3	B 2	B	2 R	2	B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	ves ves ves (can be used)	class 1	A 3	3 4	3	A/B		C/D	0 A/B		A/B	3		3 B	2	A	3	A/B	3 A/B	3	B 2	B	2 8	2	B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 4	3	A/B		C/D	0 A/B		A/B	3 4		3 B	2	A	3	A/B	3 A/B	3	B 2	B	2 8	2	B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 4	3	A/B		C/D	0 A/B		A/B	3		3 B	2	A	3	A/B	3 A/B	3	B 2		2 8	2	B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)		class 1	A 3	2 1		A/B		C/D	0 A/B		A/B	2 1		2 0	2	A	2	A/B	3 A/B	2	B 2		2 0	2	B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)		class 1	A 3	3 A		A/B		C/D	0 A/B		A/B	3 4	<u> </u>	3	2	A		A/B	3 A/B		D 2		2 0		D 2	144	class 2	Y. Saito(JAEA) July 2007 Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)		class 1	A 3	3 A		A/B		C/D	0 A/B		A/B	2 4		2 0	2	A	3	A/B	3 A/B	2	B 2		2 0	2	B 2	144	class 2	Y. Saito(JAEA) July 2007 Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)		class 1	A 3	3 A		A/B		C/D	0 A/B		4/B	3 4	<u>}  </u>	3	2	- A		AVD	2 A/D				2 0			144	class 2		
	tamura and Tomura(2003)	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	class 1	A 3	3 A	·	A/B		C/D	0 A/B		A/B	3 4	<u> </u>	3 · · · · · · · · · · · · · · · · · · ·		A		A/B	3 A/B	3			2 <u>B</u>		···· P ···· 2	144	class 2	Y. Saito(JAEA) July 2007 Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used) ves ves ves (can be used)	class 1	A 3	3 A	· <u> </u>	A/B		C/D	0 A/B		A/B	3		2 B		A		A/B	3 A/B		<u>2</u>		2 D		- <u>P</u> - 2	142	class 2	Y. Saito(JAEA) July 2007 Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
			class 1	A 3	3 A		A/B		C/D	0 A/B		A/B	3 4	<u> </u>	3 B		A		A/B	3 A/B		2		2 D			144	class 2		Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
	tamura and Tomura(2003) tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A		A/B		C/D C/D	0 A/B		A/B	3 A	<u> </u>	3 B	<u>2</u>	A		A/B	3 A/B	3	<u></u>		2 B		- B 2	144		Y. Saito(JAEA) July 2007	
		,000 ,000 ,000 (0011 00 0000)		A 3	3 A		A/B		C/D	0 A/B		A/B	3 A	<u> </u>	3 B		A		A/B	3 A/B	3	B 2		2 B	2	- B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	amura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3							3 A	<b>\</b>	3 B	2		3			3	B 2	В	2 B	2	B 2		class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	A/B A/B		C/D	0 A/B		A/B	3 A		3 B	2	A	3	A/B A/B	3 A/B 3 A/B	3	в 2	B	2 B	2	в 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3			C/D	0 A/B		A/B	3 A	\	3 B	2	A	3			3	B 2	В	2 B	2	B 2		class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	A/B		C/D	0 A/B		A/B	3 A	۱ <u> </u>	3 B	2	A	3	A/B	3 A/B	3	В 2	B	2 B	2	В 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	A/B		C/D	0 A/B		A/B	3 A	<u>ا</u>	3 B	2	A	3	A/B	3 A/B	3	B 2	B	2 B	2	B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	A/B		C/D	0 A/B		A/B	3 A	\	3 B	2	A	3	A/B	3 A/B	3	B 2	B	2 B	2	B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	A/B		C/D	0 A/B		4/B	3 C/		0 B	2	A	3	A/B	3 A/B	3	B 2	В	2 B	2	B 2	138	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)
64974 K	tamura and Tomura(2003)	yes yes yes (can be used)	class 1	A 3	3 A	3	A/B	3 (	C/D	0 A/B	3	4/B	3 A		3   B	2	A	3	A/B	3 A/B	3	B 2	B	2 B	2	B 2	144	class 2	Y. Saito(JAEA) July 2007	Revision 4b (May 19, 2005)

N.C. = not conclusive,	N.E. = not evaluated, m.i. = miss	ing information																																
Se		I - Complete	ness of documentation and	type of K _d					·····	·····				heckpointe	II - Te	chnical an	d scientific qu	uality of rep	orted data Rating I & class 1-6	/upreliable											III - Co	onsistency O	Operator Date	Classification Guideline
Datapoint	Reference		-a.2 Rating I-a	Rating I-b	ll.a solid pl		II.b pH		I.c onditions	II.d solution composition		II.e perature	1	I.f /W	II.	9	II.h inital [R		II.i phase separation	II.j reaction time		II.k litation	II. RN loa		II.m reaction ve		II.n error estimate		II.o er variation	Rating II	eva	aluation		
Bentonite																													total v	alue class	5			
44766	Barney(1981)	yes	yes yes (can be used)	class 1	C/D	0	C 3	unreliable					+										-							unreliat	ble /		Kunze/M.Ochs, BMG November 2	
44767 44768	Barney(1981) Barney(1981)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	C/D	0	C 3	unreliable unreliable			_											_								unreliat unreliat		N.E. S.	Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
44769	Barney(1981)	yes	yes yes (can be used)	class 1	C/D	0	C 3	unreliable															_							unreliat	ble /	N.E. S.	Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
44770 44771	Barney(1981) Barney(1981)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	C/D C/D	0	C 3 C 3	unreliable unreliable				-																		unreliat unreliat		N.E. S. N.E. S.	Kunze/M.Ochs, BMG November 2 Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
44772	Barney(1981) Barney(1981)	yes	yes yes (can be used)	class 1 class 1	C/D C/D	0	C 3	unreliable unreliable																						unreliat unreliat	ble /	N.E. S.	Kunze/M.Ochs, BMG November 2	006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)
44774	Barney(1981) Barney(1981)	yes	yes yes (can be used) yes yes (can be used)	class 1		0	C 3 C 3 C 3					-	+									_								unreliat	ble /	N.E. S.	Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005)
44775 44776	Barney(1981) Barney(1981)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	C/D C/D	0	C 3 C 3																							unreliat unreliat	ble /	N.E. S.	Kunze/M.Ochs, BMG November 2 Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005)
44777	Barney(1981)	yes	yes yes (can be used)	class 1	C/D	0	C 3	unreliable																						unreliat	ble /	N.E. S.	Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005)
44778	Barney(1981) Barney(1981)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	C/D C/D	0	C 3 C 3	unreliable unreliable					-																	unreliat unreliat		N.E. S. N.E. S.	Kunze/M.Ochs, BMG November 2 Kunze/M.Ochs, BMG November 2	006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)
44780 44781	Barney(1981)	yes	yes yes (can be used)	class 1	C/D C/D	0	C 3	unreliable																						unreliat	ble /	N.E. S.	Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005)
44782	Barney(1981) Barney(1981)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	C/D C/D	0	C 3	unreliable unreliable						1																unreliat unreliat		N.E. S. N.E. S.	Kunze/M.Ochs, BMG November 2 Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
44783 44784	Barney(1981) Barney(1981)	yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	C/D C/D	0	C 3 C 3	unreliable unreliable																						unreliat unreliat		N.E. S.	Kunze/M.Ochs, BMG November 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
60751	Ticknor(1988)	yes yes(t	entative) yes (can be used)	class 1	A	3	B 2	C/D C/D	0	C/D(tentative) 0	A/B A/B	3	A/B	3	A	3	A	3	B 2	C/D 0	A/B	3	C/D	0	C/D C/D	0	D 0	c	1 88	dass 4		N.E. S.	Kunze/M.Ochs, BMG December 2 Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
60752 60753	Ticknor(1988) Ticknor(1988)	yes yes(t yes yes(t	entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	в 2 В 2	C/D A/B		C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	в 2 В 2	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	C C	1 88	2 class 4			Kunze/M.Ochs, BMG December 2 Kunze/M.Ochs, BMG December 2	
60754	Ticknor(1988) Ticknor(1988)	yes yes(t	entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	B 2 B 2	A/B A/B	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	B 2	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	C	1 11 11 11	2 class 2	3 /	N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
60756	Ticknor(1988)	yes yes(t	entative) yes (can be used)	class 1	A	3	B 2	A/B	3	C/D(tentative) 0	A/B	3	A/B	3	A	3	A	3	B 2	C/D 0	A/B	3	C/D	0	C/D	0	D 0	C	1 11	2 class 3	3 /	NE S	Kunze/M Ochs BMG December 2	006 Revision 4b (May 19, 2005)
60757	Ticknor(1988) Ticknor(1988)	yes yes(t	entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	B 2 B 2	A/B A/B	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	B 2 B 2	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	C C	1 11	2 class 3 2 class 3		N.E. S.	Kunze/M.Ochs, BMG December 2 Kunze/M Ochs, BMG December 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
60759	Ticknor(1988)		entative) yes (can be used)		A	3	B 2			C/D(tentative) 0	A/B	3	A/B	3	A	3	A	3	B 2	C/D 0	A/B	3	C/D	0	C/D	0	D 0	c	1 86					006 Revision 4b (May 19, 2005)
Other Minerals (Chlo	rite, Muscovite)			_																		-									_			
60706	Ticknor(1988)		entative) yes (can be used)	class 1	A	3	B 2	C/D	0	C/D(tentative) 0	A/B	3	A/B	3	A	3	A	3	B 2	C/D 0	A/B	3	C/D	0	C/D	0	D 0	С	1 86	dass 4	4 /	N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
60707	Ticknor(1988) Ticknor(1988)		entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	B 2 B 2	A/B A/B	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	B 2 B 2	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	C C		2 class 2 2 class 2		N.E. S.	Kunze/M.Ochs, BMG December 2 Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
60709 60710	Ticknor(1988) Ticknor(1988)	yes yes(t	entative) yes (can be used)	class 1	A	3	B 2	A/B A/B	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	B 2	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	c	1 11		3 /	N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
60710	Ticknor(1988)		entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	B 2 B 2	C/D	0	C/D(tentative) 0 C/D(tentative) 0	A/B	3	A/B A/B	3	A	3	A	3	B 2	C/D 0 C/D 0 C/D 0	A/B	3	C/D C/D	0	C/D C/D C/D	0	D 0	c	1 88	2 class 3 class 4		N.E. S. N.E. S.	Kunze/M.Ochs, BMG December 2	006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)
60712	Ticknor(1988) Ticknor(1988)	yes yes(t	entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	B 2 B 2	A/B C/D	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	B 2	C/D 0	A/B A/B	3	C/D C/D	0	C/D	0	D 0	C	1 11	2 class 3 class 4		N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
60714	Ticknor(1988)	yes yes(t	entative) yes (can be used)	class 1	A	3	B 2	A/B	3	C/D(tentative) 0	A/B	3	A/B	3	Â	3	A	3	B 2	C/D 0 C/D 0	A/B	3	C/D	0	C/D C/D	0	D 0	c	1 11	2 class 3	3 /	N.E. S.	Kunze/M.Ochs, BMG December 2	006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)           006         Revision 4b (May 19, 2005)
60760 60761	Ticknor(1988) Ticknor(1988)	yes yes(t ves ves(t	entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	B 2 B 2	A/B A/B	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	B 2 B 2	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	<u>с</u>	1 11	2 class 2 2 class 2	-	N.E. S. N.E. S.	Kunze/M.Ochs, BMG December 2 Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
60762	Ticknor(1988)	yes yes(t	entative) yes (can be used)	class 1	A	3	B 2	A/B C/D	0	C/D(tentative) 0	A/B	3	A/B A/B	3	A	3	A	3	B 2	C/D 0 C/D 0	A/B	3	C/D	0	C/D	0	D 0	C	1 88	3 class 4	4 /	N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
60763 60764	Ticknor(1988) Ticknor(1988)		entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	B 2 B 2	C/D A/B	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	B 2 B 2	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	C C	1 88	2 class 4		N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
60765 60766	Ticknor(1988) Ticknor(1988)		entative) yes (can be used) entative) yes (can be used)	class 1 class 1	A	3	B 2	A/B C/D	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B	3	A	3	A	3	B 2	C/D 0	A/B A/B	3	C/D	0	C/D	0	D 0	C	1 11	2 class 3 class 4			Kunze/M.Ochs, BMG December 2 Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
60767	Ticknor(1988)	yes yes(t	entative) yes (can be used)	class 1	A	3	B 2 B 2	A/B	3	C/D(tentative) 0	A/B	3	A/B A/B	3	A	3	A	3	B 2	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	C C	1 11 11 11	2 class 3	3 /	N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
60768 Other Minerals (Illite-	Ticknor(1988)	yes yes(	entative) yes (can be used)	class 1	A	3	В 2	A/B	3	C/D(tentative) 0	A/B	3	A/B	3	A	3	A	3	8 2	C/D 0	A/B	3	C/D	0	C/D	0	0		1 11	2 class	3 1	N.E. <u>S</u>	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
67055	Ticknor(1988)	- ves(t	entative) yes (can be used)	class 1	В	2	В 3	A/B	3	C/D(tentative) 0	A/B	3	A/B	3	A	3	A	3	B 0	C/D 0	A/B	3	C/D	0	C/D	0	D 0	C	1 10	2 class 3	3 /	NE S	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
67056 67057	Ticknor(1988)	- yes(t	entative) yes (can be used)	class 1	В	2	B 3	A/B	3	C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A A	3	A	3	B 0 B 0	C/D 0	A/B A/B	3	C/D C/D	0	C/D	0	D 0	С	1 10	2 class 3	3 /	N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
67058	Ticknor(1988) Ticknor(1988)		entative) yes (can be used) entative) yes (can be used)	class 1 class 1	В	2	B 3	A/B A/B	3	C/D(tentative) 0 C/D(tentative) 0	A/B A/B	3	A/B A/B	3	A	3	A	3	B 0 B 0	C/D 0 C/D 0	A/B A/B	3	C/D C/D	0	C/D C/D	0	D 0	C	1 10	2 class 3 2 class 3		N.E. S.	Kunze/M.Ochs, BMG December 2 Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
67059	Ticknor(1988) Ticknor(1988)		entative) yes (can be used) entative) yes (can be used)	class 1 class 1	B	2	B 3	A/B C/D C/D	0	C/D(tentative) 0 C/D(tentative) 0	A/B	3	A/B	3	A	3	A	3	B 0	C/D 0	A/B	3	C/D	0	C/D	0	D 0	C	1 78	3 class 4 3 class 4		N.E. S.	Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
67061	Ticknor(1988)	- yes(t	entative) yes (can be used)	class 1	В	2	D   3	A/D	3	unreliable		-								0,0 0			0.5		0.0					unreliat	ble /	N.E. S.	Kunze/M.Ochs, BMG December 2 Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005)
67062 67063	Ticknor(1988) Ticknor(1988)		entative) yes (can be used) entative) yes (can be used)	class 1 class 1	B	2	B 3 B 3	A/B C/D	0	unreliable unreliable																				unreliat unreliat		N.E. S. N.E. S.	Kunze/M.Ochs, BMG December 2 Kunze/M.Ochs, BMG December 2	006 Revision 4b (May 19, 2005) 006 Revision 4b (May 19, 2005)
63027	Fujikawa and Fukui (1997b)	yes	no no (can not be used	I)									1	1																unreliat			Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005)
63028 63029	Fujikawa and Fukui (1997b) Fujikawa and Fukui (1997b)	yes yes	no no (can not be used no no (can not be used	l) l)							-																			unreliat unreliat			Y. Saito(JAEA) October 200 Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005) 7 Revision 4b (May 19, 2005)
63030	Fujikawa and Fukui (1997b)	ves	no no (can not be used	0																										unreliat	ble		Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005)
63032	Fujikawa and Fukui (1997b) Fujikawa and Fukui (1997b)	yes	no no (can not be used no no (can not be used	l)								-		1																unreliat unreliat	ble		Y. Saito(JAEA) October 200 Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005)
63033	Fujikawa and Fukui (1997b) Fujikawa and Fukui (1997b)	yes	no no (can not be used no no (can not be used	) D																										unreliat unreliat			Y. Saito(JAEA) October 200 Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005) 7 Revision 4b (May 19, 2005)
63035	Fujikawa and Fukui (1997b)	yes	no (can not be used	I)									1	1																unreliat	ble		Y. Saito(JAEA) October 200	17 Revision 4b (May 19, 2005)
63036 63037	Fujikawa and Fukui (1997b) Fujikawa and Fukui (1997b)	yes yes	no no (can not be used no no (can not be used	1) 1)						<u> </u>			+	+									+						+	unreliat unreliat			Y. Saito(JAEA) October 200 Y. Saito(JAEA) October 200	
63038	Fujikawa and Fukui (1997b)	yes	no (can not be used	)									-																	unreliat	ble		Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005)
63039 63040	Fujikawa and Fukui (1997b) Fujikawa and Fukui (1997b)	yes	no (can not be used	I)							_		+	1									+							unreliat unreliat	ble		Y. Saito(JAEA) October 200 Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005)
63041 63042	Fujikawa and Fukui (1997b) Fujikawa and Fukui (1997b)	yes	no no (can not be used no no (can not be used	)				_				_									_					_				unreliat unreliat	ble		Y. Saito(JAEA) October 200 Y. Saito(JAEA) October 200	17 Revision 4b (May 19, 2005)
63043	Fujikawa and Fukui (1997b)	yes	no no (can not be used	)								1	1	1																unreliat	ble		Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005)
63044 63045	Fujikawa and Fukui (1997b) Fujikawa and Fukui (1997b)	yes yes	no no (can not be used no no (can not be used	l)   l)						<u> </u>		-											+							unreliat unreliat			Y. Saito(JAEA) October 200 Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005) 7 Revision 4b (May 19, 2005)
63046	Fujikawa and Fukui (1997b)	yes	no no (can not be used	0									1																	unreliat			Y. Saito(JAEA) October 200	
56039	Legoux et al. (1992) Legoux et al. (1992)		yes yes (can be used) yes yes (can be used)	class 1	A	3	A 3	C/D	0	A/B 3 A/B 3 A/B 3 A/B 3 A/B 3	C/D	0	A/B	3	A	3	B		A 3		A/B		C/D	0	C/D	0	A 3	D	0 11	5 class	3		Y. Saito(JAEA) October 200 Y. Saito(JAEA) October 200	7 Revision 4b (May 19, 2005)
56041	Legoux et al. (1992)	yes	yes yes (can be used)	class 1	A	3	A 3	C/D C/D	0	A/B 3	C/D C/D	0	A/B	3	A	3	unreliable B		A 3 A 3		A/B A/B		C/D C/D	0	C/D	0	A 3	D	0 11	5 class 3	3		Y. Saito(JAEA) October 200	17 Revision 4b (May 19, 2005)
56042	Legoux et al. (1992)	yes	yes (can be used)	ciass 1	A	3	n 3	1 0/0	U	A/B 3	1 U/D	U	A/B	1 3	A	3	в	2	A   3		A/B	3	U/D	U	C/D	U	n   3	1 0	U   11	ວ   CIass ໂ	3		<ol> <li>JOCIODEF 200</li> </ol>	7 Revision 4b (May 19, 2005)

