

Benchmark Calculations by the Nuclear  
Criticality Safety Analysis Code System

JACS(MGCL, KENO-IV)

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Benchmark Calculations by the Nuclear Criticality  
Safety Analysis Code System JACS(MGCL, KENO-IV)

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**Abstract**

Since 1980, as many as 1394 cases of benchmark calculations on criticality problems have been performed by the KENO-IV Monte Carlo calculation code with the MGCL cross section data library. The code system is a part of the criticality safety evaluation code system JACS developed at JAERI. The code validation results have been published in a series of JAERI-M reports and others. This report summarizes these results and the reliability of the code system systematically.

The number of the calculated cases briefly described in this report together with their experimental systems and data are 502 for 17 kinds of homogeneous single-unit systems, 331 for 8 kinds of homogeneous multi-unit systems and 561 for 16 kinds of heterogeneous systems.

Discussions and interpretations are made on the calculated  $k_{eff}$ 's (neutron multiplication factors) with their bias errors. The factors related to the bias errors are confirmed together with their causes and trends.

**Keywords:** Criticality Safety, Analysis Code System, Benchmark Calculation, Monte Carlo Code, Data Library, Code Validation, Homogeneous System, Heterogeneous System, Single Unit, Multi Unit, Bias Error, Neutron Multiplication Factor

## 臨界安全解析コードシステム JACS(MGCL, KENO-IV) によるベンチマーク計算

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### 要 旨

原研で開発された臨界安全解析コードシステム JACS のうち, KENO-IV モンテカルロ・コードと MGCL データ・ライブラリーの組合せによる計算の精度を検証するため, 1980 年以来多種多様な核燃料施設を模擬した臨界実験のデータを用いてベンチマーク計算が実施してきた。これらの結果は、その都度 JAERI-M レポートその他に発表されてきた。本報告は、これらの計算精度検証に関する成果について、既発表のものを含め、他の結果を追加し、総合的に評価し、まとめたものである。

記載した実験体系及び計算ケースは、均質單一ユニット体系の 17 種類 502 ケース、均質複数ユニット体系 8 種類 331 ケース及び非均質体系 16 種類 561 ケースに上り、それぞれ実験体系、実験データ及び計算結果について簡明に記述してある。また、 $k_{eff}$  計算値にバイアス誤差を生ずる各因子を摘出し、その原因、影響について考察した。

## CONTENTS

<b>1.</b>	<b>Introduction . . . . .</b>	<b>1</b>
<b>2.</b>	<b>Calculational Method . . . . .</b>	<b>2</b>
<b>2.1</b>	Outline of the Code System . . . . .	2
<b>2.2</b>	Calculational Method . . . . .	3
<b>3.</b>	<b>Experimental Data and Calculated Results . . . . .</b>	<b>8</b>
<b>A.</b>	Homogeneous Single-Unit System . . . . .	8
<b>3.A.1</b>	Pile of $\text{PuO}_2\text{-UO}_2\text{-Polystyrene}$ Compacts with Various Poison-Plate Materials . . . . .	8
<b>3.A.2</b>	Pile of $\text{PuO}_2\text{-UO}_2\text{-Polystyrene}$ Compacts with Various Plutonium Enrichments . . . . .	9
<b>3.A.3</b>	Pile of $\text{PuO}_2\text{-UO}_2\text{-Polythene}$ Compacts . . . . .	9
<b>3.A.4</b>	Pile of $\text{PuO}_2\text{-}(\text{Polystyrene})$ Compacts . . . . .	9
<b>3.A.5</b>	Single Tank Containing Uranyl-Nitrate Solution . . . . .	10
<b>3.A.6</b>	Plutonium Fuel in Various Shapes . . . . .	10
<b>3.A.7</b>	Intersecting Cylinders Containing Uranyl-Fluoride Solution . . . . .	11
<b>3.A.8</b>	Sphere or Cylinder Containing Uranyl-Fluoride Solution . . . . .	11
<b>3.A.9</b>	Cylinder Containing Uranyl-Fluoride Solution with Combination of Reflectors . . . . .	12
<b>3.A.10</b>	Pile of $\text{UF}_4\text{-Paraffin}$ Compacts . . . . .	12
<b>3.A.11</b>	Sphere Containing Plutonium-Nitrate Solution with Various Reflectors . . . . .	12
<b>3.A.12</b>	Annular Cylinder Containing Plutonium-Nitrate Solution . . . . .	12
<b>3.A.13</b>	Cylinder or Sphere Containing Plutonium-Uranyl-Nitrate Solution Reflected by Water . . . . .	13
<b>3.A.14</b>	Assembly of Low-Enriched Uranium-Oxide( $\text{U}_3\text{O}_8$ ) Compacts . . . . .	13
<b>3.A.15</b>	Cylinder Containing Plutonium-Nitrate Solution of High-Fissile Content Poisoned with Gadolinium . . . . .	14
<b>3.A.16</b>	Cylinder Containing Plutonium-Uranyl-Nitrate Solution with Soluble Neutron Poison . . . . .	14
<b>3.A.17</b>	Homogeneous Systems with Boron Containing Glass Raschig Rings . . . . .	14
<b>B.</b>	Homogeneous Multi-Unit(Interacting) System . . . . .	15
<b>3.B.1</b>	Array of Cylinders Containing $\text{UF}_6$ . . . . .	15
<b>3.B.2</b>	Slab-Cylinder Configurations of Uranyl-Nitrate Solution . . . . .	15
<b>3.B.3</b>	Array of Cylinders Containing Uranyl-Nitrate Solution . . . . .	16
<b>3.B.4</b>	Two Parallel Cylinders Containing Uranyl-Fluoride Solution . . . . .	16
<b>3.B.5</b>	Three-Dimensional Array of Uranium-Metal Cylinders . . . . .	16
<b>3.B.6</b>	Two- and Three-Dimensional Array of Plutonium Metal Cylinders . . . . .	17
<b>3.B.7</b>	Three-Dimensional Array of Cylinders Containing Uranyl-Nitrate Solution . . . . .	17
<b>3.B.8</b>	Array of Tin Cans Containing $\text{UO}_2$ or $\text{UO}_2\text{-Alcohol}$ Slurry . . . . .	17
<b>C.</b>	Heterogeneous System . . . . .	18

<b>3.C.1</b>	LWR Critical Assemblies with Various Lattice Pitches and Patterns .....	18
<b>3.C.2</b>	UO <sub>2</sub> Rod Clusters Partially Immersed in Water with Various Lattice Pitches and Patterns .....	18
<b>3.C.3</b>	UO <sub>2</sub> Rod Clusters Fully Immersed in Water at Near-Optimum Neutron Moderation with Various Poison Plates .....	19
<b>3.C.4</b>	UO <sub>2</sub> Rod Clusters Fully Immersed in Water at Near-Optimum Neutron Moderation with Reflecting Walls .....	19
<b>3.C.5</b>	UO <sub>2</sub> Rod Clusters Fully Immersed in Water at Under-Moderation with Various Poison Plates .....	20
<b>3.C.6</b>	UO <sub>2</sub> Rod Clusters Fully Immersed in Water at Under-Moderation with Reflecting Walls .....	21
<b>3.C.7</b>	UO <sub>2</sub> Rod Clusters Fully Immersed in Water at Various Pitches with Poison Plates and Reflecting Walls .....	21
<b>3.C.8</b>	Triangular Lattice of Plutonium Alloy or Mixed Oxide Fuel Moderated by Borated Water .....	22
<b>3.C.9</b>	Mixed Oxide Fuel Lattice Moderated by Borated Water .....	22
<b>3.C.10</b>	Water Storage of Power Reactor Fuel in Close Proximity .....	22
<b>3.C.11</b>	Array of UO <sub>2</sub> -BeO Fuel Pins in Uranyl-Nitrate Solution .....	23
<b>3.C.12</b>	UO <sub>2</sub> Rod Clusters Fully Immersed in Borated Water .....	23
<b>3.C.13</b>	UO <sub>2</sub> Rod Clusters in Sodium-Nitrate Solution .....	24
<b>3.C.14</b>	Triangular Lattice of UO <sub>2</sub> Fuel Moderated by Poisoned Water .....	24
<b>3.C.15</b>	UO <sub>2</sub> Rod Clusters Partially Immersed in Uranyl-Nitrate Solution .....	24
<b>3.C.16</b>	UO <sub>2</sub> -PuO <sub>2</sub> Rod Clusters Partially Immersed in Uranyl-Plutonium- Nitrate Solution with Soluble Poisons .....	24
<b>4.</b>	Discussions and Interpretations .....	140
<b>4.1</b>	Effect of Reflecting Condition .....	140
<b>4.2</b>	Effect of Moderating Condition .....	140
<b>4.3</b>	Effect of Soluble Neutron Absorber .....	141
<b>4.4</b>	Effect of Fuel Composition .....	141
<b>4.5</b>	Effect of Distance Between Units .....	141
<b>5.</b>	Conclusions .....	150
	References .....	151

## 目 次

1.はじめに .....	1
2.計算方法 .....	2
2.1 コードシステムの概要 .....	2
2.2 計算方法 .....	3
3.実験データと計算結果 .....	8
A. 均質単一ユニット体系 .....	8
3. A.1 種々の中性子毒物板つき $\text{PuO}_2$ - $\text{UO}_2$ -ポリスチレン・コンパクトのブロック積み .....	8
3. A.2 種々のプルトニウム富化度の $\text{PuO}_2$ - $\text{UO}_2$ -ポリスチレン・コンパクトのブロック積み .....	9
3. A.3 $\text{PuO}_2$ - $\text{UO}_2$ -ポリシン・コンパクトのブロック積み .....	9
3. A.4 $\text{PuO}_2$ (-ポリスチレン)・コンパクトのブロック積み .....	9
3. A.5 硝酸ウラニル水溶液入り單一タンク .....	10
3. A.6 各種形状のプルトニウム燃料 .....	10
3. A.7 弗化ウラニル水溶液入り交差円筒 .....	11
3. A.8 弗化ウラニル水溶液入り球及び円筒 .....	11
3. A.9 組合せ反射体つき弗化ウラニル水溶液入り円筒 .....	12
3. A.10 $\text{UF}_4$ -バラフィン・コンパクトのブロック積み .....	12
3. A.11 種々の反射体つき硝酸プルトニウム水溶液入り球 .....	12
3. A.12 硝酸プルトニウム水溶液入り円環 .....	12
3. A.13 水反射の硝酸プルトニウム・ウラニル水溶液入り円筒又は球 .....	13
3. A.14 低濃縮 $\text{U}_3\text{O}_8$ ・コンパクトの集合体 .....	13
3. A.15 高 fissile, ガトリニウム中性子毒を含む硝酸プルトニウム水溶液入り円筒 .....	14
3. A.16 可溶性中性子毒を含む硝酸プルトニウム・ウラニル水溶液入り円筒 .....	14
3. A.17 ボロンを含むガラス製ラシヒ・リングを充てんした均質系 .....	14
B. 均質複数ユニット(相互干渉)体系 .....	15
3. B.1 $\text{UF}_6$ 入り円筒の配列 .....	15
3. B.2 硝酸ウラニル水溶液入りスラブ・シリンダーの組合せ .....	15
3. B.3 硝酸ウラニル水溶液入り円筒の配列 .....	16
3. B.4 弗化ウラニル水溶液入り平行2円筒 .....	16
3. B.5 ウラン金属円柱の3次元配列 .....	16
3. B.6 プルトニウム金属円柱の2次元, 3次元配列 .....	17
3. B.7 硝酸ウラニル水溶液入り円筒の3次元配列 .....	17
3. B.8 $\text{UO}_2$ 及び $\text{UO}_2$ -アルコール入り錫缶の配列 .....	17
C. 非均質体系 .....	18
3. C.1 種々の格子ピッチ及びパターンの軽水炉臨界集合体 .....	18
3. C.2 種々の格子ピッチ及びパターンの半水没の $\text{UO}_2$ 燃料棒集合体 .....	18
3. C.3 種々の中性子毒物板つき最適減速に近い完全水没の $\text{UO}_2$ 燃料棒集合体 .....	19
3. C.4 反射壁つき最適減速に近い完全水没の $\text{UO}_2$ 燃料棒集合体 .....	19
3. C.5 種々の中性子毒物板つき減速不足状態の完全水没の $\text{UO}_2$ 燃料棒集合体 .....	20

3. C.6 反射壁つき減速不足状態の完全水没の UO <sub>2</sub> 燃料棒集合体	21
3. C.7 種々の格子ピッチの中性子毒物板及び反射壁つき 完全水没の UO <sub>2</sub> 燃料棒集合体	21
3. C.8 ボロンを含む水で減速されたプルトニウム金属あるいは 混合酸化物燃料棒の三角格子	22
3. C.9 ボロンを含む水で減速された混合酸化物燃料棒の格子	22
3. C.10 動力炉燃料の水内近接貯蔵	22
3. C.11 硝酸ウラニル水溶液中の UO <sub>2</sub> -BeO 燃料棒の配列	23
3. C.12 ボロンを含む水中に完全に沈められた UO <sub>2</sub> 燃料棒集合体	23
3. C.13 硝酸ナトリウム水溶液中の UO <sub>2</sub> 燃料棒集合体	24
3. C.14 中性子毒を含む水で減速された UO <sub>2</sub> 燃料棒の三角格子	24
3. C.15 硝酸ウラニル水溶液中に一部を没した UO <sub>2</sub> 燃料棒集合体	24
3. C.16 硝酸プルトニウム・ウラニル水溶液中に一部を没した UO <sub>2</sub> -PuO <sub>2</sub> 燃料棒集合体	24
<b>4. 検討</b>	140
4.1 反射条件の影響	140
4.2 減速条件の影響	140
4.3 可溶性中性子毒の影響	141
4.4 燃料組成の影響	141
4.5 ユニット間の距離の影響	141
<b>5. 結論</b>	150
参考文献	151

## List of Tables and Figures

### **List of Tables**

Table 2.1.1	Energy group structure of MGCL
Table 3.A.1.1	Experimentally determined critical heights for 30.6 H/(Pu+U) fuel systems
Table 3.A.1.2	Experimentally determined critical heights for 2.8 H/(Pu+U) fuel (partly replaced with 30.6 H/(Pu+U) fuel) systems
Table 3.A.1.3	Calculated $k_{eff}$ 's for 30.6 H/(Pu+U) fuel systems
Table 3.A.1.4	Calculated $k_{eff}$ 's for 2.8 H/(Pu+U) fuel (partly replaced with 30.6 H/(Pu+U) fuel) systems
Table 3.A.2.1	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> -Polystyrene fuel compacts of 29.3 wt% Pu and 2.8 H/(Pu+U)
Table 3.A.2.2	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> -Polystyrene fuel compacts of 15.0 wt% Pu and 2.86 H/(Pu+U)
Table 3.A.2.3	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> -Polystyrene fuel compacts of 8.1 wt% Pu and 7.3 H/(Pu+U)
Table 3.A.2.4	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> -Polystyrene fuel compacts of 30.0 wt% Pu and 47.7 H/(Pu+U)
Table 3.A.2.5	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> -Polystyrene fuel compacts of 14.62 wt% Pu and 30.6 H/(Pu+U)
Table 3.A.2.6	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> -Polystyrene fuel compacts of 7.89 wt% Pu and 51.8 H/(Pu+U)
Table 3.A.3.1	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> -Polythene fuel compacts of 23.1 atomic plutonium enrichment percent and 4.67 H/(Pu+U) atomic ratio
Table 3.A.4.1	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -Polystyrene fuel compacts of 2.2 wt% <sup>240</sup> Pu and those of 8.0 wt% <sup>240</sup> Pu
Table 3.A.4.2	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> -Polystyrene fuel compacts of 11.46 wt% <sup>240</sup> Pu and 5.0 H/Pu
Table 3.A.4.3	Critical assembly configurations and calculated $k_{eff}$ 's for PuO <sub>2</sub> fuel compacts of 18.35 wt% <sup>240</sup> Pu and 0.04 H/Pu
Table 3.A.5.1	Calculated $k_{eff}$ 's for the minimally reflected single tank
Table 3.A.5.2	Calculated $k_{eff}$ 's for the concrete-reflected single tank
Table 3.A.5.3	Calculated $k_{eff}$ 's for the plastic-reflected single tank
Table 3.A.6.1	Summary of characteristics of homogeneous plutonium benchmark experiments
Table 3.A.6.2	Calculated results for homogeneous plutonium benchmark experiments
Table 3.A.7.1	Calculated $k_{eff}$ 's for "30 degrees lateral" system
Table 3.A.7.2	Calculated $k_{eff}$ 's for "cross" system
Table 3.A.8.1	Calculated results with experimental conditions for cylinders or spheres containing uranyl-fluoride solution
Table 3.A.9.1	Calculated results with experimental conditions for cylinders containing uranyl-fluoride solution with combination of reflectors
Table 3.A.10.1	Critical dimensions of piles of the UF <sub>4</sub> -paraffin compact fuel

Table 3.A.10.2	Calculated results for piles of the $\text{UF}_4$ -paraffin compact fuel
Table 3.A.11.1	Experimental configurations for spherical vessels containing uranyl-nitrate solution with various reflectors
Table 3.A.11.2	Calculated results for spherical vessels containing uranyl-nitrate solution with various reflectors
Table 3.A.12.1	Calculated results with experimental conditions for annular cylinders containing plutonium-nitrate solution with inner void region
Table 3.A.12.2	Calculated results with experimental conditions for annular cylinders containing plutonium-nitrate solution with inner cadmium-paraffin region
Table 3.A.13.1	Calculated results with experimental conditions for cylinders containing plutonium-nitrate solution of 5.6 wt% $^{240}\text{Pu}$ in Pu
Table 3.A.13.2	Calculated results with experimental conditions for spheres containing plutonium-uranyl-nitrate solution of 4.7 wt% $^{240}\text{Pu}$ in Pu
Table 3.A.13.3	Calculated results with experimental conditions for cylinders containing plutonium-uranyl-nitrate solution of 23.0 wt% $^{240}\text{Pu}$ in Pu
Table 3.A.14.1	Principal critical parameters and calculated $k_{eff}$ 's for $\text{U}_3\text{O}_8$ package arrays
Table 3.A.15.1	Principal critical parameters and calculated $k_{eff}$ 's
Table 3.A.16.1	Experimental condition for cylinders containing plutonium-uranyl-nitrate solution with soluble neutron absorbers
Table 3.A.16.2	Calculated results for cylinders containing plutonium-uranyl-nitrate solution with soluble neutron absorbers
Table 3.A.17.1	Calculated $k_{eff}$ 's with experimental conditions for raschig rings in plutonium-uranyl solution
Table 3.B.1.1	Calculated $k_{eff}$ 's for unreflected $3 \times 4$ arrays
Table 3.B.1.2	Calculated $k_{eff}$ 's for unreflected $4 \times 4$ arrays
Table 3.B.1.3	Calculated $k_{eff}$ 's for $2 \times 2$ arrays reflected on 5 sides by polyethylene and moderated by Plexiglas
Table 3.B.1.4	Calculated $k_{eff}$ 's for $3 \times 3$ arrays reflected on 5 sides by polyethylene
Table 3.B.1.5	Calculated $k_{eff}$ 's for $4 \times 4$ arrays with various polyethylene reflector conditions
Table 3.B.1.6	Calculated $k_{eff}$ 's for $2 \times 2$ arrays reflected on 5 sides by concrete and complete reflection
Table 3.B.1.7	Calculated $k_{eff}$ 's for $2 \times 2$ arrays with various concrete reflector conditions
Table 3.B.1.8	Calculated $k_{eff}$ 's for $3 \times 3$ arrays with various concrete reflector conditions
Table 3.B.1.9	Calculated $k_{eff}$ 's for $1 \times 12$ arrays with various concrete reflector conditions
Table 3.B.1.10	Calculated $k_{eff}$ 's for $1 \times 5$ arrays with various concrete reflector conditions
Table 3.B.1.11	Calculated $k_{eff}$ 's for $1 \times 4$ arrays with various concrete reflector conditions
Table 3.B.1.12	Calculated $k_{eff}$ 's for $1 \times 3$ arrays with various reflector conditions and with Plexiglas separating the cylinders
Table 3.B.1.13	Calculated $k_{eff}$ 's for two linear three-cylinder arrays with close fitting concrete reflectors

Table 3.B.1.14	Calculated $k_{eff}$ 's for two linear three-cylinder arrays with concrete reflectors
Table 3.B.1.15	Calculated $k_{eff}$ 's for two linear three-cylinder arrays with various reflector conditions
Table 3.B.2.1	Experimental data and calculated $k_{eff}$ 's for the critical slab systems
Table 3.B.2.2	Experimental data and calculated $k_{eff}$ 's for the critical slab-cylinder array configuration with the minimum reflector
Table 3.B.2.3	Experimental data and calculated $k_{eff}$ 's for the critical slab-cylinder configuration with the array suspended above the slab
Table 3.B.2.4	Experimental data and calculated $k_{eff}$ 's for the critical slab-cylinder configuration with Plexiglas reflectors
Table 3.B.3.1	Calculated $k_{eff}$ 's for the array of cylinders reflected by concrete
Table 3.B.3.2	Calculated $k_{eff}$ 's for the array of cylinders reflected by plastic
Table 3.B.4.1	Calculated results for two aluminum cylindrical reactors
Table 3.B.5.1	Calculated results for cubic arrays of uranium metal cylinders without reflectors
Table 3.B.5.2	Calculated results for cubic arrays of uranium metal cylinders with reflectors
Table 3.B.6.1	Calculated results for arrays of plutonium metal billets
Table 3.B.7.1	Calculated results for cubic arrays of cylinders containing uranyl-nitrate solution with reflectors of various thicknesses
Table 3.B.7.2	Calculated results for cubic arrays of cylinders containing uranyl-nitrate solution with reflectors of various thicknesses on five sides and a 15.24 cm thick paraffin reflector on the bottom
Table 3.B.8.1	Description of parameters of U(93.15)O <sub>2</sub> container units
Table 3.B.8.2	Description of cells and experimental multiplication factors for U(93)O <sub>2</sub> moderated and unmoderated
Table 3.B.8.3	Principal critical parameters and calculated $k_{eff}$ 's
Table 3.C.1.1	Calculated results for U(2.6)O <sub>2</sub> -H <sub>2</sub> O systems
Table 3.C.1.2	Calculated results for PuO <sub>2</sub> -UO <sub>2</sub> -H <sub>2</sub> O systems
Table 3.C.1.3	Calculated results for the water gap pattern
Table 3.C.1.4	Calculated results for UO <sub>2</sub> lattices with boric acid solution
Table 3.C.1.5	Calculated results for UO <sub>2</sub> lattices with the poison sheet
Table 3.C.2.1	Calculated results and experimental conditions for UO <sub>2</sub> rod clusters partially immersed in water without removal of rods
Table 3.C.2.2	Calculated results and experimental conditions for UO <sub>2</sub> rod clusters with systematic removal of rods
Table 3.C.2.3	Calculated results and experimental conditions for UO <sub>2</sub> rod clusters with removal of rods in a small region
Table 3.C.3.1	Calculated results for UO <sub>2</sub> rod clusters ( <sup>235</sup> U enrichment: 2.35 wt%, lattice pitch: 20.32 mm) submerged in water with various poison plates
Table 3.C.3.2	Calculated results for UO <sub>2</sub> rod clusters ( <sup>235</sup> U enrichment: 4.31 wt%, lattice pitch: 25.40 mm) submerged in water with various poison plates
Table 3.C.4.1	Calculated results for UO <sub>2</sub> rod clusters submerged in water with reflecting walls
Table 3.C.5.1	Calculated results and critical experimental data for three lined rod clusters in water with various poison plates
Table 3.C.5.2	Calculated results and critical experimental data for four cross-arranged

	rod clusters in water with various poison plates
Table 3.C.6.1	Calculated results and critical experimental data for three aligned UO <sub>2</sub> rod clusters in water with side reflectors
Table 3.C.7.1	Calculated results and critical experimental data for three UO <sub>2</sub> rod clusters in water with absorbers and reflecting walls
Table 3.C.8.1	Summary of characteristics of heterogeneous plutonium benchmark experiments
Table 3.C.8.2	Calculated results for heterogeneous plutonium benchmark experiments
Table 3.C.9.1	Principal critical parameters and calculated results
Table 3.C.10.1	Measured and calculated $k_{eff}$ values
Table 3.C.10.2	Comparison of the group averaged $k_{eff}$ 's for different spacings between assemblies
Table 3.C.11.1	Principal critical parameters and calculated $k_{eff}$
Table 3.C.11.2	Principal critical parameters and calculated $k_{eff}$
Table 3.C.12.1	Critical experimental data for UO <sub>2</sub> rod clusters fully immersed in borated water
Table 3.C.12.2	Calculated $k_{eff}$ 's with different Monte Carlo calculational conditions
Table 3.C.13.1	Calculated results for UO <sub>2</sub> fuel rods in NaNO <sub>3</sub> solution
Table 3.C.14.1	Calculated results with experimental conditions for each case of triangular lattices in poisonous water
Table 3.C.15.1	Calculated results with experimental conditions for UO <sub>2</sub> rod clusters partially immersed in uranyl-nitrate solution
Table 3.C.16.1	Calculated results with experimental conditions for mixed-oxide fuel rods in plutonium-uranyl-nitrate solution containing gadolinium and boron
Table 4.1	Validation results for reflected simple geometry systems with the (MGCL, KENO-IV) code system

## List of Figures

Fig. 2.1.1	General calculation flow of JACS
Fig. 2.1.2	General flow diagram of a computer code system MGCL-ACE
Fig. 3.A.1.1	Configuration of PuO <sub>2</sub> -UO <sub>2</sub> polystyrene compacts
Fig. 3.A.1.2	Calculational model for PuO <sub>2</sub> -UO <sub>2</sub> polystyrene compacts
Fig. 3.A.1.3	Histogram of calculated $k_{eff}$ 's for the two different fuel systems
Fig. 3.A.2.1	Calculational model for PuO <sub>2</sub> -UO <sub>2</sub> polystyrene fuel compacts
Fig. 3.A.2.2	Histogram of calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> polystyrene fuel compacts of low (2–8)H/(Pu+U) atomic ratio (17 cases)
Fig. 3.A.2.3	Histogram of calculated $k_{eff}$ 's for PuO <sub>2</sub> -UO <sub>2</sub> polystyrene fuel compacts of high (30–50)H/(Pu+U) atomic ratio (35 cases)
Fig. 3.A.4.1	Histogram of calculated $k_{eff}$ 's for PuO <sub>2</sub> polystyrene fuel compacts
Fig. 3.A.5.1	Calculational model of a reflected single tank
Fig. 3.A.5.2	Histogram of calculated $k_{eff}$ 's for a single tank (45 cases)
Fig. 3.A.6.1	Calculational model of B.M. No.2
Fig. 3.A.6.2	Calculational model of B.M. No.3
Fig. 3.A.6.3	Calculational model of B.M. No.4
Fig. 3.A.6.4	Calculational model of B.M. No.5
Fig. 3.A.6.5	Calculational model of B.M. No.6 and No.13

- Fig. 3.A.6.6 Calculational model of B.M. No.8  
Fig. 3.A.6.7 Calculational model of B.M. No.9  
Fig. 3.A.6.8 Calculational model of B.M. No.11  
Fig. 3.A.6.9 Calculational model of B.M. No.10 and No.14  
Fig. 3.A.6.10 Calculational model of B.M. No.12  
Fig. 3.A.6.11 Decreasing trend of calculated  $k_{eff}$  with increasing H/Pu  
Fig. 3.A.7.1 Cut-out view of the experimental arrangement of "30 degrees lateral"  
Fig. 3.A.7.2 Cut-out view of the experimental arrangement of "cross"  
Fig. 3.A.7.3 Calculational model for "30 degrees lateral"  
Fig. 3.A.7.4 Calculational model for "cross"  
Fig. 3.A.7.5 Histogram of calculated  $k_{eff}$ 's for "30 degrees lateral" systems (18 cases)  
Fig. 3.A.7.6 Histogram of calculated  $k_{eff}$ 's for "cross" systems (27 cases)  
Fig. 3.A.8.1 Calculational model for a typical water-reflected cylinder containing uranyl-fluoride solution  
Fig. 3.A.8.2 Calculational model for a typical unreflected cylinder containing uranyl-fluoride solution  
Fig. 3.A.8.3 Histogram of calculated  $k_{eff}$ 's for cylinders or spheres containing uranyl-fluoride solution  
Fig. 3.A.9.1 Calculational model for a typical experimental case of U(4.98)O<sub>2</sub>F<sub>2</sub> solution, cylinder surrounded by composite reflector of steel/water  
Fig. 3.A.9.2 Histogram of calculated  $k_{eff}$ 's for cylinders containing uranyl-fluoride solution surrounded by composite reflector  
Fig. 3.A.10.1 Calculational model for a typical reflected pile of UF<sub>4</sub>-paraffin compact fuel  
Fig. 3.A.10.2 Calculational model for a typical unreflected pile of UF<sub>4</sub>-paraffin compact fuel  
Fig. 3.A.10.3 Histogram of calculated  $k_{eff}$ 's for reflected pile of UF<sub>4</sub>-paraffin compact fuel  
Fig. 3.A.10.4 Histogram of calculated  $k_{eff}$ 's for unreflected pile of UF<sub>4</sub>-paraffin compact fuel  
Fig. 3.A.11.1 Calculational model for a typical experimental system, case No.1  
Fig. 3.A.11.2 Histogram of calculated  $k_{eff}$ 's for 48 spherical vessels containing uranyl-nitrate solution with various reflectors  
Fig. 3.A.12.1 Experimental arrangement of annular cylinders containing plutonium-nitrate solution  
Fig. 3.A.12.2 Calculational model for a typical case of annular cylinder containing plutonium-nitrate solution  
Fig. 3.A.12.3 Histogram of calculated  $k_{eff}$ 's for annular cylinders containing plutonium-nitrate solution with or without inner cadmium/paraffin  
Fig. 3.A.13.1 Experimental set-up for cylinder containing plutonium-uranyl nitrate solution  
Fig. 3.A.13.2 Experimental set-up for sphere containing plutonium-uranyl nitrate solution  
Fig. 3.A.13.3 Calculational model for a typical cylinder containing plutonium-uranyl nitrate solution  
Fig. 3.A.13.4 Histogram of calculated  $k_{eff}$ 's for cylinders containing (Pu+U) nitrate solution of 5.6 wt% <sup>240</sup>Pu in Pu  
Fig. 3.A.13.5 Histogram of calculated  $k_{eff}$ 's for cylinders containing (Pu+U) nitrate

- solution of 23.0 wt%  $^{240}\text{Pu}$  in Pu
- Fig. 3.A.14.1 Experimental arrangement of the array of compacted low-enriched uranium-oxide packages
- Fig. 3.A.14.2 Histogram of calculated  $k_{eff}$ 's for  $\text{U}_3\text{O}_8$  package arrays
- Fig. 3.A.15.1 Calculational model for plutonium-nitrate solution with gadolinium poison in a cylindrical vessel
- Fig. 3.A.15.2 Histogram of calculated  $k_{eff}$ 's for plutonium-nitrate solution with gadolinium
- Fig. 3.A.16.1 Calculational model for a cylinder containing (U+Pu) nitrate solution with soluble neutron absorbers
- Fig. 3.A.16.2 Histogram of 34 calculated  $k_{eff}$ 's for cylinders containing plutonium-uranyl nitrate solution with soluble neutron absorbers
- Fig. 3.A.17.1 Schema of experimental set up for raschig rings in uranium-plutonium-nitrate solution
- Fig. 3.A.17.2 Calculational model for raschig rings in uranium-plutonium-nitrate solution
- Fig. 3.B.1.1 Calculational models of a  $\text{UF}_6$  cylinder and its typical arrangement
- Fig. 3.B.1.2 Histogram of calculated  $k_{eff}$ 's for the unreflected  $\text{UF}_6$  cylinder systems (16 cases)
- Fig. 3.B.1.3 Histogram of calculated  $k_{eff}$ 's for the reflected  $\text{UF}_6$  cylinder systems (51 cases)
- Fig. 3.B.2.1 Typical reflected experimental configuration
- Fig. 3.B.2.2 Calculational model for a reflected  $4 \times 4$  cylinder array
- Fig. 3.B.2.3 Histogram of calculated multiplication factor ( $k_{eff}$ ). Calculated  $k_{eff}$ 's are divided into two groups; one has the values lower than 0.965 corresponding to the unreflected systems (70 cases); the other has the values greater than 0.965 corresponding to the reflected systems (26 cases)
- Fig. 3.B.3.1 Calculational model of reflected sleeved cylinders
- Fig. 3.B.3.2 Histogram of calculated  $k_{eff}$ 's for the reflected array of cylinders (31 cases)
- Fig. 3.B.4.1 Calculational model for two aluminum reactors without water reflector
- Fig. 3.B.4.2 Calculational model for two aluminum reactors with water reflector
- Fig. 3.B.4.3 Histogram of calculated  $k_{eff}$ 's for two aluminum reactors (29 cases)
- Fig. 3.B.5.1 Calculational model for a  $2 \times 2 \times 2$  array of uranium-metal cylinders with paraffin reflector
- Fig. 3.B.5.2 Histogram of calculated  $k_{eff}$ 's for cubic arrays of uranium-metal cylinders (31 cases)
- Fig. 3.B.6.1 Calculational model for a 3 kg plutonium-metal billet
- Fig. 3.B.6.2 Calculational model for a 6 kg plutonium-metal billet
- Fig. 3.B.6.3 Calculational model for a  $2 \times 2 \times 2$  array of plutonium-metal billets unmoderated with partial reflection
- Fig. 3.B.6.4 Calculational model for a  $2 \times 2 \times 2$  array of plutonium-metal billets moderated without reflectors
- Fig. 3.B.6.5 Histogram of calculated  $k_{eff}$ 's for arrays of plutonium (19 cases)
- Fig. 3.B.7.1 Calculational model for a  $2 \times 2 \times 2$  array of  $\text{UO}_2(\text{NO}_3)_2$  aqueous solution with reflectors
- Fig. 3.B.7.2 Histogram of calculated  $k_{eff}$ 's for cubic arrays of uranyl-nitrate cylinders

- (29 cases)
- Fig. 3.B.8.1 Modeling of fuel unit
- Fig. 3.B.8.2 Example of calculational model for array
- Fig. 3.C.1.1 Vertical cross-sectional view of core tank
- Fig. 3.C.1.2 2.6 wt% UO<sub>2</sub> and 3.0 wt% PuO<sub>2</sub>-natural UO<sub>2</sub> fuel rods
- Fig. 3.C.1.3 Calculational model for UO<sub>2</sub> fuel lattice
- Fig. 3.C.1.4 Histogram of calculated  $k_{eff}$ 's for U(2.6)O<sub>2</sub>-H<sub>2</sub>O systems
- Fig. 3.C.1.5 Histogram of  $k_{eff}$ 's for PuO<sub>2</sub>-U(Natural)O<sub>2</sub>-H<sub>2</sub>O systems
- Fig. 3.C.1.6 Histogram of  $k_{eff}$ 's for U(2.6)O<sub>2</sub>-poison sheet-H<sub>2</sub>O systems
- Fig. 3.C.2.1 Calculational models for representative cases of UO<sub>2</sub> rod hexagonal lattices.  
Solid circle indicates the removal of rods
- Fig. 3.C.2.2 Calculational models for representative cases of UO<sub>2</sub> rod square lattices.  
Solid circle indicates the removal of rods
- Fig. 3.C.2.3 Experimental arrangement of UO<sub>2</sub> rod clusters partially immersed in water
- Fig. 3.C.2.4 Histogram of calculated  $k_{eff}$ 's for UO<sub>2</sub> rod clusters partially immersed in water (26 cases)
- Fig. 3.C.3.1 4.29 wt% <sup>235</sup>U enriched UO<sub>2</sub> rods
- Fig. 3.C.3.2 2.35 wt% <sup>235</sup>U enriched UO<sub>2</sub> rods
- Fig. 3.C.3.3 Experimental arrangement of simulated shipping containers with poison plates
- Fig. 3.C.3.4 Calculational model for UO<sub>2</sub> lattice cell  
a) 2.35 wt% <sup>235</sup>U fuel rod cell model  
b) 4.31 wt% <sup>235</sup>U fuel rod cell model
- Fig. 3.C.3.5 A typical KENO-IV calculational model for UO<sub>2</sub> rod clusters submerged in water with boral poison plates
- Fig. 3.C.3.6 Histogram of calculated  $k_{eff}$ 's for UO<sub>2</sub> rod clusters (<sup>235</sup>U enrichment: 4.31 wt%, lattice pitch: 25.40 mm) in water with various poison plates (31 cases)
- Fig. 3.C.3.7 Histogram of calculated  $k_{eff}$ 's for UO<sub>2</sub> rod clusters (<sup>235</sup>U enrichment: 2.35 wt%, lattice pitch: 20.32 mm) in water with various poison plates (44 cases)
- Fig. 3.C.4.1 Graphical arrangement of simulated shipping container critical experiments
- Fig. 3.C.4.2 A typical KENO-IV calculational model for UO<sub>2</sub> rod clusters submerged in water with reflecting walls
- Fig. 3.C.4.3 Histogram of calculated  $k_{eff}$ 's for UO<sub>2</sub> rod clusters submerged in water with reflecting walls (23 cases)
- Fig. 3.C.5.1 Bird's eye view of typical four cross arranged clusters in water pool
- Fig. 3.C.5.2 Typical calculational model for four cross arranged clusters (case No.100)
- Fig. 3.C.5.3 Histogram of calculated  $k_{eff}$ 's for three lined clusters (18 cases)
- Fig. 3.C.5.4 Histogram of calculated  $k_{eff}$ 's for four cross arranged clusters (92 cases)
- Fig. 3.C.6.1 Histogram of calculated  $k_{eff}$ 's for three aligned UO<sub>2</sub> rod clusters in water with side reflectors
- Fig. 3.C.7.1 Experimental arrangement of three UO<sub>2</sub> rod clusters in water with absorbers and reflecting walls

- Fig. 3.C.7.2      Calculational model for three  $\text{UO}_2$  rod clusters in water with absorbers and reflecting walls
- Fig. 3.C.7.3      Histogram of calculated  $k_{eff}$ 's for three  $\text{UO}_2$  rod clusters ( $^{235}\text{U}$  enrichment: 2.35 wt%) in water with neutron absorbers and reflecting walls (25 cases)
- Fig. 3.C.7.4      Histogram of calculated  $k_{eff}$ 's for three  $\text{UO}_2$  rod clusters ( $^{235}\text{U}$  enrichment: 4.31 wt%) in water with neutron absorbers and reflecting walls (24 cases)
- Fig. 3.C.8.1      Calculational model for B.M. No.26,  $N_c = 288$
- Fig. 3.C.8.2      Calculational model for B.M. No.27,  $N_c = 320$
- Fig. 3.C.8.3      Histogram of calculated  $k_{eff}$ 's for the heterogeneous plutonium benchmark experiments (20 cases)
- Fig. 3.C.9.1      Example of fuel pins loading pattern
- Fig. 3.C.9.2       $\text{UO}_2$ -2.35 wt%  $^{235}\text{U}$  fuel description
- Fig. 3.C.9.3       $\text{UO}_2$ -2 wt%  $\text{PuO}_2$  (8%  $^{240}\text{Pu}$ ) fuel description
- Fig. 3.C.10.1     An example of core loading patterns, nine arrays separated by one pin pitch with 84  $\text{B}_4\text{C}$  pins
- Fig. 3.C.10.2     KENO calculational model for Core I-IX
- Fig. 3.C.10.3     KENO calculational model for Core X-XXI
- Fig. 3.C.11.1     Lattices of EBOR (Experimental Beryllium Oxide Reactor) fuel pins in water
- Fig. 3.C.11.2     Schema of the fuel pin model
- Fig. 3.C.11.3     Geometrical model of fuel pins in water
- Fig. 3.C.11.4     Geometrical model of fuel pins in solution
- Fig. 3.C.12.1     Schema of the fuel rod used in the experiment
- Fig. 3.C.12.2     Experimental arrangement of the  $\text{UO}_2$  rod clusters immersed in borated water
- Fig. 3.C.12.3     Calculational model for  $\text{UO}_2$  rod cluster fully immersed in borated water
- Fig. 3.C.12.4     Histogram of calculated  $k_{eff}$ 's for the  $\text{UO}_2$  rod clusters fully immersed in borated water (1)
- Fig. 3.C.12.5     Histogram of calculated  $k_{eff}$ 's for the  $\text{UO}_2$  rod clusters fully immersed in borated water (2)
- Fig. 3.C.13.1     Main dimensional figures of the fuel rods and the experimental basket (in millimeters)
- Fig. 3.C.13.2     Calculational model for a fuel rod
- Fig. 3.C.13.3     Calculational model for a typical experimental system
- Fig. 3.C.14.1     Experimental arrangement of the triangular lattice in poisonous water tank
- Fig. 3.C.14.2     Calculational model for  $\text{UO}_2$  fuel rods, triangularly arranged in poisonous water tank
- Fig. 3.C.14.3     Histogram of calculated  $k_{eff}$ 's for triangular lattice in poisonous water
- Fig. 3.C.15.1     Experimental arrangement for  $\text{UO}_2$  rod clusters partially immersed in uranyl-nitrate solution
- Fig. 3.C.15.2     Calculational model for  $\text{UO}_2$  rod clusters partially immersed in uranyl-nitrate solution
- Fig. 3.C.15.3     Histogram of 12 calculated  $k_{eff}$ 's for rod clusters partially immersed in uranyl-nitrate solution

- Fig. 3.C.16.1 Experimental arrangement for  $\text{UO}_2\text{-PuO}_2$  fuel rods in fissile solution containing gadolinium and boron
- Fig. 3.C.16.2 Calculational model for  $\text{PuO}_2\text{-}\text{UO}_2$  fuel rods in plutonium-uranyl-nitrate solution containing gadolinium and boron
- Fig. 3.C.16.3 Histogram of calculated  $k_{eff}$ 's for  $\text{PuO}_2\text{-}\text{UO}_2$  fuel rods in plutonium-uranyl-nitrate solution containing gadolinium and boron
- Fig. 4.1 Variation of calculated  $k_{eff}$  with reflector thickness. Closed circles indicate the change of  $k_{eff}$  with the top and side reflector thickness. Bottom reflector thickness is fixed to be 10.2 cm  
Open circles indicate the change of  $k_{eff}$  with the bottom reflector thickness. There are no reflectors on the top and the side of the experimental apparatus
- Fig. 4.2 Calculated  $k_{eff}$  as a function of reflector thickness surrounding the array of uranyl-nitrate solution cylinders
- Fig. 4.3 Variation of  $k_{eff}$  with center separation of fuel rods in water in the experiment (3.C.11)
- Fig. 4.4 Variation of calculated  $k_{eff}$  bias with gadolinium concentration in the experiment (3.A.16)
- Fig. 4.5 Variation of calculated  $k_{eff}$  bias with boron concentration in the experiment (3.A.16)
- Fig. 4.6 Variation of calculated  $k_{eff}$  bias with Ft/At ratio in the experiment (3.C.14)
- Fig. 4.7 Variation of calculated  $k_{eff}$  with boron concentration in the experiment (3.C.8)
- Fig. 4.8 Variation of calculated  $k_{eff}$  with boron concentration in the experiment (3.C.10)
- Fig. 4.9 Calculated  $k_{eff}$  as a function of center-to-center separation of plutonium metal cylinders in array disposition  
Array size :  $4^3$   
Unit : 6 kg Pu billet  
Moderator : None  
Reflector : None
- Fig. 4.10 Calculated  $k_{eff}$  as a function of surface separation of uranyl-nitrate solution cylinders in array disposition  
Array size :  $2^3 - 5^3$   
Unit : Uranyl-nitrate solution  
Moderator : None  
Reflector : None
- Fig. 4.11 Calculated  $k_{eff}$  v.s. surface separation of two aluminum cylinders containing uranyl-fluoride solution

## 1. Introduction

A computer code system named JACS<sup>1)</sup> has been developed at JAERI to evaluate criticality safety of nuclear fuel cycle facilities such as fuel fabrication plants, reprocessing plants, transportation casks, storage pools and so on.

Since physical and chemical properties of fissile material appearing in nuclear facilities cover so many varieties that calculated values of  $k_{eff}$  are supposed to have a wide range of bias errors according to the properties of fissile material:

$$\Delta k_{eff} = k_{eff} \text{ (calculated neutron multiplication factor)} \\ - k_{eff, true} \text{ (true neutron multiplication factor)}$$

with their particular trends and uncertainties. In order to obtain the bias error mentioned above, in other words, the bias correction to the calculated  $k_{eff}$  for any kind of nuclear facility, it is considered appropriate to assess the accuracy of the code system with experimental data of as many variety as possible. The bias corrections are determined by:

$$\Delta k_{eff, exp} = k_{eff, exp} \text{ (calculated neutron multiplication factor for an experimental system)} \\ - 1.0 \text{ (true neutron multiplication factor for the experiment).}$$

$\Delta k_{eff}$  for any system is, then, derived from analysing  $\Delta k_{eff, exp}$  through some mathematical treatment.

At JAERI, as a part of the development of the JACS code system, some 1400 cases of benchmark calculations of criticality problems for the suitably devised experimental systems simulating various nuclear facilities have been made with the Monte Carlo code KENO-IV and the multigroup constant library MGCL.

Some results of the benchmark calculations have been already published separately in several JAERI-M reports. The calculated results published in each report, however, are concerned with some restricted area of experimental data. The present report has gathered all the available benchmark calculations including those published up to now, and discusses the reliability of the code system systematically. Referring to these discussions, one can supposedly give a reasonable bias correction to the calculated  $k_{eff}$  in his criticality analysis for a nuclear fuel cycle facility.

As for the contents of this report, outline of the JACS code system and calculational method adopted in this report are given in Chapter 2. In Chapter 3, three subchapters are provided. Subchapter A is for homogeneous single-unit systems, Subchapter B for homogeneous multi-unit systems and Subchapter C for heterogeneous systems. Each subchapter is then divided to several sections. Each section deals with experiments and calculations considered to be grouped in the same kind of benchmark calculations. Discussions and interpretations of the calculated results are given in Chapter 4.

## 2. Calculational Method

### 2.1 Outline of the Code System

General flow diagram of the JACS code system for nuclear criticality safety evaluation is shown in **Fig. 2.1.1**. The multigroup constant library MGCL<sup>2)</sup> has been generated from the evaluated nuclear data file ENDF/B-IV<sup>3)</sup> or JENDL (Japanese Evaluated Nuclear Data Library)<sup>4)</sup>. The MGCL Library is the same type of library as the ABBN set<sup>5)</sup> or the JAERI-FAST set<sup>6)</sup>, consisting of one- and two-dimensional cross section data with a Bondarenco type shielding table. One-dimensional data such as  $\sigma_t$ ,  $\sigma_a$ ,  $\sigma_f$ ,  $\nu$ ,  $\sigma_e$ ,  $\sigma_{in}$ ,  $\sigma_{n,2n}$  has been tabulated as infinite dilution cross sections and self-shielding factors. Two-dimensional data are matrices of elastic and inelastic scattering kernels and  $(n, 2n)$  reaction cross section data. Angular dependent elastic scattering kernels are tabulated as Legendre Expansion coefficients. These scattering matrices are obtained by utilizing the SUPERTOG code<sup>7)</sup> and the PIXSE code<sup>8)</sup>. In these matrices, up-scattering of thermal neutrons is considered for energies up to 1.855 ev.

General flow diagram of the MGCL library production code system is shown in **Fig. 2.1.2**. Temperature dependent ultra fine cross sections (some 70,000 energy point data) are produced by the RESEND-D code developed at JAERI from the RESEND code<sup>9)</sup>. To collapse the energy groups of the ultra fine data points, the following weighting function is provided.

$$\phi(E) = \phi_s / (\sigma_t^i + \sigma_o^i),$$

here,  $\phi_s$  is a standard neutron-energy spectrum consisting of fission spectrum,  $1/E$  and Maxwellian functions, and

$$\sigma_o^i = (\text{Sum}_{j \neq i} N^j \sigma_t^j + \alpha G/l) / N^i,$$

$N^i$  : Atomic number density of the nuclide  $i$ ,

$l$  : Mean cord length of the fuel lump,

$G$  : Dancoff correction factor,

$\alpha$  : Bell factor.

The 137 energy group structure of the MGCL master library is shown in **Table 2.1.1**. The number of the energy groups of the MGCL master library can be collapsed into any fewer number of energy groups by the COLLAPS code using the standard neutron energy spectrum described above. In this way, a sub-master library is generated.

The MGCL master or sub-master library data together with atomic number densities and geometrical data are used in the MAIL code to generate effective macroscopic cross section data as follows:

$$\begin{aligned}\Sigma_{x, eff} &= \sum_j N^j \sigma_{x, eff}^j \\ \sigma_{x, eff}^j &= \sigma_{x, in}^j f_x(\sigma_o^j) \\ \sigma_o^j &= (\sum_{i \neq j} N^i \sigma_t^i - \alpha G/l) / N^j\end{aligned}$$

where,

- $\ell$  : Mean cord length of the fuel lump,
- $G$  : Dancoff correction factor,
- $a$  : Bell factor,
- $i, j$  : Nuclide indices,
- $N^i$  : Atomic number density of the nuclide  $i$ ,
- $\sigma_{x, \text{eff}}^j$  : Effective microscopic cross section of the reaction  $x$  and the nuclide  $j$ ,
- $\sigma_{x, \infty}^j$  : Infinite dilution cross section of the reaction  $x$  and the nuclide  $j$ ,
- $\sigma_o^j$  : Background cross section of the nuclide  $j$ ,
- $\sigma_t^i$  : Effective total microscopic cross section of the nuclide  $i$ ,
- $f_x(\sigma_o^j)$  : Self-shielding factor of the reaction  $x$ ,
- $\Sigma_{x, \text{eff}}$  : Effective macroscopic cross section of the reaction  $x$ .

The background cross section  $\sigma_o^j$  is determined by the iteration method. The self-shielding factor  $f_x(\sigma_o^j)$  is interpolated over the one-dimensional data table in the MGCL library. The effective microscopic cross section is obtained as the product of the infinite dilution cross section and the self-shielding factor. Macroscopic cross sections are obtained by multiplying thus obtained effective microscopic cross sections with atomic number densities.

An effective neutron multiplication factor  $k_{\text{eff}}$  for a given problem is calculated by an appropriate transport code such as the KENO-IV code<sup>10)</sup>, the MULTI-KENO code<sup>11)</sup>, the ANISN-JR code<sup>12)</sup>, or the DOT3.5 code<sup>13)</sup> with macroscopic cross section data and geometry data. A diffusion code such as CITATION<sup>14)</sup> is also available in the code system. The Monte Carlo code KENO-IV, for example, is very useful in analyses of nuclear fuel cycle facilities with three-dimensional complex geometries. If necessary, cell averaged cross section data are obtained by the CELL-CAL code prior to  $k_{\text{eff}}$  calculations. The reliability of the calculated  $k_{\text{eff}}$  is evaluated by the YENMA code referring to various benchmark calculations stored in a data file.

## 2.2 Calculational method

Some 1400 cases of benchmark calculations were performed with Monte Carlo code KENO-IV and multi-group constants library MGCL. In each calculation, the history of neutrons were taken to be 103 batches of 300 neutrons. In this case, about 0.4% of standard deviation is obtained on average. This value of standard deviation is admitted to be adequate for code validation. The first three batches, in each calculation, were skipped and the results of these batches were discarded to obtain reliable calculation results. Number of neutron energy groups is 137, since the MGCL master library has been used in the calculations. Starting neutron distributions were taken to be flat. Most of the geometrical data and existing material were taken into the KENO-IV input as exactly as possible. In tracking a neutron in a reflector, albedo and adjoint biasing techniques were not used.

**Table 2.1.1** Energy group structure of MGCL

137 group master library			Ultra fine group	109 G.L.		67 G.L.		26 G.L.	
Group No.	Upper Energy Boundary	Lethargy Width $\Delta u$	Lethargy Width $\Delta u$	Group No.	$\Delta u$	Group No.	$\Delta u$	Group No.	$\Delta u$
1	16.487 MeV	0.125	0.001 (5000)	1	0.25	1	0.5		
2	14.550			2					
3	12.840			3					
4	11.331			4		2	0.75	1	1.25
5	10.000			5					
6	8.825			6					
7	7.788			7		3	0.75		
8	6.8729			8					
9	6.0653			9					
10	5.3526			10		4	0.75		
11	4.7237			11					
12	4.1686			12					
13	3.6788			13					
14	3.2465			14					
15	2.8650			15					
16	2.5284			16				2	1.50
17	2.2313			17					
18	1.9691			18					
19	1.7377			19					
20	1.5335			20					
21	1.3533			21					
22	1.1943			22					
23	1.0540			23					
24	930.14 keV			24					
25	820.85			25		5	0.75		
26	724.40			26					
27	639.28			27					
28	564.16			28					
29	497.87			29				3	1.75
30	439.37			30					
31	387.74			31					
32	342.18			32					
33	301.97			33		6	1.00		
34	266.49			34					
35	235.18			35					
36	207.54			36					
37	183.16			37					
38	161.63			38		7	0.50		
39	142.64			39					
40	125.88			40					
41	111.09		0.00025 (12000)	41				4	1.25
42	98.037			42					
43	86.517			43					
44	76.351			44					
45	67.379			45					
46	59.462			46					

Table 2.1.1 (continued)

137 group master library			Ultra fine group	109 G.L.		67 G.L.		26 G.L.	
Group No.	Upper Energy Boundary	Lethargy Width $\Delta u$	Lethargy Width $\Delta u$	Group No.	$\Delta u$	Group No.	$\Delta u$	Group No.	$\Delta u$
47	52.475 keV	0.125	0.00025	24	0.25				
48	46.309								
49	40.868			25		9	0.75		
50	36.066								
51	31.828			26				5	1.25
52	28.088								
53	24.788			27					
54	21.875					10	0.50		
55	19.305			28					
56	17.036								
57	15.034	0.25		29					
58	11.709			30		11	1.00	6	1.00
59	9.1188			31					
60	7.1017			32					
61	5.5308		0.000125 (30000)	33					
62	4.3075			34		12	0.75		
63	3.3546			35					
64	2.6126			36					
65	2.0347			37		13	0.75	7	2.25
66	1.5846			38					
67	1.2341			39					
68	961.12 eV			40		14	0.75		
69	748.52			41					
70	582.95			42					
71	454.00			43		15	0.75		
72	353.58			44				8	1.50
73	275.36			45					
74	214.45			46		16	0.75		
75	167.02			47					
76	130.07		0.00025 (17000)	48					
77	101.30			49		17	0.75		
78	78.893			50				9	1.50
79	61.442			51					
80	47.851			52		18	0.75		
81	37.267			53					
82	29.023			54					
83	22.603			55		19	0.75		
84	17.603			56				10	1.50
85	13.710			57					
86	10.677			58		20	0.75		
87	8.3153			59					
88	6.4760			60					
89	5.0435			61		21	0.75		
90	3.9279			62				11	1.25
91	3.0590			63					
92	2.3824			64		22	0.50		

**Table 2.1.1** (continued)

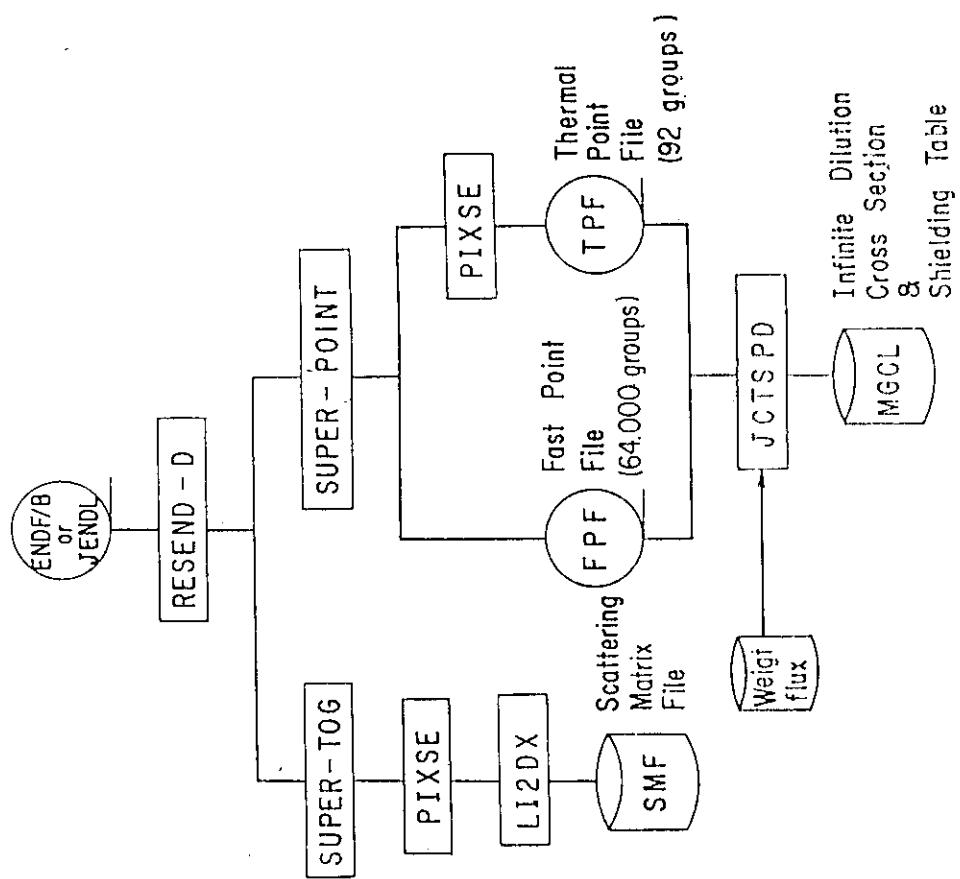


Fig. 2.1.2 General flow diagram of a computer code system MGCL-ACE.

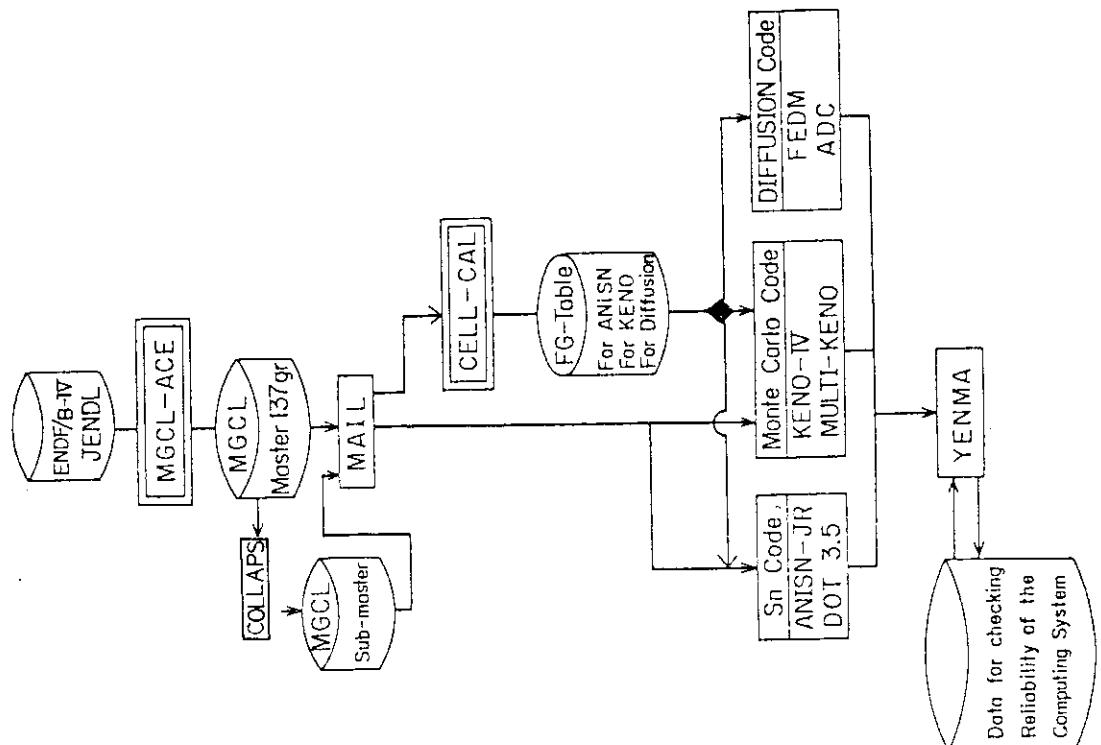


Fig. 2.1.1 General calculation flow of JACS.

### 3. Experimental Data and Calculated Results

Most of the experimental data has been taken from some 40 reports published by American laboratories. A part of them are from French reports or English reports. They cover a wide range of fuel compositions, moderating conditions, reflecting conditions, fuel shape/dimensions and so on, due to their variety in simulating nuclear fuel facilities.

In this chapter, three subchapters are provided corresponding to the differences of experimental systems. Subchapter A is for homogeneous single-unit systems, Subchapter B for homogeneous multi-unit systems and Subchapter C for heterogeneous systems. Each subchapter is then divided into several sections. Each section deals with experiments and calculations considered to be in the same group of benchmark calculations. In each section, brief descriptions are presented on the experimental system and the calculational model together with the calculated results made easy to grasp by the aid of graphs and tables. More concerned readers should refer to original documents listed at the end of this report.

#### A. Homogeneous Single-Unit System

##### 3.A.1 Pile of PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene Compacts with Various Poison-Plate Materials

Experiments were performed at BNWL, USA, simulating the storage of mixed oxide fuel.<sup>15)</sup> Fuel compacts are made of homogeneous mixture of PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene. There are two sets of fuels: one for an atomic H/(Pu+U) ratio of 30.6 and a plutonium enrichment of 14.6 wt%, the other for an atomic H/(Pu+U) ratio of 2.8 and a plutonium enrichment of 29.3 wt%. Geometry of a compact is either a 5.09 cm × 5.09 cm × 5.09 cm cube or a 5.09 cm × 5.09 cm × 1.38 cm slab, clad with 0.12 mm thick vinyl tape. These compacts are piled up to form a rectangular parallelepiped assembly of which base consists of 8 × 8 fuel compacts. One of the following poison materials is used as neutron absorber plates in each case.

- 304L Stainless steel,
- 304L Stainless steel containing 1.1 wt% natural Boron,
- 304L Stainless steel containing 1.6 wt% natural Boron,
- Uranium depleted to 0.2 wt% <sup>235</sup>U,
- Boral,
- Cadmium,
- Lead.

The assembly is fully reflected with at least 15 cm thick Plexiglas (methacrylate plastic : C<sub>5</sub>H<sub>8</sub>O<sub>2</sub>). The experimental configuration is shown in Fig. 3.A.1.1. Criticality approach was made by adding fuel compacts, one by one, to the top face of the assembly. Critical heights are given in Tables 3.A.1.1 and 3.A.1.2.

The KENO-IV calculational models are shown in Fig. 3.A.1.2 for the two different basic fuel assemblies.<sup>16)</sup> Thickness of the reflector is assumed to be 15 cm in all directions. Cladding tape and void existing between fuel compacts are neglected in geometry modelling, but they are taken in calculating atomic number densities of the fuel region.

Calculated results are listed in Tables 3.A.1.3 and 3.A.1.4<sup>16)</sup>. Histogram of the calculated  $k_{eff}$ 's for 51 cases in all is shown in Fig. 3.A.1.3. One can see that the averaged  $k_{eff}$  values

are 1.023 for the 30.6 H/(Pu+U) fuel system, and 1.026 for the 2.8 H/(Pu+U) fuel (partly with the 30.6 H/(Pu+U) fuel). These calculated  $k_{eff}$ 's are about 2–3% higher than unity.

### 3.A.2 Pile of PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene Compacts with Various Plutonium Enrichments

Experiments were performed at BNWL, USA, using compacts of homogeneous PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene with various plutonium enrichments.<sup>17),18)</sup> There are six batches of fuels having various plutonium enrichments and atomic H/(Pu+U) ratios as follows:

- 29.3 wt% Pu and 2.8 H/(Pu+U),
- 15.0 wt% Pu and 2.86 H/(Pu+U),
- 8.1 wt% Pu and 7.3 H/(Pu+U),
- 30.0 wt% Pu and 47.4 H/(Pu+U),
- 14.62 wt% Pu and 30.6 H/(Pu+U),
- 7.89 wt% Pu and 51.8 H/(Pu+U).

A compact is a 5.09 cm cube or a slab of 5.09 cm square with 1.3 cm thickness, clad in 0.12 mm thick vinyl tape. These compacts are piled up to form a rectangular parallelepiped assembly with or without Plexiglas reflector of 150 mm thickness at least. Criticality approach was made by adding the compacts, one by one, to the top face of the assembly.

The basic KENO-IV calculational model is shown in Fig. 3.A.2.1. Thickness of Plexiglas reflector is assumed to be 15 cm in all directions. Cladding tape and stacking void existing between the compacts are not expressed explicitly in the calculational model. Corrections have been made to the measured critical heights of the assembly to compensate reactivity effects of the cladding tape, the stacking voids and the structural supports.

Calculated results together with critical dimensions and masses for 52 assemblies in all are listed in Tables 3.A.2.1 through 3.A.2.6. Histograms of the calculated  $k_{eff}$ 's are shown in Figs. 3.A.2.2 and 3.A.2.3, respectively, corresponding to the different H/(Pu+U) atomic ratios. The calculated  $k_{eff}$ 's for low (2 to 8) H/(Pu+U) systems are seen to be between 0.980 and 1.025 with the average value of 1.005. On the other hand, the calculated  $k_{eff}$ 's for high (30 to 50) H/(Pu+U) systems are between 0.980 and 1.030 with the average value of 1.010, about 1% higher than unity.

### 3.A.3 Pile of PuO<sub>2</sub>-UO<sub>2</sub>-Polythene Compacts

Experiments were performed in Britain using compacts of homogeneous PuO<sub>2</sub>-U(natural)O<sub>2</sub>-Polythene fuel.<sup>19)</sup> Plutonium enrichment is 23.1 atomic percent, and H/(Pu+U) atomic ratio is 4.67. A compact is a slab of 4.28 cm square with 2.15, 2.69 or 3.25 cm height, clad with 0.2 mm thick polythene tape. These compacts are piled up to form a rectangular parallelepiped assembly having a base of 5×5 to 8×8 compacts. Criticality approach was made by adding compacts, one by one, to the top face of the assembly.

The basic KENO-IV calculational model is the same as that shown in Fig. 3.A.2.1 except for the fuel material. Thickness of Polythene reflector is assumed to be 20 cm in all directions. Fuel region is assumed to contain atoms of the cladding tape homogeneously.

Calculated results together with critical dimensions and masses for 4 assemblies in all are listed in Table 3.A.3.1. Average value of the four calculated  $k_{eff}$ 's is 1.018, about 2% higher than unity.