		eteness of documentation a Rating ♥ 1-a: yes/no, 1-b: clas	s 1-6						<u>т</u>		1			Checkpoin	its 🕊 level: A	A-D (numeric	cal value: 3-	0) / unreliable	reported data Rating I ♥	class 1-6 / un							1						II - Consistency		Date	Classification Gui
nt Reference	I-a.1	I-a.2 Rating I-a	Rating I	I-b II.a solic phas	d	II.b pH		II.c redox condition		II.d solution compositi		.e eratu e	II.f S/W		II.g sorption value		II.h inital [RN]		II.i phase separatio	rea	II.j action ime	ll.k agitat		II.I RN Ioading	re	II.m eaction essels		II.n error stimates	II.o parame variatio	er	Rating II	e class	comment/rating			
Barney and Anderson (1979) Barney and Anderson (1979)	yes	no no (can not be used	(b																													unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007	Revision 4b (May 1 Revision 4b (May 1
Barney and Anderson (1979) Barney and Anderson (1979)	yes yes	no no (can not be used	1) 1)											-										-							-	unreliable unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007	Revision 4b (May 1
Barney and Anderson (1979)		no no (can not be used no no (can not be used																														unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 1 Revision 4b (May 1
Barney and Brown (1979) Barney and Brown (1979) Barney and Brown (1979)	yes	no no (can not be used no no (can not be used no no (can not be used	(1																													unreliable unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007	Revision 4b (May 1 Revision 4b (May 1 Revision 4b (May 1
Daniels (1981) Daniels (1981)	yes	yes yes (can be used) yes yes (can be used)	class 1	1 C/D 1 C/D		A	3	A/B A/B	3	C/D C/D	0 A	/B 3 /B 3	C/D C/D	0	C/D C/D	0	A		unreliable													unreliable			August 2007	Revision 4b (May 1
Daniels (1981) Daniels (1981)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	1 C/D 1 C/D		C A	1	A/B A/B	3	C/D C/D	0 A 0 A	/B 3 /B 3	C/D C/D	0	C/D C/D	0	A	3	C/D unreliable		C/D	0 A/B	3	В	2	В	2	D	0 B	2	84	class 4 unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May Revision 4b (May
Daniels (1981) Daniels (1981) Daniels (1981)		yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	1 C/D	0 0	C C	1	A/B A/B	3	C/D C/D C/D		/B 3 /B 3 /B 3	C/D C/D C/D	0	C/D C/D C/D	0	A A A	3 3	C/D C/D C/D	0 0	C/D C/D C/D	0 A/B 0 A/B	3	B	2 2 2	B	2 2	D	0 B 0 B	2	84 84 84	class 4 class 4 class 4		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007	Revision 4b (May Revision 4b (May
Daniels (1981) Daniels (1981) Daniels (1981)		yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	1 C/D 1 C/D 1 C/D		C A C	1 3	A/B A/B A/B	3 3 3	C/D C/D C/D	0 A 0 A	/B 3 /B 3 /B 3	C/D C/D C/D	0	C/D C/D C/D	0	A A A		C/D unreliable C/D		C/D C/D	0 A/B		B	2	B	2	D	0 B	2	84	class 4 unreliable class 4		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007 August 2007	Revision 4b (May Revision 4b (May Revision 4b (May
Daniels (1981) Daniels (1981)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	1 C/D 1 C/D 1 C/D		A	3	A/B A/B A/B	3	C/D C/D C/D	0 A	/B 3 /B 3 /B 3	C/D C/D	0	C/D C/D	0	A A A		unreliable unreliable			0 A/B			-					-		unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (Ma Revision 4b (Ma
Daniels (1981) Daniels (1981) Daniels (1981)	yes yes yes	yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1	1 C/D 1 C/D	0 0	C C	1	A/B A/B A/B	3	C/D C/D	0 A 0 A	/B 3 /B 3	C/D C/D C/D	0	C/D C/D C/D	0	A	3 3 3	C/D C/D C/D	0 0	C/D C/D C/D	0 A/B 0 A/B	3	B	2 2 2	B	2 2 2	D D	0 B 0 B	2 2	84 84 84	class 4 class 4 class 4		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (Ma Revision 4b (Ma
Daniels (1981) Duursma (1973)	yes yes	yes yes (can be used)	class 1	1 C/D	0	С	1	A/B	3	C/D	0 A	/B 3	C/D	0	C/D	0	A	3	C/D	0 0	C/D	0 A/B	3	В	2	В	2	D	0 B	2	84	unreliable		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (Ma
Duursma (1973) Duursma (1973) Duursma (1973)	yes yes	no no (can not be used no no (can not be used no no (can not be used	1) 1)	_	_																											unreliable unreliable unreliable		Y. Saito(JAEA)	August 2007 August 2007 August 2007	Revision 4b (Ma
Duursma (1973) Duursma (1973)	yes yes	no no (can not be used no no (can not be used	(t (t																													unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (Ma Revision 4b (Ma
Duursma (1973) Duursma (1973) Duursma (1973)	yes yes yes	no no (can not be used no no (can not be used no no (can not be used	(1	_		_																										unreliable unreliable unreliable		Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (Ma Revision 4b (Ma
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Duursma (1973) Duursma (1973)	yes	no no (can not be used no no (can not be used	1) 1)			_	_										<u> </u>						_			-						unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007 August 2007	Revision 4b (Ma Revision 4b (Ma
Duursma (1973) Duursma (1973) Duursma (1973)	yes yes yes	no no (can not be used no no (can not be used no no (can not be used	4) (t						++																							unreliable unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007 August 2007	Revision 4b (M Revision 4b (M Revision 4b (M
Duursma (1973) Duursma (1973) Duursma (1973)	yes yes	no no (can not be used no no (can not be used no no (can not be used	1) 1)																													unreliable unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007	Revision 4b (M
Duursma (1973) Duursma (1973)	yes yes yes	no no (can not be used no no (can not be used	4) (t																													unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (M Revision 4b (M
Duursma (1973) Duursma (1973) Duursma (1973)	yes yes yes	no no (can not be used no no (can not be used no no (can not be used	d) (E																													unreliable unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (M Revision 4b (M Revision 4b (M
Duursma (1973) Duursma (1973)	yes yes	no no (can not be used no no (can not be used	(t (t																													unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (M Revision 4b (M
Duursma (1973) Duursma (1973) Duursma (1973)	yes	no no (can not be used no no (can not be used no no (can not be used	4)																													unreliable unreliable unreliable		Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (M
Duursma (1973) Duursma (1973)	yes yes	no no (can not be used no no (can not be used	1) 1)		_		_																									unreliable unreliable		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007	
Erdal et al.(1977) Erdal et al.(1977)		yes yes (can be used) yes yes (can be used)	class 1 class 1	1 B 1 B	2	A	3	A/B A/B	3	C/D C/D		/B 3 /B 3	A/B A/B	3	C/D C/D	0	A	3	B		A/B A/B	3 C/D 3 C/D		C/D C/D	0	B	2	D D	0 D 0 D	0	109 109	class 3 class 3		Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (M Revision 4b (M
Erdal et al.(1977) Erdal et al.(1977) Erdal et al.(1977)	yes	yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	1 B 1 B 1 B	2	A	3	A/B A/B A/B	3	C/D C/D C/D	0 A 0 A	/B 3 /B 3 /B 3	A/B A/B A/B	3	C/D C/D C/D	0	A A A	3	B	2 /	A/B A/B A/B	3 C/D 3 C/D 3 C/D	0	C/D C/D C/D	0	B	2 2	D	0 D 0 D	0	109 109 109	class 3 class 3 class 3		Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (M Revision 4b (M
Erdal et al.(1977) Erdal et al.(1977) Erdal et al.(1977)	yes yes	yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	1 B	2	A	3	A/B A/B	3	C/D C/D		/B 3 /B 3	A/B A/B A/B	3	C/D C/D C/D	0	A A A	3	B	2 /	A/B A/B A/B	3 C/D 3 C/D 3 C/D	0	C/D C/D	0	B	2	D	0 D 0 D	0	109 109 109	class 3 class 3 class 3		Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (M Revision 4b (M
Erdal et al.(1977) Erdal et al.(1977)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	1 B	2	A	3	A/B A/B	3	C/D C/D	0 A		A/B A/B	3	C/D C/D	0	A	3	B	2 /	A/B A/B	3 C/D 3 C/D	0	C/D C/D	0	B	2 2	D	0 D 0 D	0	109 109	class 3 class 3		Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (N Revision 4b (N
Erdal et al.(1977) Erdal et al.(1977) Erdal et al.(1977)	yes	yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	1 B	2	A	3	A/B A/B A/B	3	C/D C/D C/D		/B 3 /B 3 /B 3	A/B A/B A/B	3	C/D C/D C/D	0	A A A	3 3 3	B B	2 /	A/B A/B A/B	3 C/D 3 C/D 3 C/D		C/D C/D C/D	0	B	2	D D	0 D 0 D	0	109 109 109	class 3 class 3 class 3		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007	Revision 4b (M Revision 4b (M Revision 4b (M
Erdal et al.(1977) Erdal et al.(1977) Erdal et al.(1977)	yes	yes yes (can be used) yes yes (can be used)	class 1	1 B	2	A		A/B A/B	3			/B 3 /B 3				0	A		B	2 /		3 C/D		C/D C/D		B	2		0 D 0 D					Y. Saito(JAEA) Y. Saito(JAEA)	August 2007	Revision 4b (N
Erdal et al.(1979b) Erdal et al.(1979b)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1		2	A	3	A/B A/B	3	C/D C/D		/B 3	A/B A/B	3	C/D C/D	0	A	3	B		C/D C/D	0 A/B 0 A/B	3	C/D C/D	0	A	3	D D	0 C 0 C	1	115 115			Y. Saito(JAEA) Y. Saito(JAEA)	November 200	07 Revision 4b (N
Erdal et al.(1979b) Erdal et al.(1979b) Erdal et al.(1979b)	yes	yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1	1 B	2	A	3	A/B A/B A/B	3	C/D C/D C/D	0 A	/B 3 /B 3 /B 3	A/B A/B A/B	3	B C/D C/D	2	A A A	3	BB	2 0	C/D C/D C/D	0 A/B 0 A/B	3	C/D C/D C/D	0	A A A	3	D	0 C 0 C 0 C	1	115	class 3 class 3		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	November 200	07 Revision 4b (N
Erdal et al.(1979b) Erdal et al.(1979b)	yes yes	yes (can be used) yes (can be used)	class 1 class 1	1 B 1 B	2	A	3	A/B A/B	3	C/D C/D	0 A 0 A	/B 3 /B 3	A/B A/B	3	C/D C/D	0	A	3	B	2 0	C/D C/D	0 A/B 0 A/B	3	C/D C/D	0	A	3	D	0 C 0 C	1	115 115	class 3 class 3		Y. Saito(JAEA) Y. Saito(JAEA)	November 200 November 200	07 Revision 4b (N 07 Revision 4b (N
Erdal et al.(1979b) Erdal et al.(1979b) Erdal et al.(1979b)		yes yes (can be used) yes yes (can be used) yes yes (can be used)		1 B	2	A A A	3	A/B A/B A/B		C/D C/D C/D	0 A	/B 3 /B 3 /B 3	A/B A/B A/B	3	C/D C/D C/D		A A A	3 3 3	BB	2 (	C/D C/D C/D	0 A/B 0 A/B 0 A/B		C/D C/D C/D		A A A	3 3	D D D	0 C 0 C			class 3 class 3 class 3		Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	November 2007	07 Revision 4b (M 07 Revision 4b (M 07 Revision 4b (M
Erdal et al. (1979b) Erdal et al. (1979b) Erdal et al. (1979b)	yes yes	yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1		2	A	3	A/B A/B A/B	3	C/D C/D C/D		/B 3 /B 3	A/B A/B A/B	3	C/D C/D C/D	0	A	3	B	2 0	C/D C/D C/D	0 A/B	3	C/D C/D C/D	0	A	3	D	0 C 0 C	1	115 115 115	class 3 class 3 class 3		Y. Saito(JAEA) Y. Saito(JAEA)	November 200 November 200	07 Revision 4b (N
Erdal et al.(1979b) Erdal et al.(1979b)	yes yes	yes yes (can be used) yes yes (can be used)	class 1 class 1	1 B 1 B	2	A	3	A/B A/B		C/D C/D	0 A 0 A	/B 3 /B 3	A/B A/B		C/D C/D	0	A	3	BB	2 0	C/D C/D	0 A/B 0 A/B	3	C/D C/D		A A A	3		0 C 0 C	1	115 115	class 3 class 3		Y. Saito(JAEA) Y. Saito(JAEA)	November 200 November 200	07 Revision 4b (M 07 Revision 4b (M
Erdal et al.(1979b) Erdal et al.(1979b) Erdal et al.(1979b)	yes	yes yes (can be used) yes yes (can be used) yes yes (can be used)	class 1 class 1 class 1	1 B	2	A A A	3	A/B A/B A/B	3	C/D C/D C/D	0 A	/B 3 /B 3 /B 3	A/B A/B A/B	3	C/D C/D C/D	0	A A A	3 3 3	B B B	2 (	C/D C/D C/D	0 A/B 0 A/B 0 A/B	3	C/D C/D C/D	0	A	3	D D D	0 C 0 C	1	115 115 115	class 3 class 3 class 3		Y, Saito(JAEA)	November 200	07 Revision 4b (M 07 Revision 4b (M 07 Revision 4b (M
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				unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007 August 2007 August 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
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				unreliable unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007 August 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
				unreliable unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007 August 2007	Revision 4b (May 19, 2005) Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)