### 3.A.4 Pile of PuO<sub>2</sub> (-Polystyrene) Compacts

Experiments were performed at BNWL, USA, using compacts of homogeneous PuO<sub>2</sub> or PuO<sub>2</sub>-Polystyrene fuel.<sup>20),21),22)</sup> There are four batches of fuels having various plutonium

isotopic compositions and atomic H/(Pu+U) ratios as follows:

- 2.2 wt%  $^{240}\text{Pu}$  and 15.0 H/Pu,
- 8.0 wt%  $^{240}\text{Pu}$  and 15.0 H/Pu,
- 11.46 wt%  $^{240}\text{Pu}$  and 5.0 H/Pu,
- 18.35 wt%  $^{240}\text{Pu}$  and 0.04 H/Pu.

A compact is a 5.1 cm cube or a slab of 5.1 cm square with 3.8 cm height, clad with 0.3 mm polymer coating. These compacts are piled up to form a rectangular parallelepiped assembly with or without Plexiglas reflector of 150 mm thickness at least. Criticality approach was made by adding the compacts, one by one, to the top face of the assembly.

The basic KENO-IV calculational model is the same as that shown in **Fig. 3.A.2.1** except for the fuel material. Thickness of Plexiglas reflector is assumed to be 15 cm in all directions. Corrections have been made to the measured critical heights of the assemblies to compensate voids existing between compacts and temperature effect.

Calculated results together with critical dimensions and masses are listed in **Tables 3.A.4.1** through **3.A.4.3**. Histogram of the calculated  $k_{eff}$ 's for 30 cases in all is shown in **Fig. 3.A.4.1**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.975 and 1.035 with the average value of 1.015, about 1.5% higher than unity.

### 3.A.5 Single Tank Containing Uranyl-Nitrate Solution

Experiments were performed at Rocky Flats, USA, using single tank containing 93.17%  $^{235}\text{U}$  enriched uranyl-nitrate solution.<sup>23)</sup> Uranium concentration is ranged from 54.89 to 357.71 gU/l. Unreflected experiments have been reported, as well as measurements within thick-walled cubical reflector shells composed of concrete or plastics. The single tank is composed of either Type 6061 aluminum or Type 304 stainless steel with inner diameter ranging from 27.88 to 50.69 cm, and has a coaxial tailpipe about 30 cm long, welded to its bottom. Uranyl-nitrate solution was introduced into the single tank through the tailpipe. Critical height inside the single tank was decided by linear interpolation between a slightly supercritical and a slightly subcritical height.

In the calculational model, the tailpipe is neglected, as shown in **Fig. 3.A.5.1**.

Calculated  $k_{eff}$ 's are listed in **Tables 3.A.5.1** through **3.A.5.3<sup>24)</sup>**. Histogram of the calculated  $k_{eff}$ 's for 45 cases in all is shown in **Fig. 3.A.5.2**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.960 and 0.995 with the average value of 0.973. The average value of 0.970 for minimally reflected systems is about 0.8% lower than the average value of 0.978 for concrete reflected systems.

### 3.A.6 Plutonium Fuel in Various Shapes

The US NUREG report<sup>25)</sup> contains several different experiments concerning homogeneous plutonium fuels having various shapes as listed in **Table 3.A.6.1**. These experiments have been selected to provide data for benchmark calculations to validate the accuracy of the neutron cross sections for the isotopes of plutonium. They cover a wide range of H/Pu ratio (a wide variation of neutron spectrum) and a wide range of  $^{240}\text{Pu}$  content in plutonium.

Calculational models for 15 cases selected from these experiments are shown in **Figs. 3.A.6.1** through **3.A.6.10**, respectively.

Calculated results are summarized in **Table 3.A.6.2<sup>26)</sup>**. The calculated  $k_{eff}$ 's distribute from 0.984 to 1.031 with the average value of 1.008. **Figure 3.A.6.11** shows that the calculated  $k_{eff}$  decreases with the increase of H/Pu atomic ratio.

### 3.A.7 Intersecting Cylinders Containing Uranyl-Fluoride Solution

Experiments were performed at ORNL, USA, simulating intersecting pipes or vessels in fuel manufacturing plants.<sup>27)</sup> Aqueous U(5)O<sub>2</sub>F<sub>2</sub> solution was used as fissile material. Two types of intersections were studied. They are nominally called “30 degrees lateral” and “cross”. Critical approach was made by increasing the level of fissile solution.

Experimental arrangement of “30 degrees lateral” is shown in **Fig. 3.A.7.1**. The lateral and the vertical cylinders have the inner diameter of 27.9 cm. A vent tube of 2.2 cm i.d. connects the top of the lateral cylinder with the vertical cylinder. The top of the lateral cylinder is closed by welding a 0.64 cm thick plate, and the top of the vertical cylinder is open. The bottom plate of the vertical cylinder is 1.3 cm thick and penetrated by a 5.1 cm i.d. coaxial tube. Through this tube, the fissile solution was introduced into the vertical cylinder. The bottom plate is located at 22.9 cm above the floor of reflector tank.

Experimental arrangement of “cross” is shown in **Fig. 3.A.7.2**. The cylinders and the supporting plates are made of Plexiglas having density of 1.18 g/cm<sup>3</sup>. Four types of cylinders, of which inner diameters are 26.7, 27.3, 27.9 and 28.6 cm, are used depending on the experimental phase. The crossing cylinders are fixed rigidly with the Plexiglas plates so that the centerlines of the cylinders intersect at a central point. The upper and the lower vertical cylinders are held coaxial, but the horizontal cylinders are slightly tilted down by 2 degrees toward the central point to drain the solution. The ends of the horizontal cylinders are cemented closely by 3.17 cm thick Plexiglas discs with vent lines for air escaping. The end of the lower vertical cylinder is cemented by a bottom flange of 1.27 cm thick stainless steel. The top of the upper vertical cylinder is open.

Experimental configurations have been modelled for KENO calculations as shown in **Figs. 3.A.7.3** and **3.A.7.4**<sup>28)</sup>. Detailed structures such as vent-lines or lead-tubes are neglected in the models. Surface equations expressed by quadratic functions for KENO general geometry input data are shown in **Fig. 3.A.7.3**. As shown in **Fig. 3.A.7.4**, the Plexiglas plates supporting the cylinders are neglected in the model.

Calculated results for the “30 degrees lateral” systems are listed in **Table 3.A.7.1** and those for “cross” are listed in **Table 3.A.7.2**. For the former, the calculated  $k_{eff}$ 's for 18 cases in all are between 0.96 and 0.99 with the average value of 0.978, about 2% lower than unity as shown in the histogram of **Fig. 3.A.7.5**. For the latter, the calculated  $k_{eff}$ 's for 27 cases in all distribute from 0.97 and 0.99 with the average value of 0.985, about 1.5% lower than unity as shown in the histogram of **Fig. 3.A.7.6**.

### 3.A.8 Sphere or Cylinder Containing Uranyl-Fluoride Solution

Critical dimensions of aqueous solutions of UO<sub>2</sub>F<sub>2</sub> were measured at ORNL, USA, to provide data useful in criticality calculations.<sup>29)</sup> The solution is contained in a cylindrical or spherical vessel made of aluminum or stainless steel, unreflected or reflected with water. Uranium is enriched to 4.9 wt% <sup>235</sup>U. High solubility of uranyl fluoride (nearly 890 gU/l) allows experiments with solutions having an H/<sup>235</sup>U atomic ratio as low as some 500.

Calculational models for typical experimental systems are shown in **Figs. 3.A.8.1** and **3.A.8.2**. One can see that structural materials other than the cylinder vessel are neglected in the calculational models.

**Table 3.A.8.1** lists calculated results together with experimental conditions. Histogram of the calculated  $k_{eff}$ 's for 23 critical experiments (except for 4 subcritical cases) is shown in **Fig. 3.A.8.3**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.950 and 1.005 with the average value of 0.981, about 2% lower than unity.

### 3.A.9 Cylinder Containing Uranyl-Fluoride Solution with Combination of Reflectors

Experiments were performed at ORNL, USA, to study the effect of steel/water composite reflectors on reactivity of a cylinder containing aqueous  $\text{UO}_2\text{F}_2$  solution.<sup>30)</sup> A stainless-steel cylinder having 0.079 cm thick wall with inner diameter of 33.09 cm or 39.09 cm was utilized to contain low  $^{235}\text{U}$  enriched (4.98 wt%) uranyl-fluoride solution. A lateral steel reflector adjacent to the solution region ranges up to 5.08 cm in thickness, and is surrounded by an infinitely thick water reflector.

**Figure 3.A.9.1** shows the calculational model for a typical experimental system.

**Table 3.A.9.1** lists calculated results together with experimental conditions for 24 cases in all. Histogram of the calculated  $k_{eff}$ 's is shown in **Fig. 3.A.9.2**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.970 and 1.000 with the average value of 0.983, about 2% lower than unity.

### 3.A.10 Pile of $\text{UF}_4$ -Paraffin Compacts

Experiments were performed at ORNL, USA, with homogeneous mixtures of finely divided  $\text{U}(2)\text{F}_4$  or  $\text{U}(3)\text{F}_4$  dispersed in paraffin.<sup>31)</sup> The mixture was pressed into rectangular parallelepipeds with dimensions ranging from 0.63 cm × 2.54 cm × 2.54 cm to 10.16 cm cubes and covered with the 0.02 cm thick aluminum foil. The H/U atomic ratio varies from 4 to 20. The uranium in the fuel mixtures contains either 2 or 3 wt%  $^{235}\text{U}$ . The assembly was constructed in rectangular geometry with or without paraffin/Plexiglas reflectors. Critical dimensions of the assembly were measured in the experiment. Measured data are listed in **Table 3.A.10.1**.

**Figure 3.A.10.1** shows the calculational model for a typical reflected experimental pile of  $\text{UF}_4$ -paraffin compacts. The calculational model for a typical unreflected experimental pile is shown in **Fig. 3.A.10.2**.

Calculated results for these experimental systems are listed in **Table 3.A.10.2**. Histogram of the calculated  $k_{eff}$ 's for 22 reflected piles is shown in **Fig. 3.A.10.3**, and histogram for 30 unreflected piles is shown in **Fig. 3.A.10.4**. As seen from these histograms, the average value of 1.007 for the reflected piles is slightly higher than 1.004 for the unreflected piles.

### 3.A.11 Sphere Containing Plutonium-Nitrate Solution with Various Reflectors

Experiments were performed at BNWL, USA, with plutonium (4.6%  $^{240}\text{Pu}$ ) nitrate solution in stainless steel spheres of 11.5, 14.0, and 15.2 in. diameters.<sup>32)</sup> Reflectors are of water, concrete, paraffin, and stainless steel. In some cases, unreflected spheres of 15.2 in. diameter were used in the experiment. The critical conditions of the spheres were obtained at various plutonium concentrations from 24 to 435 g plutonium/liter and various molalities from 0.2 to 7.7. The effect of a 0.030 in. cadmium shell and a 4 in. air gap between the reflector and the vessel was also studied in the experiment. **Table 3.A.11.1** lists experimental conditions for 48 cases in all.

The calculational model is shown in **Fig. 3.A.11.1**. Calculated results for all experimental cases are listed in **Table 3.A.11.2**. Histogram of the calculated  $k_{eff}$ 's is shown in **Fig. 3.A.11.2**.

As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.980 and 1.025 with the average value of 1.003, about 0.3% higher than unity.

### 3.A.12 Annular Cylinder Containing Plutonium-Nitrate Solution

Experiments were performed in France using stainless steel annular cylinders containing plutonium-nitrate solution immersed in a larger water-reflector tank.<sup>33)</sup> The annular vessel with 3 mm thick wall, 200 mm i.d. and 500 mm o.d., has either an inner void region for some

cases of experiments, or an inner cadmium-paraffin region for other cases of experiments. The  $^{240}\text{Pu}$  content in Pu of  $\text{PU}(\text{NO}_3)_4$  is about 19%. The reflecting water level is adjusted to be equal to the solution level in the vessel. Criticality approach was made by increasing the solution height gradually, as well as by varying the plutonium concentration from 28.7 to 165 g/l, and the nitrate concentration from 1.85 to 3.85 mol..

The calculational model for a typical case of these experiments is shown in **Fig. 3.A.12.2**.

**Tables 3.A.12.1** and **3.A.12.2** list the calculated results together with the experimental conditions. Histogram of the calculated  $k_{\text{eff}}$ 's for 17 cases in all is shown in **Fig. 3.A.12.3**. As seen from the histogram, the calculated  $k_{\text{eff}}$ 's are between 0.980 and 1.005 with the average value of 0.994, about 0.6% lower than unity.

### 3.A.13 Cylinder or Sphere Containing Plutonium-Uranyl-Nitrate Solution Reflected by Water

Experiments were performed at BNWL, USA, with plutonium-uranyl nitrate solution in a stainless steel cylinder of 61.0 cm i.d. reflected by water.<sup>34)</sup> Stainless steel spheres of 35.7 and 38.6 cm diameter were also utilized in the experiment. For the experiment in cylindrical geometry, the  $^{240}\text{Pu}$  content in Pu is either 5.6 wt% or 23.0 wt%. For the water reflected spheres, the  $^{240}\text{Pu}$  content in Pu is 4.7 wt%. The uranium in the mixture is slightly depleted, containing 0.66 wt%  $^{235}\text{U}$ . The plutonium concentration is ranged from 12.4 to 97.3 gPu/l (uranium plus plutonium concentration from 42 to 432 g (U+Pu)/l). Experimental arrangements for cylindrical and spherical geometries are shown in **Figs. 3.A.13.1** and **3.A.13.2**, respectively.

The calculational model for a typical cylindrical geometry is shown in **Fig. 3.A.13.3**.

Calculated results together with experimental conditions for 29 cases in all are listed in **Tables 3.A.13.1** and **3.A.13.2**. Histograms of the calculated  $k_{\text{eff}}$ 's for the cylindrical geometries are shown in **Figs. 3.A.13.4** and **3.A.13.5**, respectively, corresponding to the  $^{240}\text{Pu}$  contents in Pu in the plutonium-uranyl-nitrate solution. As seen from the histograms, the calculated  $k_{\text{eff}}$ 's are between 0.985 and 1.020. The average value of  $k_{\text{eff}}$  for the 29 cases in all is almost equal to unity.

### 3.A.14 Assembly of Low-Enriched Uranium-Oxide ( $\text{U}_3\text{O}_8$ ) Compacts

Experiments were performed at Rocky Flats, USA, using low-enriched damp uranium oxide ( $\text{U}_3\text{O}_8$ ) systems.<sup>35)</sup> The experimental core consists of 152 mm cubical aluminum cans containing low enriched (4.46 wt%  $^{235}\text{U}$ ) uranium oxide. The uranium oxide was compacted to a density of 4.68 g/cm<sup>3</sup> and moistened to an H/U atomic ratio of 0.77. These aluminum cans were piled up to be a 770 mm cubical assembly. Since the uranium oxide in an assembly of this size could not go critical state by itself, a highly enriched (about 93 wt%  $^{235}\text{U}$ ) uranium metal or solution "driver" shown in **Fig. 3.A.14.1** was used. As shown in this figure, the experiments were performed on a horizontal split table. Each half of the table supports a portion of the experimental core. Criticality approach was made by decreasing the distance between the two interfaces of the core. Measurements are reported for systems having the minimum reflector and for systems reflected by 254 mm thick concrete or plastic.

Calculated results together with principal critical conditions are listed in **Table 3.A.14.1**<sup>36)</sup>. Histogram of the calculated  $k_{\text{eff}}$ 's for 10 cases in all is shown in **Fig. 3.A.14.2**. The calculated  $k_{\text{eff}}$ 's are between 0.970 and 1.010 with the average value of 0.991, about 1% lower than unity.

### 3.A.15 Cylinder Containing Plutonium-Nitrate Solution of High-Fissile Content Poisoned with Gadolinium

Experiments were performed at BNWL, USA, utilizing gadolinium as a soluble neutron absorber in the plutonium-nitrate solution.<sup>37)</sup> Plutonium concentration of the solution is either 116 gPu/l with 1.85 mol. of nitric acid, or 363 gPu/l with 4.1 mol. of nitric acid. Gadolinium concentration ranges between 0.00 and 20.25 gGd/l.

Experimental vessel is a stainless steel cylinder of 61.03 cm i.d. and 0.079 cm wall thickness, as shown in **Fig. 3.A.13.1**. The <sup>239</sup>Pu content in Pu is 90.7 wt%.

The KENO-IV calculational model is shown in **Fig. 3.A.15.1**. Calculated results together with the critical conditions are listed in **Table 3.A.15.1**.<sup>36)</sup> Histogram of the calculated  $k_{eff}$ 's for 15 cases in all is shown in **Fig. 3.A.15.2**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.980 and 1.005 with the average value of 0.988, about 1.2% lower than unity.

### 3.A.16 Cylinder Containing Plutonium-Uranyl-Nitrate Solution with Soluble Neutron Poison

Experiments were performed at BNWL, USA, simulating dissolution of spent fuel elements in a dissolver vessel.<sup>38)</sup> A schema of the experimental system is the same as that shown in **Fig. 3.A.13.1**. Plutonium-(uranyl)-nitrate solution in the stainless steel cylindrical vessel of 61.03 cm i.d. contains gadolinium and/or boron as a soluble neutron absorber. The <sup>239</sup>Pu content in Pu is about 93 wt%, and uranium is slightly depleted, containing 0.66 wt% <sup>235</sup>U. Experimental conditions are listed in **Table 3.A.16.1**. As listed in this table, plutonium concentration ranges between 75.6 and 363 gPu/l, and uranium concentration between 0.0 and 182.6 gU/l. Gadolinium concentration was varied up to 20.25 gGd/l, and natural boron concentration up to 1.54 gB/l. Criticality approach was made by increasing the solution height incrementally and the critical height was measured.

The calculational model for a typical case is shown in **Fig. 3.A.16.1**.

Calculated results for all 34 cases are listed in **Table 3.A.16.2**. Histogram of the calculated  $k_{eff}$ 's is shown in **Fig. 3.A.16.2**. As shown in this figure, the calculated  $k_{eff}$ 's are between 0.965 and 1.025 with the average value of 0.987, about 1.3% lower than unity.

### 3.A.17 Homogeneous Systems with Boron Containing Glass Raschig Rings

Experiments were performed at BNWL, USA, to obtain data for checking calculational models for raschig ring poisoned systems containing mixture of uranium and plutonium in solution.<sup>38)</sup> Borosilicate glass rings, 1.5 in. o.d., 1.25 in. i.d., and 1.7 in. long, fill a cylindrical vessel containing the plutonium-uranyl-nitrate solution of the concentration ranging from 79 to 92 gPu/l and from 147 to 182 gU/l. The plutonium contains 5.6 wt% <sup>240</sup>Pu. The <sup>235</sup>U enrichment is 0.66 wt%. The raschig ring contains about 0.5 wt% natural boron. The experimental vessel is 107 cm in height and 61 cm in i.d., and is reflected with water, as shown in **Fig. 3.A.17.1**.

The calculational model is shown in **Fig. 3.A.17.2**. As shown in the figure, the stack of raschig rings at random disposition is modelled to be an uniformly arranged triangular lattice with zig-zag boundary. The experimental vessel is neglected in the calculational model.

Calculated results together with principal experimental conditions are listed in **Table 3.A.17.1**.<sup>36)</sup> The average value of the calculated  $k_{eff}$ 's for 8 cases in all is 0.9833, about 2% lower than unity.

## B. Homogeneous Multi-Unit (Interacting) System

### 3.B.1 Array of Cylinders Containing UF<sub>6</sub>

Experiments were performed at ORNL, USA, simulating a storage of UF<sub>6</sub> transport container.<sup>39)</sup> The 97.7 wt% <sup>235</sup>U enriched UF<sub>6</sub> is contained in 20.3 cm i.d./125.0 cm high/0.48 cm thick Monel steel cylinders. The H/U atomic ratio of UF<sub>6</sub> fuel is reported to be less than 0.088. A split table carrying the array of the cylinders was utilized in the experiment. The table top measures 180 cm × 300.0 cm at the closure and is of 2.5 cm thick mild steel. Criticality approach was done by gradually narrowing the spacing between each halves of the table. Unreflected experiments are reported, as well as measurements with concrete and/or polyethylene reflectors. Plexiglas was used as a moderator in some cases.

**Figure 3.B.1.1** shows a typical calculational model for the experiment. The split table is not taken into consideration explicitly in the calculational model of the reflected experiments. On the other hand, in the calculational model of the unreflected experiment, the table is taken into consideration to be located beneath UF<sub>6</sub> cylinders. The H/U atomic ratio is assumed to be 0.044, just a half of the measured maximum value.

Calculated results together with critical conditions are listed in **Tables 3.B.1.1 through 3.B.1.15.**<sup>40)</sup> Histograms of the calculated  $k_{eff}$ 's are shown in **Fig. 3.B.1.2** for the 16 unreflected systems and in **Fig. 3.B.1.3** for the 51 reflected systems. The calculated  $k_{eff}$ 's for the unreflected systems are between 0.95 and 0.975 with the average value of 0.963. Those for the reflected systems are between 0.975 and 1.02 with the average value of 0.996.

### 3.B.2 Slab-Cylinder Configurations of Uranyl-Nitrate Solution

Experiments were performed at Rocky Flats, USA, simulating tanks and pipes containing uranyl-nitrate solution in spent fuel reprocessing plants.<sup>41)</sup> A slab tank and several cylindrical vessels containing 93.2 wt% <sup>235</sup>U enriched uranyl-nitrate aqueous solution were arranged with or without Plexiglas reflector, as typically shown in **Fig. 3.B.2.1**. Vertical cylinders are arrayed on the bottom of slab tank and supported rigidly by two mild steel grid plates. These plates, 152 cm square and 0.159 cm thick each, are fastened to a framework. The cylinders are uniformly spaced in a square pattern with equal center-to-center distance. The space between the centerline of the outermost cylinders and the inner surface of the side walls of the tank was adjusted to be a half of the center-to-center distance of the cylinders. Inner size of the stainless steel slab tank is 120.7 cm square and 20.3 cm high, and the thickness of the side wall is 0.15 cm and that of the bottom is 0.635 cm. This tank is supported by six mild steel pipes. These pipes stand on a mild steel table, 152 cm square and 1.9 cm thick. This table is elevated to 137 cm above concrete floor. Concrete walls of the experimental room are located more than 3 m away from the side walls of the slab tank. Uranium concentration in the uranyl-nitrate solution increases from 465 to 525 gU/l during the experiments because of the evaporation of water. This uranium concentration was measured at each step of the experiments. The H/<sup>235</sup>U ratio was measured at both the beginning and the end of the experiments. Criticality approach was made by increasing the solution height in the cylinders and/or in the slab tank.

In the calculational model, the drain hose in the slab tank, the mild steel support-pipes and -plates are not included due to the lack of information of their geometric configuration. A typical calculational model for the KENO-IV code is illustrated in **Fig. 3.B.2.2.**<sup>42)</sup> Measured uranium concentrations were used for the calculation. The H/<sup>235</sup>U ratios for the calculation were estimated by the interpolation principle.

Calculated results together with critical conditions for reflected and unreflected slab

cylinder systems are listed in **Tables 3.B.2.1** through **3.B.2.4**.<sup>42)</sup> Histogram of the calculated  $k_{eff}$ 's for 96 cases in all is shown in **Fig. 3.B.2.3**. The calculated  $k_{eff}$ 's for the unreflected systems are between 0.905 and 0.965 with the average value of 0.929, about 7% lower than unity. Those for the reflected systems are between 0.955 and 1.015 with the average value of 0.989, about 1% lower than unity.

### 3.B.3 Array of Cylinders Containing Uranyl-Nitrate Solution

Experiments were performed at Rocky Flats, USA, using the array of cylinders containing 93.17%  $^{235}\text{U}$  enriched uranyl-nitrate solution.<sup>23)</sup> Uranium concentration is ranged from 60.32 to 369.96 gU/l. The cylinders are positioned in a cubical plastic or a concrete reflector. Cylindrical containers used in the experiment are made of aluminum with inner diameters from 16.12 cm to 21.12 cm. Each has a coaxial tail pipe, about 30 cm long, welded to the bottom of the container, through which uranyl-nitrate solution is introduced. Stainless steel sleeves are used to cover the cylinders in some experiments. A critical height inside the cylinders is determined by interpolation between a slightly supercritical height and a slightly subcritical height.

In the calculational model, the tail pipes leading uranyl-nitrate solution into the cylinders are neglected, as shown in **Fig. 3.B.3.1**.

Calculated  $k_{eff}$ 's are listed in **Tables 3.B.3.1** and **3.B.3.2**.<sup>24)</sup> Histogram of the calculated  $k_{eff}$ 's for 31 cases in all is shown in **Fig. 3.B.3.2**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.965 and 1.000 with the average value of 0.982, about 2% lower than unity.

### 3.B.4 Two Parallel Cylinders Containing Uranyl-Fluoride Solution

Experiments were performed at ORNL, USA, using two parallel cylinders containing 93.4%  $^{235}\text{U}$  enriched uranyl-fluoride solution.<sup>43)</sup> Uranium concentration ranges from 78.72 gU/l to 758.84 gU/l. Cylindrical containers used in the experiments are made of aluminum with inner diameters of 8, 10 and 15 in.. An experimental system consists of two parallel such cylinders, with or without a water reflector. Criticality approach was done by increasing the solution height inside the cylinders. The uranyl-fluoride solution was introduced into the inside of cylinder through a feeding pipe attached to the bottom of each cylinder.

In the calculational model, a plastic feeding pipe attached to one of the cylinders is neglected, as shown in **Figs. 3.B.4.1** and **3.B.4.2**, due to the lack of information.

Calculated  $k_{eff}$ 's are listed in **Table 3.B.4.1** for 29 cases in all.<sup>44)</sup> Histogram of these calculated  $k_{eff}$ 's is shown in **Fig. 3.B.4.3**. As seen from this figure, the calculated  $k_{eff}$ 's are between 0.945 and 1.01 with the average value of 0.976, about 2% lower than unity.

### 3.B.5 Three-Dimensional Array of Uranium-Metal Cylinders

Experiments were performed at ORNL, USA, using uranium metal cylindrical units arranged in a three-dimensional array.<sup>45)</sup> A unit is a cylinder of 11.5 cm in diameter and 5.40 to 13.47 cm in height. Uranium is enriched to 93.2%  $^{235}\text{U}$ . The units are supported, with their axes vertical, by stainless steel rods passing through two holes of 0.51 cm diameter in each cylinder. The holes are parallel to the cylinder axis and are separated 8.55 cm each other on a diameter. Paraffin reflectors of 1.3 cm, 2.5 cm, 3.8 cm, 7.6 cm or 15.2 cm thickness were provided for some experimental systems.

In the calculational model, the supporting rods and holes of uranium cylinders are neglected and filled with uranium fuel. **Figure 3.B.5.1** shows a typical calculational model for this array experiment.

**Tables 3.B.5.1** and **3.B.5.2** list calculated  $k_{eff}$ 's for 31 cases in all.<sup>44)</sup> Histogram of these  $k_{eff}$ 's is shown in **Fig. 3.B.5.2**. As seen from this histogram, the calculated  $k_{eff}$ 's are between 0.995 and 1.025 with the average value of 1.012, about 1% higher than unity.

### 3.B.6 Two- and Three-Dimensional Array of Plutonium-Metal Cylinders

Experiments were performed at LRL, USA, using plutonium-metal billets in two- or three-dimensional array.<sup>46)</sup> A unit of array is a cylinder of 6.53 cm diameter and 4.63 cm height, weighing 3 kg. The  $^{239}\text{Pu}$  content in Pu is 93.6%. Doubled billets (6 kg) are also used as a unit of array. The plutonium billets are placed inside supporting tubes. These tubes are fastened upright and arrayed on the assembly table. Vertical positioning within the tube was accomplished with tubular aluminum spacers. Plastic moderator discs are attached to the supporting tubes in some cases. Polyethylene reflector slabs are arranged along the surface of the fissile array in other cases. The parameter used for criticality approach is the distance between units.

The calculational model for a 3 kg plutonium-metal unit is shown in **Fig. 3.B.6.1**. **Figure 3.B.6.2** shows the calculational model for a 6 kg plutonium-metal unit. Calculational models for typical array dispositions are shown in **Figs. 3.B.6.3** and **3.B.6.4**.

**Table 3.B.6.1** lists calculated results for 19 experimental cases in all.<sup>44)</sup> Histogram of these  $k_{eff}$ 's is shown in **Fig. 3.B.6.5**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.975 and 1.02 with the average value of 0.999, nearly equal to unity.

### 3.B.7 Three-Dimensional Array of Cylinders Containing Uranyl-Nitrate Solution

Experiments were performed in USA using Plexiglas cylinders containing 93.2%  $^{235}\text{U}$  enriched uranyl-nitrate solution.<sup>47)</sup> The 5 liter of uranium solution having concentration of 415 gU/l and specific gravity of 1.555 is contained in each of 0.64 cm thick Plexiglas cylinders of 20.32 cm o.d. and 19.05 cm outside height. These cylinders are arranged to be a cubical array of  $2 \times 2 \times 2$ ,  $3 \times 3 \times 3$ ,  $4 \times 4 \times 4$  or  $5 \times 5 \times 5$  units with or without a Plexiglas or paraffin reflector of various thicknesses. The parameter used for criticality approach is the distance between units.

The calculational model for a typical experimental arrangement is shown in **Fig. 3.B.7.1**.

**Tables 3.B.7.1** and **3.B.7.2** list calculated results for 29 cases in all.<sup>44)</sup> Histogram of the calculated  $k_{eff}$ 's is shown in **Fig. 3.B.7.2**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 0.955 and 1.01 with the average value of 0.989, about 1% lower than unity.

### 3.B.8 Array of Tin Cans Containing $\text{UO}_2$ or $\text{UO}_2$ -Alcohol Slurry

Experiments were performed at ORNL, USA, using highly enriched (93.2 wt%  $^{235}\text{U}$ )  $\text{UO}_2$  and  $\text{UO}_2$ -alcohol slurry canned in tin and positioned on a split table.<sup>48)</sup> Descriptions of four types of fissile units utilized in the experiment are summarized in **Table 3.B.8.1**.

**Table 3.B.8.2** lists experimental conditions for 29 cases in all. As shown in this table, experimental cases, from No.1 to No.4, are related to Type I unit. Experiments No.5 to No.19 are related to Type II, and experiments No.20 to No.29 to Type III and Type IV. Type III and Type IV units contain alcohol, which is intended to simulate such conditions as wet oxide powders in processing, transport or storage.

The KENO-IV calculational models for the fissile units are shown in **Fig. 3.B.8.1**.<sup>36)</sup> Canning material, tin, is assumed to be steel, since the MGCL data library does not provide cross section data for tin. A calculational array model is shown in **Fig. 3.B.8.2**. As shown in this figure, the reflector thickness is assumed to be 18.1 cm.

Calculated results together with principal experimental parameters<sup>36)</sup> are listed in **Table 3.B.8.3**. As seen from this table, C/E value ranges between 0.976 and 1.015.

### C. Heterogeneous System

#### 3.C.1 LWR Critical Assemblies with Various Lattice Pitches and Patterns

Experiments were performed using the Tank-type Critical Assembly (TCA) at the Japan Atomic Energy Research Institute.<sup>49)</sup> Critical assemblies of TCA consist essentially of fuel rods, grid plates and a core tank with a water moderator and reflector. The core tank is 1.83 m in diameter and 2.08 m in height as shown in **Fig. 3.C.1.1**. Experimental lattices were built in the core tank, and made critical by raising the water level incrementally. No control-rod operation was needed to attain criticality. Fuel is either of 2.6%  $^{235}\text{U}$  enriched  $\text{UO}_2$  or of 3.0%  $\text{PuO}_2\text{-U(natural)}\text{O}_2$ , as shown in **Fig. 3.C.1.2**. Water-to-fuel volume ratio in a lattice cell ranges from 1.50 to 3.00 for the  $\text{UO}_2$  fuel, and from 2.42 to 5.55 for the  $\text{PuO}_2\text{-UO}_2$  fuel. Fuel rods are held vertical in the core tank by the grid plates. In some experiments, the water gap was formed by removing several rows from a uniform square-pitch lattice array. In some cases,  $\text{H}_3\text{BO}_3$  soluble poison was added into the water in the core tank. In other cases, poison plates such as Al-Cd sheets were inserted between fuel-rod rows. There are five types of experimental systems as follows:

- (1)  $\text{UO}_2\text{-H}_2\text{O}$  systems,
- (2)  $\text{PuO}_2\text{-UO}_2\text{-H}_2\text{O}$  systems,
- (3)  $\text{UO}_2\text{-Water Gap}$  systems,
- (4)  $\text{UO}_2\text{-H}_3\text{BO}_3\text{-H}_2\text{O}$  systems,
- (5)  $\text{UO}_2\text{-poison plate-H}_2\text{O}$  systems.

The KENO calculational model for  $\text{UO}_2\text{-H}_2\text{O}$  systems is shown as a typical case in **Fig. 3.C.1.3**.<sup>50)</sup> As shown in this figure, a  $\text{UO}_2$  fuel rod is modeled to be a cylinder of 12.50 mm o.d. with 1441.5 mm height, clad with an aluminum cylinder of 14.17 mm o.d. with 0.76 mm thickness. For  $\text{PuO}_2\text{-UO}_2\text{-H}_2\text{O}$  systems, a fuel rod is modelled to be a cylinder of 10.65 mm o.d. and 706 mm height, clad with a zircaloy-2 cylinder of 12.23 mm o.d. with 0.70 mm thickness. The gap between pellet and cladding is replaced by fuel. Water reflector thickness around submerged lattices is assumed to be more than 40 cm in these calculational models.

Calculated  $k_{\text{eff}}$ 's for 125 cases in all are listed in **Tables 3.C.1.1** through **3.C.1.5**,<sup>50)</sup> respectively, corresponding to five experimental systems. Histograms of the calculated  $k_{\text{eff}}$ 's are shown in **Figs. 3.C.1.4** through **3.C.1.6**. The average values of these calculated  $k_{\text{eff}}$ 's are about 0.99 for all systems, in good agreement with experimental values.

#### 3.C.2 $\text{UO}_2$ Rod Clusters Partially Immersed in Water with Various Lattice Pitches and Patterns

Experiments were performed in France, simulating water storage of  $\text{UO}_2$  fuel assemblies with variation of lattice pitches and patterns.<sup>51)</sup> Uranium is enriched to 4.75 wt%  $^{235}\text{U}$ . The lattice pattern is either hexagonal or square as shown in **Figs. 3.C.2.1** and **3.C.2.2**. There are three types of  $\text{UO}_2$  rod clusters, i.e., without removal of rods, with systematic removal of rods and with removal of rods in a small zone. The experimental arrangement of fuel rods in water is shown in **Fig. 3.C.2.3**. Criticality approach was made by raising the water level incrementally.

The calculational model consists of a part of fuel assembly submerged in water, bottom

plugs, support plate, lower grid and water.

Calculated  $k_{eff}$ 's together with experimental conditions are listed in **Tables 3.C.2.1** through **3.C.2.3**. Histogram of the calculated  $k_{eff}$ 's is shown in **Fig. 3.C.2.4**. The average value of these 26  $k_{eff}$ 's is 0.990, about 1% lower than unity.

### **3.C.3 $\text{UO}_2$ Rod Clusters Fully Immersed in Water at Near-Optimum Neutron Moderation with Various Poison Plates**

Experiments were performed at BNWL, USA, simulating a spent fuel storage in water pools or shipping casks with various neutron absorbing materials.<sup>52),53)</sup> Three identical subcritical  $\text{UO}_2$  clusters aligned in a row were used in the experiment. Uranium in  $\text{UO}_2$  is enriched to either 2.35 wt%  $^{235}\text{U}$  or 4.31 wt%  $^{235}\text{U}$ . There are two types of  $\text{UO}_2$  rod, as are depicted in **Figs. 3.C.3.1** and **3.C.3.2**. The  $\text{UO}_2$  rods are arrayed in a water pool in a square pitch of 2.032 cm or 2.450 cm, providing near-optimum neutron moderation. The experimental arrangement is shown in **Fig. 3.C.3.3**. Critical separation  $X_c$ , as shown in this figure, was determined with various poison plate materials/thicknesses located between the clusters. The following materials are utilized for the poison plates in the experiment:

- Boral,
- 304L Stainless Steel with 0.0, 1.1 and 1.6 wt% Boron,
- Copper with 0.0 and 1.0 wt% Cadmium,
- Cadmium,
- Aluminum,
- Zircaloy-4.

The KENO-IV calculational model for the  $\text{UO}_2$  lattice cell is shown in **Fig. 3.C.3.4**.<sup>54)</sup> The calculational model for a typical experimental arrangements is shown in **Fig. 3.C.3.5**. As shown in this figure, the calculational model consists of three  $\text{UO}_2$  rod clusters submerged in water and two poison plates. Thickness of the water reflector surrounding the clusters is assumed to be more than 20 cm. Supporting materials such as 12.7 mm thick acrylic plates, aluminum angles, aluminum spacer rods and acrylic grid plates are all substituted by water to simplify the model.

Calculated  $k_{eff}$ 's together with critical experimental data are listed in **Tables 3.C.3.1** and **3.C.3.2**.<sup>54)</sup> Histograms of the calculated  $k_{eff}$ 's are shown in **Figs. 3.C.3.6** and **3.C.3.7**. The average value of these 75 calculated  $k_{eff}$ 's is 0.990, about 1% lower than unity, irrespective of the difference in the uranium enrichment.

### **3.C.4 $\text{UO}_2$ Rod Clusters Fully Immersed in Water at Near-Optimum Neutron Moderation with Reflecting Walls**

Experiments were performed at BNWL, USA, simulating spent fuel storage in water pool or shipping casks with reflecting walls.<sup>55)</sup> Three identical subcritical  $\text{UO}_2$  rod clusters aligned in a row were used in the experiment. Uranium in  $\text{UO}_2$  is enriched to either 2.35 wt%  $^{235}\text{U}$  or 4.31 wt%  $^{235}\text{U}$ . The  $\text{UO}_2$  rods are the same as depicted in **Figs. 3.C.3.1** and **3.C.3.2**. These  $\text{UO}_2$  rods are arrayed in a water pool in a square pitch of 2.032 cm or 2.450 cm to be a rectangular paralleliped cluster, providing near-optimum neutron moderation. The experimental arrangement of the clusters is shown in **Fig. 3.C.4.1**. As shown in this figure, lead or depleted uranium reflecting walls are set up on both sides of the three aligned clusters. The critical separation ( $X_c$ ) between the clusters was determined at various distances ( $Y$ ) between the reflecting walls and the side of the clusters.

The KENO-IV calculational model for two types of  $\text{UO}_2$  lattice cell is shown in **Fig.**

**3.C.3.4.** The calculational model for a typical experimental system is shown in **Fig. 3.C.4.2**. The calculational model consists of three UO<sub>2</sub> rod clusters submerged in water, lead or depleted uranium side reflectors and the water reflector of more than 20 cm thickness around the clusters. Supporting materials such as 12.7 cm thick acrylic plates, aluminum angles, aluminum spacer rods and acrylic grid plates are all substituted by water in the simplified model.

Calculated  $k_{eff}$ 's together with critical experimental data are listed in **Table 3.C.4.1**. Histogram of the calculated  $k_{eff}$ 's is shown in **Fig. 3.C.4.3**. The average value of these 23 calculated  $k_{eff}$ 's is 0.990, about 1% lower than unity, irrespective of the difference in the uranium enrichment and reflecting wall.

### 3.C.5 UO<sub>2</sub> Rod Clusters Fully Immersed in Water at Under-moderation with Various Poison Plates

Experiments were performed at BNWL, USA, simulating a spent fuel storage in water pool or shipping casks with various neutron absorbing materials.<sup>56)</sup> Three subcritical UO<sub>2</sub> rod clusters aligned in a row and four cross arranged subcritical UO<sub>2</sub> rod clusters are used in the experiment. Uranium in UO<sub>2</sub> is enriched to either 2.35 wt% <sup>235</sup>U or 4.31 wt% <sup>235</sup>U. The UO<sub>2</sub> rods are the same as those depicted in **Figs. 3.C.3.1** and **3.C.3.2**. These UO<sub>2</sub> rods are arrayed in a water pool in a square pitch of 1.684 cm or 1.892 cm to be a rectangular parallelepiped cluster, providing typical LWR fuel moderating conditions at under-moderation. For the experiments of three aligned clusters, the critical separation ( $X_c$ ), as shown in **Fig. 3.C.3.3**, was measured with various poison plate materials of various thicknesses inserted between each of the three clusters. In the experiments of four cross arranged clusters, the poison plates were attached to the cell boundaries on one side of the clusters, as shown in **Fig. 3.C.5.1**, and fuel rods were added, one by one, in the opposite side of the poison plates to reach criticality. The following materials are used for the poison plates:

- Aluminum,
- Zircaloy-4,
- 304L Stainless steel (0.0, 1.1 and 1.6 wt% Boron),
- Boral (A, B, C),
- Boroflex,
- Cadmium,
- Copper,
- Copper-Cadmium.

The KENO-IV calculational models for the two types of UO<sub>2</sub> lattice cells are shown in **Fig. 3.C.3.4**. The calculational model for a typical four cross arranged clusters is shown in **Fig. 3.C.5.2**. The calculational model for three aligned clusters is the same as that shown in **Fig. 3.C.3.5** except for the lattice pitch. The calculational model consists of three or four rod clusters submerged in water, two poison plates parallel or crossed, and the water reflector of more than 20 cm thickness around the clusters. Supporting materials such as 12.7 cm thick acrylic plates, aluminum angles, aluminum spacer rods and acrylic grid plates are all substituted by water in the simplified model.

Calculated  $k_{eff}$ 's together with critical experimental data are listed in **Tables 3.C.5.1** and **3.C.5.2**. Histograms of calculated  $k_{eff}$ 's are shown in **Figs. 3.C.5.3** and **3.C.5.4**. The average value of these 110 calculated  $k_{eff}$ 's is 0.995, about 0.5% lower than unity, irrespective of the difference in uranium enrichment and the poison plates.

### 3.C.6 UO<sub>2</sub> Rod Clusters Fully Immersed in Water at Under-moderation with Reflecting Walls

Experiments were performed at BNWL, USA, simulating a spent fuel storage in water pools or shipping casks with reflecting walls.<sup>57)</sup> Three identical subcritical UO<sub>2</sub> rod clusters aligned in a row were used in the experiment. Uranium of UO<sub>2</sub> is enriched to either 2.35 wt% or 4.31 wt% <sup>235</sup>U. The UO<sub>2</sub> rods are the same as those depicted in Fig. 3.C.3.1 or Fig. 3.C.3.2. These UO<sub>2</sub> rods are arrayed in a water pool in a square pitch of 1.684 cm or 1.892 cm to be a rectangular parallelepiped cluster, providing typical LWR neutron moderation. Lead or depleted uranium reflecting walls are set up on both sides of the three clusters. The critical separation ( $X_c$ ) is measured at various distances ( $Y$ ) between the reflecting wall and the side of the cluster, as shown in Fig. 3.C.4.1.

The KENO-IV calculational model for two types of UO<sub>2</sub> lattice cells is shown in Fig. 3.C.3.4. The calculational model for a typical experimental system is the same as that shown in Fig. 3.C.4.2 except for the lattice pitch. The calculational model consists of three UO<sub>2</sub> rod clusters submerged in water, lead or depleted uranium side reflectors and the water reflector of more than 20 cm thickness around the clusters. Supporting materials such as 12.7 cm thick acrylic plates, aluminum angles, aluminum spacer rods and acrylic grid plates are all substituted by water in the simplified model.

Calculated  $k_{eff}$ 's together with critical experimental data are listed in Table 3.C.6.1. Histogram of the calculated  $k_{eff}$ 's is shown in Fig. 3.C.6.1. The average value of these 22 calculated  $k_{eff}$ 's is 0.990, about 1% lower than unity, irrespective of the difference in the uranium enrichment and the reflecting wall.

### 3.C.7 UO<sub>2</sub> Rod Clusters Fully Immersed in Water at Various Pitches with Poison Plates and Reflecting Walls

Experiments were performed at BNWL, USA, simulating a spent fuel storage in water pools or shipping casks with poison plates and reflecting walls.<sup>58)</sup> Three subcritical UO<sub>2</sub> rod clusters aligned in a row were used in the experiment. Uranium in UO<sub>2</sub> is enriched to either 2.35 wt% or 4.31 wt% <sup>235</sup>U. The UO<sub>2</sub> rods are the same as those depicted in Fig. 3.C.3.1 or Fig. 3.C.3.2. These UO<sub>2</sub> rods are arrayed in a water pool in a square pitch of 1.684, 1.892, 2.032 and 2.540 cm to be a rectangular parallelepiped cluster. The critical separation  $X_c$ , as shown in Fig. 3.C.7.1, is measured with various poison plate materials of various thicknesses inserted between each of the three rod clusters and with steel reflecting walls set up on both sides of the three clusters. The following materials are used for the poison plates:

- 304L Stainless steel (0.0 and 1.1 wt% Boron),
- Boral B,
- Boroflex,
- Cadmium,
- Copper,
- Copper-Cadmium.

The KENO-IV calculational models for two types of UO<sub>2</sub> lattice cell are shown in Fig. 3.C.3.4. Calculational model for a typical experimental system is shown in Fig. 3.C.7.2. The calculational model consists of three rod clusters submerged in water, two poison plates, two steel side reflectors and the water reflector of more than 20 cm thickness around the clusters. Supporting materials such as 12.7 cm thick acrylic plates, aluminum angles, aluminum spacer rods and acrylic grid plates are all substituted by water in the simplified model.

Calculated  $k_{eff}$ 's together with critical experimental data are listed in Table 3.C.7.1.

Histograms of the calculated  $k_{eff}$ 's are shown in **Figs. 3.C.7.3** and **3.C.7.4**. The average values of the 49 calculated  $k_{eff}$ 's are 0.990 for 2.35 wt%  $^{235}\text{U}$  enriched  $\text{UO}_2$  systems and 0.992 for 4.31 wt%  $^{235}\text{U}$  enriched  $\text{UO}_2$  systems.

### 3.C.8 Triangular Lattice of Plutonium Alloy or Mixed Oxide Fuel Moderated by Borated Water

The US NUREG report<sup>25)</sup> contains 20 critical experiments concerning water moderated triangular lattices of plutonium alloy or mixed oxide fuel. These experiments were selected to provide data for benchmark calculations to evaluate the accuracy of neutron cross sections for the plutonium isotopes. They cover a wide range of H/Pu ratio, i.e., a wide variation in neutron spectrum. They also cover a wide range of  $^{240}\text{Pu}$  content as listed in **Table 3.C.8.1**. Fuel rods are arrayed in a water pool in a triangular lattice to be a cylindrical assembly. The assembly is surrounded by water of more than 20 cm thickness.

Since the configuration of fuel rods is not precisely given in the report, the fuel assembly is assumed to have a cylindrical periphery as perfect as possible in calculational models. Typical calculational models are shown in **Figs. 3.C.8.1** and **3.C.8.2**.

Calculated results are listed in **Table 3.C.8.2**.<sup>59)</sup> Histogram of the calculated  $k_{eff}$ 's for 20 cases in all is shown in **Fig. 3.C.8.3**. One can see that the calculation predicts an average value of 1.004, close to unity.

### 3.C.9 Mixed Oxide Fuel Lattice Moderated by Borated Water

Experiments were performed at EPRI, USA, at the Plutonium Recycle Facility (PRCF) to obtain the critical number of fuel rods in water moderator with various boron concentration.<sup>60)</sup> The PRCF is an experimental reactor generating low power (15 kW thermal), designed for studying neutronics phenomena in liquid moderated reactor cores. **Figure 3.C.9.1** shows an example of loading pattern of fuel pins. The fuel is either of 2.35 wt%  $^{235}\text{U}$  enriched  $\text{UO}_2$  or  $\text{UO}_2$  2 wt%  $\text{PuO}_2$  (8 wt%  $^{240}\text{Pu}$ ). Descriptions of these fuel rods are shown in **Figs. 3.C.9.2** and **3.C.9.3**. Lattice pitch is determined to provide under-moderation, near-optimum moderation or over-moderation for each of the fuel types.

In the KENO-IV calculational model, supporting structure is neglected, so that the calculational model consists of the lattice of fuel rods surrounded borated water as moderator or reflector.

Calculated results together with critical experimental data are listed in **Table 3.C.9.1**.<sup>36)</sup> The calculated  $k_{eff}$ 's for 12 cases in all range from 0.983 to 1.022 with the average value of just 1.000. The calculated  $k_{eff}$ 's for the  $\text{UO}_2\text{-PuO}_2$  fuel systems are about 1% larger on average than those for the  $\text{UO}_2$  fuel systems.

### 3.C.10 Water Storage of Power Reactor Fuel in Close Proximity

Experiments were performed at BAW, USA, supporting close proximity water storage of spent fuel assemblies of power reactor fuel.<sup>61)</sup> Uranium of  $\text{UO}_2$  is enriched to 2.46 wt%  $^{235}\text{U}$ . The  $\text{UO}_2$  fuel pins are arrayed in a square pitch of 1.636 cm to be a  $14 \times 14$  pin assembly. These fuel assemblies are arranged to make a  $3 \times 3$  square assembly array with assembly-to-assembly gaps of zero to four pin pitches, i.e., 0.0, 3.27, 4.91 and 6.54 cm. The  $3 \times 3$  array is set up in a water tank of 152.4 cm i.d., 198 cm height, 1.27 cm wall thickness. The following materials are inserted in the gaps between the fuel assemblies, providing various experimental loadings:

$\text{B}_4\text{C}$  filled pins,

Stainless steel sheets,  
Borated aluminum sheets.

An example of core loading patterns is shown in **Fig. 3.C.10.1**.

Typical calculational models are shown in **Figs. 3.C.10.2** and **3.C.10.3**. As shown in these figures, the experimental core loadings patterns are classified into two groups according to the supporting method of fuel pins. Structural materials such as aluminum top grids and aluminum core frames are neglected in the calculational model.

Calculated  $k_{eff}$ 's together with measured values for 21 cases in all are listed in **Table 3.C.10.1**.<sup>36)</sup> It is seen that the calculated  $k_{eff}$ 's range from 0.99 to 1.02 with the average value of 1.003. **Table 3.C.10.2** shows inter-comparison of the measured and calculated  $k_{eff}$ 's for the different spacings between the assemblies.

### 3.C.11 Array of UO<sub>2</sub>-BeO Fuel Pins in Uranyl-Nitrate Solution

Experiments were performed at ORNL, USA, simulating a dissolver in spent fuel reprocessing facility.<sup>62)</sup> A homogeneous mixture of UO<sub>2</sub>-BeO is compacted to ceramic pellets. Uranium is enriched to 62.4 wt% <sup>235</sup>U. The pellets are clad with Hastelloy 280 tubings of 0.952 cm o.d. and 196.8 cm length. The effective length of fuel is 143 cm. These fuel pins are arrayed in a square pitch in a moderator/reflector tank, filled with water for one series of experiments and with borated uranyl-nitrate solution for another series of experiments. Fuel pins are fixed by Plexiglas spacers and aluminum struts. The lattice pitch and its pattern are varied from case to case, as shown in **Fig. 3.C.11.1**. Criticality approach was made by increasing the aqueous moderator/reflector height in the tank.

The calculational model for the fuel pin is shown in **Fig. 3.C.11.2**. **Figure 3.C.11.3** shows calculational model for a water reflected system, and **Fig. 3.C.11.4** for a uranyl-nitrate solution system. In these models, the Plexiglas spacers and the aluminum struts are neglected.

Calculated results together with experimental conditions are listed in **Tables 3.C.11.1** and **3.C.11.2**.<sup>36)</sup> The average  $k_{eff}$  values are 0.965 for the 12 water reflected systems and 0.976 for the 6 uranyl-nitrate solution systems, about 3 to 4% lower than unity.

### 3.C.12 UO<sub>2</sub> Rod Clusters Fully Immersed in Borated Water

Experiments were performed at BNWL, USA, simulating a close proximity water storage of spent fuel assemblies.<sup>63)</sup> Uranium is enriched to 4.31 wt% <sup>235</sup>U. The schema of fuel rod is shown in **Fig. 3.C.12.1**. Maximum of 1237 UO<sub>2</sub> fuel rods are arrayed in a square pitch of 1.890 cm or 1.715 cm in a rectangular fashion. This fuel assembly is set up in a Plexiglas box of 79.7 cm width, 83.41 cm length and 116.8 cm height with 1.905 cm wall thickness. Borated water is poured into the box to reach a level of 15.2 cm above the top end of the fuel rods, as shown in **Fig. 3.C.12.2**. With the array width remaining constant as 40 or 44 rods depending on the pitch, the array length is increased symmetrically one by one to reach criticality. Critical experimental data such as the number of fuel rods required to attain criticality, are shown in **Table 3.C.12.1**.

The calculational model is shown in **Fig. 3.C.12.3**. Supporting materials such as upper and bottom polypropylene templates are neglected in the model. The KENO-IV calculation was performed by varying the number of neutrons per batch.

Calculated  $k_{eff}$ 's for 9 cases in all are listed in **Table 3.C.12.2**. Histograms of the calculated  $k_{eff}$ 's are shown in **Figs. 3.C.12.4** and **3.C.12.5**. The average value is 0.98, about 2% lower than unity, irrespective of the difference in the number of neutrons per batch.

### 3.C.13 UO<sub>2</sub> Rod Clusters in Sodium-Nitrate Solution

Experiments were performed in France, simulating UO<sub>2</sub> fuel assemblies in a dissolver of reprocessing plant.<sup>64)</sup> Maximum of 600 fuel rods are laid out in a square pitch lattice by means of a stainless steel basket made up of two spacing grids and a support plate. The square pitch is selected from 1.260 cm, 1.600 cm, 2.100 cm and 2.520 cm. Uranium is enriched to 4.74 wt% <sup>235</sup>U. The fuel assembly is put into an experimental tank. Sodium-nitrate solution, acting as a moderator/reflector, is introduced through an inlet into the bottom of the tank. The solution height is the critical approach parameter. The experimental set up is shown in Fig. 3.C.13.1.

The calculational model for a fuel rod is shown in Fig. 3.C.13.2. Figure 3.C.13.3 shows the calculational model for a typical experimental set up. Structural materials except for the lower grid and the support plate are neglected, as shown in this figure.

Calculated  $k_{eff}$ 's for 6 cases in all are listed in Table 3.C.13.1. The average value of the calculated  $k_{eff}$ 's is 0.986, about 1% lower than unity.

### 3.C.14 Triangular Lattice of UO<sub>2</sub> Fuel Moderated by Poisoned Water

Experiments were performed at BNWL, USA, using soluble neutron absorbers (poisons) to determine their effectiveness for criticality prevention of UO<sub>2</sub> fuel lattice assemblies.<sup>65)</sup> UO<sub>2</sub> rods enriched to 4.3 wt% <sup>235</sup>U are assembled in a triangular pitch in a water tank with soluble neutron absorbers such as boron, cadmium, and gadolinium. The triangular pitch is selected from 2.29, 2.79 and 3.30 cm. Experimental systems are set up as shown in Fig. 3.C.14.1. Criticality approach was done by adding fuel rods, one by one, on the cylindrical periphery of the assembly.

The calculational model is shown conceptually in Fig. 3.C.14.2. As shown in this figure, structural materials other than the lower lattice plate are neglected.

Table 3.C.14.1 lists calculated results together with experimental conditions for 20 cases in all. Histogram of the calculated  $k_{eff}$ 's is shown in Fig. 3.C.14.3. The calculated  $k_{eff}$ 's distribute between 0.955 and 1.035, with the average value of 0.989, about 1% lower than unity.

### 3.C.15 UO<sub>2</sub> Rod Clusters Partially Immersed in Uranyl-Nitrate Solution

Experiments were performed at BNWL, USA, to determine the effect of gadolinium as a soluble neutron absorber on the criticality of UO<sub>2</sub> fuel rod assemblies in uranyl-nitrate solution.<sup>66)</sup> Uranium in UO<sub>2</sub> is enriched to 4.3 wt% <sup>235</sup>U. The UO<sub>2</sub> rods are cylindrically arrayed in an experimental vessel in a triangular pitch selected from 2.29, 2.79 and 3.30 cm. Uranyl-nitrate solution with the same <sup>235</sup>U enrichment as that of UO<sub>2</sub> was subsequently added to approach criticality. Outside the experimental vessel is the water provided by a steel tank, as shown in Fig. 3.C.15.1.

The calculational model is shown in Fig. 3.C.15.1. Structural materials other than the experimental vessel are neglected in the calculational model.

Calculated results together with experimental conditions are listed in Table 3.C.15.1. Histogram of the calculated  $k_{eff}$ 's for all 12 cases is shown in Fig. 3.C.15.2. As seen from the histogram, the calculated  $k_{eff}$ 's are distributed between 0.895 and 0.975 with the average value of 0.938, about 6% lower than unity.

### 3.C.16 UO<sub>2</sub>-PuO<sub>2</sub> Rod Clusters Partially Immersed in Uranyl-Plutonium-Nitrate Solution with Soluble Poisons

Experiments were performed at BNWL, USA, simulating spent fuel elements in a

dissolver vessel.<sup>38)</sup> Total of 301 UO<sub>2</sub>-PuO<sub>2</sub> mixed oxide fuels are assembled to be a triangular lattice array with cylindrical boundary, and positioned in a 55.5 cm i.d. stainless steel vessel. The vessel is, in turn, included in a larger cylindrical tank, providing a water reflector. The lattice pitch is specified to be 3.048 cm. Subsequently, plutonium-uranyl-nitrate solution, containing gadolinium and/or boron at various concentrations as soluble neutron absorbers, is added to the loaded lattice assembly to obtain criticality. **Figure 3.C.16.1** gives schema of the experimental system, showing water reflector, vents, dump lines, etc.

In the calculational model, the stainless steel vessel containing fuel rods and solution together with water reflector is taken into consideration, but other structural materials are neglected, as shown in **Fig. 3.C.16.2**. The vessel diameter is assumed to be 67.16 cm, since the diameter of 55.5 cm specified in the experiment report cannot accomodate 301 fuel rods in the specified pitch. The MULTI-KENO code, a modified version of the KENO-IV code, was utilized, since cylindrical boundary, in which “CUBOID” type fuels are arrayed, cannot be treated easily by the KENO-IV code. The reason is that, in this case, intersection, which is not allowed in KENO-IV, is occured between the cylindrical boundary and the fuel array.

**Table 3.C.16.1** lists calculated results together with experimental conditions. Histogram of the calculated  $k_{eff}$ 's for 13 cases in all is shown in **Fig. 3.C.16.3**. As seen from the histogram, the calculated  $k_{eff}$ 's are between 1.00 and 1.025 with the average value of 1.010, about 1% higher than unity.

**Table 3.A.1.1** Experimentally determined critical heights for 30.6 H/(Pu+U) fuel systems

Serial No.	Neutron poison plate		Plate mass (kg)	Plate thickness (mm)	Plate+Void thickness (mm)	Layers of fuel		Total layers of fuel (50.9 mm)
	Layers of fuel (50.9 mm)	Type				(50.9 mm)	(13.84 mm)	
001	3	None (g)	0	0	--	1	3.578	4.921 ± 0.006
002	3	Steel (g)	3.902 ± 0.010	3.10 ± 0.05	3.5 ± 0.5	2	0.781	5.201 ± 0.002
003	3	Steel (g)	7.804 ± 0.014	6.20 ± 0.07	6.9 ± 0.8	2	1.655	5.426 ± 0.002
004	3	Steel (g)	19.510 ± 0.022	15.50 ± 0.11	15.9 ± 0.5	2	3.667	5.944 ± 0.505
005	3	Steel (g)	31.216 ± 0.028	24.80 ± 0.14	25.2 ± 0.5	3	1.317	6.339 ± 0.003
006	3	Steel (g)	46.824 ± 0.034	37.20 ± 0.17	38.0 ± 1.0	3	2.770	6.713 ± 0.003
007	3	Steel (g)	62.432 ± 0.040	49.64 ± 0.19	51.2 ± 0.8	4	0.012	7.003 ± 0.004
008	3	1.1WT% B-Steel	3.908 ± 0.028	2.98 ± 0.06	3.8 ± 0.9	3	1.251	6.3221 ± 0.004
009	3	1.1WT% B-Steel	7.816 ± 0.040	5.95 ± 0.08	7.0 ± 0.8	3	2.937	6.756 ± 0.006
010	3	1.1WT% B-Steel	15.632 ± 0.056	11.91 ± 0.11	13.2 ± 0.7	4	0.866	7.2231 ± 0.006
011	3	1.1WT% B-Steel	23.448 ± 0.069	17.86 ± 0.14	18.9 ± 0.6	4	1.663	7.4281 ± 0.002
012	3	1.1WT% B-Steel	42.998 ± 0.094	32.75 ± 0.18	35.1 ± 0.5	4	2.813	7.7241 ± 0.006
013	3	1.6WT% B-Steel	3.834 ± 0.008	2.98 ± 0.05	3.4 ± 0.5	3	2.074	6.534 ± 0.004
014	3	1.6WT% B-Steel	3.834 ± 0.008	2.98 ± 0.05	3.4 ± 0.5	3	2.024	6.521 ± 0.005
015	3	1.6WT% B-Steel	15.336 ± 0.017	11.94 ± 0.11	12.8 ± 1.0	4	1.336	7.3441 ± 0.003
016	3	1.6WT% B-Steel	30.672 ± 0.024	23.87 ± 0.15	25.6 ± 1.9	4	2.661	7.6851 ± 0.005
017	3	Uranium	20.165 ± 0.021	6.58 ± 0.13	6.7 ± 0.2	2	1.445	5.3721 ± 0.002
018	3	Uranium	61.370 ± 0.030	19.53 ± 0.18	20.2 ± 0.5	2	3.586	5.9231 ± 0.002
019	3	Uranium	141.195 ± 0.042	45.05 ± 0.24	46.5 ± 0.7	3	2.762	6.7111 ± 0.003
020	3	Uranium	182.370 ± 0.047	57.99 ± 0.30	60.1 ± 1.5	4	0.109	7.0281 ± 0.004
021	3	Boral	1.552 ± 0.002	3.68 ± 0.06	4.2 ± 0.6	4	0.967	7.2491 ± 0.005
022	3	Boral	3.142 ± 0.002	7.52 ± 0.08	8.2 ± 0.7	4	2.479	7.6381 ± 0.005
023	3	Boral	7.840 ± 0.002	18.71 ± 0.12	19.5 ± 0.8	5	0.047	8.0121 ± 0.006
024	3	Boral	10.907 ± 0.002	25.98 ± 0.15	26.8 ± 0.8	5	0.548	8.1411 ± 0.006
025	3	Boral	15.393 ± 0.002	36.60 ± 0.18	38.2 ± 0.8	5	3.885	8.2461 ± 0.006
026	3	Cadmium	1.442 ± 0.001	1.058 ± 0.005	1.28 ± 0.2	3	1.736	6.4471 ± 0.004
027	3	Cadmium	2.847 ± 0.001	2.079 ± 0.005	2.30 ± 0.2	3	2.350	6.6051 ± 0.005
028	3	Lead	12.231 ± 0.021	6.44 ± 0.03	6.5 ± 0.1	2	0.190	5.0491 ± 0.002
029	3	Lead	30.496 ± 0.030	16.02 ± 0.02	16.1 ± 0.1	2	0.921	5.2371 ± 0.002
030	3	Lead	61.386 ± 0.036	32.00 ± 0.02	32.8 ± 0.5	2	1.989	5.5121 ± 0.003
031	3	Lead	86.136 ± 0.042	44.86 ± 0.02	45.7 ± 0.8	2	2.984	5.7681 ± 0.003
032	3	Lead	110.896 ± 0.047	57.71 ± 0.03	58.0 ± 0.3	3	0.062	6.0161 ± 0.002

**Table 3.A.1.2** Experimentally determined critical heights for 2.8 H/(Pu+U) fuel (partly replaced with 30.6 H/(Pu+U) fuel) systems

Serial no.	Neutron poison plate		Plate mass (kg)	Plate thickness (mm)	Plate+Void thickness (mm)	Layers of fuel		Total layers of fuel (50.9 mm)
	Layers of 2.8H:Pu+U fuel (50.9 mm)	Type				(50.9 mm)	(50.9 mm)	
033	2	None	0	0	--	2	2	0.233
034	2	Steel	19.510 ± 0.022	15.50 ± 0.11	16.4 ± 0.5	2	2	1.309
035	2	Steel	42.922 ± 0.033	34.11 ± 0.16	34.8 ± 0.3	2	2	6.590 ± 0.003
036	2	1.1WT% B-Steel	3.908 ± 0.028	2.98 ± 0.06	3.1 ± 0.3	2	2	0.995
037	2	1.1WT% B-Steel	19.540 ± 0.063	14.88 ± 0.12	15.8 ± 0.1	2	2	2.754
038	2	1.1WT% B-Steel	42.988 ± 0.094	32.75 ± 0.18	34.7 ± 1.5	2	2	3.966
039	2	1.6WT% B-Steel	3.834 ± 0.008	2.98 ± 0.005	3.2 ± 0.3	2	2	1.173
040	2	1.6WT% B-Steel	30.672 ± 0.024	23.87 ± 0.15	25.2 ± 1.2	2	2	3.733
041	2	Uranium	20.165 ± 0.021	6.58 ± 0.13	7.1 ± 0.7	2	2	0.761
042	2	Uranium	61.370 ± 0.030	19.53 ± 0.18	20.1 ± 0.7	2	2	1.472
043	2	Uranium	182.370 ± 0.047	57.99 ± 0.30	59.8 ± 0.7	2	2	3.302
044	2	Boral	1.552 ± 0.002	3.68 ± 0.06	3.9 ± 0.3	2	2	2.300
045	2	Boral	4.747 ± 0.002	11.38 ± 0.10	11.7 ± 0.4	2	2	3.908
046	2	Boral	15.393 ± 0.002	36.60 ± 0.18	39.2 ± 1.9	2	2	1.670
047	2	Cadmium	2.847 ± 0.001	2.079 ± 0.005	2.30 ± 0.2	2	2	1.049
048	2	Lead	18.265 ± 0.021	9.58 ± 0.03	10.9 ± 1.4	2	2	0.719
049	2	Lead	49.265 ± 0.030	25.55 ± 0.04	26.3 ± 0.8	2	2	1.395
050	2	Lead	79.761 ± 0.042	41.58 ± 0.06	43.2 ± 1.4	2	2	1.872
051	2	Aluminum	11.910 ± 0.056	26.78 ± 0.06	27.1 ± 0.4	2	2	1.620