	0.0700																						
	62724	Kamel and Ibrahim(1998)								3	A 3 A 3	A	3	C/D 0 C/	0 A/B	3 B 3 B	2 B 2 B	2	D 0 D 0	B	2		
			yes yes	yes (can be used)			C/D 0				A 3	A	3			3 B	2 B	2	D 0	B	2		
	62727	Kamel and Ibrahim(1998)	yes yes	yes (can be used)	class 1 C/D 0	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3		A	3	C/D 0 C/I	0 0 A/B	3 B	2 B	2	D 0	В	2	104	class 3
										3	A 3 A 3		3			3 B 3 B	2 B 2 B	2	D 0 D 0	B	2		
		Kamel and Ibrahim(1998) Kamel and Ibrahim(1998)	yes yes	yes (can be used)			C/D 0			3	A 3 A 3	A	3		D 0 A/B 0 A/B	3 B	2 B 2 B	2	D 0	B	2	104	
	62732	Kamel and Ibrahim(1998)	yes yes	yes (can be used)			C/D 0			3	A 3		3	C/D 0 C/I	0 0 A/B	3 B	2 B	2	D 0	B	2	112	
	62734	Kamel and Ibrahim(1998)	yes yes	yes (can be used)	class 1 C/D 0	A 3 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A	3	C/D 0 C/	0 0 A/B	3 B	2 B	2	D 0	B	2	112	class 3
		Kamel and Ibrahim(1998) Kamel and Ibrahim(1998)	yes yes yes yes	yes (can be used) yes (can be used)	class 1 C/D 0 class 1 C/D 0	A 3 A/B 3 A 3 A/B 3	C/D 0 C/D 0	A/B 3 A/B 3		3	A 3 A 3	A	3	C/D 0 C/I C/D 0 C/I		3 B 3 B	2 B 2 B	2	D 0 D 0	B	2	112	
	62772	Strezov et al.(2000)			class 1 C/D 0	C 1 A/B 3	A/B 3	A/B 3	3 A/B	3	A 3	A	3	C/D 0 unreli	able								unreliable
	62773	Strezov et al.(2000)	yes yes	yes (can be used)	class 1 C/D 0	C 1 A/B 3	A/B 3	A/B 3	3 A/B		A 3		3	C/D 0 unreli	able						-		unreliable
	62775	Strezov et al.(2000)	yes yes	yes (can be used)	class 1 C/D 0	C 1 A/B 3	A/B 3	A/B 3	3 A/B	3	A 3		3	C/D 0 A/	3 3 A/B	3 B	2 B	2	D 0	В	2		class 2
	62777	Strezov et al.(2000)			class 1 C/D 0 class 1 C/D 0					3	A 3 A 3	A	3			3 B 3 B	2 B 2 B	2	D 0 D 0	B	2	126	
	62778 62779	Strezov et al.(2000) Strezov et al.(2000)	yes yes yes yes	yes (can be used) yes (can be used)			A/B 3 A/B 3			3	A 3 A 3		3			3 B 3 B	2 B 2 B	2	D 0 D 0	B	2		
		Strezov et al.(2000)	yes yes	yes (can be used)	class 1 C/D 0		A/B 3			3			3			3 B	2 B	2	D 0	B	2	142	class 2
	62782	Strezov et al.(2000)	yes yes	yes (can be used)	class 1 C/D 0	A 3 A/B 3	A/B 3	A/B 3	3 A/B	3	A 3	A	3	C/D 0 A/	3 3 A/B	3 B	2 B	2	D 0	B	2	142	class 2
1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1         1	62784	Strezov et al.(2000)	yes yes	yes (can be used)	class 1 C/D 0	A 3 A/B 3	A/B 3	A/B 3	3 A/B		A 3 A 3	A	3	C/D 0 A/	3 3 A/B	3 B 3 B	2 B 2 B	2	D 0	B	2	142	class 2
			yes yes	yes (can be used) yes (can be used)						3			3			3 B 3 B	2 B 2 B	2	D 0 D 0	B	2		
Image         Image <th< th=""><th>62892</th><th></th><th></th><th></th><th>class 5 C/D 0</th><th></th><th></th><th></th><th>3 C/D</th><th>0</th><th>A 3</th><th>A</th><th>3</th><th></th><th></th><th>3 C/D</th><th>0 B</th><th>2</th><th>D 0</th><th>D</th><th>0</th><th>108</th><th>class 3</th></th<>	62892				class 5 C/D 0				3 C/D	0	A 3	A	3			3 C/D	0 B	2	D 0	D	0	108	class 3
											A 3	A	3				0 B	2	D 0	D	0		
	63346	Inoue and Morisawa(1975)	yes no	no (can not be used)				1				1											unreliable
	Bentonite																						
		Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0		3 A/B	3	B 2	A	3		3 3 C/D	0 A	3 B	2	B 2	A	3		class 2
	65510	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3		C/D 0	A/B 3			A 3		3	B 2 A/	3 3 C/D	0 A 0	3 B 3 B	2	B 2 B 2		3	143	class 2
	65512	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B		A 3	A	3	B 2 A/	3 3 C/D	0 A	3 B	2	B 2	A	3	143	class 2
	65514	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	B 2	A	3	B 2 A/	3 3 C/D	0 A 0	з в 3 В	2	B 2	A	3	141	class 2
	65516	Silva et al. (1979)				B 2 A/B 3	C/D 0	A/B 3	3 A/B		A 3 A 3		3	B 2 A/	3 3 C/D	0 A 0 A	3 B 3 B	2	B 2 B 2		3	143	class 2
			yes yes	yes (can be used)						3	A 3 B 2	A	3			0 A 0 A	3 B 3 B	2	B 2 B 2	A	3		
	65519	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	B 2 A 3		3	B 2 A/	3 3 C/D	0 A 0	3 B 3 P	2	B 2 B 2		3	141	class 2
	65521	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A	3	B 2 A/	3 3 C/D	0 A	3 B	2	- 2 B 2	A	3	143	class 2
	65523	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3 B 2		3	B 2 A/	3 3 C/D	0 A 0	3 B	2	B 2		3	141	class 2
	65525	Silva et al. (1979)	yes yes yes yes	yes (can be used) yes (can be used)		B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	B 2 A 3	A	3	B 2 A/	3 3 C/D	0 A 0 A	3 B 3 B	2	B 2 B 2	A	3	143	
			yes yes	yes (can be used)		B 2 A/B 3	C/D 0	A/B 3		3	A 3 A 3	A	3		3 3 C/D	0 A 0 A	3 B 3 B	2	B 2 B 2	A	3		
	65528	Silva et al. (1979) Silva et al. (1979)	yes yes	yes (can be used)						3	B 2	A	3			0 A	3 B	2	B 2	A	3		
	65530	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A	3	B 2 A/	3 C/D	0 A	3 B	2	B 2	A	3	143	class 2
						B 2 A/B 3	C/D 0			3	A 3 A 3	A	3	B 2 A/	3 3 C/D	0 A 0 A	3 B 3 B	2	B 2 B 2	A	3		
No. 0         No. 0 <th< th=""><th></th><th></th><th>yes yes</th><th>yes (can be used)</th><th></th><th></th><th>C/D 0</th><th></th><th></th><th>3</th><th>B 2 B 2</th><th>A</th><th>3</th><th></th><th></th><th>0 A 0 A</th><th>3 B 3 B</th><th>2</th><th>B 2 B 2</th><th></th><th>3</th><th></th><th></th></th<>			yes yes	yes (can be used)			C/D 0			3	B 2 B 2	A	3			0 A 0 A	3 B 3 B	2	B 2 B 2		3		
	65535	Silva et al. (1979)	yes yes	yes (can be used)		B 2 A/B 3	C/D 0	A/B 3		3	A 3	A	3	B 2 A/	3 3 C/D	0 A	3 B	2	B 2 B 2		3	143	
Image         Image        Image	65537	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A	3	B 2 A/	3 3 C/D	0 A	3 B	2	B 2	A	3	143	class 2
Image         Image <th< th=""><th>65539</th><th>Silva et al. (1979)</th><th>yes yes</th><th>yes (can be used)</th><th>class 1 A 3</th><th>B 2 A/B 3</th><th>C/D 0</th><th>A/B 3</th><th></th><th>3</th><th>B 2</th><th></th><th>3</th><th></th><th>3 3 C/D</th><th>0 A</th><th>3 B</th><th>2</th><th>B 2</th><th>A</th><th>3</th><th>141</th><th>class 2</th></th<>	65539	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3		3	B 2		3		3 3 C/D	0 A	3 B	2	B 2	A	3	141	class 2
Desc         Desc        Desc        Desc        Desc        Des									3 A/B 3 A/B	3	A 3 A 3	A	3	B 2 A/ B 2 A/		0 A 0 A	3 B 3 B	2	B 2 B 2	A	3		
			yes yes	yes (can be used)						3	A 3 B 2		3			0 A 0 A	3 B 3 B	2	B 2 B 2	A	3		
Desc         Desc        Desc        Desc        Desc        Desc        Desc        Desc        Desc        Desc        Desc <th></th> <th>Silva et al. (1979)</th> <th>yes yes</th> <th>yes (can be used)</th> <th></th> <th></th> <th>C/D 0</th> <th>A/B 3</th> <th></th> <th>3</th> <th>B 2</th> <th>A</th> <th>3</th> <th></th> <th>3 3 C/D</th> <th>0 A</th> <th>3 B</th> <th>2</th> <th>B 2 B 2</th> <th>A</th> <th>3</th> <th></th> <th>class 2</th>		Silva et al. (1979)	yes yes	yes (can be used)			C/D 0	A/B 3		3	B 2	A	3		3 3 C/D	0 A	3 B	2	B 2 B 2	A	3		class 2
Image         Image        Image         Image        I	65546	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3		3	B 2 A/	3 C/D	0 A	3 B	2	B 2	A	3	143	class 2
No. 1         No. 1 <th< th=""><th>65548</th><th>Silva et al. (1979)</th><th>yes yes</th><th>yes (can be used)</th><th>class 1 A 3</th><th>B 2 A/B 3</th><th>C/D 0</th><th>A/B 3</th><th>3 A/B</th><th>3</th><th></th><th>A</th><th>3</th><th>B 2 A/</th><th>3 3 C/D</th><th>0 A 0</th><th>3 B 3 B</th><th>2</th><th>B 2 B 2</th><th>A</th><th>3</th><th>137</th><th>class 2</th></th<>	65548	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3		A	3	B 2 A/	3 3 C/D	0 A 0	3 B 3 B	2	B 2 B 2	A	3	137	class 2
Image	65550	Silva et al. (1979)	yes yes	yes (can be used)		B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	B 2 A 3	A	3	B 2 A/	3 3 C/D	0 A 0 A	3 B 3 B	2	B 2 B 2	A	3	143	class 2
Image         Mark         Mark        Mark        Mark										3	A 3 A 3	A	3			0 A 0 A	3 B 3 B	2	B 2 B 2	A	3		
No.es         No.es <th< th=""><th></th><th></th><th>yes yes</th><th>yes (can be used)</th><th></th><th></th><th></th><th></th><th></th><th>3</th><th>B 2 B 2</th><th>A</th><th>3</th><th></th><th></th><th>0 A 0 A</th><th>3 B</th><th>2</th><th>B 2 B 2</th><th>A</th><th>3</th><th></th><th></th></th<>			yes yes	yes (can be used)						3	B 2 B 2	A	3			0 A 0 A	3 B	2	B 2 B 2	A	3		
No. 4         No. 5         No. 5 <th< th=""><th>65555</th><th>Silva et al. (1979)</th><th>yes yes</th><th>yes (can be used)</th><th>class 1 A 3</th><th>B 2 A/B 3</th><th>C/D 0</th><th>A/B 3</th><th>3 A/B</th><th>3</th><th>A 3</th><th></th><th>3</th><th>B 2 A/</th><th>3 3 C/D</th><th>0 A</th><th>3 B</th><th>2</th><th>B 2</th><th>A</th><th>3</th><th>143</th><th>class 2</th></th<>	65555	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3		3	B 2 A/	3 3 C/D	0 A	3 B	2	B 2	A	3	143	class 2
No.e         No.e        No.e        No.e         N	65557	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B		A 3	A	3	B 2 A/	3 C/D	0 A	3 B	2	B 2		3	143	class 2
Nord         Nord        Nord        Nord        No	65559	Silva et al. (1979)	yes yes	yes (can be used)		B 2 A/B 3	C/D 0	A/B 3	3 A/B	3			3	B 2 A/	3 3 C/D	0 A 0 A	3 B 3 B	2	B 2 B 2	A	3	141	class 2
Disc         Dis         Disc        Disc         D			yes yes	yes (can be used)									3				3 B 3 B	2	B 2 B 2		3		
Bird 4         Bird 4<		Silva et al. (1979) Silva et al. (1979)	yes yes	yes (can be used)			C/D 0 C/D 0			3	A 3 B 2	A	3			0 A 0 A	3 B 3 B	2	B 2 B 2	A	3		
brace         brace<	65564	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	B 2	A	3	B 2 A/	3 3 C/D	0 A	3 B 3 P	2	B 2 B 2	A	3	141	class 2
Booke (19)         Ma         Ma        Ma        Ma         <	65566	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A	3	B 2 A/	3 3 C/D	0 A	3 B	2	B 2 B 2	A	3	143	class 2
Dial         Dial <th< th=""><th>65568</th><th>Silva et al. (1979)</th><th>yes yes</th><th>yes (can be used)</th><th>class 1 A 3</th><th>B 2 A/B 3</th><th>C/D 0</th><th>A/B 3</th><th>3 A/B</th><th>3</th><th>A 3 B 2</th><th>A</th><th>3</th><th>B 2 A/</th><th>3 3 C/D</th><th>0 A</th><th>з в 3 В</th><th>2</th><th>B 2</th><th>A</th><th>3</th><th>141</th><th>class 2</th></th<>	65568	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3 B 2	A	3	B 2 A/	3 3 C/D	0 A	з в 3 В	2	B 2	A	3	141	class 2
bbs         bbs<	65570	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	в 2 А 3	A	3	B 2 A/	3 3 C/D	0 A	3 B	2	B 2	A	3	143	class 2
Base dt (17)         Ym         Ym        Ym         Ym        Ym       <	65572	Silva et al. (1979)	yes yes yes yes	yes (can be used) yes (can be used)		B 2 A/B 3	C/D 0	A/B 3	3 A/B					B 2 A/	3 3 C/D		3 B 3 B	2	B 2 B 2	A	3	143	class 2
Bised a (17)         Yet         Prec (not books)         Bised a (17)        Yet         Prec (not books) </th <th>65573</th> <th>Silva et al. (1979)</th> <th>yes yes</th> <th>yes (can be used)</th> <th>class 1 A 3</th> <th>B 2 A/B 3</th> <th>C/D 0</th> <th>A/B 3</th> <th>3 A/B</th> <th>3</th> <th>B 2 B 2</th> <th>A</th> <th>3</th> <th>B 2 A/</th> <th>3 3 C/D</th> <th></th> <th>3 B 3 B</th> <th>2 2</th> <th>B 2 B 2</th> <th>A</th> <th>3</th> <th>141</th> <th>class 2</th>	65573	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	B 2 B 2	A	3	B 2 A/	3 3 C/D		3 B 3 B	2 2	B 2 B 2	A	3	141	class 2
Bold of (1)         ym         ym      <	65575	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A	3	B 2 A/	3 3 C/D	0 A	3 B 3 P	2	B 2 B 2	A	3	143	class 2
black         black         clask         clask <th< th=""><th>65577</th><th>Silva et al. (1979)</th><th>yes yes</th><th>yes (can be used)</th><th>class 1 A 3</th><th>B 2 A/B 3</th><th>C/D 0</th><th>A/B 3</th><th>3 A/B</th><th></th><th></th><th>A</th><th>3</th><th>B 2 A/</th><th>3 3 C/D</th><th></th><th>3 B</th><th>2</th><th>- 2 B 2</th><th>A</th><th>3</th><th>143</th><th>class 2</th></th<>	65577	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B			A	3	B 2 A/	3 3 C/D		3 B	2	- 2 B 2	A	3	143	class 2
bbs         bbs <th>65579</th> <th>Silva et al. (1979)</th> <th>yes yes</th> <th>yes (can be used)</th> <th>class 1 A 3</th> <th>B 2 A/B 3</th> <th>C/D 0</th> <th>A/B 3</th> <th>3 A/B</th> <th></th> <th></th> <th>A</th> <th>3</th> <th>B 2 A/</th> <th>3 C/D</th> <th>0 A 0</th> <th>з в 3 В</th> <th>2</th> <th>B 2</th> <th>A</th> <th>3</th> <th>141</th> <th>class 2</th>	65579	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B			A	3	B 2 A/	3 C/D	0 A 0	з в 3 В	2	B 2	A	3	141	class 2
bested         wite         yee	65581	Silva et al. (1979)	yes yes yes	yes (can be used) yes (can be used)		B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A	3	B 2 A/	3 3 C/D		з В 3 В	2	в 2 В 2		3	143	class 2
best         No         No        No        No        No </th <th>65582</th> <th>Silva et al. (1979)</th> <th>yes yes yes yes</th> <th>yes (can be used) yes (can be used)</th> <th></th> <th>B 2 A/B 3</th> <th>C/D 0 C/D 0</th> <th></th> <th></th> <th></th> <th></th> <th></th> <th></th> <th>B 2 A/</th> <th>3 C/D</th> <th></th> <th>3 B 3 B</th> <th>2</th> <th>B 2 B 2</th> <th></th> <th>3</th> <th>143</th> <th></th>	65582	Silva et al. (1979)	yes yes yes yes	yes (can be used) yes (can be used)		B 2 A/B 3	C/D 0 C/D 0							B 2 A/	3 C/D		3 B 3 B	2	B 2 B 2		3	143	
bits         yes         yes <th>65584</th> <th>Silva et al. (1979)</th> <th>yes yes</th> <th>yes (can be used)</th> <th>class 1 A 3</th> <th></th> <th>C/D 0</th> <th></th> <th></th> <th>3</th> <th>A 3</th> <th></th> <th>3</th> <th>B 2 A/</th> <th>3 3 C/D</th> <th>0 A</th> <th>3 B 3 R</th> <th>2</th> <th>B 2 B 2</th> <th>A</th> <th>3</th> <th>143</th> <th>class 2</th>	65584	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3		C/D 0			3	A 3		3	B 2 A/	3 3 C/D	0 A	3 B 3 R	2	B 2 B 2	A	3	143	class 2
bbs         bbs         yes         yes <th>65586</th> <th>Silva et al. (1979)</th> <th>yes yes</th> <th>yes (can be used)</th> <th>class 1 A 3</th> <th>B 2 A/B 3</th> <th>C/D 0</th> <th>A/B 3</th> <th>3 A/B</th> <th></th> <th></th> <th></th> <th>3</th> <th>B 2 A/</th> <th>3 3 C/D</th> <th></th> <th>3 B</th> <th>2</th> <th>- 2 B 2</th> <th></th> <th>3</th> <th>143</th> <th>class 2</th>	65586	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B				3	B 2 A/	3 3 C/D		3 B	2	- 2 B 2		3	143	class 2
bits at at (1979)         yes	65588	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B				3	B 2 A/	3 3 C/D		з в 3 В	2	B 2		3	141	class 2
bits         bits <th>65590</th> <th>Silva et al. (1979)</th> <th>yes yes</th> <th>yes (can be used)</th> <th>class 1 A 3</th> <th>B 2 A/B 3</th> <th>C/D 0</th> <th>A/B 3</th> <th>3 A/B</th> <th></th> <th>A 3</th> <th>A</th> <th>3</th> <th>B 2 A/</th> <th>3 3 C/D</th> <th></th> <th>3 B 3 B</th> <th>2 2</th> <th>в <u>2</u> В <u>2</u></th> <th>A</th> <th>3</th> <th>143</th> <th>class 2</th>	65590	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B		A 3	A	3	B 2 A/	3 3 C/D		3 B 3 B	2 2	в <u>2</u> В <u>2</u>	A	3	143	class 2
Bits add. (1979)         yes		Silva et al. (1979) Silva et al. (1979)	yes yes yes yes	yes (can be used) yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3			3	B 2 A/	3 3 C/D		3 B 3 B	2	B 2 B 2		3	143	class 2
Silva etal. (1979)       yes       yes       yes (an bu sed)       diast       A       3       B       2       AB       3       CD       0       AB       3       B       2       AB       3       CD       0       A       3       B       2       A       3       B       2       AB       3       CD       0       A       3       B       2       A       3       B       2       AB       3       CD       0       A       3       B       2       A       3       B       2       AB       3       AB       3 <th>65593</th> <th>Silva et al. (1979)</th> <th>yes yes</th> <th>yes (can be used)</th> <th>class 1 A 3</th> <th>B 2 A/B 3</th> <th>C/D 0</th> <th>A/B 3</th> <th>3 A/B</th> <th>3</th> <th>B 2</th> <th>A</th> <th></th> <th>B 2 A/</th> <th>3 3 C/D</th> <th>0 A</th> <th>3 B 3 P</th> <th>2</th> <th>B 2 B 2</th> <th>A</th> <th>3</th> <th>141</th> <th>class 2</th>	65593	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	B 2	A		B 2 A/	3 3 C/D	0 A	3 B 3 P	2	B 2 B 2	A	3	141	class 2
65507         Silva etal. (1979)         yes	65595	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A	3	B 2 A/	3 3 C/D	0 A	3 B	2	- 2 B 2	A	3	143	class 2
Silva etal. (1979)         yes         yes         yes (a bu soue)         class1         A         3         B         2         AB         3         CD         AB         3         AB         3         CD         AB         3         B         2         AB         3         AB         3         CD         AB         3         B         2         AB         3         CD         0         A         3         B         2         AB         3         CD         3         AB         3         CD         0         A         3         B         2         AB         3         CD         3         AB         3         CD         3         AB         3         CD         3         AB         3         AB<	65597	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3		A	3	B 2 A/	3 3 C/D	0 A	з в 3 В	2	B 2	A	3	143	class 2
66600         Silva etal. (1979)         yes         yes         yes (an b used)         class 1         A         3         B         2         AB         3         CD         0         AB         3         CD         0         AB         3         CD         0         AB         3         CD         0         A         3         B         2         AB         3         CD         0         A         3         B         2         AB         3         CD         0         AB         3         CD         0         A         3         B         2         B         2         A         3         CD         0         AB         3         CD         0         A         3         B         2         A         3         CD         0         AB         3         CD         0         A         3         B         2         AB         3         CD         0         AB         3         AB         3         AB         3         AB         3         CD         0         A         3         B         2         AB         3         CD         0         AB         3         AB         3         AB	65599	Silva et al. (1979)	yes yes yes yes	yes (can be used) yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	B 2 B 2	A	3	B 2 A/	3 3 C/D		3 B 3 B	2	B 2 B 2		3	141	class 2
66602       Silva al. (1979)       yes       yes (an bused)       class 1       A       3       B       2       AB       3       C/D       0       AB       3       AB       3       AB       3       AB       3       AB       3       AB       3       B       2       AB       3       C/D       0       AB       3       AB       3       AB       3       B       2       AB       3       C/D       0       AB       3       AB       3       B       2       AB       3       C/D       0       AB       3       AB       3       AB       3       AB      3       AB       3       AB       3       C/D       0       AB       3       A		Silva et al. (1979) Silva et al. (1979)	yes yes	yes (can be used)			C/D 0 C/D 0				A 3 A 3		3		3 3 C/D 3 3 C/D		3 B 3 B	2	B 2 B 2		3	143	
66604       Silva etal. (1979)       yes       yes       yes (anb bused)       class 1       A       3       B       2       AB       3       ADB       3       AB       3       ADB       3       B       2       AB       3       CD       0       AB       3       B       2       AB       3       B       2       AB       3       B       2       AB       3       CD       0       AB       3       B       2       AB       3       CD       0       AB       3       CD       0       AB       3       B       2       AB       3       CD       0       AB       3       AB       3       AB       3       AB       3       AB       3       AB       3       CD       0       AB       3       CD       0       AB       3	65602	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B		A 3 B 2	A	3	B 2 A/	3 3 C/D		3 B	2	B 2 B 2	A	3	143	class 2
66606         Silva et al. (1979)         yes         yes         yes (an be used)         class 1         A         3         B         2         AB         3         CD         0         AB         3         B         2         AB         3         CD         0         AB         3         AB         3         AB         3         B         2         AB         3         B         2         AB         3         CD         0         AB         3         CD         0         AB         3         B         2         AB         3         B         2         AB         3         CD         0         AB         3         AB         3         B         2         AB         3         CD         0         AB         3         B         2         AB         3         CD         0         AB         3         B         2         AB         3         CD         0         AB         3         CD         0	65604	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B		B 2	A	3	B 2 A/	3 3 C/D		3 B	2	- 2 B 2	A		141	class 2
65609         Silva et al. (1979)         yes	65606	Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3	B 2 A/B 3	C/D 0	A/B 3	3 A/B	3	A 3	A		B 2 A/	3 3 C/D	0 A	з В 3 В	2	B 2	A	3	143	class 2
65609         Silva et al. (1979)         yes	65608	Silva et al. (1979)	yes yes yes yes	yes (can be used) yes (can be used)	class 1 A 3	B 2 A/B 3		A/B 3	3 A/B	3		A	3	B 2 A/	3 3 C/D		3 B 3 B	2	B 2 B 2	A	3	141	class 2
		Silva et al. (1979) Silva et al. (1979)	yes yes	yes (can be used)	class 1 A 3		C/D 0			3			3	B 2 A/	3 3 C/D 3 3 C/D		3 B 3 B	2 2	B 2 B 2		3	143 143	class 2 class 2