**Table 3.A.1.3** Calculated  $k_{eff}$ 's for 30.6 H/(Pu+U) fuel systems

No.	$k_{eff}$	$\sigma$	No.	$k_{eff}$	$\sigma$						
None											
001	1.0248	0.0054	017	1.0215	0.0046						
Steel			018	1.0206	0.0049						
002	1.0205	0.0047	019	1.0206	0.0050						
003	1.0260	0.0046	020	1.0276	0.0044						
004	1.0181	0.0044	Average	1.0236							
005	1.0214	0.0047	Borax								
006	1.0210	0.0043	021	1.0276	0.0049	033	1.0212	0.0047			
007	1.0246	0.0047	022	1.0285	0.0050	034	1.0224	0.0045			
Average	1.0219		023	1.0266	0.0046	035	1.0239	0.0037			
1.1 wt% boron steel						Average	1.0231				
008	1.0197	0.0049	024	1.0332	0.0048	Cadmium					
009	1.0285	0.0047	025	1.0189	0.0053	1.1 wt% boron steel	1.0279	0.0043			
Average	1.0206		Average	1.0269		036	1.0240	0.0043			
010	1.0186	0.0050	Lead								
011	1.0138	0.0045	026	1.0121	0.0051	037	1.0284	0.0043			
012	1.0226	0.0047	027	1.0214	0.0048	038	1.0281	0.0044			
Average	1.0206		Average	1.0167		Average	1.0268	0.0051			
1.6 wt% boron steel						Cadmium					
013	1.0213	0.0045	028	1.0316	0.0048	039	1.0276	0.0046			
014	1.0251	0.0051	029	1.0287	0.0046	040	1.0233	0.0049			
015	1.0152	0.0049	030	1.0294	0.0055	Average	1.0254				
016	1.0139	0.0039	031	1.0274	0.0046	041	1.0320	0.0047			
Average	1.0189		032	1.0184	0.0047	042	1.0303	0.0043			
Average						043	1.0252	0.0043			
Average						Average	1.0292				
Average						Uranium					

**Table 3.A.1.4** Calculated  $k_{eff}$ 's for 2.8 H/(Pu+U) fuel (partly replaced with 30.6 H/(Pu+U) fuel) systems

No.	$k_{eff}$	$\sigma$	No.	$k_{eff}$	$\sigma$
None					
033	1.0212	0.0047	044	1.0274	0.0046
Steel			045	1.0160	0.0042
034	1.0224	0.0045	046	1.0292	0.0047
Average	1.0239	0.0037	Average	1.0242	
Cadmium					
047	1.0329	0.0043	048	1.0283	0.0043
Lead			049	1.0271	0.0043
050	1.0254		050	1.0208	0.0051
Aluminum					
051	1.0284	0.0044	051	1.0284	0.0044

**Table 3.A.2.1** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 29.3 wt% Pu and 2.8 H/(Pu+U)

No.	Reflector	Critical Number of Fuel Compacts					
		Length (5.090 cm)	Width (5.083 cm)	Height		Corrected Height (5.090 cm)	KENO $k_{eff} \pm \sigma$
				(5.090 cm)	(1.339 cm)		
1	None	10	10	8	3.768 ± 0.008	8.888 ± 0.019	0.9895 ± 0.0036
2	None	10	11	8	1.510 ± 0.002	8.252 ± 0.004	0.9845 ± 0.0045
3	Plexiglas	7	7	7	0.646 ± 0.001	7.060 ± 0.010	1.0085 ± 0.0036
4	Plexiglas	8	8	5	2.548 ± 0.004	5.615 ± 0.008	0.9976 ± 0.0042
5	Plexiglas	9	9	4	4.192 ± 0.011	4.997 ± 0.013	1.0112 ± 0.0045
6	Plexiglas	10	10	4	1.300 ± 0.002	4.316 ± 0.007	1.0021 ± 0.0044
7	Plexiglas	10	10	4	1.297 ± 0.003	4.308 ± 0.009	1.0036 ± 0.0041
8	Plexiglas	12	10	4	0.038 ± 0.001	3.948 ± 0.006	1.0015 ± 0.0041
9	Plexiglas	12	12	3	2.791 ± 0.004	3.668 ± 0.005	1.0024 ± 0.0039
10	Plexiglas	12	13	3	2.407 ± 0.005	3.578 ± 0.006	1.0019 ± 0.0046
11	Plexiglas	14	13	3	2.068 ± 0.007	3.487 ± 0.009	1.0068 ± 0.0036
12	Plexiglas	∞	∞	—	---	2.540 ± 0.028	1.0070 ± 0.0042
						Average	1.0014

**Table 3.A.2.2** Critical assembly configurations and calculated  $k_{eff}$  for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 15.0 wt% Pu and 2.86 H/(Pu+U)

No.	Reflector	Critical Number of Fuel Compacts					
		Length (5.090 cm)	Width (5.090 cm)	Height		Corrected Height (5.082 cm)	KENO $k_{eff} \pm \sigma$
				(5.082 cm)	(1.265 cm)		
13	Plexiglas	10	10	9	6.651 ± 0.016	10.749 ± 0.031	0.9955 ± 0.0042

**Table 3.A.2.3** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 8.1 wt% Pu and 7.3 H/(Pu+U)

No.	Reflector	Critical Number of Fuel Compacts					
		Length (5.090 cm)	Width (5.090 cm)	Height		Corrected Height (5.081 cm)	KENO $k_{eff} \pm \sigma$
				(5.081 cm)	(1.274 cm)		
14	Plexiglas	10	9	9	1.135 ± 0.001	9.274 ± 0.003	1.0220 ± 0.0046
15	Plexiglas	10	11	7	3.239 ± 0.001	7.780 ± 0.002	1.0197 ± 0.0046
16	Plexiglas	12	12	6	2.526 ± 0.003	6.609 ± 0.007	1.0119 ± 0.0042
17	Plexiglas	14	13	5	3.733 ± 0.005	5.919 ± 0.008	1.0154 ± 0.0041
						Average	1.017

**Table 3.A.2.4** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 30.0 wt% Pu and 47.7 H/(Pu+U)

No.	Reflector	Critical Dimensions, cm			Critical Mass		$k_{eff}$	$\pm \sigma$
		Length	Width	Height	kg of Pu	kg of U		
1	Plexiglas	30.54±0.03	30.54±0.03	30.89±0.02	3.23±0.05	7.53±0.12	1.00370	0.00466
2	Plexiglas	35.63±0.04	35.63±0.04	23.95±0.10	3.40±0.05	7.94±0.12	1.00268	0.00493
3	Plexiglas	40.72±0.04	40.72±0.04	20.22±0.05	3.75±0.05	8.76±0.13	1.01852	0.00496
4	Plexiglas	50.90±0.05	45.81±0.05	17.14±0.07	4.48±0.06	10.44±0.16	1.00809	0.00458
5	Plexiglas	61.08±0.06	50.90±0.05	15.53±0.11	5.44±0.08	12.69±0.20	1.01107	0.00537
6	Plexiglas	61.08±0.06	55.99±0.06	15.16±0.02	5.80±0.09	13.51±0.21	1.01092	0.00550
7	Plexiglas	66.17±0.06	61.08±0.06	14.43±0.10	6.53±0.09	15.24±0.22	1.01265	0.00453
8	Plexiglas	50.90±0.05	50.90±0.05	16.49±0.04	4.78±0.08	11.14±0.18	1.00578	0.00490
9	Plexiglas	30.60±0.02	30.60±0.02	30.60±0.02	3.21±0.05	7.49±0.12	1.00305	0.00517
10	Plexiglas	∞	∞	10.80±0.11	—	—	1.02113	0.00423
11	Bare	45.81±0.05	40.72±0.04	37.98±0.06	7.93±0.12	18.51±0.78	0.98832	0.00651
12	Bare	40.72±0.04	40.72±0.04	42.24±0.03	7.84±0.11	18.30±0.77	0.99719	0.00579
13	Bare	45.81±0.05	50.90±0.05	32.49±0.02	8.48±0.12	19.80±0.84	0.99243	0.00548
14	Bare	41.20±0.05	41.20±0.05	41.20±0.05	7.83±0.11	18.28±0.77	0.98490	0.00572

**Table 3.A.2.5** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 14.62 wt% Pu and 30.6 H/(Pu+U)

No.	Reflector	Critical Dimensions, cm			Critical Mass		$k_{eff}$	$\pm \sigma$
		Length	Width	Height	kg of Pu	kg of U		
15	Plexiglas	30.54±0.03	40.72±0.04	29.81±0.15	3.14±0.04	18.35±0.35	1.01453	0.00558
16	Plexiglas	40.72±0.04	40.72±0.04	23.84±0.10	3.35±0.04	19.57±0.37	1.02512	0.00511
17	Plexiglas	45.81±0.04	50.90±0.05	19.82±0.11	3.92±0.05	22.89±0.44	1.01600	0.00513
18	Plexiglas	50.90±0.05	50.90±0.05	18.92±0.09	4.16±0.05	24.28±0.46	1.01860	0.00484
19	Plexiglas	61.08±0.06	50.90±0.05	17.72±0.09	4.67±0.06	27.28±0.52	1.02585	0.00490
20	Plexiglas	61.08±0.06	61.08±0.06	16.63±0.09	5.26±0.06	30.73±0.59	1.01231	0.00508
21	Plexiglas	33.30±0.17	33.30±0.17	33.30±0.17	3.13±0.05	18.28±0.40	1.01960	0.00501
22	Plexiglas	∞	∞	11.56±0.09	—	—	1.00941	0.00447
23	Bare	40.72±0.04	40.76±0.17	52.39±0.07	7.37±0.09	43.06±0.82	1.01296	0.00475
24	Bare	40.72±0.04	45.86±0.19	45.10±0.06	7.14±0.08	41.70±0.79	1.00897	0.00495
25	Bare	50.90±0.005	45.86±0.19	36.99±0.05	7.32±0.08	42.75±0.81	1.01258	0.00497
26	Bare	43.78±0.07	43.78±0.07	43.78±0.07	7.12±0.08	41.55±0.80	1.01379	0.00484

**Table 3.A.2.6** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 7.89 wt% Pu and 51.8 H/(Pu+U)

No.	Reflector	Critical Dimensions, cm			Critical Mass		$k_{eff}$	$\pm \sigma$
		Length	Width	Height	kg of Pu	kg of U		
27	Plexiglas	40.72±0.04	45.72±0.09	32.89±0.12	1.74±0.01	20.29±0.22	1.02113	0.00429
28	Plexiglas	50.90±0.05	50.80±0.10	26.40±0.10	1.94±0.02	22.63±0.25	1.01579	0.00453
29	Plexiglas	61.08±0.06	60.96±0.12	22.66±0.08	2.39±0.02	27.88±0.31	1.01035	0.00413
30	Plexiglas	61.08±0.06	66.04±0.13	22.09±0.18	2.53±0.03	29.53±0.39	1.01510	0.00442
31	Plexiglas	61.08±0.06	55.88±0.11	23.22±0.08	2.25±0.02	26.26±0.28	1.00651	0.00470
32	Plexiglas	61.08±0.06	50.80±0.10	24.37±0.09	2.15±0.02	25.06±0.28	1.02532	0.00414
33	Plexiglas	40.72±0.05	40.64±0.08	36.42±0.13	1.71±0.01	19.97±0.22	1.01814	0.00455
34	Plexiglas	39.12±0.28	39.12±0.28	39.12±0.28	1.70±0.01	19.84±0.32	1.01560	0.00440
35	Plexiglas	∞	∞	14.83±0.60	—	—	0.98698	0.00438

**Table 3.A.3.1** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 23.1 atomic plutonium enrichment percent and 4.67 H/(Pu+U) atomic ratio

No.	Base Side (cm)	Height (cm)	Critical Mass (kg)			Calculatrd $k_{eff} \pm \sigma$
			Oxide	Pu	Pu-239	
1	25.7	32.7	92.4	20.5	19.2	1.0158 ± 0.0042
2	30.0	24.3	94.2	20.9	19.6	1.0169 ± 0.0036
3	34.3	20.7	104.9	23.2	21.7	1.0189 ± 0.0048
4	30.0 × 21.5	37.5	104.1	23.0	21.6	1.0215 ± 0.0044
			Average			1.0183

**Table 3.A.4.1** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 2.2 wt% <sup>240</sup>Pu and those of 8.0 wt% <sup>240</sup>Pu

No.	Reflector	Critical Dimensions, cm			Critical Mass, kg of Pu		Calculated $k_{eff} \pm \sigma$
		Length	Width	Height	Experiment	Corrected	
		2.2 % <sup>240</sup> Pu Compacts					
1	Plexiglas	51.69	46.13	9.04 $\pm$ 0.06	23.39 $\pm$ 0.17	24.14 $\pm$ 0.18	1.0273 $\pm$ 0.0046
2	Plexiglas	41.35	38.46	10.34 $\pm$ 0.05	17.84 $\pm$ 0.09	18.42 $\pm$ 0.09	1.0201 $\pm$ 0.0048
3	Plexiglas	31.01	31.01	13.13 $\pm$ 0.02	13.70 $\pm$ 0.02	14.14 $\pm$ 0.02	1.0280 $\pm$ 0.0052
4	Plexiglas	25.86	25.86	16.43 $\pm$ 0.06	11.92 $\pm$ 0.04	12.30 $\pm$ 0.04	1.0218 $\pm$ 0.0050
5	Plexiglas	23.27	23.27	19.79 $\pm$ 0.05	11.63 $\pm$ 0.03	12.01 $\pm$ 0.03	1.0203 $\pm$ 0.0055
6	Plexiglas	20.68	20.68	24.87 $\pm$ 0.06	11.54 $\pm$ 0.03	11.91 $\pm$ 0.03	1.0193 $\pm$ 0.0046
7	Plexiglas	15.52	18.08	50.04 $\pm$ 0.13	15.23 $\pm$ 0.04	15.72 $\pm$ 0.04	1.0205 $\pm$ 0.0047
8	Bare	31.01	31.01	31.87 $\pm$ 0.06	33.25 $\pm$ 0.06		1.0017 $\pm$ 0.0052
9	Bare	30.68	30.68	32.84 $\pm$ 0.52		34.62 $\pm$ 0.55	1.0052 $\pm$ 0.0050
		8.0 % <sup>240</sup> Pu Compacts					
10	Plexiglas	51.31	68.25	10.36 $\pm$ 0.03	37.19 $\pm$ 0.11	38.09 $\pm$ 0.11	1.0078 $\pm$ 0.0045
11	Plexiglas	35.92	35.92	15.42 $\pm$ 0.03	20.39 $\pm$ 0.04	20.89 $\pm$ 0.04	1.0038 $\pm$ 0.0053
12	Plexiglas	30.78	30.78	18.56 $\pm$ 0.02	18.07 $\pm$ 0.02	18.46 $\pm$ 0.02	1.0040 $\pm$ 0.0045
13	Plexiglas	25.65	25.65	25.03 $\pm$ 0.05	16.88 $\pm$ 0.03	17.29 $\pm$ 0.03	1.0041 $\pm$ 0.0051
14	Plexiglas	25.65	25.65	25.13 $\pm$ 0.03	16.95 $\pm$ 0.02	17.36 $\pm$ 0.02	1.0053 $\pm$ 0.0044
15	Plexiglas	20.52	20.52	49.15 $\pm$ 0.01	21.22 $\pm$ 0.01	21.73 $\pm$ 0.01	1.0162 $\pm$ 0.0043
16	Bare	35.92	35.92	35.54 $\pm$ 0.02	46.93 $\pm$ 0.03		0.9798 $\pm$ 0.0050
17	Bare	35.63	35.63	36.04 $\pm$ 0.60		48.04 $\pm$ 0.80	0.9865 $\pm$ 0.0050

**Table 3.A.4.2** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene fuel compacts of 11.46 wt% <sup>240</sup>Pu and 5.0 H/Pu

No.	Reflector	Critical Dimensions, cm			Critical Mass, kg of Pu		Calculated $k_{eff} \pm \sigma$
		Length	Width	Height	Experi- ment	Corrected	
		11.46 wt% <sup>240</sup> Pu					
1	Plexiglas	25.88	25.88	19.04 $\pm$ 0.01	27.66 $\pm$ 0.01	27.66 $\pm$ 0.01	1.0184 $\pm$ 0.0038
2	Plexiglas	31.24	30.96	14.77 $\pm$ 0.02	30.94 $\pm$ 0.04	30.94 $\pm$ 0.04	1.0231 $\pm$ 0.0047
3	Plexiglas	41.66	41.28	11.03 $\pm$ 0.01	41.20 $\pm$ 0.03	41.20 $\pm$ 0.03	1.0180 $\pm$ 0.0048
4	Plexiglas	52.07	51.60	9.38 $\pm$ 0.01	54.78 $\pm$ 0.02	54.78 $\pm$ 0.02	1.0209 $\pm$ 0.0043
5	Bare	36.59	36.11	28.60 $\pm$ 0.05	82.88 $\pm$ 0.13	---	0.9998 $\pm$ 0.0038
6	Bare	35.84	35.84	28.82 $\pm$ 0.05	---	85.03 $\pm$ 0.2	1.0161 $\pm$ 0.0042
		Average					1.0160

**Table 3.A.4.3** Critical assembly configurations and calculated  $k_{eff}$ 's for PuO<sub>2</sub> fuel compacts of 18.35 wt% <sup>240</sup>Pu and 0.04 H/Pu

No.	Reflector	Critical Dimensions (cm)			Critical Mass (kg Pu)	Calculated $k_{eff} \pm \sigma$
		Length	Width	Height		
		18.35 wt% <sup>240</sup> Pu				
1	Plexiglas	25.65	25.65	10.03 $\pm$ 0.01	38.3 $\pm$ 0.3	1.0344 0.0048
2	Plexiglas	25.65	30.78	8.98 $\pm$ 0.01	41.1 $\pm$ 0.3	1.0262 $\pm$ 0.0046
3	Plexiglas	30.78	30.78	7.97 $\pm$ 0.03	43.8 $\pm$ 0.3	1.0252 $\pm$ 0.0053
4	Plexiglas	30.78	41.05	6.86 $\pm$ 0.0	40.3 $\pm$ 0.7	1.0282 $\pm$ 0.0051
5	Plexiglas	41.05	41.05	5.95 $\pm$ ----	58.2 $\pm$ 0.3	1.0130 $\pm$ 0.0053
6	Bare	30.78	30.78	20.90 $\pm$ 0.0	113.8 $\pm$ 0.9	1.0247 $\pm$ 0.0049
7	Bare	30.78	30.78	21.12 $\pm$ 0.06	113.1 $\pm$ 0.8	1.0350 $\pm$ 0.0042
		Average			1.0267	

**Table 3.A.5.1** Calculated  $k_{eff}$ 's for the minimally reflected single tank

No.	Tank		Location of tank	Uranium concentration (g/l)	Critical solution level(cm)	Calculated $k_{eff}$
	Material	Inner dia.(cm)				
1	Stainless steel	27.92	At center	145.68	31.20	$0.9602 \pm 0.0058$
2	"	27.92	"	346.73	28.93	$0.9654 \pm 0.0061$
3	Al	28.01	"	142.92	33.55	$0.9721 \pm 0.0060$
4	"	28.01	"	357.71	30.91	$0.9702 \pm —$
5	"	33.01	"	54.89	39.48	$0.9609 \pm 0.0054$
6	"	33.01	"	59.65	36.67	$0.9831 \pm 0.0054$
7	"	33.01	"	137.40	23.96	$0.9687 \pm 0.0058$
8	"	33.01	"	145.68	23.67	$0.9734 \pm 0.0062$
9	"	33.01	"	357.71	22.57	$0.9712 \pm 0.0057$
10	"	50.69	"	63.95	20.48	$0.9721 \pm 0.0048$
				Average	0.9697	

**Table 3.A.5.2** Calculated  $k_{eff}$ 's for the concrete-reflected single tank

No.	Tank		Location of tank	Uranium concentration (g/l)	Critical solution level(cm)	Calculated $k_{eff}$
	Material	Inner dia.(cm)				
11	Stainless steel	27.92	At center	144.38	29.79	$0.9691 \pm 0.0058$
12	"	"	"	334.77	27.23	$0.9642 \pm 0.0058$
13	Al	28.01	"	144.38	31.37	$0.9768 \pm 0.0063$
14	"	"	"	334.77	28.60	$0.9790 \pm —$
15	"	33.01	"	59.65	34.01	$0.9657 \pm 0.0051$
16	"	"	"	144.38	22.85	$0.9833 \pm 0.0058$
17	"	"	"	334.77	21.50	$0.9733 \pm 0.0057$
18	Stainless steel	27.92	In corner	144.38	24.19	$0.9753 \pm 0.0056$
19	"	"	"	334.77	21.79	$0.9766 \pm 0.0062$
20	Al	28.01	"	144.38	24.70	$0.9947 \pm 0.0061$
21	"	"	"	334.77	22.33	$0.9764 \pm 0.0059$
22	"	33.01	"	59.65	27.27	$0.9866 \pm 0.0053$
23	"	"	"	144.38	18.24	$0.9822 \pm 0.0052$
24	"	"	"	334.77	16.78	$0.9779 \pm 0.0060$
				Average	0.9772	

**Table 3.A.5.3** Calculated  $k_{eff}$ 's for the plastic-reflected single tank

No.	Tank		Location of tank	Uranium concentration (g/l)	Critical solution level (cm)	Calculated $k_{eff}$
	Material	Inner dia.(cm)				
25	Stainless steel	27.92	At center	147.66	29.71	$0.9674 \pm 0.0056$
26	"	"	"	345.33	27.60	$0.9615 \pm 0.0055$
27	Al	27.88	"	60.32	78.1	$0.9746 \pm —$
28	"	28.01	"	147.66	31.26	$0.9756 \pm 0.0058$
29	"	"	"	345.33	28.84	$0.9691 \pm 0.0063$
30	"	33.01	"	60.32	34.33	$0.9775 \pm 0.0047$
31	"	"	"	147.66	22.78	$0.9773 \pm 0.0060$
32	"	"	"	345.33	21.67	$0.9785 \pm 0.0056$
33	Stainless steel	27.93	In corner	60.32	50.52	$0.9734 \pm 0.0052$
34	"	27.92	"	147.66	25.03	$0.9650 \pm 0.0057$
35	"	"	"	345.33	22.75	$0.9697 \pm 0.0060$
36	Al	27.88	"	60.32	51.67	$0.9843 \pm 0.0048$
37	"	28.01	"	147.66	25.26	$0.9683 \pm 0.0050$
38	"	"	"	345.33	22.87	$0.9711 \pm 0.0067$
39	"	33.01	"	60.32	27.70	$0.9683 \pm 0.0046$
40	"	"	"	66.33	25.10	$0.9639 \pm 0.0051$
41	"	"	"	147.66	18.49	$0.9851 \pm 0.0061$
42	"	"	"	345.33	17.20	$0.9735 \pm 0.0051$
43	Stainless steel	27.93	Centered on floor	60.32	67.48	$0.9707 \pm 0.0052$
44	Al	27.88	"	"	77.1	$0.9787 \pm —$
45	"	33.01	"	"	31.75	$0.9691 \pm 0.0054$
					Average	0.9725

**Table 3.A.6.1** Summary of characteristics of homogeneous plutonium benchmark experiments

No.	H/Pu atom ratio	% <sup>240</sup> Pu	Pu density ,g/cm	Geometry
1	3695	0	0.007	Infinite
2	125	5	0.172	Sphere
3	980	1	0.026	Sphere
4	758	5	0.034	Sphere
5	15	2	1.12	Parallelepiped
6	0	5	15.6	Sphere
7	422	5	0.058	Slab (Infinite)
8	910	14	0.028	Cylinder
9	50	18	0.37	Parallelepiped
10	210	8	0.116	Cylinder
11	0	18	5.8	Parallelepiped
12	5	11	2.3	Parallelepiped
13	0	20	15.7	Sphere
14	623	43	0.041	Cylinder

**Table 3.A.6.2** Calculated results for homogeneous plutonium benchmark experiments

No.	H/Pu atom ratio	% $^{240}\text{Pu}$	Pu density g/cm <sup>3</sup>	Calculated $k_{\text{eff}}$
2	125	5	0.172	$1.0001 \pm 0.0064$
3	980	1	0.026	$0.9935 \pm 0.0046$
4	758	5	0.034	$0.9844 \pm 0.0046$
5A	15	2	1.12	$1.0181 \pm 0.0030$
5B	15	2	1.12	$1.0243 \pm 0.0047$
6	0	5	15.6	$0.9948 \pm 0.0052$
8	910	14	0.028	$0.9882 \pm 0.0042$
9A	50	18	0.37	$1.0007 \pm 0.0030$
9B	50	18	0.37	$1.0101 \pm 0.0044$
11A	0.04	18	5.8	$1.0308 \pm 0.0024$
11B	0.04	18	5.8	$1.0303 \pm 0.0048$
12A	5	11	2.3	$1.0206 \pm 0.0029$
12B	5	11	2.3	$1.0275 \pm 0.0041$
13	0	20	15.7	$1.0045 \pm 0.0048$
14	623	43	0.041	$0.9942 \pm 0.0028$
Average				1.0081

**Table 3.A.7.1** Calculated  $k_{\text{eff}}$ 's for "30 degrees lateral" system

No.	Height of water (cm)	Critical height (cm)	Solution density (g/cm <sup>3</sup> )	Calculated $k_{\text{eff}}$
1	210.1	128.2	—	—
2	209.3	128.4	—	—
3	209.8	128.3	2.0289	$0.9814 \pm 0.0040$
4	144.9	128.55	—	$0.9683 \pm —$
5	144.9	128.45	—	—
6	129.2	129.1	—	$0.9831 \pm 0.0038$
7	210.2	131.35	—	$0.9860 \pm 0.0044$
8	147.2	131.6	2.0048	$0.9787 \pm 0.0045$
9	132.8	132.2	—	$0.9781 \pm —$
10	210.2	132.1	—	$0.9760 \pm 0.0089$
11	147.5	132.4	1.9952	$0.9764 \pm 0.0042$
12	132.8	132.2	—	$0.9800 \pm 0.0041$
13	209.9	141.65	—	$0.9880 \pm 0.0036$
14	157.5	141.9	1.9367	$0.9696 \pm 0.0040$
15	143.2	142.75	—	$0.9762 \pm 0.0038$
16	210.1	155.25	—	$0.9763 \pm 0.0042$
17	171.9	155.35	1.9108	$0.9738 \pm 0.0044$
18	155.9	155.85	—	$0.9733 \pm 0.0039$
19	210.1	169.10	—	$0.9857 \pm 0.0041$
20	185.5	168.00	1.9059	$0.9857 \pm 0.0045$
21	168.1	168.20	—	$0.9736 \pm 0.0045$
22	209.5	204.60	1.8823	Subcritical
Average				0.9781

**Table 3.A.7.2** Calculated  $k_{eff}$ 's for "cross" system

No.	Height of water (cm)	Critical height (cm)	Solution density (g/cm <sup>3</sup> )	Inside diameter (cm)	Calculated $k_{eff}$
23	210.5	115.45	2.020	26.67	0.9898±0.0043
24	210.5	115.6			—
25	115.9	115.9			0.9812±0.0037
26	116.0	116.0			—
27	117.3	117.2	2.015		0.9826±0.0042
28	210.5	133.9	1.970		0.9944±0.0038
29	134.1	134.1	0.9796±0.0041		
30	210.5	109.6	2.026		0.9858±0.0045
31	110.6	110.6	1.971	27.30	0.9838±0.0040
32	210.5	115.55			0.9835±0.0048
33	116.05	116.05			0.9869±0.0041
34	210.5	106.3	2.024		0.9781±0.0041
35	107.9	107.9		27.94	0.9857±0.0042
36	210.5	110.0			0.9851±0.0041
37	110.95	110.95			0.9776±0.0039
38	210.5	117.0	1.920		0.9910±0.0043
39	117.5	117.5		28.57	0.9745±0.0043
40	210.5	104.45	2.023		0.9874±0.0040
41	105.7	105.7	0.9897±0.0039		
42	210.5	107.8	1.967		0.9915±0.0040
43	109.0	109.0		28.57	0.9751±0.0044
44	210.5	111.0	1.921		0.9829±0.0034
45	111.9	111.9	0.9892±0.0038		
46	210.5	115.4	1.892		0.9841±0.0043
47	115.95	115.95			0.9821±0.0039
48	210.5	122.65	1.866		0.9859±0.0037
49	123.15	123.15	0.9905±0.0037		
50	210.5	150.05	1.844		0.9837±0.0043
51	150.05	150.05		Average	0.9848

**Table 3.A.8.1** Calculated results with experimental conditions for cylinders or spheres containing uranyl-fluoride solution

No.	Solution Concentration		Container (0.16-cm-thick wall)	Cylindrical Critical Dimensions				Calculated Results	
	H/ <sup>235</sup> U	g of U per liter		Diameter (cm)	Height (cm)	Volume (Liters)	<sup>235</sup> U Mass (kg)	k <sub>eff</sub>	$\sigma$
Water-Reflected Units									
1	495	890	Aluminum	38.1	41.7	47.6	2.11	0.9743	0.0042
2	495	890	Aluminum	33.0	72.4	62.1	2.75	0.9740	0.0040
3	495	890	Aluminum	31.1	139.6	106.0	4.70	0.9741	0.0040
4	495	890	Aluminum	30.7	173.2	128.1	5.67	0.9657	0.0040
5	524	870	Stainless steel	50.8	29.3	59.4	2.53	0.9957	0.0043
6	524	870	Aluminum	38.1	44.8	51.1	2.17	0.9876	0.0045
7	524	870	Stainless steel	38.1	50.4	57.4	2.44	1.0015	0.0037
8	524	870	Stainless steel	30.5	>153	>111	>4.73	0.9546 <sup>a</sup>	0.0043
9	643	728	Aluminum	76.2	23.9	109.2	3.89	0.9972	0.0036
10	643	728	Stainless steel	50.8	34.7	70.3	2.50	0.9903	0.0040
11	643	728	Stainless steel	38.1	75.5	86.1	3.06	0.9859	0.0041
12	735	650	Aluminum	76.2	24.2	110.4	3.50	0.9728	0.0034
13	735	650	Stainless steel	50.8	40.1	81.1	2.58	0.9859	0.0039
14	735	650	Stainless steel	38.1	153.0	174.4	5.54	0.9906	0.0035
15	994	496	Aluminum	76.2	37.9	173.1	4.20	0.9882	0.0035
16	994	496	Stainless steel	50.8	85.7	173.7	4.21	0.9815	0.0034
17	1099	452	Aluminum <sup>b</sup>	69.3		170.5	3.77	0.9850	0.0030
Unreflected Units									
18	524	870	Stainless steel	50.8	38.7	78.3	3.33	0.9666	0.0044
19	524	870	Stainless steel	38.1	>147	>168	>7.13	0.9742 <sup>a</sup>	0.0043
20	643	870	Aluminum	76.2	28.9	132	4.69	0.9594	0.0038
21	643	728	Stainless steel	50.8	45.7	92.6	3.30	0.9773	0.0041
22	643	728	Stainless steel	38.1	>166	>189	>6.73	0.9470 <sup>a</sup>	0.0037
23	735	650	Aluminum	76.2	31.5	144	4.58	0.9530	0.0038
24	735	650	Stainless steel	50.8	54.3	110	3.50	0.9850	0.0045
25	994	496	Aluminum	76.1	44.8	204	4.96	0.9812	0.0034
26	994	496	Stainless steel	50.8	>140	>283	>6.84	0.9659 <sup>a</sup>	0.0037
27	1002	492	Aluminum <sup>c</sup>	69.3		172.0	4.14	0.9836	0.0035

<sup>a</sup> Subcritical case.<sup>b</sup> This was a spherical container 98 % filled when critical.<sup>c</sup> This was a spherical container 99 % filled when critical.**Table 3.A.9.1** Calculated results with experimental conditions for cylinders containing uranyl-fluoride solution with combination of reflectors

Case No.	Cylinder Diameter (cm)	Combination of Reflector	Steel Thickness (cm)	Critical Height (cm)	Calculated Results	
					$k_{eff} \pm \sigma$	$k_{eff} \pm \sigma$
1	33.02	Steel-Water	0	84.60	0.97886 ± 0.00457	
2	33.02	Steel-Water	0.64	158.85	0.98556 ± 0.00426	
3	33.02	Steel-Water	1.27	201.60	0.98106 ± 0.00451	
4	33.02	Steel-Water	1.90	172.80	0.98547 ± 0.00462	
5	33.02	Steel-Water	2.54	143.00	0.98936 ± 0.00389	
6	33.02	Steel-Water	3.81	109.10	0.97785 ± 0.00424	
7	33.02	Steel-Water	5.08	91.95	0.98536 ± 0.00399	
8	39.09	Steel	0	98.80	0.99218 ± 0.00447	
9	39.09	Steel	0.64	74.65	0.98724 ± 0.00401	
10	39.09	Steel	1.27	64.75	0.98752 ± 0.00407	
11	39.09	Steel	1.90	59.30	0.97833 ± 0.00455	
12	39.09	Steel	2.54	55.70	0.97287 ± 0.00413	
13	39.09	Steel	3.81	50.90	0.97928 ± 0.00473	
14	39.09	Steel	5.08	48.20	0.98514 ± 0.00390	
15	39.09	Steel-Water	0	44.70	0.97908 ± 0.00489	
16	39.09	Steel-Water	0.64	48.70	0.98203 ± 0.00389	
17	39.09	Steel-Water	1.27	49.40	0.97292 ± 0.00414	
18	39.09	Steel-Water	1.90	48.95	0.99218 ± 0.00423	
19	39.09	Steel-Water	2.54	48.20	0.97526 ± 0.00399	
20	39.09	Steel-Water	3.81	46.50	0.97717 ± 0.00412	
21	39.09	Steel-Water	5.08	45.10	0.98310 ± 0.00440	
22	39.09	Cadmium-Steel-Water	0	64.90	0.98888 ± 0.00416	
23	39.09	Cadmium-Steel-Water	2.54	54.45	0.99879 ± 0.00445	
24	39.09	Cadmium-Steel-Water	5.08	50.30	0.98464 ± 0.00453	

**Table 3.A.10.1** Critical dimensions of piles of UF<sub>4</sub>-paraffin compact fuel

Critical Dimensions*		Worth of Support Structures of Unreflected Assemblies (cm of fuel height)	
Reflected (cm)	Unreflected (cm)		
Mixture U(2)F <sub>4</sub> -1			
( 1) 56.22 X 56.22 X 112.88	(23) 71.47 X 71.47 X 94.14		0.89
( 2) 61.33 X 61.33 X 79.46	(24) 76.65 X 76.65 X 78.08		0.77
( 3) 71.55 X 71.55 X 58.03	(25) 81.75 X 86.75 X 66.71		0.63
( 4) 74.11 X 74.11 X 54.71			
Mixture U(2)F <sub>4</sub> -2			
( 5) 51.11 X 51.11 X 73.87	(26) 56.22 X 56.22 X 122.47		
( 6) 61.33 X 61.33 X 49.12	(27) 61.33 X 61.33 X 79.60	c	
( 7) 71.55 X 71.55 X 40.70	(28) 71.55 X 71.55 X 57.92		
	(29) 92.00 X 92.00 X 46.18		
Mixture U(2)F <sub>4</sub> -3			
( 8) 56.22 X 56.22 X 49.54	(30) 45.97 X 61.21 X 144.5		2.64
( 9) 53.67 X 53.67 X 54.29	(31) 51.07 X 56.13 X 110.5		0.85
	(32) 51.07 X 61.21 X 88.27		0.60
	(33) 61.21 X 61.21 X 64.85		0.46
	(34) 61.21 X 71.57 X 56.68		0.61
Mixture U(2)F <sub>4</sub> -4			
(10) 46.00 X 46.00 X 96.57	(35) 45.99 X 61.3 X 144.83		2.37
(11) 51.11 X 51.11 X 62.97	(36) 51.1 X 56.23 X 112.09		1.11
(12) 61.33 X 61.33 X 44.53	(37) 51.1 X 61.3 X 88.74		0.59
	(38) 56.23 X 61.3 X 72.38		0.63
	(39) 61.3 X 61.3 X 64.64		0.42
	(40) 61.3 X 66.5 X 59.68		0.50
Mixture U(2)F <sub>4</sub> -5			
(13) 56.32 X 61.29 X 54.08	(41) 53.80 X 56.30 X 126.98		0.42
	(42) 61.30 X 66.54 X 66.52		0.43
	(43) 62.28 X 112.64 X 41.59		0.59
Mixture U(2)F <sub>4</sub> -6			
(14) 76.51 X 76.44 X 82.42	(44) 76.50 X 76.45 X 119.45		0.58
(15) 76.43 X 81.53 X 77.66	(45) 81.45 X 86.70 X 88.22		0.52
(16) 81.52 X 81.60 X 73.02	(46) 101.97 X 101.85 X 66.64		0.65
Mixture U(3)F <sub>4</sub> -1			
(17) 43.47 X 43.47 X 86.39	(47) 56.47 X 56.47 X 86.64		0.83
(18) 46.02 X 46.02 X 67.57	(48) 56.25 X 51.36 X 74.38		0.76
(19) 51.14 X 51.14 X 51.27	(49) 61.4 X 61.4 X 66.00		0.65
(20) 56.25 X 56.25 X 43.41			
(21) 61.36 X 61.36 X 38.67			
Mixture U(3)F <sub>4</sub> -2			
(22) 40.81 X 40.80 X 39.49	(50) 40.90 X 40.93 X 116.80		0.37
	(51) 48.59 X 51.14 X 48.53		0.42
	(52) 81.71 X 81.66 X 31.34		0.53

\* The bottom reflector for all mixtures was 15.2 cm of methacrylate plastic (Plexiglass) while the top and sides of mixtures U(2)F<sub>4</sub>-1, -2, -3, and -4 and U(3)F<sub>4</sub>-1 were reflected with 15.2 cm of paraffin and mixtures U(2)F<sub>4</sub>-5 and -6 and U(3)F<sub>4</sub>-2 with 15.2 cm of polyethylene.