2	104	class 3	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
2	104	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	104 104	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	104	class 3	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
2	104	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	112	class 3	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
2	112 112	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	112	class 3 class 3	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	112	class 3	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
		unreliable	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
		unreliable	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
2	126	unreliable class 2	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	126	class 2	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
2	142	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	142	class 2	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
2	142	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	142	class 2	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
2	142	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
2	142 142	class 2 class 2	Y. Saito(JAEA)	August 2007 August 2007	Revision 4b (May 19, 2005
			Y. Saito(JAEA)		Revision 4b (May 19, 2005
0	108 108	class 3 class 3	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
U	108	class 3	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
		unreliable	Y. Saito(JAEA)	August 2007	Revision 4b (May 19, 2005
3	141	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	141	class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005
3	141 143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	141	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	141 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
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3	141	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143 143	class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005
3	143	class 2 class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	141	class 2 class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	141	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	141	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143 141	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	141	class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	141 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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3	143 141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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3	143 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2008 Revision 4b (May 19, 2008
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3	141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2008 Revision 4b (May 19, 2008
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3	143 141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3	143 141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2008 Revision 4b (May 19, 2008
3	141	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143 143	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 200 Revision 4b (May 19, 200
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	141	class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2008 Revision 4b (May 19, 2008
	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3	143 141	class 2 class 2	Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 200 Revision 4b (May 19, 200
3 3 3 3		class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005
3 3 3 3 3	141	class 2	Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007	Revision 4b (May 19, 2008 Revision 4b (May 19, 2008
3 3 3 3	141 143 143	class 2		July 2007	
3 3 3 3 3 3 3 3 3	143 143 143	class 2	Y. Saito(JAEA)		Revision 4b (Way 19, 200;
3 3 3 3 3 3 3 3	143 143	class 2 class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007 July 2007	Revision 4b (May 19, 2005
3 3 3 3 3 3 3 3 3 3 3 3 3	143 143 143 141 141 141 143	class 2 class 2 class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3 3 3 3 3 3 3 3 3 3 3 3 3 3	143 143 143 141 141	class 2 class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007 July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	143 143 143 141 141 141 143 143	class 2 class 2 class 2 class 2 class 2 class 2	Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA) Y. Saito(JAEA)	July 2007 July 2007 July 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005