**Table 3.A.10.2** Calculated results for piles of the UF<sub>4</sub>-paraffin compact fuel

No.	$k_{\text{eff}} \pm \sigma$	No.	$k_{\text{eff}} \pm \sigma$
1	1.00976 ± 0.00414	23	1.01369 ± 0.00365
2	1.01015 ± 0.00374	24	1.00745 ± 0.00394
3	1.01606 ± 0.00406	25	1.00914 ± 0.00362
4	1.01374 ± 0.00369	26	1.00975 ± 0.00355
5	1.01050 ± 0.00357	27	1.00910 ± 0.00373
6	1.00360 ± 0.00354	28	1.00685 ± 0.00395
7	1.01518 ± 0.00419	29	1.00548 ± 0.00417
8	1.00576 ± 0.00367	30	1.00282 ± 0.00321
9	1.00410 ± 0.00358	31	0.99595 ± 0.00428
10	1.00074 ± 0.00362	32	0.99657 ± 0.00343
11	0.99733 ± 0.00367	33	1.00496 ± 0.00342
12	1.00031 ± 0.00344	34	1.00746 ± 0.00336
13	0.99438 ± 0.00310	35	1.00533 ± 0.00377
14	0.98392 ± 0.00289	36	1.00433 ± 0.00321
15	0.98879 ± 0.00284	37	0.99918 ± 0.00400
16	0.99010 ± 0.00287	38	0.99648 ± 0.00347
17	1.02039 ± 0.00381	39	0.99919 ± 0.00393
18	1.01887 ± 0.00398	40	1.00120 ± 0.00365
19	1.02039 ± 0.00432	41	0.99528 ± 0.00345
20	1.03095 ± 0.00367	42	0.99772 ± 0.00334
21	1.01969 ± 0.00413	43	0.99556 ± 0.00324
22	1.00754 ± 0.00382	44	0.98518 ± 0.00277
		45	0.99446 ± 0.00316
		46	0.99148 ± 0.00306
		47	1.02719 ± 0.00452
		48	1.01357 ± 0.00477
		49	1.01750 ± 0.00443
		50	1.00676 ± 0.00425
		51	1.02505 ± 0.00479
		52	1.00582 ± 0.00399

**Table 3.A.11.1** Experimental configurations for spherical vessels containing uranyl-nitrate solution with various reflectors

No.	Reflector	Sp. Cr.	Crit. Conc. ( $\text{g Pu/Liter}$ )	Molarity	Total Nitrate ( $\text{g NO}_3^-/\text{Liter}$ )	$\text{H}_2\text{O}$ ( $\text{g/Liter}$ )	$\text{H}/\text{Pu}$	Crit. <sup>a</sup> Atomic Mass Ratio (( $\text{kg}/\text{Pu}$ ))
11.5-in. Stainless-Steel Sphere: 12.95 Liters (Vessel Wall Thickness 0.049 in.)								
1	Water	1.130	73	0.2	86	971	354	0.95
2		1.143	74.5	0.4	105	963	344	0.96
3		1.230	100	1.9	230	898	243	1.30
4		1.282	126	2.2	262	892	192	1.63
5		1.484	269	1.1	346	868	87	3.48
6		1.492	295 <sup>b</sup>	0.8	303	893	81	3.82
7		1.680	435 <sup>b</sup>	0.8	372	872	54	5.63
8	Concrete: 11.25-in.	1.145	75	0.44	109	961	341	0.97
9		1.433	236	1.16	318	878	100	3.06
14-in. Stainless-Steel Sphere: 23.22 Liters (Vessel Wall Thickness 0.044 in.)								
10	$\frac{1}{2}$ -in. Paraffin <sup>c</sup>		53.6	2.2	190	930	470	1.24
11			62.9	2.8	238	912	396	1.48
12			63.8	3.5	278	886	382	1.48
13			89	4.0	338	847	253	2.07
14			96.7	4.2	354	841	241	2.25
15			97.4	4.5	377	833	238	2.26
16	1-in. Paraffin <sup>c</sup>		53.6	4.3	323	855	443	1.24
17			95.1	5.4	482	763	256	1.98
18			97.6	6.1	473	765	223	2.27
19	4-in. Concrete and 0.03-in. Cadmium <sup>c</sup>		40	1.1	110	948	634	0.93
20			50.9	4.6	336	834	457	1.18
21			232	6.7	650	689	86	5.39
22	4-in. Concrete	1.093	32.8	1.1	118	949	770	0.75
23		1.143	35.5	2.9	216	889	684	0.82
24		1.253	45.2	5.62	410	792	495	1.05
25	10-in. Concrete	1.085	29.6	1.07	118	936	848	0.69
26		1.209	36.6	4.39	310	858	651	0.85
27		1.285	43.4	6.37	445	790	518	1.01
28	4-in. Air Gap;	1.250	67.9	4.29	336	842	344	1.58
29	6-in. Concrete	1.175	50.4	2.28	194	933	500	1.17
30	4-in. Air Gap; 6-in. Concrete Hemisphere	1.193	75.0	2.10	209	907	328	1.74
31	10-in. Concrete Hemisphere	1.146	46.4	2.05	175	923	538	1.08
32	4-in. Concrete +0.036-in. Stainless Steel	1.252	46.4	5.62	404	796	484	1.08
33	4-in. Concrete+0.72-in. Stainless Steel	1.258	46.9	5.62	416	790	476	1.09
34	Water	1.123	33.2	2.08	164	924	754	0.77
35		1.196	38.6	4.07	292	881	618	0.90
36		1.290	47.5	6.56	462	774	466	1.10
37	Water and 0.030-in.	1.126	46.9	1.4	136	942	540	1.09
38	Cadmium	1.238	69.0	3.9	313	852	341	1.60
39	Water + 0.072-in. Stainless Steel	1.268	49.5	6.01	424	789	452	1.15
15.2-in. Stainless-Steel Sphere: 30.2 Liters (Vessel Wall Thickness 0.048 in.)								
40	Unreflected <sup>d</sup>	1.081	39.0	0.4	64	978	668	1.18
41		1.429	172.3	4.9	485	766	125	5.20
42	Stainless-Steel Shell 0.26-in. Thick	1.076	34.3	0.5	66	975	758	1.04
43	Concrete Shell: Air Gap (6-in. Thick Concrete, 3.63-in. Air Gap)	1.071	29.0	0.5	64	977	900	0.88
44	Water	1.060	24.4	0.5	58	977	1058	0.74
45		1.300	38.7	7.7	517	737	553	1.17
46	Water and 0.08-in. Stainless Steel	1.060	25.2	0.5	60	974	1032	0.76
47	Water Only <sup>e</sup>		23.6	0.5	55	979	1107	0.71
48	Unreflected <sup>e</sup>		41.0	0.4	67	976	634	1.24

<sup>a</sup> Includes 4.6%  $^{240}\text{Pu}$ ; experimental values not corrected for vessel walls or neck supports, except where noted. Correction values are discussed in the text.

<sup>b</sup> Contains Pu(VI) and plutonium polymer, increasing the uncertainty ±5%

<sup>c</sup> Values of nitrate corrected to full sphere from measurements close to full. Water calculated from nitric acid content and corrected by comparison to a known similar solution.

<sup>d</sup> These same solutions would be critical in spherical geometry at a volume of 31.1 liters with no stainless-steel walls or neck supports.

<sup>e</sup> Corrected for effect of vessel wall and neck supports.

**Table 3.A.11.2** Calculated results for spherical vessels containing uranyl-nitrate solution with various reflectors

No.	$k_{\text{eff}}$	$\pm\sigma$	No.	$k_{\text{eff}}$	$\pm\sigma$
1	1.00059	0.00481	25	0.98562	0.00469
2	0.99331	0.00559	26	1.00285	0.00461
3	0.98141	0.00471	27	1.01424	0.00416
4	0.99620	0.00479	28	1.01460	0.00454
5	0.99997	0.00430	29	1.00981	0.00474
6	1.01679	0.00530	30	0.99243	0.00590
7	1.01845	0.00546	31	1.00839	0.00581
8	1.00975	0.00582	32	1.00871	0.00537
9	1.02079	0.00504	33	1.00546	0.00519
10	1.00054	0.00521	34	0.99789	0.00393
11	1.01352	0.00536	35	0.98364	0.00466
12	0.99792	0.00478	36	0.99520	0.00507
13	1.00934	0.00611	37	0.98357	0.00538
14	1.01286	0.00510	38	0.99087	0.00490
15	1.00342	0.00511	39	0.99520	0.00507
16	0.99881	0.00520	40	0.99324	0.00461
17	1.00714	0.00575	41	1.00440	0.00660
18	1.01158	0.00548	42	0.99786	0.00503
19	0.99971	0.00507	43	1.00301	0.00406
20	0.99852	0.00483	44	1.00917	0.00463
21	0.99851	0.00523	45	1.01300	0.00434
22	1.00293	0.00490	46	1.01295	0.00467
23	1.00129	0.00470	47	1.00461	0.00463
24	1.00212	0.00474	48	1.00122	0.00481

**Table 3.A.12.1** Calculated results with experimental conditions for annular cylinders containing plutonium-nitrate solution with inner void region

No.	C(g/l of Pu total) $\pm 1.5\%$	$\bar{N}\text{O}_3^-$ (N)	$H^+$ (N)	Fer (g/l)	Density	Tempe- rature (°C)	Critical Height $H_c$ (cm)	Calculated Results	
								$k_{\text{eff}}$	$\sigma$
1	152	6.68	3.85	3.4	1.386	21.0	$36.88 \pm 0.04$	0.9919	0.0058
2	104	4.57	2.65	2.12	1.271	20.5	$34.31 \pm 0.03$	0.9998	0.0049
3	62	3.02	1.88	1.28	1.171	20.5	$37.13 \pm 0.02$	0.9910	0.0048
4	51	2.80	1.85	1.13	1.151	20.5	$41.08 \pm 0.06$	1.0024	0.0049
5	40.9	2.63	1.87	0.92	1.135	20.5	$49.16 \pm 0.03$	0.9927	0.0046
6	36.0	2.58	1.90	0.88	1.127	20.5	$57.65 \pm 0.03$	1.0016	0.0037
7	33.1	2.51	1.90	0.71	1.121	20.5	$66.64 \pm 0.02$	0.9968	0.0040
8	30.8	2.45	1.88	0.69	1.118	20.8	$77.31 \pm 0.02$	0.9883	0.0034
9	28.7	2.47	1.94	0.65	1.114	20.3	$93.34 \pm 0.02$	0.9916	0.0038

**Table 3.A.12.2** Calculated results with experimental conditions for annular cylinders containing plutonium-nitrate solution with inner cadmium-paraffin region

No.	C(g/l of Pu total) ±1.5%	NO <sub>3</sub> (N)	H <sup>+</sup> (N)	Per (g/l)	Density	Temper- ature °C	Critical Height Hc(cm)	Calculated Results	
								k <sub>eff</sub>	σ
10	165	6.17	3.11	3.8	1.393	21.8	60.72 ± 0.02	1.0058	0.0045
11	136	5.30	2.76	3.2	1.329	21.0	54.33 ± 0.02	0.9969	0.0049
12	105	4.14	2.17	2.5	1.261	21.5	51.46 ± 0.02	0.9858	0.0049
13	86	3.82	2.21	1.87	1.226	21.0	54.24 ± 0.03	0.9956	0.0049
14	74	3.59	2.22	1.56	1.202	21.5	59.33 ± 0.02	0.9910	0.0044
15	61	3.34	2.19	1.41	1.180	21.5	70.46 ± 0.02	0.9930	0.0044
16	55.8	3.17	2.12	1.33	1.168	21.0	83.61 ± 0.03	0.9849	0.0046
17	53.7	3.11	2.11	1.25	1.164	21.5	92.09 ± 0.04	0.9853	0.0044

**Table 3.A.13.1** Calculated results with experimental conditions for cylinders containing plutonium-nitrate solution of 5.6 wt% <sup>240</sup>Pu in Pu

No.	Plutonium Concentration (g/l)	Uranium Concentration (g/l)	Acid Molarity	Specific Gravity	Critical Height (cm)	Critical Volume (l)	Calculated Results	
							k <sub>eff</sub>	±σ
1	97.3	200.4	4.8	1.562	19.96	58.37	1.0163	0.0049
2	87.9	184.3	4.4	1.512	19.48	57.00	0.9912	0.0050
3	75.4	168.5	3.7	1.450	18.82	55.04	1.0027	0.0045
4	63.1	147.7	2.9	1.389	18.72	54.76	1.0043	0.0044
5	49.6	122.3	2.4	1.320	19.35	56.59	1.0039	0.0055
6	39.9	96.7	2.2	1.257	20.32	59.44	1.0009	0.0049
7	30.2	72.5	2.1	1.213	22.89	66.97	1.0146	0.0041
8	25.3	60.9	2.0	1.182	25.12	73.48	0.9984	0.0041
9	20.3	48.9	1.8	1.154	29.92	87.53	0.9983	0.0038
10	17.2	41.5	1.7	1.132	36.04	105.44	1.0024	0.0042
11	14.1	34.4	1.5	1.117	52.60	153.91	0.9906	0.0036
12	13.3	32.2	1.4	1.110	64.01	187.24	1.0057	0.0033
13	12.4	29.9	1.3	1.103	95.20	278.50	0.9862	0.0039

**Table 3.A.13.2** Calculated results with experimental conditions for spheres containing plutonium-uranyl-nitrate solution of 4.7 wt% <sup>240</sup>Pu in Pu

No.	Approximate Percent Plutonium	Sphere Wall Thickness (cm)	Plutonium Concentration (g/l)	Uranium Concentration (g/l)	Gadolinium Concentration (g/l)	Acid Molarity	Specific Gravity	Critical Radius (cm)	Critical Volume (l)	Calculated Results	
										k <sub>eff</sub>	±σ
1	30	0.112	70.93	157.1	0.051	3.12	1.429	17.87	23.90	0.9945	0.0045
2	30	0.122	35.05	75.7	0.025	1.49	1.215	19.30	30.14	1.0096	0.0039
3	15	0.122	45.6	264.9	0.005	2.1	1.491	19.31	30.18	1.0120	0.0045

**Table 3.A.13.3** Calculated results with experimental conditions for cylinders containing plutonium-uranyl-nitrate solution of 23.0 wt% <sup>240</sup>Pu in Pu

No.	Plutonium Concentration (g/liter)	Uranium Concentration (g/liter)	Acid Molarity	Specific Gravity	Critical Height (cm)	Critical Volume (liter)	Calculated Results	
							k <sub>eff</sub>	±σ
1	30.63	390.2	0.45	1.5744	50.27	147.03	1.0046	0.0039
2	29.00	394.5	0.44	1.5799	54.66	159.89	0.9992	0.0040
3	27.32	399.0	0.44	1.5838	61.04	178.56	0.9928	0.0034
4	25.71	403.3	0.37	1.5840	70.49	206.20	0.9929	0.0034
5	24.28	407.1	0.36	1.5878	84.86	248.23	0.9909	0.0034
6	23.66	408.7	0.36	1.5889	94.56	276.62	0.9972	0.0033
7	30.44	390.7	0.43	1.5782	51.41	150.38	1.0036	0.0039
8	29.13	373.8	0.47	1.5536	52.29	152.98	0.9996	0.0034
9	27.28	351.1	0.54	1.5214	56.87	166.36	1.0019	0.0033
10	25.76	331.3	0.56	1.4914	60.99	178.39	1.0075	0.0035
11	24.35	312.5	0.56	1.4672	67.59	197.71	0.9955	0.0037
12	23.04	296.7	0.54	1.4416	76.96	225.13	0.9910	0.0031
13	21.85	281.9	0.56	1.4212	89.61	262.13	0.9925	0.0033

**Table 3.A.14.1** Principal critical parameters and calculated  $k_{eff}$ 's for  $\text{U}_3\text{O}_8$  package arrays

Type of Driver	Total Driver Mass (kg)	Metal Diameter of Solution Height*) (mm)	Critical Core Separation (mm)	Calculated $k_{eff}$
				MGCL(137-Group)
93.12 % Enriched Uranium Metal	29.870	0.0 (ID), 146.68(OD)	12.60±1.19 <sup>1)</sup>	0.9997±0.0063
	29.870	0.0 (ID), 146.68(OD)	8.01±1.67 <sup>2)</sup>	0.9855±0.0047
	33.543	40.02(ID), 153.32(OD)	14.64±2.26 <sup>3)</sup>	0.9858±0.0057
High Concentration Solution (351.2 kg U/m <sup>3</sup> )	14.066	105.47(S), 106.22(N)	6.68±1.79 <sup>1)</sup>	1.0059±0.0045
	14.844	111.30(S), 112.10(N)	9.42±3.01 <sup>2)</sup>	0.9874±0.0049
	16.143	121.04(S), 121.91(N)	9.04±1.44 <sup>3)</sup>	0.9727±0.0052
Low Concentration Solution (86.4 kg U/m <sup>3</sup> )	12.446	124.08(S), 124.42(N)	8.64±1.31 <sup>1)</sup>	0.9998±0.0041
	12.875	128.28(S), 128.74(N)	9.12±1.42 <sup>2)</sup>	0.9870±0.0045
	13.001	129.54(S), 129.99(N)	11.17±1.42 <sup>2)</sup>	0.9942±0.0044
	13.999	139.46(S), 139.99(N)	6.30±1.66 <sup>3)</sup>	0.9868±0.0050

\*) ID = Inner diameter; OD = Outer diameter; S = South; N = North

1) Concrete reflector

2) Plastic reflector

3) Minimal reflector

**Table 3.A.15.1** Principal critical parameters and calculated  $k_{eff}$ 's

No.	Concentration		Critical Values			Calculated $k_{eff}$
	Pu (g Pu/l)	Gd (g Gd/l)	Height (cm)	Volume (l)	Mass (kg Pu)	
1	116	0.00	15.44	45.17	5.240	1.0009±0.0047
2	116	0.48	20.16	58.98	6.842	1.0008±0.0044
3	116	0.96	25.98	75.99	8.815	0.9838±0.0035
4	116	1.42	34.48	100.85	11.698	0.9833±0.0037
5	116	1.92	48.28	141.24	16.383	0.9879±0.0030
6	116	2.38	80.37	235.10	27.272	0.9853±0.0030
7	363	4.40	26.52	77.56	28.192	0.9938±0.0038
8	363	5.28	28.07	82.08	29.838	0.9842±0.0041
9	363	6.28	29.62	86.65	31.497	0.9858±0.0036
10	363	8.21	32.84	96.10	34.932	0.9860±0.0038
11	363	9.88	36.04	105.40	38.311	0.9875±0.0039
12	363	12.85	41.02	120.00	43.608	0.9883±0.0039
13	363	15.55	46.76	136.76	49.714	0.9840±0.0039
14	363	18.4	53.70	157.05	57.089	0.9818±0.0037
15	363	20.25	59.36	173.63	63.113	0.9881±0.0027

**Table 3.A.16.1** Experimental condition for cylinders containing plutonium-uranyl-nitrate solution with soluble neutron absorbers

CASE No.	CRITICAL HEIGHT (cm)	PU CONC. (g/l)	U CONC. (g/l)	ACID MOLARITY (g/l)	TOTAL NO <sub>3</sub> (g/l)	SPECIFIC GRAVITY	Cd (g/l)	B (g/l)	
01	15.44	116	0	1.85	238.9	1.255	0.00	0	0.00451
02	20.16	116	0	1.85	238.9	1.255	0.48	0	0.00413
03	25.98	116	0	1.85	238.9	1.255	0.96	0	0.00361
04	34.48	116	0	1.85	238.9	1.255	1.42	0	0.00379
05	48.28	116	0	1.85	238.9	1.255	1.92	0	0.00316
06	80.37	116	0	1.85	238.9	1.255	2.38	0	0.00448
07	26.52	363	0	4.10	634.1	1.701	4.40	0	0.00418
08	28.07	363	0	4.10	634.1	1.701	5.28	0	0.00438
09	29.62	363	0	4.10	634.1	1.701	6.28	0	0.00324
10	32.84	363	0	4.10	634.1	1.701	8.21	0	0.00392
11	36.04	363	0	4.10	634.1	1.701	9.88	0	0.00376
12	41.02	363	0	4.10	634.1	1.701	12.85	0	0.00327
13	46.76	363	0	4.10	634.1	1.701	15.55	0	0.00341
14	53.70	363	0	4.10	634.1	1.701	18.4	0	0.00341
15	59.36	363	0	4.10	634.1	1.701	20.25	0	0.00368
16	19.10	75.6	171.9	3.27	371.4	1.446	0.042	0	0.00448
17	21.93	76.3	173.4	3.26	377.0	1.452	0.180	0	0.00365
18	24.61	76.3	174.6	3.42	383.8	1.450	0.288	0	0.00400
19	29.47	76.6	174.7	3.07	375.8	1.456	0.459	0	0.00399
20	33.42	76.6	175.3	3.05	365.2	1.454	0.581	0	0.00374
21	37.15	77.0	175.9	3.14	370.8	1.456	0.679	0	0.00407
22	43.71	77.1	175.1	3.13	380.7	1.456	0.800	0	0.00333
23	53.12	77.3	177.7	3.24	376.4	1.461	0.923	0	0.00317
24	62.12	77.2	176.3	3.27	381.3	1.455	1.01	0	0.00270
25	75.32	77.3	176.8	3.25	383.8	1.462	1.06	0	0.00466
26	18.13	85.0	182.5	1.55	320	1.433	0.04	0.0	0.00454
27	18.65	84.9	182.2	1.61	311	1.434	0.04	0.1	0.00446
28	19.68	84.8	182.6	1.67	290	1.433	0.04	0.3	0.00415
29	21.13	84.5	182.1	1.75	296	1.438	0.128	0.27	0.00383
30	82.16	82.8	180.6	1.95	309	1.443	0.293	0.9	0.00330
31	43.03	82.2	179.5	2.12	319	1.446	0.388	1.2	0.00341
32	52.12	81.2	180.0	2.18	321	1.447	0.424	1.35	0.00332
33	67.83	81.0	180.4	2.09	316	1.451	0.519	1.5	0.00314
34	75.44	81.0	180.3	2.21	318	1.452	0.537	1.54	

**Table 3.A.16.2** Calculated results for cylinders containing plutonium-uranyl-nitrate solution with soluble neutron absorbers

CASE No.	k <sub>eff</sub>	±σ
01	1.02072	0.00451
02	1.00647	0.00413
03	0.99143	0.00361
04	0.99306	0.00379
05	0.98055	0.00316
06	0.98359	0.00324
07	0.99212	0.00448
08	0.98198	0.00418
09	0.98422	0.00438
10	0.98072	0.00392
11	0.98515	0.00376
12	0.98039	0.00327
13	0.97649	0.00341
14	0.97682	0.00341
15	0.98086	0.00368
16	1.00265	0.00509
17	1.00160	0.00448
18	0.99027	0.00365
19	0.99116	0.00400
20	0.98706	0.00414
21	0.97924	0.00374
22	0.97894	0.00407
23	0.97926	0.00333
24	0.97369	0.00317
25	0.97767	0.00270
26	1.00387	0.00466
27	0.98723	0.00454
28	0.99799	0.00446
29	1.00001	0.00415
30	0.98427	0.00383
31	0.98168	0.00330
32	0.97416	0.00341
33	0.96792	0.00332
34	0.96670	0.00314

**Table 3.A.17.1** Calculated  $k_{eff}$ 's with experimental conditions for raschig rings in plutonium-uranyl solution

Exp. No.	Critical height (cm)	Pu Conc. (g/l)	U Conc. (g/l)	Calculated $k_{eff}$
106	57.56	92.3	147	$0.9795 \pm 0.0041$
107	63.98	89.9	157	$0.9891 \pm 0.0042$
108	70.46	86.1	173	$0.9852 \pm 0.0042$
109	76.07	83.7	181	$0.9765 \pm 0.0037$
110	79.78	82.4	191	$0.9880 \pm 0.0035$
111	79.93	81.7	186	$0.9800 \pm 0.0038$
112	81.51	80.8	185	$0.9789 \pm 0.0034$
113	85.70	79.4	182	$0.9893 \pm 0.0040$
Average				0.9833

**Table 3.B.1.1** Calculated  $k_{eff}$ 's for unreflected 3X4 arrays

No.	Cylinder center separation(cm)	Table separation at critical (cm)	Calculated $k_{eff}$
1	22.52	2.56	$0.9548 \pm 0.0040$
2	22.68	0.74	$0.9599 \pm 0.0043$
3	22.73	0.61	$0.9588 \pm 0.0037$
4	22.81	0.12	$0.9673 \pm 0.0039$
5	22.81	0.19	$0.9597 \pm 0.0040$
6	22.81	1.01	$0.9662 \pm 0.0042$
7	22.81	0.13	$0.9695 \pm 0.0044$
8	22.81	1.26	$0.9579 \pm 0.0040$
9	22.81	1.48	$0.9585 \pm 0.0042$
10	22.81	1.21	$0.9527 \pm 0.0036$
Average			0.9605

**Table 3.B.1.2** Calculated  $k_{eff}$ 's for unreflected 4X4 arrays

No.	Cylinder center separation (cm)	Table separation at critical (cm)	Calculated $k_{eff}$
11	24.50	4.10	$0.9728 \pm 0.0044$
12	24.82	1.16	$0.9652 \pm 0.0040$
13	24.91	0.88	$0.9690 \pm 0.0041$
14	24.96	0.34	$0.9688 \pm 0.0043$
15	24.96	1.14	$0.9658 \pm 0.0042$
16	24.96	0.33	$0.9640 \pm 0.0040$
Average			0.9676

**Table 3.B.1.3** Calculated  $k_{eff}$ 's for 2X2 arrays reflected on 5 sides by polyethylene and moderated by Plexiglas

No.	Cylinder center separation (cm)	Table separation (cm) at critical	Calculated $k_{eff}$
17	22.52	—	Sub-critical
18	23.66	—	"
19	25.05	1.22	$0.9901 \pm 0.0041$
20	25.61	—	Sub-critical
21	26.24	—	"
22	25.13	0.59	$0.9794 \pm 0.00046$
Average			0.9848

**Table 3.B.1.5** Calculated  $k_{eff}$ 's for 4X4 arrays with various polyethylene reflector conditions

No.	Cylinder center separation (cm)	Table separation (cm) at critical	Calculated $k_{eff}$
27	29.19	0.44	$0.9990 \pm 0.00045$
28	29.19	0.60	$0.9960 \pm 0.00040$
29	29.59	0.00	$0.9909 \pm 0.0004^*)$
Average			0.9953

\* $) k_{eff}$  of assembly was 0.9993

**Table 3.B.1.6** Calculated  $k_{eff}$ 's for 2X2 arrays reflected on 5 sides by concrete and complete reflection

No.	Cylinder center separation (cm)	Table separation (cm) at critical	Calculated $k_{eff}$
30	22.52	0.96	$0.9874 \pm 0.00040$
31	23.12	—	Sub-critical
32	23.12	1.52	$0.9924 \pm 0.00042$
33	23.12	1.86	$0.9964 \pm 0.00042$
34	23.12	2.01	$1.0002 \pm 0.00043$
35	23.12	1.65	$0.9960 \pm 0.00044$
36	22.84	0.18	$0.9787 \pm 0.00048$
37	22.84	0.12	$0.9809 \pm 0.00044$
38	22.68	0.54	$0.9860 \pm 0.00040$
39	22.68	0.62	$0.9862 \pm 0.00043$
Average			0.9894

**Table 3.B.1.4** Calculated  $k_{eff}$ 's for 3X3 arrays reflected on 5 sides by polyethylene

No.	Cylinder center separation (cm)	Table separation (cm) at critical	Calculated $k_{eff}$
23	23.56	0.56	$1.0069 \pm 0.00043$
24	24.17	~10.9	$0.9880 \pm 0.00044$
25	24.84	~7.1	$0.9934 \pm 0.00040$
26	26.09	0.31	$1.0102 \pm 0.00045$
Average			0.9996

**Table 3.B.1.7** Calculated  $k_{eff}$ 's for  $2 \times 2$  arrays with various concrete reflector conditions

No.	Cylinder center separation (cm)	Table separation at critical(cm)	Calculated $k_{eff}$
40	25.70	—	Subcritical
41	25.70	1.04	$1.0013 \pm 0.0039$
42	25.70	1.35	$0.9933 \pm 0.0044$
43	25.70	1.56	$0.9952 \pm 0.0045$
44	25.70	1.26	$1.0065 \pm 0.0038$
45	25.01	—	Subcritical
46	25.01	—	Subcritical
Average			0.9991

**Table 3.B.1.8** Calculated  $k_{eff}$ 's for  $3 \times 3$  arrays with various concrete reflector conditions

No.	Cylinder center separation (cm)	Table separation at critical(cm)	Calculated $k_{eff}$
47	30.78	1.35	$0.9975 \pm 0.0043$
48	31.18	—	$0.9901 \pm 0.0036^*)$
49	31.18	2.31	$0.9927 \pm 0.0045$
50	31.18	1.83	$0.9989 \pm 0.0040$
Average			0.9948

\*)  $k_{eff}$  of assembly at table closure was 1.0009**Table 3.B.1.9** Calculated  $k_{eff}$ 's for  $1 \times 12$  arrays with various concrete reflector conditions

No.	Cylinder center separation (cm)	Table separation at critical(cm)	Calculated $k_{eff}$
51	22.52	0.69	$0.9994 \pm 0.0038$
52	22.52	0.62	$0.9922 \pm 0.0038$
53	22.52	0.61	$0.9996 \pm 0.0052$
54	22.52	—	Subcritical
Average			0.9971