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65612 Silva et al. (1979)	yes	ves v	res (can be used)	class 1	Α	3	В	3	2	A/B	3	C/D	0	A/B	3	A/B	3		A	3	A	3	В	2	A/B	3	C/D	0	A	3	В	2	В	2	A	3	143	class 2	Y	Saito(JAEA)	) July 2007	Revision 4b (May 19, 2005
65613 Silva et al. (1979)	yes		es (can be used)	class 1	A	3	B		2	A/B	3	C/D	0	A/B	3	A/B	3		B	2	A	3	B	2	A/B	3	C/D	0	A	3	B	2	B	2	A	3	141	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65614 Silva et al. (1979)	yes	yes y	es (can be used)	class 1	A	3	B	3	2	A/B	3	C/D	0	A/B	3	A/B	3		A	3	A	3	В	2	A/B	3	C/D	0	A	3	В	2	В	2	A	3	143	class 2	Y	Saito(JAEA)	) July 2007	Revision 4b (May 19, 2005
65615 Silva et al. (1979)	yes		es (can be used)	class 1	A	3	B	3	2	A/B	3	C/D	0	A/B	3	A/B	3		A	3	A	3	B	2	A/B	3	C/D	0	A	3	B	2	B	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65616 Silva et al. (1979)	yes		es (can be used)	class 1	Δ	3	B	3	2	A/B	3	C/D	0	A/B	3	A/B	3		Δ	3	Δ	3	B	2	A/B	3	C/D	0	Δ	3	B	2	B	2	Δ	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65617 Silva et al. (1979)	ves		es (can be used)	class 1	Â	3	B	<u></u>	2	A/B	3	C/D	0	A/B	3	A/B	- 3			3	A	3	B	2	A/B	3	C/D	0		3	B	2	B	2		3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65618 Silva et al. (1979)	yes		es (can be used)	class 1	A	3	B		2	A/B	-	C/D	0	A/B	3	A/B	3			2	A	3	B	2	A/B	3	C/D	0	Â	3	B	2	B	2		3	141	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65619 Silva et al. (1979)	yes		es (can be used)	class 1	A .	3		2	2	A/B		C/D	0	A/B	3	A/B	- 3			2	A	3	B	2	A/B	3	C/D	0	-	3	8	2	B	2		3	141	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65620 Silva et al. (1979)			es (can be used)	class 1		3	8		2	A/B		C/D	0	A/B	3	A/B	3		A	3	A	3	B	2	A/B	3	C/D	0	-	3	8	2	B	2	Â	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65621 Silva et al. (1979)	yes		es (can be used)	class 1				,	2	A/B		C/D		A/B		A/B					A			2	A/B		C/D				-	2	B	2	+ 2		143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65622 Silva et al. (1979)	yes		es (can be used)	class 1	A .	3	D	2	2	A/B		C/D	0	A/B	3	A/B	3		A		A	3	B	2	A/B	3	C/D	0	A	3	B	2	B	2		3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
65623 Silva et al. (1979)	yes		es (can be used)	class 1	A			<u></u>	2	A/B		C/D	0	A/B	0	A/B			<u> </u>		Δ			2	A/B A/B		C/D	0	A			2	B	2	- <u>^</u>		143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
65624 Silva et al. (1979)	yes		es (can be used)	class 1	A			<u>}</u>	2	A/B		C/D	0	A/B		A/B			B	2	A			2	A/B		C/D	0				- 2	B	2	A		141	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65625 Silva et al. (1979)	yes			class 1	A			<u></u>	2	A/B		C/D	0	A/B	0	A/B		-		2	Δ			2	A/B A/B		C/D	0	A			2	B	2	- <u>^</u>		141	class 2		Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
65626 Silva et al. (1979)	yes		res (can be used)	class 1	A			<u> </u>	2	A/B		C/D	0	A/B A/B	3	A/B A/B	- 3			3	Δ	3	B	2	A/B A/B	3	C/D	0	A	3	- B	2	В	2	Α Δ		143	class 2 class 2		Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
65627 Silva et al. (1979)	yes		res (can be used) res (can be used)	class 1 class 1	A	2	B		2	A/B	-	C/D C/D	0	A/B A/B	2	A/B A/B	- 3		а.   А.	3	A	3	B	2	A/B A/B	3	C/D C/D	0	A .	3	B	2	B	2	Α Δ	3	143	class 2 class 2		. Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
65628 Silva et al. (1979)	yes				A			<u></u>	2	A/B		C/D	0	A/B	0	A/B			<u> </u>		A			2	A/B		C/D	0	A			2	B	2	- <u>^</u>		143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
65629 Silva et al. (1979)	yes		res (can be used)	class 1 class 1	A	3	B		2	A/B A/B	-		0	A/B A/B	3	A/B A/B	- 3			2	A	3	В	2	A/B A/B	3		0	A	3	B	2	В	2	A	3	141	class 2 class 2		. Saito(JAEA) . Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	yes		es (can be used)		A	- 3		<u> </u>	2	A/B A/B		C/D	0	A/B A/B	- 3		- 3		Δ		A	3	в	2	A/B A/B	3	C/D	0	A	3	-	2	в	2	A .	3						
	yes		res (can be used) res (can be used)	class 1	A	3	8	<u>.</u>	2	A/B		C/D C/D	0	A/B A/B	3	A/B A/B	- 3	······ '	A		A	3	В	2	A/B A/B	3	C/D	0	A	3	в	2	в	2	A	3	143	class 2 class 2		Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	yes	, , , , ,			A	- 3		· · · · ·	2	A/B	3	C/D	0	A/B A/B	- 3	A/B A/B	- 3		· · · · · · · · · · · · · · · · · · ·	3	Δ	3	в	2	A/B A/B	3	C/D C/D	0	A	3	-	2	в	2	A .	3	143					
	yes		res (can be used)	class 1	A	3	B	<u> </u>	2	A/B	3	0/0	0		3	A/B A/B	- 3		A	3	A	3	В	2		3		0	A	3	<u>в</u>	2	в	2	A	3		class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65633 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	8		2	A/B	-	C/D	0	A/B	3		3		8	2	A	3	B	2	A/B A/B	3	C/D	0	A	3	в	2	в	2	A	3	141	class 2			) July 2007	Revision 4b (May 19, 2005
65634 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B	<u> </u>	2	A/B		C/D	0	A/B	3	A/B	- 3		A	3	A	3	В	2	A/B	3	C/D	0	A	3	<u> </u>	2	в	2	A	3	143	class 2			) July 2007	Revision 4b (May 19, 2005
65635 Silva et al. (1979) 65636 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B		2	A/B	•	0/0	0	745	3	A/B	3		A	3	A	3	в	2	A/B	3	C/D	0	A	3	8	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
	yes		res (can be used)	class 1	A	- 3		<u> </u>	2	A/B		C/D	0	A/B	- 3	A/B	- 3		A	3	A	3	в	2	A/B A/B	3	C/D	0	A	3	-	2	в	2	A	3	143	class 2			) July 2007	Revision 4b (May 19, 2005
65637 Silva et al. (1979) 65638 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	8		2	A/B	-	C/D	0	A/B	3	A/B	3		A	3	A	3	в	2		3	C/D	0	A	3	8	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
	yes		res (can be used)	class 1	A	3	B	5	2	A/B		C/D	0	A/B A/B	3	A/B A/B	- 3		A	3	A	3	В.	2	A/B A/B	3	C/D C/D	0	A	3	в	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65639 Silva et al. (1979) 65640 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B	<u> </u>	2	A/B		C/D	0	A/B A/B	3		- 3	······ '	· · · · · · · · · · · · · · · · · · ·	3	A	3	В	2		3		0	A	3	<u>в</u>	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
	yes		res (can be used)	class 1	A	- 3		<u> </u>	2	A/B	3	C/D	0		- 3	A/B	- 3		A	3	A	3	в	2	A/B	3	C/D	0	A	3	-	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65641 Silva et al. (1979) 65642 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B	5	2	A/B	3	C/D	0	A/B A/B	3	A/B A/B	- 3		A	3	A	3	В	2	A/B A/B	3	C/D C/D	0	A	3	<u>в</u>	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
	yes		res (can be used)	class 1	A	3	B		2	A/B		C/D	0		3		3		-	3	A	3	B	2	,,,,,	3		0	A	3	в	2	в	2	A	3	1.10	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65643 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B	<u> </u>	2	A/B		C/D	0	A/B A/B	3	A/B	- 3	*	A	3	A	3	В	2	A/B A/B	3	C/D	0	A	3	<u> </u>	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65644 Silva et al. (1979) 65645 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B		2	A/B	3	C/D	0		3	A/B	- 3		A	-	A	3	В	2	A/B A/B	3	C/D	0	A	3	B	2	В	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	yes		res (can be used)	class 1	A	- 3		<u> </u>	2	A/B	3	C/D	0	A/B A/B	- 3	A/B A/B	- 3			3	A	3	в	2	A/B A/R	3	C/D C/D	0	A	3	-	2	в	2	A	3	143	class 2		Saito(JAEA)		
	yes		res (can be used)	class 1	A	3	B		2	A/B	<u> </u>	0/0	0		3		3			3	A	3	8	2	A/B A/B	3		0	A	3	B	2	в	2	A	3		class 2		Saito(JAEA)		Revision 4b (May 19, 2005
	yes		res (can be used)	class 1	A	- 3		<u> </u>	2	A/B A/B		C/D	0	A/B	- 3	A/B A/B	- 3		A	3	A	3	в	2	A/B A/B	3	C/D	0	A	3	-	2	в	2	A .	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65648 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B	<u> </u>	2	A/B	3	C/D	0	A/B	3	A/B A/B	- 3		<u> </u>	2	A	3	В	2	A/B	3	C/D	0	A	3	<u>в</u>	2	в	2	A	3	141	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65649 Silva et al. (1979)	yes		es (can be used)	class 1	A	3	B	5	2	A/B	3	C/D	0	A/B	3		- 3		8	2	A	3	В.	2	A/B	3	C/D	0	A	3	в	2	в	2	A .	3		class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65650 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B	3	2	A/B	3	C/D	0	A/B	3	A/B	- 3		<u>A</u>	3	A	3	B	2	A/B	3	C/D	0	A	3	B	2	В	2	<u> </u>	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65651 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	В	5	2	A/B		C/D	0	A/B	3	A/B	3	_	A	3	A	3	в	2	A/B	3	C/D	0	A	3	в	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65652 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B	3	2	A/B		C/D	0	A/B A/B	3	A/B A/B	3		A	3	A	3	B	2	A/B A/B	3	C/D	0	A	3	В	2	В	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65653 Silva et al. (1979) 65654 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B		2	A/B		C/D	0	A/B A/B	3		3		8	2	A	3	в	2	A/B A/B	3	C/D	0	A	3	8	2	в	2	A	3		class 2		Saito(JAEA)		Revision 4b (May 19, 2005
	yes		res (can be used)	class 1	A	3	B	5	2	A/B		C/D	0		3	A/B	- 3			3	A	3	В.	2		3	C/D	0	A	3	в	2	в	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65655 Silva et al. (1979)	yes		res (can be used)	class 1	A	3		<u>.</u>	2	A/B	3	C/D	0	A/B	3	A/B	3			3	A	3	8	2	A/B	3	C/D		A	3	<u> </u>	2	8	2	A	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65656 Silva et al. (1979)	yes		es (can be used)	class 1	A	3	- B	5	2	A/B	3	C/D	0	A/B	3	A/B	3		A	3	A	3	В	2	A/B	3	C/D C/D	0	A	3	в	2	8	2	<u> </u>	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65657 Silva et al. (1979)	yes		res (can be used)	class 1	A	3	B	5	2	AVB	3	0/0	0	A/B	3	A/B	- 3	······ '	A	3	A	3	В	2	A/B	3		0	A	3	8	2	8	2	<u> </u>	3	143	class 2		Saito(JAEA)		Revision 4b (May 19, 2005
65658 Silva et al. (1979)	yes	yes y	res (can be used)	class 1	A	3	B	5	2	A/B	3	C/D	U	A/B	3	A/B	3		B	2	A	3	В	2	A/B	3	C/D	0	A	3	в	2	в	2	A	3	141	class 2	Y	. saito(JAEA)	) July 2007	Revision 4b (May 19, 2005

I.C. = not cor	nclusive, N.E. = not eval	luated, m.i.	= missing info	ormation																																				
Pu				documentation and	type of K _d											01		II - Technica	l and scientifi	c quality of	f reported dat															Operator	Date	Classific	cation Gui	deline
			Rating • I-a	-a: yes/no, I-b: class 1-	6											Check	coints 🕊 le	vel: A-D (nume	rical value: 3-0	i) / unreliable	e Rating 🕈	class 1-6/	inreliable						1						III - Consiste	ncy			_	_
Datapoint	Reference	I-a.1	I-a.2	Rating I-a	Rating I-b	l solid	ll.a I phase		II.b pH	ll redox co	l.c onditions	II.d solution com	position	II.e temperatu	re	II.f S/W	so	II.g rption value	II. inital		II.i phase sepa	ration	II.j reaction t	time	II. agita	k ition	II.I RN loading	rea	II.m Inction vessels	error e	II.n stimates		I.o r variation	Rati	ng II comment/ra	ting				
Bentonite																																		total value	class		-			
65922	Baston et al.(1998)	yes	yes y	yes (can be used)	class 1	A	3	A	3	C/D	0	A/B	3	A/B	3 A	B 3	C/	0 0	В	2	A	3	C/D	0	A/B	3	C/D	) A	3	A	3	D	0	115	class 3	Y. Saito(JAEA	July 200	7 Revision	4b (May 1	9, 2005)
65923	Baston et al.(1998)	yes	yes y	yes (can be used)	class 1	А	3	A	3	C/D	0	A/B	3	A/B	3 A	B 3	C/	0 (	В	2	A	3	C/D	0	A/B	3	C/D	) A	3	A	3	D	0	115	class 3	Y. Saito(JAEA	July 200	7 Revision	4b (May 1	9, 2005)
65924	Baston et al.(1998)	yes	yes y	yes (can be used)	class 1	А	3	A	3	C/D	0	A/B	3	A/B	3 A	B 3	C/	0 (	В	2	A	3	C/D	0	A/B	3	C/D	) A	3	A	3	D	0	115	class 3	Y. Saito(JAEA	July 200	7 Revision	4b (May 1	9, 2005)
65925	Baston et al.(1998)	yes	yes y	yes (can be used)	class 1	A	3	A	3	C/D	0	A/B	3	A/B	3 A	B 3	C/	0 0	В	2	A	3	C/D	0	A/B	3	C/D	) A	3	A	3	D	0	115	class 3	Y. Saito(JAEA				
65926	Baston et al.(1998)	yes	yes y	yes (can be used)	class 1	Α	3	A	3	C/D	0	A/B	3	A/B	3 A	B 3	C/	0 0	В	2	A	3	C/D	0	A/B	3	C/D	) A	3	A	3	D	0	115	class 3	Y. Saito(JAEA)				
65927	Baston et al.(1998)	yes	yes y	es (can be used	class 1	A	3	A	3	C/D	0	A/B	3	A/B	3 A	B 3	C/	0 (	В	2	A	3	C/D	0	A/B	3	C/D	A C	3	A	3	D	0	115	class 3	Y. Saito(JAEA	July 200	7 Revision	4b (May 1	9,2005

		n, N.E. = not evaluated, m.i. =			of documentation	and type	of K.										II - Technico	and scienti	fic quality of	reported data												III - Consistence	Operator	Date	Classification Guideline
Norm         Norm        Norm        Norm        No		rence		Rating ¥	I-a: yes/no, I-b: clas	ss 1-6		ll.a		II.b		II.c	ll.d	II.e		eckpoints	<ul> <li>level: A-D (nume)</li> </ul>	ical value: 3-	-0) / unreliable	e Rating 🖝 cla	iss 1-6 / unre	eliable II.j	ll.k		11.1	ll.m		ll.n	-	.0	Ratin		Operator	Date	Classification Guideline
														temperature	s/w			inita	I [RN]	phase separati	ion re	reaction time	agitati	ion RN	loading	reaction ves	ssels e	ror estimates		r variation					
																																unreliable		October 2007	Revision 4b (May 19, 2005
	240 Barne 241 Barne	ey and Anderson (1979) ey and Anderson (1979)	yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
																																			Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	323 Barne	ey and Brown (1979)	yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	664 Basto	on et al. (1995b)	yes	yes	yes (can be used)	cl			A	A 3		3		C/D 0	C/D	0	C/D 0		3	C/D 0	) C/	/D 0	C/D	0 B	2				В		125		Y. Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	666 Basto	on et al. (1995b)	yes	yes yes	yes (can be used) yes (can be used)	cl cl	ass 1 C	/D 0			A/B	3	A/B 3		C/D C/D C/D	0	C/D 0 C/D 0 C/D 0	A	3		C/			0 B 0 B	2 2 2	A	3	A 3	B	2	125	class 2	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	002 Danie	els (1981)	yes	yes	yes (can be used)	cl	lass 1 C	/D 0	A	A 3		3	C/D 0	A/B 3	C/D	0	C/D 0	unreliable	1													unreliable	Y. Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	004 Daniel 005 Daniel	als (1981) als (1981)	yes	yes yes	yes (can be used) yes (can be used)	cl	class 1 C	/D 0 /D 0	A	A 3 A 3 A 3	A/B	3	C/D 0 C/D 0	A/B 3 A/B 3	C/D	0	C/D 0	B	2 2						0	B	2	0 0	B		88	class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	007 Daniel 008 Daniel	els (1981) els (1981)	yes	yes yes	yes (can be used) yes (can be used)	cl cl	ass 1 C		A A A	A 3 A 3 A 3	A/B A/B	3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0	C/D 0 C/D 0			C/D 0 C/D 0	) C/	/D 0 /D 0	A/B A/B	3 C/D 3 C/D	0	B	2	0 0	B	2	88 88	class 4	Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	009 Danie 010 Danie	als (1981) als (1981)	yes yes	yes yes	yes (can be used) yes (can be used)	cl	ass 1 C	/D 0	A	A 3 A 3	A/B	3	C/D 0	A/B 3 A/B 3	C/D		C/D 0 C/D 0	unreliable	·				Δ/B	3 C/D	0	В	2	) 0	В	2	88		Y. Saito(JAEA) Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	012 Daniel 013 Daniel	als (1981) als (1981)	yes yes	yes yes	yes (can be used) yes (can be used)	cl cl	class 1 C	/D 0 /D 0	A	A 3 A 3	A/B A/B	3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0	C/D 0 C/D 0	unreliable							0	В	2	0	B	2	88	class 4 unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	015 Daniel	els (1981)	yes	yes	yes (can be used)	cl	ass 1 C	/D 0	A	A 3 A 3 A 3	A/B A/B A/B	3	C/D 0	A/B 3		0	C/D 0 C/D 0 C/D 0	В	2		) C/	/D 0	A/B		0	В	2	0 0	В	2	88		Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	017 Danie 018 Danie	əls (1981) əls (1981)	yes yes	yes yes	yes (can be used) yes (can be used)	cl	class 1 C	/D 0 /D 0	A	A 3 A 3		3	C/D 0 C/D 0	A/B 3 A/B 3	C/D		C/D 0	B unreliable	2	C/D 0	) C/	/D 0	A/B	3 C/D	0	В	2	0 0	В	2		unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	020 Danie 021 Danie	əls (1981) əls (1981)	yes yes	yes yes	yes (can be used) yes (can be used)	cl	ass 1 C	/D 0 /D 0	A	A 3 A 3	A/B A/B	3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0	C/D 0 C/D 0	B	2	C/D 0	) C/	/D 0	A/B	3 C/D	0	B	2	0 0	B	2 2	88	class 4 class 4	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	023 Danie	els (1981)	yes yes	yes yes	yes (can be used) yes (can be used)	cl	ass 1 C	/D 0	A A A	A 3 A 3 A 3	A/B	3	C/D 0 C/D 0 C/D 0	A/B 3 A/B 3 A/B 3	C/D	0	C/D 0	unreliable	1	C/D 0				3 C/D			2	0 0	B	2		unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	025 Daniel	els (1981)	yes	yes	yes (can be used)	cl			A	A 3		3				0													_			unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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	981 Erdal	(1980)	yes	no	no (can not be use	ed)						_																				unreliable	Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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	986 Erdal 987 Erdal	(1980)	yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
Image         Image <th< td=""><td>989 Erdal</td><td>(1980)</td><td>yes</td><td>no</td><td>no (can not be use</td><td>ed)</td><td></td><td></td><td></td><td></td><td></td><td>-</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>unreliable</td><td>Y. Saito(JAEA)</td><td>October 2007 October 2007 October 2007</td><td>Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005</td></th<>	989 Erdal	(1980)	yes	no	no (can not be use	ed)						-																				unreliable	Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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		(1980)	yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	996 Erdal 997 Erdal	(1980) (1980)	yes yes	no	no (can not be use no (can not be use	ed) ed)																										unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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Mo	002 Erdal	(1980)	yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	004 Erdal	(1980) (1980)	yes	no	no (can not be use no (can not be use	ed) ed)																										unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	007 Erdal	(1980)	yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA)	October 2007 October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
No.         No. <td>009 Erdal 010 Erdal</td> <td>(1980) (1980)</td> <td>yes yes</td> <td>no</td> <td>no (can not be use no (can not be use</td> <td>ed) ed)</td> <td></td> <td>unreliable unreliable</td> <td>Y. Saito(JAEA) Y. Saito(JAEA)</td> <td>October 2007 October 2007</td> <td>Revision 4b (May 19, 2005 Revision 4b (May 19, 2005</td>	009 Erdal 010 Erdal	(1980) (1980)	yes yes	no	no (can not be use no (can not be use	ed) ed)																										unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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	015 Erdal	(1980)	yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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Bin         Bin <td>023 Erdal</td> <td>(1980) (1980)</td> <td>yes yes</td> <td>no</td> <td>no (can not be use</td> <td>ed)</td> <td></td> <td>unreliable</td> <td>Y. Saito(JAEA)</td> <td>October 2007</td> <td>Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005</td>	023 Erdal	(1980) (1980)	yes yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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9585         Balf 1980         948         749         600 (and deteed)         7400 (and deted)         7400 (and deted)         7400 (and	033 Erdal 034 Erdal	(1980) (1980)	yes	no	no (can not be use no (can not be use	ed) ed)																										unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
9168         944         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94         94       94       94         94 </td <td>035 Erdal 036 Erdal 037 Erdal</td> <td>(1980) (1980) (1980)</td> <td>yes</td> <td>no</td> <td>no (can not be use</td> <td>ed)</td> <td></td> <td></td> <td>_</td> <td></td> <td>unreliable</td> <td>Y. Saito(JAEA) Y. Saito(JAEA)</td> <td>October 2007 October 2007</td> <td>Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005</td>	035 Erdal 036 Erdal 037 Erdal	(1980) (1980) (1980)	yes	no	no (can not be use	ed)			_																							unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
5044       Edd       Edd       Celu       <	038 Erdal 039 Erdal	(1980) (1980)	yes yes	no	no (can not be use no (can not be use	ed) ed)		_	_			+															_					unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
5104         End (198)         98         70         70(an obsues)         70        70(an obsues)         70	041 Erdal 042 Erdal	(1980)	yes	no	no (can not be use	ed)												+														unreliable	Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
5104       End (198)       yes       no       no (an nobe used)                                                                                                                              <	043 Erdal 044 Erdal	(1980) (1980)	yes yes	no no	no (can not be use no (can not be use	ed) ed)																										unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
5100 5101 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104 5104<	046 Erdal 047 Erdal	(1980) (1980)	yes yes	no	no (can not be use no (can not be use	ed) ed)												<b> </b>														unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
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5105 cha(1980) yes no no (can robe used)<	051 Erdal 052 Erdal	(1980) (1980)	yes yes	no	no (can not be use no (can not be use	ed) ed)						-																				unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
51056       Erda(1980)       yes       no       no (can tob busd)       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -	054 Erdal 055 Erdal	(1980) (1980)	yes	no	no (can not be use no (can not be use	ed) ed)																										unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
51059       Erdal (1980)       yes       no       no (can nob e used)	057 Erdal	(1980)	yes	no	no (can not be use	ed)																										unreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
51062 Erda(1980) yes no no(canotbeused) and a constructional description of a construction of a constr	059 Erdal 060 Erdal	(1980) (1980)	yes yes	no	no (can not be use no (can not be use	ed) ed)						+																				unreliable unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
	061 Erdal 062 Erdal	(1980) (1980)	yes	no	no (can not be use	ed)						+																				unreliable	Y. Saito(JAEA) Y. Saito(JAEA)	October 2007 October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005
53185       Hggo et al. (1987)       yes       yes       yes (can be used)       class 1       C/D       0		o et al. (1987)	yes yes	yes yes	yes (can be used) yes (can be used)	cl cl	ass 1 C	/D 0 /D 0	C	C 1 C 1	A/B A/B	3	C/D 0 C/D 0	A/B 3 A/B 3	C/D C/D	0	C/D 0 C/D 0	B	2	C/D 0 C/D 0	) C/	/D 0 /D 0	A/B A/B	3 C/D 3 C/D	0	C/D C/D	0		B	2	72 72	class 4 class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005 Revision 4b (May 19, 2005 Revision 4b (May 19, 2005