**Table 3.B.1.10** Calculated  $k_{eff}$ 's for  $1 \times 5$  arrays with various concrete reflector conditions

No.	Cylinder center separation (cm)	Table separation at critical(cm)	Calculated $k_{eff}$
55	26.66	—	—
56	26.66	0.00	$0.9817 \pm 0.0041^*)$
57	26.33	0.38	$0.9859 \pm 0.0043$
58	26.33	0.72	$1.0000 \pm 0.0043$
59	26.33	0.94	$0.9892 \pm 0.0044$
60	26.33	—	Subcritical
61	26.33	0.28	$0.9880 \pm 0.0038$
62	26.33	0.67	$0.9920 \pm 0.0040$
Average			0.9895

\*)  $k_{eff}$  of assembly was 0.9997

**Table 3.B.1.11** Calculated  $k_{eff}$ 's for  $1 \times 4$  arrays with various concrete reflector conditions

No.	Cylinder center separation (cm)	Table separation at critical (cm)	Calculated $k_{eff}$
63	24.72	0.56	0.9985 ± 0.0047
64	24.72	0.65	0.9912 ± 0.0045
65	24.72	0.23	0.9964 ± 0.0047
		Average	0.9954

**Table 3.B.1.12** Calculated  $k_{eff}$ 's for  $1 \times 3$  arrays with various reflector conditions and with Plexiglas separating the cylinders

No.	Cylinder center separation(cm)	Table separation at critical (cm)	Calculated $k_{eff}$
66	22.52	—	Subcritical
67	23.66	0.69	0.9997 ± 0.0043
68	23.66	0.42	1.0010 ± 0.0049
69	23.66	0.12	0.9998 ± 0.0040
70	23.66	0.11	0.9965 ± 0.0046
		Average	0.9993

**Table 3.B.1.13** Calculated  $k_{eff}$ 's for two linear three-cylinder arrays with close fitting concrete reflectors

No.	Cylinder center separation(cm)	Table separation at critical (cm)	Calculated $k_{eff}$
71	22.52	2.49	1.0057 ± 0.0041
72	22.52	3.54	1.0008 ± 0.0042
73	22.52	5.17	0.9967 ± 0.0042
74	22.52	1.06	1.0081 ± 0.0043
75	22.52	2.47	1.0100 ± 0.0041
		Average	1.0043

**Table 3.B.1.14** Calculated  $k_{eff}$ 's for two linear three-cylinder arrays with concrete reflectors

No.	Cylinder center separation(cm)	Table separation at critical (cm)	Calculated $k_{eff}$
76	22.52	—	Subcritical
77	22.52	0.22	1.0134 ± 0.0039
78	22.52	0.13	1.0075 ± 0.0050
		Average	1.0105

**Table 3.B.1.15** Calculated  $k_{eff}$ 's for two linear three-cylinder arrays with various reflector conditions

No.	Cylinder center separation (cm)	Table separation (cm) at critical	Calculated $k_{eff}$
79	22.52	—	Subcritical
80	22.52	0.39	$1.0170 \pm 0.0042$
81	22.52	7.36	$1.0135 \pm 0.0045$

**Table 3.B.2.2** Experimental data and calculated  $k_{eff}$ 's for the critical slab cylinder array configuration with the minimum reflector

No.	Cylinder Number	Slab thickness (cm)	Array height (cm)	Uranium concentration (g/liter)	Calculated $k_{eff}$
11	1	12.6	87		$0.9247 \pm 0.0057$
12	1	12.6	37		$0.9191 \pm 0.0060$
13		12.3	88		$0.9133 \pm 0.0051$
14	4	12.4	5		$0.9058 \pm 0.0055$
15		12.5	0		$0.9061 \pm 0.0055$
16		12.1	88		$0.9214 \pm 0.0057$
17	9	12.2	39	46.5	$0.9224 \pm 0.0052$
18		12.3	18		$0.9135 \pm 0.0062$
19		12.6	2		$0.9288 \pm 0.0054$
20		11.4	88		$0.9164 \pm 0.0057$
21	16	11.7	38		$0.9214 \pm 0.0061$
22		11.9	19		$0.9092 \pm 0.0050$
23		12.5	2		$0.9198 \pm 0.0058$
24		12.7	0		$0.9278 \pm 0.0058$
25		10.0	108		$0.9227 \pm 0.0055$
26	16	10.4	69		$0.9240 \pm 0.0058$
27		11.3	26	520	$0.9187 \pm 0.0051$
28		12.3	4		$0.9103 \pm 0.0057$

**Table 3.B.2.1** Experimental data and calculated  $k_{eff}$ 's for the critical slab systems

No.	Slab with minimum reflector	Estimated b) accuracy (cm)	Uranium concentrations (g/liter)	Calculated $k_{eff}$
1	12.7	$\pm 0.3$	465	$0.9179 \pm 0.0065$
2	12.7	$\pm 0.3$	465	—
3	12.7	$\pm 0.3$	465	—
4	12.6	$\pm 0.3$	465	$0.9192 \pm 0.0053$
5	12.7	$\pm 0.3$	465	—
6	12.8	$\pm 0.2$	495	—
7	13.0	$\pm 0.2$	500	$0.9174 \pm 0.0056$
8	13.0	$\pm 0.2$	510	$0.9326 \pm 0.0054$
9	13.0	$\pm 0.2$	520	—
10	10.3	$\pm 0.2$	Average	0.9218
			Reflected slab c)	
			Reflected slab c)	
			Reflected slab c)	

a) Includes any precipitate present in the slab tank.

b) Repeatability of measurements was  $\pm 0.1$  cm.

c) Reflected slab tank without cylinders.

Table 3.B.2.2 (continued)

No.	Cylinder diameter (cm)	Number of cylinders	Slab thickness (cm)	Array height (cm)	Uranium concentration (g/ $\Omega$ )	Calculated $k_{\text{eff}}$	Cylinder No.	Number of cylinders	Slab thickness (cm)	Array height (cm)	Uranium concentration (g/ $\Omega$ )	Calculated $k_{\text{eff}}$	
29			12.2	88		0.9294±0.0052	49	16.3	16	12.4	3	495	0.9308±0.0054
30			12.2	40		0.9300±0.0055	50			12.6	0		0.9070±0.0057
31	1	12.3	19	470	0.9180±0.0057		51			11.1	88		0.9384±0.0058
32	12.3	14			0.9210±0.0053		52	1	11.1	39			0.9269±0.0068
33		12.5	6		0.9279±0.0051		53		12.4	3			0.9158±0.0059
34		12.0	88		0.9311±0.0057		54			8.3	91		0.9070±0.0059
35	4	12.2	25	480	0.9314±0.0054		55			10.1	39		0.9299±0.0064
36		12.3	12		0.9612±0.0051		56	4	11.0	19			0.9348±0.0059
37		12.6	1		0.9198±0.0055		57		11.8	10			0.9232±0.0053
38	16.3	10.5	90		0.9328±0.0054		58		12.7	2			0.9329±0.0064
39		11.0	40		0.9302±0.0062		59	21.3		0	47	500	0.9484±0.0060
40	9	11.8	14	485	0.9243±0.0058		60			1.5	43		0.9624±0.0053
41		12.4	3		0.9271±0.0060		61			5.9	32		0.9578±0.0068
42		12.7	0		0.9142±0.0058		62			9	8.6	23	
43		0	78		0.9404±0.0065		63			10.0	15		0.9167±0.0055
44		2.6	64		0.9459±0.0058		64			14.6	7		0.9299±0.0062
45	16	5.5	50	495	0.9394±0.0056		65			12.7	1		0.9171±0.0054
46		7.7	37		0.9384±0.0056		66			0	26		0.9371±0.0063
47		10.0	20		0.9254±0.0061		67	16	2.7	22			0.9425±0.0051
48		11.4	8		0.9176±0.0053		68		7.5	14			0.9259±0.0051

Table 3.B.2.2 (continued)

Cylinder No.	Number of cylinders	Slab thickness (cm)	Array height (cm)	Uranium concentration (g / l)	Calculated $k_{eff}$
69		9.1	11		0.9383±0.0051
70	21.3	16	9.5	500	0.9089±0.0054
71		11.8	3		0.9274±0.0054
72	22.4	1	10.8	505	0.9404±0.0057
73	22.9	1	10.1	110	0.9495±0.0062
74		8.9	111		0.9481±0.0060
75	23.4	1	10.3	33	0.9407±0.0057
76		11.4	15		0.9243±0.0052
77		0	112	525	0.9533±0.0055
78	23.9	1	7.7	66	0.9568±0.0052
79			9.7	34	0.9493±0.0057
80			11.9	8	0.9155±0.0061
				Average	0.9289

Table 3.B.2.3 Experimental data and calculated  $k_{eff}$ 's for the critical slab-cylinder configuration with the array suspended above the slab

No. of array	Spacing between slab to bottom of array (cm)	Slab thickness (cm)	Solution height in cylinders a) (cm)	Calculated $k_{eff}$
81	9.4	4.7	66	0.9491±0.0063
82	6.9	7.3	49	0.9353±0.0053
83	3.6	10.5	19	0.9173±0.0052
84	2.7	11.4	8	0.9069±0.0061
85	20.4	7.8	67	0.9617±0.0056
86	18.2	10.0	49	0.9267±0.0059
87	16.7	11.4	19	0.9065±0.0059
88	16.4	11.7	10	0.8906±0.0058
89	28.2	12.4	0.3	0.9115±0.0050
				Average 0.9228

a) This height is measured from the bottom of the cylinders of the array.

**Table 3.B.2.4** Experimental data and calculated  $k_{eff}$ 's for the critical slab-cylinder configuration with Plexiglas reflectors

No.	Number of cylinders	Top and side reflector thickness(cm)	Slab tank bottom reflector thickness (cm)	Slab thickness (cm)	Array height (cm)	Uranium concentration (g/l)	Calculated $k_{eff}$
90	9	10.2	10.2	7.1 ± 0.2	112 ± 1	505	1.0138±0.0043
91				8.8 ± 0.2	42 ± 1		1.0040±0.0056
92				9.5 ± 0.2	20 ± 1		1.0070±0.0055
93				10.3 ± 0.2	0 ± 0		—
94	16	10.2	10.2	0 ± 0.2	92 ± 1		1.0085±0.0049
95				3.8 ± 0.2	61 ± 4		1.0070±0.0054
96				8.1 ± 0.2	26 ± 1		1.0057±0.0049
97				10.3 ± 0.2	0 ± 0		—
98	16	7.6	10.2	0 ± 0.5	96 ± 1		1.0049±0.0049
99				4.7 ± 0.2	59 ± 2		1.0000±0.0054
100				7.9 ± 0.2	26 ± 2		0.9850±0.0055
101				10.5 ± 0.2	0 ± 2		0.9826±0.0055
102	16	5.1	10.2	0 ± 0.5	105 ± 1		1.0034±0.0057
103				3.4 ± 0.2	75 ± 1		0.9969±0.0053
104				7.4 ± 0.2	36 ± 1		0.9928±0.0057
105				10.6 ± 0.2	2 ± 1		1.0056±0.0055
106	16	2.5	10.2	0 ± 0.5	149		0.9736±0.0059
107				3.3 ± 0.2	114 ± 1		0.9846±0.0055
108				6.0 ± 0.2	72 ± 1		0.9696±0.0058
109				8.6 ± 0.2	30 ± 1		0.9887±0.0051
110	16	0	10.2	9.8 ± 0.2	9 ± 1		0.9678±0.0052
111				8.3 ± 0.2	110 ± 0.6		0.9577±0.0055
112				8.7 ± 0.2	67.5±0.6		0.9715±0.0058
113				9.6 ± 0.2	27.5±0.6		0.9755±0.0054
114	16	0	5.1	8.7 ± 0.2	109.0 ± 0.5		0.9738±0.0056
115				9.0 ± 0.2	69.5 ± 0.5		0.9738±0.0056
116				9.9 ± 0.2	260 ± 0.5		0.9766±0.0057
117				11.0 ± 0.2	7.5 ± 0.5		0.9892±0.0059
118	16	0	0	10.0 ± 0.2	108 ± 1	520	—
119				10.4 ± 0.2	69 ± 1		—
120				11.3 ± 0.2	26 ± 1		—
121				12.3 ± 0.2	4 ± 1		—

**Table 3.B.3.1** Calculated  $k_{eff}$ 's for the array of cylinders reflected by concrete

No.	Cylinder inner dia.(cm)	Stainless steel sleeve	Array size	Uranium concentration (g/l)	Critical solution level(cm)	Calculated $k_{eff}$
1	21.12	Yes	4 × 4	67.28	28.63	0.9914±0.0050
2	21.12	No	4 × 4	67.28	27.15	0.9896±0.0043
3	21.12	Yes	2 × 2	76.09	60.70	0.9940±0.0054
4	21.12	No	2 × 2	76.09	62.34	0.9757±0.0051
5	21.12	No	2 × 2	80.72	57.88	0.9849±0.0052
6	16.12	Yes	4 × 4	83.49	57.34	0.9873±0.0045
7	16.12	No	4 × 4	83.49	51.21	0.9954±0.0054
8	21.12	Yes	4 × 4	369.96	17.24	0.9828±0.0063
9	21.12	No	4 × 4	364.11	17.13	0.9896±0.0065
10	21.12	Yes	2 × 2	360.37	29.49	0.9817±0.0051
11	21.12	No	2 × 2	364.11	31.11	0.9791±0.0058
12	16.12	Yes	4 × 4	360.37	32.32	0.9841±0.0054
13	16.12	No	4 × 4	359.55	31.82	0.9891±0.0059
14	16.12	No	2 × 4	359.55	51.45	0.9872±0.0055
15	16.12	No	2 × 3	359.55	65.49	0.9822±0.0057
16	16.12	Yes	2 × 2	359.55	101.45	0.9768±0.0049
17	16.12	No	2 × 2	359.55	104.04	0.9843±0.0052
					Average	0.9856

**Table 3.B.3.2** Calculated  $k_{eff}$ 's for the array of cylinders reflected by plastic

**Table 3.B.4.1** Calculated results for two aluminum cylindrical reactors

	Case No.	H/ <sup>235</sup> U Atomic Ratio	Surface Separation (cm)	Critical Height (cm)	Calculated Results	
					$k_{eff}$	$\sigma$
Cylinder Dia. = 8in	1- 1 <sup>R</sup>	29.9	0	13.4	1.0009*	0.0052
	2 <sup>R</sup>	52.9	0	13.2	0.9891**	0.0028
	3 <sup>R</sup>	52.9	3.5	14.9	0.9879*	0.0052
	4 <sup>R</sup>	52.9	7.0	17.0	0.9877*	0.0053
	5 <sup>R</sup>	52.9	14.7	19.1	0.9952*	0.0052
	6 <sup>R</sup>	52.9	20.1	18.6	0.9803**	0.0031
Cylinder Dia. = 10in	1- 7	169	15.6	35.8	0.9719*	0.0067
	8	329	16.6	64.7	0.9717*	0.0059
	9 <sup>R</sup>	329	0	16.9	0.9931*	0.0049
	10 <sup>R</sup>	329	8.0	21.1	0.9860**	0.0040
	11 <sup>R</sup>	52.9	13.0	12.9	0.9983*	0.0054
	12 <sup>R</sup>	52.9	20.0	13.0	0.9924*	0.0052
	13 <sup>R</sup>	29.9	0	11.0	1.0074	0.0055
Cylinder Dia. = 15in	1-14	169	0.3	17.3	0.9493*	0.0060
	15	169	5.0	17.8	0.9574	0.0058
	16	169	15.0	18.0	0.9543*	0.0056
	17	329	0.2	20.1	0.9546*	0.0052
	18	329	5.0	20.8	0.9567*	0.0053
	19	329	9.7	21.0	0.9566*	0.0054
	20	329	31.3	21.3	0.9589*	0.0059
	21 <sup>R</sup>	52.9	0	7.3	0.9722*	0.0054
	22 <sup>R</sup>	52.9	5.8	7.65	0.9789*	0.0051
	23 <sup>R</sup>	52.9	11.6	7.7	0.9711*	0.0053
	24 <sup>R</sup>	169	0	9.0	0.9709*	0.0031
	25 <sup>R</sup>	169	3.0	9.3	0.9881*	0.0057
	26 <sup>R</sup>	169	22.0	9.7	0.9672*	0.0053
	27 <sup>R</sup>	329	0	11.5	0.9747*	0.0055
	28 <sup>R</sup>	329	5.0	12.3	0.9754*	0.0049
	29 <sup>R</sup>	329	20.0	12.6	0.9685**	0.0027

R: Experiment with water reflector.

\*: Calculated with 30,000 neutron histories.

\*\*: Calculated with 100,000 neutron histories.

**Table 3.B.5.1** Calculated results for cubic arrays of uranium metal cylinders without reflectors

Case No.	Array	Cylinder		Critical Separation		Calculated Results	
		Radius (cm)	Height (cm)	Hori.	Vert.	$k_{eff}$	$\sigma$
2-1	3 <sup>3</sup>	5.7545	5.382	0	6.127	0.9957***	0.0017
2	2 <sup>3</sup>	5.747	8.077	0	3.417	0.9992***	0.0017
3	2 <sup>3</sup>	5.753	10.765	1.997	2.738	1.0105*	0.0044
4	3 <sup>3</sup>	5.742	10.765	6.118	6.837	1.0162**	0.0025
5	2 <sup>3</sup>	5.7545	13.459	4.269	2.319	1.0169*	0.0042

\*: Calculated with 30,000 neutron histories.

\*\*: Calculated with 100,000 neutron histories.

\*\*\*: Calculated with 200,000 neutron histories.

**Table 3.B.5.2** Calculated results for cubic arrays of uranium metal cylinders with reflectors

Case No.	Array	Cylinder Radius (cm)	Reflector Height (cm)	Reflector Thickness (cm)	Critical Separation (cm)	Calculated Results
		$k_{\text{eff}}$	$\sigma$	$k_{\text{eff}}$	$\sigma$	
2- 6	3 <sup>3</sup>	5.7545	5.382	0	2.007	1.0185*
7				1.3	2.992	1.0050*
8				3.8	5.872	1.0035*
9				7.6	8.258	1.0090*
10				15.2	8.689	1.0140*
11	2 <sup>3</sup>	5.747	8.077	0	0.902	1.0144*
12				1.3	1.905	1.0198*
13				3.8	4.961	1.0058*
14				7.6	7.391	1.0094*
15				15.2	7.823	1.0117*
16	2 <sup>3</sup>	5.753	10.765	0	2.248	1.0188*
17				1.3	3.678	1.0130**
18				2.5	5.710	1.0193**
19				3.8	8.207	1.0153**
20				7.6	11.509	1.0158**
21				15.2	11.986	1.0121**
22	3 <sup>3</sup>	5.742	10.765	0	6.363	1.0166**
23				1.3	8.574	1.0035**
24				3.8	14.764	1.0169**
25				7.6	18.720	1.0148**
26				15.2	19.147	1.0221**
27	2 <sup>3</sup>	5.7545	13.459	0	3.543	1.0088*
28				1.3	5.423	1.0038*
29				3.8	11.532	1.0055*
30				7.6	15.697	1.0139*
31				15.2	16.378	1.0120*

**Table 3.B.6.1** Calculated results for arrays of plutonium metal billets

Case No.	Array	Moderator Thickness (cm)	$k_{\text{eff}}$	Calculated Results
		$\sigma$	$\sigma$	
3- 1	2 <sup>3</sup>	5.40	7.30	1.0018*
	2	3 <sup>3</sup>	7.71	1.0088*
	3	4 <sup>3</sup>	7.86	1.0033*
	4 <sup>R</sup>	2 <sup>3</sup>	5.74	0.9982*
	5 <sup>R</sup>	3 <sup>3</sup>	8.24	1.0121*
	billet	6	9.63	0.9761**
		7	14.19	1.27
		8	13.64	2.54
		9	13.63	3.81
		10	11.98	0.9878**
		11	13.68	1.0055**
		12	14.51	0.0029
		13	10.91	
		14	11.93	
		15	32.12	
		16	32.12	
		17	13.09	
		18	20.19	
		19	17.50	
		20	17.28	
		21	22.12	
		22	15.23	
		23	15.23	
		24	17.12	
		25	17.50	
		26	21.24	
		27	21.24	
		28	25.82	
		29	25.82	
		30	21.24	
		31	25.82	

R: Experiment with polyethylene reflector

\*: Calculated with 30,000 neutron histories

\*\*: Calculated with 100,000 neutron histories

\*\*\*: Calculated with 200,000 neutron histories

**Table 3.B.7.1** Calculated results for cubic arrays of cylinders containing uranyl-nitrate solution with reflectors of various thicknesses

Case No.	Array	Surface Separation (cm)	Reflector Thickness (cm)	Calculated Results	
				$k_{eff}$	$\sigma$
4- 1	2 <sup>3</sup>	1.43	None	0.9763 **	0.0031
2	2 <sup>3</sup>	3.28	Paraffin 1.27	0.9760 **	0.0031
3	2 <sup>3</sup>	6.91	Paraffin 3.81	0.9919 *	0.0053
4	2 <sup>3</sup>	8.48	Paraffin 7.62	1.0044 *	0.0049
5	2 <sup>3</sup>	8.99	Paraffin 15.24	0.9967 **	0.0030
6	2 <sup>3</sup>	3.00	Plexiglas 1.27	0.9815 **	0.0029
7	3 <sup>3</sup>	6.48	None	0.9679 **	0.0031
8	3 <sup>3</sup>	9.02	Paraffin 1.27	0.9797 **	0.0030
9	3 <sup>3</sup>	13.69	Paraffin 3.81	0.9977 *	0.0060
10	3 <sup>3</sup>	16.53	Paraffin 15.24	1.0025 **	0.0032
11	3 <sup>3</sup>	8.76	Plexiglas 1.27	0.9756 **	0.0032
12	4 <sup>3</sup>	10.67	None	0.9578 **	0.0031
13	5 <sup>3</sup>	14.40	None	0.9584 **	0.0031

\* Calculated with 30,000 neutron histories

\*\* Calculated with 100,000 neutron histories

**Table 3.B.7.2** Calculated results for cubic arrays of cylinders containing uranyl-nitrate solution with reflectors of various thicknesses on five sides and a 15.24 cm thick paraffin reflector on the bottom

Case No.	Array	Surface Separation (cm)	Reflector Thickness	Calculated Results	
				$k_{eff}$	$\sigma$
4-14	2 <sup>3</sup>	3.86	Paraffin 1.27	0.9935 *	0.0067
15	2 <sup>3</sup>	7.26	Paraffin 3.81	0.9870 **	0.0031
16	2 <sup>3</sup>	8.71	Paraffin 7.62	1.0002 *	0.0056
17	2 <sup>3</sup>	8.99	Paraffin 15.24	0.9967 **	0.0030
18	2 <sup>3</sup>	3.61	Plexiglas 1.27	0.9842 **	0.0030
19	2 <sup>3</sup>	5.41	Plexiglas 2.54	0.9963 *	0.0053
20	2 <sup>3</sup>	7.39	Plexiglas 4.45	1.0061 *	0.0061
21	2 <sup>3</sup>	8.64	Plexiglas 6.35	0.9974 *	0.0053
22	2 <sup>3</sup>	9.53	Plexiglas 11.43	1.0008 *	0.0058
23	2 <sup>3</sup>	9.60	Plexiglas 15.24	1.0026 *	0.0028
24	3 <sup>3</sup>	9.88	Paraffin 1.27	0.9849 **	0.0031
25	3 <sup>3</sup>	14.27	Paraffin 3.81	0.9950 *	0.0055
26	3 <sup>3</sup>	15.85	Paraffin 7.62	1.0042 *	0.0055
27	3 <sup>3</sup>	16.53	Paraffin 15.24	0.9917 *	0.0058
28	3 <sup>3</sup>	9.58	Plexiglas 1.27	0.9786 **	0.0032
29	3 <sup>3</sup>	11.94	Plexiglas 2.54	0.9856 **	0.0031

\* Calculated with 30,000 neutron histories

\*\* Calculated with 100,000 neutron histories

**Table 3.B.8.1** Description of parameters of U(93.15)O<sub>2</sub> container units

Container	Container type			
	I	II	III	IV
Number of containers	196	18	9	9
Inside radius, cm	3.735	7.65	6.76	6.76
Inside height, cm	11.59	21.76	25.18	25.18
Wall thickness, cm	0.055	0.03	0.03	0.03
Average weight, g	128	360	341	341
$\text{UO}_2$				
Height, cm	4.48	19.76	17.87	17.87
Density, g/cm <sup>3</sup>	2.144	5.505	6.628	6.628
Mass, g	421	20,000	17,000	17,000
Uranium analysis, g of U/g of $\text{UO}_2$	0.8754	0.8780	0.8780	0.8780
$\text{C}_2\text{H}_6\text{O} - 5\% \text{ H}_2\text{O}$ in $\text{UO}_2$ Region				
Density, g/cm <sup>3</sup>		0.3458	0.3458	0.3458
Mass, g		711.5	711.5	711.5
$\text{H}/^{235}\text{U}$ Atom ratio		1.556	1.556	1.556
$\text{C}_2\text{H}_6\text{O} - 5\% \text{ H}_2\text{O}$ Layer above oxide				
Height, cm		3.14	0.53	0.53
Density, g/cm		0.8020	0.8020	0.8020
Mass, g		36.1	61.5	61.5
Total alcohol mass, g		1,073	773	773

**Table 3.B.8.2** Description of cells and experimental multiplication factors for U(93)O<sub>2</sub> moderated and unmoderated

**Table 3.B.8.2** (continued)

Experiment number	Array size	Material	Geometry	Cell description					Dimensions, cm	k(exp)
				R	$\pm X$	$\pm Y$	$+Z$	$-Z$		
8	3X3X2	Void	Cuboid	—	$\pm 10.22$	$\pm 10.22$	$+11.15$	$-11.15$	1.0001	
		$C_5O_2H_8$	Cuboid	—	$\pm 10.22$	$\pm 10.22$	$+1.18$	$-1.18$		
9	3X3X2	$C_5O_2H_8$	Cuboid	—	$\pm 10.22$	$\pm 10.22$	$+1.81$	$-1.81$	1.0137	
10	3X3X2	One corner unit removed from No.9							0.9996	
11	3X3X2	$C_5O_2H_8$	Cuboid	—	$\pm 10.22$	$\pm 10.22$	$+2.41$	$-2.41$	1.0129	
12	3X3X2	One corner unit removed from No.11							0.9989	
13	3X3X2	$C_5O_2H_8$	Cuboid	—	$\pm 10.22$	$\pm 10.22$	$+3.04$	$-3.04$	1.0049	
20 kg units with 2-in-thick methyl methacrylate in three dimensions										
14	3X3X2	Void	Cuboid	—	$\pm 10.27$	$\pm 10.27$	$+11.15$	$-11.15$	—	
15	3X3X2	Three units removed from top edge							1.010	
16	3X3X2	Three units removed from top edge, 2 from opposite corners, and 1 from the middle of an edge							1.0018	
17	3X3X2	Void	Cuboid	—	$\pm 10.27$	$\pm 10.27$	$+11.15$	$-11.15$	1.0012	
		$C_5O_2H_8$	Cuboid	—	$\pm 12.70$	$\pm 12.70$	$+11.15$	$-15.97$		
		Void	Cuboid	—	$\pm 12.70$	$\pm 12.70$	$+15.28$	$-15.97$		
		Void	Cuboid	—	$\pm 12.70$	$\pm 12.70$	$+18.77$	$-15.97$		
20 kg units with 1-in-thick methyl methacrylate in three dimensions										
19	3X3X2	Void	Cuboid	—	$\pm 15.30$	$\pm 10.22$	$+11.15$	$-11.15$	1.0054	
17 kg units with $C_2H_6O$ -5% $H_2O$ , Type 3										
20	2X2X1	$UO_2-C_2H_6O$	Cylinder	6.76	—	—	$+5.27$	$-12.60$	1.0065	
		$C_2H_6O$	Cylinder	6.76	—	—	$+8.41$	$-12.60$		
		Void	Cylinder	6.76	—	—	$+12.58$	$-12.60$		
		Steel	Cylinder	6.79	—	—	$+12.63$	$-12.63$		
		Void	Cuboid	—	$\pm 6.985$	$\pm 6.985$	$+12.78$	$-12.78$		
21	2X2X1	Void	Cuboid	—	$\pm 7.025$	$\pm 7.025$	$+12.78$	$-12.78$	1.0031	
22	2X2X1	Void	Cuboid	—	$\pm 7.064$	$\pm 7.064$	$+12.78$	$-12.78$	1.0018	
23	2X3X1	Void	Cuboid	—	$\pm 7.620$	$\pm 7.620$	$+12.78$	$-12.78$	1.0146	
24	2X3X1	Void	Cuboid	—	$\pm 7.938$	$\pm 7.938$	$+12.78$	$-12.78$	1.0025	
25	2X3X1	Void	Cuboid	—	$\pm 8.017$	$\pm 8.017$	$+12.78$	$-12.78$	0.9988	
26	2X4X1	Void	Cuboid	—	$\pm 8.493$	$\pm 8.493$	$+12.78$	$-12.78$	1.0003	
27	3X3X1	Void	Cuboid	—	$\pm 9.049$	$\pm 9.049$	$+12.78$	$-12.78$	0.9996	
17 kg units with $C_2H_6O$ -5% $H_2O$ , Type 4										
28	3X3X1	Void	Cuboid	—	$\pm 8.573$	$\pm 8.573$	$+12.78$	$-12.78$	1.0042	
29	3X3X1	Void	Cuboid	—	$\pm 8.652$	$\pm 8.652$	$+12.78$	$-12.78$	0.9997	

**Table 3.B.8.3** Principal critical parameters and calculated  $k_{\text{eff}}$ 's

No.	Fuel					$k_{\text{eff}}$ (Measured Value)	Calculated $k_{\text{eff}}$	C/E		
	Type of Fuel	Density (g/cm <sup>3</sup> )	Array	Pitch (cm)						
X, Y	Z	Unit Numbers								
1	I.D. = 3.735 cm	2.144	7×7×4	10.39	14.05	196	1.0167	1.0323±0.0049	1.015	
2						178	1.0012	0.9988±0.0044	0.998	
3						—	—	1.0006±0.0045	—	
4						196	—	1.0515±0.0052	—	
5	I.D. = 7.65 cm	4.999	3×3×2	20.44	22.3	16	1.0094	1.0122±0.0044	1.003	
6						12	0.9968	0.9875±0.0053	0.991	
7						18.16	1.0014	1.0002±0.0046	0.999	
8						24.66	1.0001	0.9869±0.0048	0.987	
9						25.92	1.0137	1.0017±0.0050	0.988	
10						17	0.9996	0.9759±0.0046	0.976	
11						27.12	1.0129	1.0082±0.0048	0.995	
12						17	0.9989	0.9891±0.0043	0.990	
13						28.38	1.0049	1.0099±0.0047	1.005	
14						18	—	1.0743±0.0051	—	
15						27.12	1.010	1.0204±0.0051	1.010	
16						25.4	1.0018	0.9555±0.0042	0.954	
17						31.25	1.0012	1.0221±0.0048	1.021	
18						34.74	1.009	1.0225±0.0048	1.013	
19						33.03(X) 22.87(Y)	1.0054	0.9926±0.0047	0.987	
20	I.D. = 6.76 cm	6.626	2×3×1	16.986	13.97	4	1.0065	0.9953±0.0049	0.989	
21						14.05	1.0031	0.9966±0.0041	0.994	
22						13.128	1.0018	0.9974±0.0052	0.996	
23						15.24	1.0146	1.0175±0.0051	1.003	
24						15.876	6	1.0025	1.0108±0.0043	1.008
25						16.034	—	0.9988	0.9884±0.0050	0.990
26						2×4×1	8	1.0003	0.9981±0.0049	0.998
27	$\text{C}_2\text{H}_6\text{O}$ -5% $\text{H}_2\text{O}$ = 711.5 g		3×3×1	17.146	18.098	—	—	0.9996	0.9929±0.0056	0.998
28						17.304	9	1.0042	0.9982±0.0042	0.994
29						—	—	0.9997	0.9944±0.0050	0.995

**Table 3.C.1.1** Calculated results for U(2.6)O<sub>2</sub>-H<sub>2</sub>O systems

Lattice Name	Pattern	Critical Water Level [cm]	$k_{\text{eff}}$	$\pm \sigma$
1.50U $(H/U = 4.33)$ $(\text{Pitch} = 1.849 \text{ cm})$	19 × 19	99.45	0.99597	0.00312
	20 <sup>2</sup>	73.73	1.00010	0.00282
	21 <sup>2</sup>	60.81	0.99852	0.00293
	22 <sup>2</sup>	53.23	0.99276	0.00275
	23 <sup>2</sup>	47.81	0.99510	0.00282
	24 <sup>2</sup>	43.94	0.99469	0.00300
	25 <sup>2</sup>	40.89	0.98960	0.00288
1.83U $(H/U = 5.28)$ $(\text{Pitch} = 1.956 \text{ cm})$	12 × 29	131.94	0.99405	0.00311
	14 × 27	69.01	0.99113	0.00295
	14 × 24	85.36	0.99234	0.00321
	14 × 21	135.70	0.98645	0.00283
	15 × 19	139.72	0.99175	0.00271
	15 × 20	113.95	0.99717	0.00281
	15 × 21	94.58	0.99812	0.00289
	15 × 22	83.45	0.99068	0.00305
	15 × 23	75.74	0.99311	0.00290
	16 × 18	120.73	0.98465	0.00282
	17 × 17	114.59	0.99091	0.00328
	17 × 17+20	81.17	0.99218	0.00286
	18 × 18	75.32	0.99364	0.00302
	18 × 19	66.90	0.99321	0.00307
	19 × 19	60.38	0.99050	0.00279
	19 × 20	55.53	0.99597	0.00285
	20 × 20	51.65	0.99288	0.00351
	20 × 21	48.62	0.99277	0.00278
	21 <sup>2</sup>	46.01	0.99875	0.00285
	21 × 23	43.91	0.99278	0.00291
	22 <sup>2</sup>	42.12	0.99256	0.00286
	22 × 23	40.50	0.98987	0.00299
2.48U $(H/U = 7.16)$ $(\text{Pitch} = 2.150 \text{ cm})$	16 × 16	78.67	0.99121	0.00286
	17 × 17	59.96	0.99120	0.00294
	18 × 18	50.52	0.99181	0.00258
	19 × 19	44.55	0.99235	0.00295
	20 × 20	40.44	0.98858	0.00301
3.00U $(H/U = 8.65)$ $(\text{Pitch} = 2.293 \text{ cm})$	15 × 15	90.75	0.99343	0.00304
	16 × 16	64.42	0.98876	0.00299
	17 × 17	53.87	0.98980	0.00289
	18 × 18	46.06	0.98756	0.00305
	18 × 20	41.79	0.99043	0.00274
	19 × 19	41.54	0.99038	0.00292