 $-113 \sim 114 -$ 

53193 Higgo et al. (1987)		yes yes (can be used)	class			С	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72		Y. Saito(JAEA)		Revision 4b (May 19, 2005)
53195 Higgo et al. (1987)		yes yes (can be used)	class			C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7		class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
53197 Higgo et al. (1987)		yes yes (can be used)	class			C	1	A/B	3	C/D	0	A/B		C/D		C/D 0	) В	2	C/D	0	C/D	0 A		C/D		C/D	0	С	1	B	-		class 4	Y. Saito(JAEA)	October 2007	
53199 Higgo et al. (1987)	yes	yes yes (can be used)	class			C	1	A/B	3	C/D	0	A/B		C/D		C/D 0	) В	2	C/D	0	C/D	0 A		C/D		C/D	0	С	1	В	2 7		class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
53201 Higgo et al. (1987)		yes yes (can be used)	class			C	1	A/B		C/D	0	A/B		C/D		C/D 0	) В	2	C/D		C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7		class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
53203 Higgo et al. (1987)		yes yes (can be used)	class	s 1 C/D		C	1	A/B	3	C/D	0	A/B		C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В			class 4	Y. Saito(JAEA)		
53205 Higgo et al. (1987)		yes yes (can be used)	class		0	С	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72	class 4	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
53207 Higgo et al. (1987)	yes	yes yes (can be used)	class	s 1 C/D	0	C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72	class 4	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
53209 Higgo et al. (1987)	yes	yes yes (can be used)	class	s 1 C/D	0	C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	B	2 7	72	class 4	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
53211 Higgo et al. (1987)	yes	yes yes (can be used)	class	s 1 C/D	0	C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72	class 4	Y. Saito(JAEA)		
53213 Higgo et al. (1987)	yes	yes yes (can be used)	class	s 1 C/D	0	C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) B	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72	class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
53215 Higgo et al. (1987)	yes	yes yes (can be used)	class	s 1 C/D	0	C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72	class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
53217 Higgo et al. (1987)		yes yes (can be used)	class	s 1 C/D	0	C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72	class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
53219 Higgo et al. (1987)	ves	yes yes (can be used)	class	s1 C/D	0	C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72	class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
53221 Higgo et al. (1987)		yes yes (can be used)	class	s 1 C/D	0	C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) B	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	С	1	В	2 7	72	class 4	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
53223 Higgo et al. (1987)		yes yes (can be used)	class			C	1	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) B	2	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	C	1	B	2 7	72	class 4	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
	1	<u></u> ,,,													-					-								-	-	-						
53965 Kim et al. (1994)	ves	no no (can not be used)							-		1							-															nreliable	Y. Saito(JAFA)	October 2007	Revision 4b (May 19, 2005)
53966 Kim et al. (1994)		no no (can not be used)							1	1	1	<u>     </u>							1	1	1 1				1								nreliable	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
										1	1	-								1	<u>  </u>															
54003 Konishi et al. (1988)	VAS	yes yes (can be used)	class	s1 B	2	A	3	A/B	3	C/D	0	A/B	3 (	C/D	0	C/D 0	) В	2	В	2	C/D	0 C	/D 0	C/D	0	C/D	0	D	0	C	1 9	95	class 3	Y. Saito(JAFA)	October 2007	Revision 4b (May 19, 2005)
54004 Konishi et al. (1988)		yes yes (can be used)	class		2	A	3	A/B		C/D	0	A/B		C/D		C/D 0		2	B	2		0 C		C/D		C/D	0	D	0	C	1 0	95	class 3	Y Saito(JAEA)		Revision 4b (May 19, 2005)
ordor Indinidir dr di. (1000)	100	Job Job (dan be doed)	ondoc					700									· · · · · · ·				0,0					0.0			-		`		04000	1. Outo(0/12/1)		ricitation 40 (may 10, 2000)
55991 Legoux et al. (1992)	Ves	yes yes (can be used)	class	s 1 Δ	3	Δ	3	A/B	3	A/B	3	C/D	0	VВ	3			2	Δ	3	C/D	0 A	/B 3	C/D	0	C/D	0	Δ	3	D	0 1	133	class 2	Y Saito(JAFA)	October 2007	Revision 4b (May 19, 2005)
55992 Legoux et al. (1992)		yes yes (can be used)	class	s1 A	3	A	3	A/B				C/D		VB		C/D 0		2	A	3	C/D	0 A		C/D	0	C/D	0	A	3	D			class 2			Revision 4b (May 19, 2005)
55993 Legoux et al. (1992)		yes yes (can be used)	class		3	A		A/B		A/B	3	C/D		VB		C/D 0		2	A		C/D	0 A		C/D	0	C/D	0	A	3	D			class 2	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
55994 Legoux et al. (1992)		yes yes (can be used)		s1 A	2						3						) B	2		3			/B 3		0	C/D	0		3	D		133				Revision 4b (May 19, 2005)
55994 Legoux et al. (1992)	yes	yes yes (can be used)	CidSt	51 A	3	A		AVD		AVD		G/D		-VD	3	5/0 0	, .				G/D	0 A	10 3			G/D		~	3	0	0 1	133	CidSS 2	T. Sallo(JAEA)	OCIODEI 2007	Revision 40 (May 19, 2003)
60389 Tanaka et al. (1996)	VOE	no no (can not be used)					-		-		-																						nreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
60390 Tanaka et al. (1996)	yes	no no (can not be used)																															nreliable	Y Saito(JAEA)	October 2007	
60391 Tanaka et al. (1996)		no no (can not be used)																															nreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
60392 Tanaka et al. (1996)		no no (can not be used) no no (can not be used)																															nreliable			Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
					_				_		-								_	-														Y. Saito(JAEA)		Revision 4b (May 19, 2005)
		no no (can not be used)							-																								nreliable	Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
60394 Tanaka et al. (1996)	yes	no no (can not be used)							_										_						_							L	nreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
00000 7 1 14 1 (1000)		( ) N					-	1.0	-	1.0	-	1.00						-		-					-			-						N. 0. 1. (11 E 1)	0.1.1.0007	Revision 4b (May 19, 2005)
62326 Tanaka and Muraoka (1999) 62327 Tanaka and Muraoka (1999)		yes yes (can be used)	class			A	3	A/B		A/B	3	A/B		A/B			) A	-	C/D		C/D		/B 3	C/D		C/D	0		0	D		114		Y. Saito(JAEA) Y. Saito(JAEA)		Revision 4b (May 19, 2005) Revision 4b (May 19, 2005)
		yes yes (can be used)	class			A	3	A/B		A/B	3	A/B		¥ИВ		C/D 0	) A		C/D		C/D		/B 3	C/D		C/D	0	D	0	D		114				
62328 Tanaka and Muraoka (1999)	yes	yes yes (can be used)	class	s 5 A	3	A	3	A/B	3	A/B	3	A/B	3 /	¥/В	3	C/D 0	) A	3	C/D	0	C/D	0 A	/B 3	C/D	0	C/D	0	D	0	D	0 1	114	class 3	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
															-																					
62367 Tanaka and Muraoka (1998)	yes	yes yes (can be used)	class		2	B	2	A/B	3	A/B	3	A/B		¥В		C/D 0	) B	2	В	2	A/B		/B 3	C/D		В	2	D	0	D		120		Y. Saito(JAEA)		Revision 4b (May 19, 2005)
62368 Tanaka and Muraoka (1998)		yes yes (can be used)	class		2	B	2	A/B		A/B	3	A/B		A/B		C/D 0	B	2	B	2	A/B	3 A		C/D		В	2	D	U	D			class 3	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
62369 Tanaka and Muraoka (1998)	yes	yes yes (can be used)	class	s1 B	2	В	2	A/B	3	A/B	3	A/B	3 /	∿/B	3	C/D 0	ј В	2	В	2	A/B	3 A	и 3	C/D	0	В	2	U	U	D	0 1	120	class 3	Y. Saito(JAEA)	Uctober 2007	Revision 4b (May 19, 2005)
62810 Tanaka et ai. (1999)			_				_		_										_															N. 0. 11 (11 - 11	0.11.0077	
		no no (can not be used)																			-												nreliable	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
62811 Tanaka et ai. (1999)		no no (can not be used)			_				-										-														nreliable	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
62812 Tanaka et ai. (1999)		no no (can not be used)							_	-		-																					nreliable	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
62813 Tanaka et ai. (1999)		no no (can not be used)																															nreliable	Y. Saito(JAEA)		Revision 4b (May 19, 2005)
62814 Tanaka et ai. (1999)	yes	no no (can not be used)								1										1												ι	nreliable	Y. Saito(JAEA)	October 2007	Revision 4b (May 19, 2005)
										-		-								1																
Bentonite	1		_		_				-											-																
										1	1	-																								
65940 Baston et al.(1998)		yes yes (can be used)	class	s1 A	3	A	3	A/B	3	A/B	3	A/B		∿/B		C/D 0	) В	2	A	3	C/D	0 A		C/D	0	A	3	A	3	D		139				Revision 4b (May 19, 2005)
65941 Baston et al.(1998)		yes yes (can be used)	class	s1 A	3	A	3	A/B	3	A/B	3	A/B		A/B		C/D 0	) В	2	A	3	C/D	0 A	/B 3	C/D	0	A	3	A	3	D			class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005)
65942 Baston et al.(1998)		yes yes (can be used)	class		3	A	3	A/B		A/B	3	A/B		A/B		C/D 0	) В	2	A	3	C/D	0 A	/B 3	C/D	0	A	3	A	3	D		139		Y. Saito(JAEA)		Revision 4b (May 19, 2005)
65943 Baston et al.(1998)	yes	yes yes (can be used)	class	s1 A	3	A	3	A/B	3	A/B	3	A/B		A/B	3	C/D 0	) В	2	A	3	C/D	0 A	/B 3	C/D	0	A	3	A	3	D			class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005)
65944 Baston et al.(1998)		yes yes (can be used)	class		3	A	3	A/B	3	A/B	3	A/B		A/B	3	C/D 0	) В	2	A	3	C/D	0 A	/B 3	C/D	0	A	3	A	3	D		139		Y. Saito(JAEA)		Revision 4b (May 19, 2005)
65945 Baston et al.(1998)	yes	yes yes (can be used)	class	s1 A	3	A	3	A/B	3	A/B	3	A/B	3 /	A/B	3	C/D 0	) В	2	A	3	C/D	0 A	/B 3	C/D	0	A	3	A	3	D	0 1	139	class 2	Y. Saito(JAEA)	July 2007	Revision 4b (May 19, 2005)
										*	*																									

	表1.	SI	基	本	単位	Ż
其7	本昰		SI	基	本	单位
- 48/	严里		名	称		記号
長	い	メ	I	$\mathbb{P}$	ル	m
質	量	キ	ログ	ブラ	4	kg
時	間		耟	少		S
電	流	7	ン	ペ	7	А
熱力的	学温度	ケ	$\mathcal{N}$	ビ	ン	K
物質	質量	モ			ル	mol
光	度	力	ン	デ	ラ	cd

表2. 基本単位	を用いて表されるSI組立単位	の例
組立量	SI 基本単位	
和立里	名称	記号
面 積	平方メートル	$m^2$
体 積	立法メートル	$m^3$
速さ、速度	メートル毎秒	m/s
加 速 度	メートル毎秒毎秒	$m/s^2$
波 数	毎 メ ー ト ル	m-1
密度(質量密度)	キログラム毎立法メートル	kg/m ³
質量体積(比体積)	立法メートル毎キログラム	m ³ /kg
電 流 密 度	アンペア毎平方メートル	$A/m^2$
磁界の強さ	アンペア毎メートル	A/m
(物質量の)濃度	モル毎立方メートル	$mol/m^3$
輝 度	カンデラ毎平方メートル	$cd/m^2$
屈 折 率	(数 の) 1	1

				表5.S	I 接頭語	ĥ	
乗数		接	頭語	記号	乗数	接頭語	記号
$10^{24}$		Ш	タ	Y	$10^{-1}$	デシ	d
$10^{2}$	1	ゼ	タ	Z	$10^{-2}$	センチ	с
$10^{18}$		Ι	クサ	Е	$10^{-3}$	ミリ	m
$10^{15}$	5	ペ	タ	Р	$10^{-6}$	マイクロ	μ
$10^{12}$	2	テ	ラ	Т	$10^{-9}$	ナノ	n
$10^{9}$		ギ	ガ	G	$10^{-12}$	ピコ	р
$10^{6}$		X	ガ	Μ	$10^{-15}$	フェムト	f
$10^{3}$		キ		k	$10^{-18}$	アト	а
$10^{2}$		$\sim$	クト	h	$10^{-21}$	ゼプト	Z
10 ¹		デ	力	da	$10^{-24}$	ヨクト	У

表3. 固有の名称とその独自の記号で表されるSI組立単位

			SI 組立単位	
組立量	名称	記号	他のSI単位による	SI基本単位による
			表し方	表し方
	ラジアン ^(a)	rad		$\mathbf{m} \cdot \mathbf{m}^{-1} = 1^{(b)}$
	ステラジアン ^(a)	sr ^(c)		$m^{2} \cdot m^{-2} = 1^{(b)}$
周 波 数		Hz		s ⁻¹
_ 力	ニュートン	N		m•kg•s ⁻²
压力, 応力		Pa	$N/m^2$	$\mathbf{m}_{a}^{-1} \cdot \mathbf{kg} \cdot \mathbf{s}_{a}^{-2}$
エネルギー、仕事、熱量		J	N•m	$m^2 \cdot kg \cdot s^{-2}$
工 率 , 放 射 束		W	J/s	$m^2 \cdot kg \cdot s^{-3}$
	クーロン	С		s•A
電位差(電圧),起電力		V	W/A	$m^2 \cdot kg \cdot s^{-3} \cdot A^{-1}$
	ファラド	F	C/V	$\mathbf{m}^{-2} \cdot \mathbf{kg}^{-1} \cdot \mathbf{s}^4 \cdot \mathbf{A}^2$
	オーム	Ω	V/A	$m^2 \cdot kg \cdot s^{-3} \cdot A^{-2}$
コンダクタンス		S	A/V	$\mathbf{m}^{-2} \cdot \mathbf{kg}^{-1} \cdot \mathbf{s}^3 \cdot \mathbf{A}^2$
	ウエーバ	Wb	V · s	$\mathbf{m}^2 \cdot \mathbf{kg} \cdot \mathbf{s}^{-2} \cdot \mathbf{A}^{-1}$
磁 束 密 度 インダクタンス	テスラ	Т	Wb/m ²	$kg \cdot s^{-2} \cdot A^{-1}$
		Н	Wb/A	$m^2 \cdot kg \cdot s^{-2} \cdot A^{-2}$
セルシウス温度	セルシウス度 ^(d)	°C		K
光東	ルーメン	1 m	cd • sr ^(c)	$m^2 \cdot m^{-2} \cdot cd = cd$
照度	ルクス	1 x	$1\mathrm{m/m^2}$	$m^2 \cdot m^{-4} \cdot cd = m^{-2} \cdot cd$
(放射性核種の)放射能	ベクレル	Bq		s ⁻¹
吸収線量、質量エネル	ガレイ	Gy	J/kg	$m^2 \cdot s^{-2}$
キー分与、カーマ		ay	37 118	ш · 5
線量当量,周辺線量当		C	T. /1	2 -2
量,方向性線量当量,個	シーベルト	Sv	J/kg	$m^2 \cdot s^{-2}$
人線量当量, 組織線量当				

(a) ラジアン及びステラジアンの使用は、同じ次元であっても異なった性質をもった量を区別するときの組立単位の表し方として利点がある。組立単位を形作るときのいくつかの用例は表4に示されている。
 (b) 実際には、使用する時には記号rad及びsrが用いられるが、習慣として組立単位としての記号"1"は明示されない。
 (c) 測光学では、ステラジアンの名称と記号srを単位の表し方の中にそのまま維持している。
 (d) この単位は、例としてミリセルシウス度m℃のようにSI接頭語を伴って用いても良い。

表4.単位の中に固有の名称とその独自の記号を含むSI組立単位の例

衣4.単位の中に	_ 固有の名称とその独自の		
組立量		SI 組立単	鱼位
和立里	名称	記号	SI 基本単位による表し方
粘度	パスカル秒	Pa•s	$m^{-1} \cdot kg \cdot s^{-1}$
カのモーメント	ニュートンメートル	N•m	$m^2 \cdot kg \cdot s^{-2}$
表 面 張 九	ニュートン毎メートル	N/m	$kg \cdot s^{-2}$
角 速 度	ラジアン毎秒	rad/s	$\mathbf{m} \cdot \mathbf{m}^{-1} \cdot \mathbf{s}^{-1} = \mathbf{s}^{-1}$
角 加 速 度	ラジアン毎平方秒	rad/s ²	$m \cdot m^{-1} \cdot s^{-2} = s^{-2}$
熱流密度, 放射照度	ワット毎平方メートル	$W/m^2$	$kg \cdot s^{-3}$
熱容量、エントロピー	ジュール毎ケルビン	J/K	$ \begin{array}{c} \mathbf{kg} \cdot \mathbf{s}^{-3} \\ \mathbf{m}^2 \cdot \mathbf{kg} \cdot \mathbf{s}^{-2} \cdot \mathbf{K}^{-1} \end{array} $
質量熱容量(比熱容量)	ジュール毎キログラム		
質量エントロピー	毎ケルビン	J∕(kg • K)	$\mathbf{m}^2 \cdot \mathbf{s}^{-2} \cdot \mathbf{K}^{-1}$
質量エネルギー		<b>x</b> //	2 -21
(比エネルギー)	ジュール毎キログラム	J/kg	$\mathbf{m}^2 \cdot \mathbf{s}^{-2} \cdot \mathbf{K}^{-1}$
**	ワット毎メートル毎ケ	W ( ( Y)	21
熱 伝 導 率	ルビン	₩/ (m • K)	$\mathbf{m} \cdot \mathbf{kg} \cdot \mathbf{s}^{-3} \cdot \mathbf{K}^{-1}$
	ジュール毎立方メート	<b>x</b> / 3	-1 , -2
体積エネルギー	N		$m^{-1} \cdot kg \cdot s^{-2}$
電界の強さ	ボルト毎メートル	V/m	$\mathbf{m} \cdot \mathbf{kg} \cdot \mathbf{s}^{-3} \cdot \mathbf{A}^{-1}$
体積電荷	- クーロン毎立方メート		$m^{-3} \cdot s \cdot A$
14 1 电 10	ル	C/m ³	m · · S · A
電 気 変 位	クーロン毎平方メート	0.4-2	$m^{-2} \cdot s \cdot A$
电火发性	ル	$C/m^2$	m · s · A
誘 電 率	『ファラド毎メートル		$\mathbf{m}^{-3} \cdot \mathbf{kg}^{-1} \cdot \mathbf{s}^{4} \cdot \mathbf{A}^{2}$
透磁率	「ヘンリー毎メートル	H/m	$\mathbf{m} \cdot \mathbf{kg} \cdot \mathbf{s}^{-2} \cdot \mathbf{A}^{-2}$
モルエネルギー	ジュール毎モル	J/mol	$m^2 \cdot kg \cdot s^{-2} \cdot mol^{-1}$
モルエントロピー,	ジュール毎モル毎ケル	$I \neq (mol \cdot K)$	$\mathbf{m}^2 \cdot \mathbf{kg} \cdot \mathbf{s}^{-2} \cdot \mathbf{K}^{-1} \cdot \mathbf{mol}^{-1}$
	ビン		-
照射線量 (X線及びγ線)	クーロン毎キログラム	C/kg	$kg^{-1} \cdot s \cdot A$ $m^2 \cdot s^{-3}$
吸収線量率			
放 射 強 度	ワット毎ステラジアン	W/sr	$\mathbf{m}^4 \cdot \mathbf{m}^{-2} \cdot \mathbf{kg} \cdot \mathbf{s}^{-3} = \mathbf{m}^2 \cdot \mathbf{kg} \cdot \mathbf{s}^{-3}$
放 射 輝 度	ワット毎平方メートル	$W/(m^2 - cm)$	$\mathbf{m}^2 \cdot \mathbf{m}^{-2} \cdot \mathbf{kg} \cdot \mathbf{s}^{-3} = \mathbf{kg} \cdot \mathbf{s}^{-3}$
	毎ステラジアン	₩/ (ш • ST)	ш•ш•кg•s =кg•s

表6. 国際単位系と併用されるが国際単位系に属さない単位

· 国际中位示C	- 01711 C	4.6976国际平区木に周には44平区
名称	記号	SI 単位による値
分	min	1 min=60s
時	h	1h =60 min=3600 s
日	d	1 d=24 h=86400 s
度	0	$1^{\circ} = (\pi / 180)$ rad
分	,	1' = $(1/60)^{\circ}$ = $(\pi/10800)$ rad
秒	"	1" = $(1/60)$ ' = $(\pi/648000)$ rad
リットル	1. L	$11=1 \text{ dm}^3=10^{-3}\text{m}^3$
トン	t	1t=10 ³ kg
ネーパ	Np	1Np=1
ベル	В	1B= (1/2) 1n10 (Np)

表7. 国際単位系と併用されこれに属さない単位で SI単位で表される数値が実験的に得られるもの					
名称	記号	SI 単位であらわされる数値			
電子ボルト	eV	$\begin{array}{l} 1 eV = 1. \ 60217733 \ (49) \ \times 10^{-19} J \\ 1 u = 1. \ 6605402 \ (10) \ \times 10^{-27} kg \\ 1 u a = 1. \ 49597870691 \ (30) \ \times 10^{11} m \end{array}$			
統一原子質量単位	u	1u=1.6605402 (10) ×10 ⁻²⁷ kg			
天 文 単 位	ua	1ua=1. 49597870691 (30) ×10 ¹¹ m			

表8. 国際単位系に属さないが国際単位系と

	併用されるその他の単位					
	名称		記号	SI 単位であらわされる数値		
海		里		1 海里=1852m		
ノ	ツ	ト		1ノット=1海里毎時=(1852/3600)m/s		
P		ル	а	$1 \text{ a=} 1 \text{ dam}^2 = 10^2 \text{m}^2$		
へり	7 タ ー	ル		$1 \text{ ha}=1 \text{ hm}^2=10^4 \text{m}^2$		
バ	-	ル	bar	1 bar=0. 1MPa=100kPa=1000hPa=10 ⁵ Pa		
オン	グストロ・	-4	Å	1 Å=0. 1nm=10 ⁻¹⁰ m		
バ	-	ン	b	$1 \text{ b}=100 \text{ fm}^2=10^{-28}\text{m}^2$		

表9. 固有の名称を含むCGS組立単位

表9. 固有の名称を含むUG3組立単位						
名称	記号	SI 単位であらわされる数値				
エルグダイン	erg	1 erg=10 ⁻⁷ J				
	dyn	$1  \text{dyn} = 10^{-5} \text{N}$				
ポアズ	Р	1 P=1 dyn • s/cm ² =0. 1Pa • s				
ストークス	St	1 St $=1 \text{ cm}^2/\text{s}=10^{-4}\text{m}^2/\text{s}$				
ガウス	G	$1 G = 10^{-4} T$				
エルステッド	0e	1 Oe 🚊 (1000/4π) A/m				
マクスウェル	Mx	$1 \text{ Mx} = 10^{-8} \text{Wb}$				
スチルブ	sb	$1 \text{ sb} = 1 \text{ cd/cm}^2 = 10^4 \text{ cd/m}^2$				
ホト	ph	$1 \text{ ph}=10^4 \text{lx}$				
ガル	Gal	1 Gal =1cm/s ² =10 ⁻² m/s ²				

表10. 国際単位に属さないその他の単位の例						
名称	記号	SI 単位であらわされる数値				
キュリー	Ci	1 Ci=3. $7 \times 10^{10}$ Bq				
レントゲン	R	$1 R = 2.58 \times 10^{-4} C/kg$				
ラ ド	rad	1 rad=1cGy=10 ⁻² Gy				
V 4	rem	1 rem=1 cSv=10 ⁻² Sv				
レ ム X 線 単 位		1X unit=1.002×10 ⁻⁴ nm				
ガンマ	γ	$1 \gamma = 1 nT = 10^{-9}T$				
ジャンスキー	Jy	$1 \text{ Jy}=10^{-26}\text{W} \cdot \text{m}^{-2} \cdot \text{Hz}^{-1}$				
フェルミ		1 fermi=1 fm=10 ⁻¹⁵ m				
メートル系カラット		1 metric carat = 200 mg = $2 \times 10^{-4}$ kg				
トル	Torr	1 Torr = (101 325/760) Pa				
標準大気圧	atm	1 atm = 101 325 Pa				
カロリー	cal					
ミクロン	μ	$1 \mu = 1 \mu m = 10^{-6} m$				

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