**Table 3.C.1.2** Calculated results for PuO<sub>2</sub>-UO<sub>2</sub>-H<sub>2</sub>O systems

Lattice Name	Pattern	Critical Water Level [cm]	Date	k <sub>eff</sub>	± σ
2.42PU $(H/(U+Pu) = 12.10)$ $(\text{Pitch} = 1.825 \text{ cm})$	22 <sup>2</sup>	69.41	'72-6- 5	1.00128	0.00295
	22 × 23	69.57	'74-5-14	0.99391	0.00311
	23 <sup>2</sup>	59.55	'72-6- 7	0.99998	0.00304
	23 <sup>2</sup>	61.90	'73-5-14	0.99355	0.00304
	23 <sup>2</sup>	64.06	'74-5-14	0.99493	0.00305
	23 <sup>2</sup>	66.46	'75-5-16	0.99618	0.00320
	23 × 24	60.06	'74-5-14	0.99612	0.00277
	24 <sup>2</sup>	53.30	'72-6- 7	0.99423	0.00313
	24 <sup>2</sup>	56.68	'74-5-14	0.99815	0.00310
	24 <sup>2</sup>	58.36	'75-5-16	0.99559	0.00302
2.98PU $(H/(U+Pu) = 14.87)$ $(\text{Pitch} = 1.956 \text{ cm})$	20 × 21	67.10	'72-5-18	0.99724	0.00300
	21 <sup>2</sup>	61.50	'72-5-18	0.99734	0.00294
	21 <sup>2</sup>	64.39	'73-5-22	0.99565	0.00283
	21 <sup>2</sup>	66.87	'74-5-28	0.99804	0.00297
	21 <sup>2</sup>	69.40	'75-5-21	1.00584	0.00282
	21 × 22	57.38	'72-5-18	0.99879	0.00278
	21 × 22	60.37	{'73-5-22 '73-6- 6}	1.00051	0.00282
	21 × 22	61.92	'74-5-28	0.99861	0.00306
	21 × 22	63.88	'75-5-21	0.99502	0.00311
	22 <sup>2</sup>	57.83	'74-5-28	0.99467	0.00296
	22 × 23	54.72	'74-5-28	0.99452	0.00262
	23 <sup>2</sup>	51.94	'74-5-28	0.99676	0.00290
	23 × 24	49.82	'74-5-28	0.99315	0.00303
	24 <sup>2</sup>	47.78	'74-5-28	0.99592	0.00275
	24 <sup>2</sup>	48.68	'75-5-21	1.00151	0.00306
4.24PU $(H/(U+Pu) = 21.16)$ $(\text{Pitch} = 2.225 \text{ cm})$	19 × 20	68.67	'73-5-29	0.99338	0.00283
	20 <sup>2</sup>	60.32	'72-4-13	0.99223	0.00288
	20 <sup>2</sup>	62.95	'73-5-29	0.99556	0.00267
	20 <sup>2</sup>	65.63	'74-6- 5	0.98485	0.00297
	20 <sup>2</sup>	68.18	'75-5-28	1.00048	0.00285
	20 × 21	61.07	'74-6- 5	0.99072	0.00278
	21 <sup>2</sup>	53.41	'72-4-14	0.99066	0.00279
	21 <sup>2</sup>	55.48	'73-5-29	0.99490	0.00300
	21 <sup>2</sup>	57.28	'74-6- 5	1.00086	0.00291
	21 <sup>2</sup>	59.05	'75-5-28	0.99897	0.00265
	21 × 22	54.33	'74-6- 5	0.99886	0.00282
	22 <sup>2</sup>	51.74	'74-6- 6	0.99307	0.00289
	23 <sup>2</sup>	47.72	'74-6- 6	0.99349	0.00288
	24 <sup>2</sup>	44.67	'74-6- 6	0.99109	0.00285
	24 <sup>2</sup>	45.62	'75-5-28	0.99419	0.00311
5.55PU $(H/(U+Pu) = 27.72)$ $(\text{pitch} = 2.474 \text{ cm})$	20 × 21	69.16	'73-6- 6	0.99192	0.00257
	21 <sup>2</sup>	62.05	'72-4-28	0.99662	0.00286
	21 <sup>2</sup>	64.53	'73-6- 7	0.99989	0.00279
	21 × 22	58.73	'72-4-26	0.98517	0.00271
	21 × 22	61.10	'73-6- 6	0.99048	0.00278
	22 <sup>2</sup>	58.08	'73-6- 6	0.98960	0.00248
	22 × 23	55.69	'73-6- 6	0.99557	0.00261
	23 <sup>2</sup>	53.50	'73-6- 6	0.99155	0.00286

**Table 3.C.1.3** Calculated results for the water gap pattern

Lattice name	Pattern	Critical Water Level [cm]	$k_{eff}$	$\pm \sigma$
1.50U $(H/U = 4.33)$ $(pitch = 1.849 \text{ cm})$	24 × 25	42.42	0.99447	0.00448
	24 × 26	48.80	0.99836	0.00404
	24 × 27	64.05	0.99565	0.00414
	24 × 28	100.47	0.99850	0.00387
3.00U $(H/U = 8.65)$ $(pitch = 2.293 \text{ cm})$	20 × 19	46.84	0.99365	0.00429
	20 × 20	65.81	0.99438	0.00411
	20 × 21	130.77	0.99764	0.00394

**Table 3.C.1.4** Calculated results for  $\text{UO}_2$  lattices with boric acid solution

Lattice name	Pattern	Boron Concentration ppm	Critical Water Level [cm]	$k_{eff}$	$\pm \sigma$
1.83U $(H/U = 5.28)$ $(pitch = 1.956 \text{ cm})$	18 × 19	72.3	85.78	0.99828	0.00322
	19 <sup>2</sup>	146.6	96.85	0.99579	0.00298
	22 <sup>2</sup>	345.0	84.99	1.00547	0.00301
	25 <sup>2</sup>	554.0	85.79	1.00672	0.00277

**Table 3.C.1.5** Calculated results for  $\text{UO}_2$  lattices with the poison sheet

Lattice name	Poison Name	Position	Critical Water Level (cm)	$k_{eff}$	$\pm \sigma$
1.83U $(H/U = 5.28)$ $(pitch = 1.956 \text{ cm})$	Boral	16P16	120.4	1.00058	0.00297
		13P19	100.1	0.99959	0.00275
		11P21	83.9	1.00438	0.00296
1.83U	Al-Cd (No. 1)	13P14	86.62	0.99994	0.00284
		10P17	83.92	0.99808	0.00323
		6P21	78.11	0.99762	0.00291
		3P24	73.96	0.98999	0.00315
		1P26	72.00	0.99794	0.00303
1.83U	Al-Cd (No. 2)	13P14	98.24	1.00262	0.00300
		10P17	93.27	0.99671	0.00294
		6P21	82.46	0.99705	0.00315
		3P24	75.92	1.00237	0.00275
		1P26	73.10	0.99393	0.00305
1.83U	Al-Cd (No. 3)	13P14	115.87	1.00058	0.00283
		10P17	105.85	1.00161	0.00307
		6P21	87.40	0.99796	0.00280
		3P24	78.08	0.99696	0.00292
		1P26	74.32	0.99988	0.00321
1.83U	Al-Cd (No. 4)	9P18	118.38	1.00036	0.00319
		6P21	94.04	1.00153	0.00296
		3P24	80.39	0.99392	0.00271
		1P26	75.76	0.99998	0.00306
1.83U	Al-Cd (No. 5)	8P19	126.60	0.99526	0.00295
		6P21	102.13	0.99535	0.00293
		3P24	83.32	0.99680	0.00321
		1P26	77.40	0.99616	0.00313

**Table 3.C.2.1** Calculated results and experimental conditions for UO<sub>2</sub> rod clusters partially immersed in water without removal of rods

No.	Type	Pitch (mm)	Number of Rods	Number of Rods on Each Side	Critical Water Height (cm)	$k_{eff} \pm \sigma$
1	Square	12.6	484	22×22	90.69 ±0.10	0.99090 ±0.00349
2		16.0	272	16×17	73.53 ±0.10	0.98616 ±0.00348
3		21.0	255	15×15	77.98 ±0.06	0.98527 ±0.00354
4		25.2	306	17×18	79.85 ±0.01	0.99280 ±0.00305
5	Triangular (hexagonal shape)	13.5	547	14 Each side	60.93 ±0.06	0.99141 ±0.00344
6		17.2	271	10 Each side	68.06 ±0.06	0.98307 ±0.00327
7		22.6	217	9 Each side	79.50 ±0.06	0.99182 ±0.00350
8	Triangular with irregular hexagonal shape	13.5	519	13/13/15/13/13/15	69.50 ±0.06	0.99379 ±0.00374
9			505	15/12/15/12/15/12	75.79 ±0.07	0.99573 ±0.00345
10			495	14/14/12/14/14/12	82.43 ±0.06	0.98754 ±0.00354
11	Triangular with pseudo-cylindrical shape	13.5	484	See Fig. 3.C.2.1	85.21 ±0.06	0.98532 ±0.00342
12		17.2	277	See Fig. 3.C.2.1	61.99 ±0.06	0.98319 ±0.00363
13		22.6	225	See Fig. 3.C.2.1	70.44 ±0.07	0.98929 ±0.00299

**Table 3.C.2.2** Calculated results and experimental conditions for UO<sub>2</sub> rod clusters with systematic removal of rods

No.	Square Pitch (mm)	Pattern	Number of Rods	Number of Holes	Frequency of Hole	Critical Water Height (cm)	$k_{eff} \pm \sigma$
14	12.6	22×22	459	25	1 in 5 spaces	81.36 ±0.07	0.99816 ±0.00398
15			448	36	1 in 4 spaces	77.69 ±0.07	0.99409 ±0.00337
16			420	64	1 in 3 spaces	73.05 ±0.06	0.99214 ±0.00345
17			363	121	1 in 2 spaces	58.77 ±0.06	0.99420 ±0.00382
18		21×21	392	49	1 in 3 spaces	89.07 ±0.07	0.98399 ±0.00361
19			320	121	1 in 2 spaces	84.37 ±0.06	0.98947 ±0.00292

**Table 3.C.2.3** Calculated results and experimental conditions for UO<sub>2</sub> rod clusters with removal of rods in a small region

No.	Pitch (mm)	Pattern	Number of Rods	Number of Holes	Frequency of Holes	keff ± σ	
						Alternate rods removed	
20	12.6	22×22	459	25			
		Configuration (See Fig. 3.C.2.2)			Locations of the Holes <sup>a</sup>		
					Row Number ↓	Column Number →	Critical Water Height (cm)
21							
22							
23							
24							
25							
26							

<sup>a</sup> Rows and columns are numbered from the upper left corner.

**Table 3.C.3.1** Calculated results for  $\text{UO}_2$  rod clusters ( $^{235}\text{U}$  enrichment: 2.35 wt%, lattice pitch: 20.32 mm) submerged in water with various poison plates

Serial No.	Cluster Array Size	No. Clus	Plate Material	tp mm	G mm	Xc mm	keff	$\pm \sigma$
1	20×18.08	1	—	—	—	$\infty$	0.99086	0.00294
2	20×17	3	—	—	—	119.2	0.98813	0.00276
3	20×16	3	—	—	—	83.9	—	—
4	20×16	3	—	—	—	84.1	—	—
5	20×16	3	—	—	—	84.2	0.99052	0.00299
6	20×16	3	—	—	—	84.4	—	—
7	22×16	3	—	—	—	100.5	0.98948	0.00320
8	20×15	3	—	—	—	63.9	0.98824	0.00268
9	24×14	3	—	—	—	80.1	1.00001	0.00297
10	20×16	3	—	—	—	44.6	0.99200	0.00324
11	20×16	3	304L Steel (B=0)	4.85	6.45	68.8	0.98874	0.00278
12	20×16	3	304L Steel (B=0)	4.85	27.32	76.4	0.98997	0.00301
13	20×16	3	304L Steel (B=0)	4.85	40.42	75.1	0.98686	0.00297
14	20×16	3	304L Steel (B=0)	3.02	6.45	74.2	0.98708	0.00305
15	20×16	3	304L Steel (B=0)	3.02	40.42	77.6	0.99188	0.00328
16	20×17	3	304L Steel (B=0)	3.02	6.45	104.4	0.99465	0.00279
17	20×17	3	304L Steel (B=0)	3.02	40.42	114.7	0.99699	0.00326
18	20×17	3	304L Steel (B=1.05%)	2.98	6.45	75.6	0.98370	0.00320
19	20×17	3	304L Steel (B=1.05%)	2.98	40.42	96.2	0.99023	0.00298
20	20×17	3	304L Steel (B=1.62%)	2.98	6.45	73.6	0.98839	0.00296
21	20×17	3	304L Steel (B=1.62%)	2.98	40.42	95.2	0.98983	0.00306
22	20×17	3	Boral	7.13	6.45	63.4	—	—
23	20×17	3	Boral	7.13	6.45	63.2	0.98876	0.00306
24	20×17	3	Boral	7.13	44.42	90.3	0.99262	0.00294
25	22.21×16	3	Boral	7.13	6.45	52.2	0.98828	0.00314
26	22×16	3	Boral	7.13	6.45	50.5	0.98653	0.00314
27	20×18.88	1	Boral	7.13	27.32	$\infty$	0.99194	0.00311
28	20×18.48	1	Boral	7.13	27.32	$\infty$	0.98686	0.00331
29	20×16	3	Copper (Cd=0)	6.46	6.45	66.2	0.98734	0.00313
30	20×16	3	Copper (Cd=0)	6.46	27.32	77.2	0.98574	0.00303
31	20×16	3	Copper (Cd=0)	6.46	44.42	75.1	0.98531	0.00290
32	24×15	3	Copper (Cd=0)	3.37	6.45	68.8	0.98846	0.00332
33	24×15	3	Copper (Cd=0)	3.37	40.42	70.0	0.99192	0.00276
34	24×15	3	Copper (Cd=0.989%)	3.57	6.45	51.5	0.99173	0.00300
35	24.21×15	3	Copper (Cd=0.989%)	3.57	6.45	52.2	0.98763	0.00291
36	24.83×15	3	Copper (Cd=0.989%)	3.57	40.42	52.2	0.99536	0.00300
37	20×17	3	Cadmium	0.61	6.45	67.4	0.98538	0.00294
38	20×17	3	Cadmium	0.61	14.82	76.0	0.98450	0.00296
39	20×17	3	Cadmium	0.61	40.42	98.7	0.99751	0.00294
40	20×17	3	Cadmium	0.291	14.82	77.8	0.98515	0.00294
41	20×17	3	Cadmium	0.291	40.42	94.0	0.98923	0.00286
42	20×17	3	Cadmium	0.291	14.82	75.4	0.99460	0.00288
43	20×17	3	Cadmium	0.291	40.42	93.9	0.98849	0.00288
44	20×16	3	Aluminum	6.25	6.45	86.6	0.99306	0.00299
45	20×16	3	Aluminum	6.25	40.42	87.8	0.99115	0.00311
46	20×16	3	Aluminum	6.25	44.42	88.3	0.99164	0.00290
47	20×16	3	Zircaloy 4	6.22	6.45	87.9	0.98436	0.00333
48	20×16	3	Zircaloy 4	6.52	40.42	87.8	0.99282	0.00289

**Table 3.C.3.2** Calculated results for UO<sub>2</sub> rod clusters (<sup>235</sup>U enrichment: 4.31 wt%, lattice pitch: 25.40 mm) submerged in water with various poison plates

No.	Fuel Cluster	Poison Plate			keff	$\pm \sigma$	
		Material	Thickness (mm) $t_p$	Distance G (mm)			
1	1 · 10 * 11 · 51		0.0	0.0	0.98316	0.00323	
2	1 · 9 * 13 · 35		0.0	0.0	0.99034	0.00301	
3	1 · 8 * 16 · 37		0.0	0.0	0.99087	0.00331	
4&5	3 · 15 * 8		0.0	0.0	0.99201	0.00335	
6	3 · 15 * 8	304L S.S (B=0wt%)	4.850	2.450	0.98638	0.00300	
7	3 · 15 * 8	304L S.S (B=0wt%)	4.850	3.2770	9.650	0.99050	0.00325
8	3 · 15 * 8	304L S.S (B=0wt%)	3.020	4.280	9.220	0.99441	0.00321
9	3 · 15 * 8	304L S.S (B=0wt%)	3.020	3.2770	9.760	0.98666	0.00312
10	3 · 15 * 8	304L S.S (B=1.05wt%)	2.980	4.320	61.00	0.98603	0.00348
11	3 · 15 * 8	304L S.S (B=1.05wt%)	2.980	3.2770	8.08	0.98692	0.00299
12	3 · 15 * 8	304L S.S (B=1.62wt%)	2.980	4.320	5.76	0.98673	0.00324
13	3 · 15 * 8	304L S.S (B=1.62wt%)	2.980	3.2770	7.90	0.98921	0.00339
14	3 · 15 * 8	Boron	7.130	3.2770	6.72	0.98472	0.00357
15	3 · 15 * 8	Copper (Cd=0wt%)	6.460	0.840	81.5	0.99806	0.00332
16	3 · 15 * 8	Copper (Cd=0wt%)	6.460	3.2770	94.2	0.98597	0.00357
17	3 · 15 * 8	Copper (Cd=0wt%)	3.370	-0.570	84.8	0.99475	0.00339
18	3 · 15 * 8	Copper (Cd=0wt%)	3.370	4.2410	9.64	0.99372	0.00354
19	3 · 15 * 8	Copper (Cd=0.989wt%)	3.370	-0.570	66.6	0.99036	0.00328
20	3 · 15 * 8	Copper (Cd=0.989wt%)	3.370	3.2770	83.5	0.98588	0.00322
21	3 · 15 * 8	Cadmium	0.291	7.009	5.93	0.98732	0.00360
22	3 · 15 * 8	Cadmium	0.291	3.2770	74.2	0.99196	0.00320
23	3 · 15 * 8	Cadmium	0.610	6.69	59.6	0.98962	0.00306
24	3 · 15 * 8	Cadmium	0.610	3.2770	74.2	0.98455	0.00323
25	3 · 15 * 8	Cadmium	0.901	6.40	5.87	0.98740	0.00314
26	3 · 15 * 8	Cadmium	0.901	3.2770	73.8	0.98757	0.00346
27	3 · 15 * 8	Cadmium	2.006	5.290	56.8	0.98977	0.00339
28	3 · 15 * 8	Cadmium	2.006	3.2770	72.8	0.99203	0.00282
29	3 · 15 * 8	Aluminium	6.250	1.050	107.2	0.98574	0.00330
30	3 · 15 * 8	Aluminium	6.250	3.2770	107.7	0.99135	0.00335
31	3 · 15 * 8	Zircaloy-4	6.250	0.780	109.2	0.98764	0.00342
32	3 · 15 * 8	Zircaloy-4	6.250	3.2770	108.6	0.99149	0.00350

**Table 3.C.4.1** Calculated results for UO<sub>2</sub> rod clusters submerged in water with reflecting walls

No.	Fuel Cluster	Reflector Material	Pitch (mm)	Enrichment (wt %)	Distance Y (mm)	Distance Xc (mm)	keff	$\pm \sigma$
1	3 · 19 * 16	Uranium	20.32	2.35	0.0	118.30	0.99004	0.00264
2	3 · 19 * 16	Uranium	20.32	2.35	132.1	139.3	0.98970	0.00246
3	3 · 19 * 16	Uranium	20.32	2.35	19.56	141.1	0.99366	0.00263
4	3 · 19 * 16	Uranium	20.32	2.35	26.16	137.0	0.99342	0.00262
5	3 · 19 * 16	Uranium	20.32	2.35	54.05	106.9	0.99074	0.00286
6	3 · 19 * 16	Uranium	20.32	2.35	106.76	85.6	0.98342	0.00293
7	3 · 19 * 16	Uranium	20.32	2.35	$\infty$	83.1	0.97858	0.00304
8	3 · 20 * 16	Uranium	20.32	2.35	$\infty$	91.3	0.98147	0.00318
9	3 · 19 * 16	Lead	20.32	2.35	0.0	138.4	0.98981	0.00288
10	3 · 19 * 16	Lead	20.32	2.35	6.60	137.2	0.98256	0.00292
11	3 · 19 * 16	Lead	20.32	2.35	26.16	112.5	0.99099	0.00309
12	3 · 19 * 16	Lead	20.32	2.35	$\infty$	83.1	0.98250	0.00318
13	3 · 20 * 16	Lead	20.32	2.35	$\infty$	91.3	0.98336	0.00290
14	3 · 13 * 8	Uranium	25.40	4.29	0.0	153.8	0.99111	0.00278
15	3 · 12 * 8	Uranium	25.40	4.29	19.56	153.2	0.99239	0.00339
16	3 · 13 * 8	Uranium	25.40	4.29	39.12	180.5	0.98900	0.00285
17	3 · 13 * 8	Uranium	25.40	4.29	54.05	134.9	0.98850	0.00293
18	3 · 13 * 8	Uranium	25.40	4.29	$\infty$	82.4	0.98107	0.00320
19	3 · 13 * 8	Lead	25.40	4.29	0.0	206.2	1.00077	0.00320
20	3 · 13 * 8	Lead	25.40	4.29	6.6	207.8	0.99714	0.00360
21	3 · 13 * 8	Lead	25.40	4.29	13.21	190.4	1.00175	0.00305
22	3 · 13 * 8	Lead	25.40	4.29	54.05	103.0	1.00494	0.00315
23	3 · 19 * 16	Lead	25.40	4.29	$\infty$	82.4	0.98353	0.00352

**Table 3.C.5.1** Calculated results and critical experimental data for three lined rod-clusters in water with various poison plates

Serial No.	Fuel Clusters	Pitch mm	Plate Material	$t_p$ mm	G mm	$X_c$ mm	Enrich wt%	Calculated Results	
								$k_{eff}$	$\pm \sigma$
1	3-12X16	18.92	—	—	—	122.7	4.31	0.99680	0.00366
2	3-12X16	18.92	304L STEEL	3.02	—	105.2	4.31	0.99401	0.00302
3	3-12X16	18.92	304L STEEL, 1.1 wt% B	2.98	—	72.3	4.31	0.99512	0.00332
4	3-12X16	18.92	304L STEEL, 1.6 wt% B	2.98	—	66.3	4.31	0.99466	0.00371
5	3-12X16	18.92	COPPER	3.37	—	103.6	4.31	0.99292	0.00301
6	3-12X16	18.92	COPPER-CADMIUM	3.57	—	76.1	4.31	0.99606	0.00352
7	1-25X20, 2-17X20	16.84	—	—	—	98.8	2.35	0.99156	0.00335
8	1-18X25, 2-18X20	16.84	—	—	—	67.8	2.35	1.00976	0.00298
9	1-25X20, 2-17X20	16.84	304L STEEL	3.02	—	78.0	2.35	0.99224	0.00336
10	1-25X20, 2-17X20	16.84	304L STEEL, 1.1 wt% B	2.98	—	38.6	2.35	0.99472	0.00264
11	1-25X20, 2-17X20	16.84	304L STEEL, 1.6 wt% B	2.98	—	34.6	2.35	0.99270	0.00301
12	1-25X20, 2-17X20	16.84	BORAL B	2.92	—	16.8	2.35	0.98894	0.00293
13	1-25X20, 2-17X20	16.84	BORAL C	2.31	—	19.3	2.35	0.99020	0.00294
14	1-25X20, 2-17X20	16.84	BOROFLEX	5.46	—	18.4	2.35	0.99515	0.00284
15	1-25X20, 2-17X20	16.84	BOROFLEX	4.08	—	17.3	2.35	0.99406	0.00293
16	1-25X20, 2-17X20	16.84	CADMIUM	0.61	—	30.4	2.35	0.98568	0.00341
17	1-25X18, 2-20X18	16.84	COPPER	3.37	—	52.4	2.35	0.98636	0.00341
18	1-25X18, 2-20X18	16.84	COPPER-CADMIUM	3.57	—	26.0	2.35	0.99482	0.00295

**Table 3.C.5.2** Calculated results and critical experimental data for four cross-arranged rod-clusters in water with various poison-plates

Serial No.	Fuel Clusters	Pitch (mm)	Plate Material	$t_p$ (mm)	$X_c$ (mm)	$Y_c$ (mm)	Critical Size	Calculated Results	
								$k_{eff}$	$\pm \sigma$
19	1-(18X12+7)	18.92	—	—	0.0	0.0	223	0.99031	0.00320
20	2-(9X12), $1-\{3X1\}$ , $1-\{2X1\}$	18.92	—	—	28.3	28.3	221	0.99986	0.00346
21	2-(9X12), 2-(9X2+1)	18.92	—	—	47.2	47.2	254	0.99589	0.00353
22	2-(9X12), $1-\{9X12+1\}$	18.92	—	—	66.1	66.1	433	0.99516	0.00299
23	1-(9X12+1), 1-(9X12+2)	18.92	—	—	28.3	0.0	219	0.98965	0.00312
24	2-(9X12), $1-\{6X1\}$ , $1-\{4X1\}$	18.92	—	—	28.3	83.8	225	0.99303	0.00297
25	2-(9X12), 2-(9X1)	18.92	—	—	28.3	108.6	234	0.99495	0.00391
26	2-(9X12), 2-(9X2)	18.92	—	—	28.3	112.9	252	0.99093	0.00336
27	2-(9X12), 2-(9X4)	18.92	—	—	28.3	120.2	288	0.98944	0.00292
28	2-(9X12), 2-(9X8)	18.92	—	—	28.3	136.4	360	0.99599	0.00356
29	2-(9X12), 2-(9X10)	18.92	—	—	28.3	149.8	396	0.98639	0.00357
30	4-(9X12)	18.92	—	—	28.3	198.1	432	0.99896	0.00314
31	2-(9X12), 2-(9X1)	18.92	ALUMINUM	6.25	28.3	85.0	234	0.99232	0.00341
32	2-(9X12), 2-(9X1)	18.92	ALUMINUM	6.25	28.3	90.4	234	0.99193	0.00375
33	2-(9X12), 2-(9X1)	18.92	ZIRCALOY-4	6.52	28.3	110.4	234	1.00146	0.00299
34	2-(9X10), 2-(9X8+5)	18.92	ZIRCALOY-4	6.52	28.3	85.0	334	0.99078	0.00329
35	2-(8X10), $1-\{9X9+2\}$ , $1-\{9X9+3\}$	18.92	—	—	28.3	85.0	347	0.99977	0.00360
36	2-(9X12), $1-\{9X1+6\}$ , $1-\{9X1+7\}$	18.92	304L STEEL	3.02	28.3	28.3	247	0.99786	0.00330

Table 3.C.5.2 (continued)

Serial No.	Fuel Clusters	Pitch (mm)	Plate Material	$t_p$ (mm)	$X_c$ (mm)	$Y_c$ (mm)	Critical Size	Calculated Results	
								$k_{eff}$	$\pm \sigma$
37	2-(9X12), 2-(9X2)	18.92	304L STEEL	3.02	28.3	33.8	252	0.99924	0.00370
38	2-(9X12), 2-(9X3)	18.92	304L STEEL	3.02	28.3	45.4	270	0.99395	0.00380
39	2-(9X12), 2-(9X7)	18.92	304L STEEL	3.02	28.3	64.9	342	0.99024	0.00392
40	2-(9X12), 2-(9X10)	18.92	304L STEEL	3.02	28.3	81.0	396	0.99044	0.00364
41	4-(9X12)	18.92	304L STEEL	3.02	28.3	99.6	432	1.00473	0.00320
42	2-(9X12), 2-(9X13)	18.92	304L STEEL	3.02	28.3	115.5	450	0.98971	0.00323
43	2-(9X12), 2-(9X13+2)	18.92	304L STEEL	3.02	28.3	122.9	454	0.99728	0.00387
44	2-(9X12), 2-(9X3+1)	18.92	304L STEEL	4.85	28.3	28.3	272	1.00229	0.00325
45	2-(9X12), 2-(9X5)	18.92	304L STEEL	4.85	28.3	44.7	306	0.99042	0.00346
46	4-(9X12)	18.92	304L STEEL	4.85	28.3	83.6	432	0.99613	0.00337
47	3-(9X12), $1-(9X2+2)$ $1-(9X2+3)$	18.92	COPPER	3.37	28.3	28.3	257	0.99675	0.00351
48	2-(9X12), 2-(9X7)	18.92	COPPER	3.37	28.3	59.4	342	0.99368	0.00310
49	4-(9X12)	18.92	COPPER	6.46	28.3	90.2	432	0.99169	0.00317
50	2-(9X12), 2-(9X5)	18.92	COPPER	6.46	28.3	26.7	306	0.99520	0.00398
51	2-(9X12), 2-(9X5+2)	18.92	COPPER	6.46	28.3	28.3	310	0.99922	0.00350
52	2-(9X12), 2-(9X6)	18.92	COPPER	6.46	28.3	34.7	324	0.99236	0.00385
53	2-(9X12), $1-(9X8+4)$ $1-(9X8+3)$	18.92	COPPER	6.46	28.3	47.2	367	1.00026	0.00292
54	4-(9X12)	18.92	COPPER	6.46	28.3	71.0	432	0.98935	0.00301
55	2-(9X12), 2-(9X13)	18.92	COPPER	6.46	28.3	81.2	450	0.99433	0.00304
56	2-(9X12), 2-(9X14)	18.92	COPPER	6.46	28.3	102.1	468	0.98845	0.00341
57	2-(11X14), $1-(11X15)$ $1-(11X15+1)$	18.92	BORAL-A	7.13	28.3	28.3	639	0.99910	0.00309
58	2-(11X14), 2-(11X16)	18.92	BORAL-A	7.13	28.3	48.0	660	0.99289	0.00337
59	2-(11X14), $1-(11X16+7)$ $1-(11X16+8)$	18.92	BORAL-A	7.13	28.3	85.0	675	0.99617	0.00296
60	2-(11X14), 2-(11X16+10)	18.92	BORAL-A	7.13	28.3	103.9	680	0.98829	0.00347
61	2-(11X14), 2-(11X13+5)	18.92	BORAL-B	2.92	28.3	28.3	604	0.99291	0.00343
62	4-(11X14)	18.92	BORAL-B	2.92	28.3	31.7	616	1.00013	0.00371
63	2-(11X14), 2-(11X14+9)	18.92	BORAL-B	2.92	28.3	47.2	634	0.98486	0.00338
64	2-(11X14), 2-(11X15)	18.92	BORAL-B	2.92	28.3	52.4	638	1.00120	0.00290
65	2-(11X14), 2-(11X16)	18.92	BORAL-B	2.92	28.3	85.0	660	0.99179	0.00357
66	2-(11X14), 2-(11X13+6)	18.92	BORAL-C	2.31	28.3	28.3	606	0.99689	0.00376
67	4-(11X14)	18.92	BORAL-C	2.31	28.3	35.3	616	1.00034	0.00338
68	2-(11X14), 2-(11X16+1)	18.92	BORAL-C	2.31	28.3	85.0	662	0.99838	0.00319
69	2-(11X14), $1-(11X13+3)$ $1-(11X13+4)$	18.92	BOROFLEX	5.46	28.3	28.3	601	1.00044	0.00330
70	4-(11X14)	18.92	BOROFLEX	5.46	28.3	36.0	616	0.99672	0.00355
71	2-(11X14), 2-(11X14+9)	18.92	BOROFLEX	5.46	28.3	47.2	634	0.99910	0.00337
72	2-(11X14), $1-(11X16+1)$ $1-(11X16+2)$	18.92	BOROFLEX	5.46	28.3	85.0	663	0.99530	0.00306
73	2-(11X14), 2-(11X13+2)	18.92	BOROFLEX	4.08	28.3	28.3	598	0.99623	0.00332
74	2-(11X14), $1-(11X14+7)$ $1-(11X14+8)$	18.92	BOROFLEX	4.08	28.3	47.2	631	1.00086	0.00388
75	2-(11X14), $1-(11X15+6)$ $1-(11X15+7)$	18.92	BOROFLEX	4.08	28.3	66.1	651	1.00011	0.00339
76	2-(11X14), $1-(11X15+2)$ $1-(11X15+3)$	18.92	BOROFLEX	7.72	28.3	28.3	643	0.99458	0.00312
77	2-(11X14), 2-(11X16)	18.92	BOROFLEX	7.72	28.3	49.4	660	0.99673	0.00408
78	2-(11X14), 2-(11X16+6)	18.92	BOROFLEX	7.72	28.3	66.1	672	0.99210	0.00340

Table 3.C.5.2 (continued)

Serial No.	Fuel Clusters	Pitch (mm)	Plate Material	$t_p$ (mm)	$X_c$ (mm)	$Y_c$ (mm)	Critical Size	Calculated Results	
								$k_{eff}$	$\pm \sigma$
79	2-(11X14), $1\begin{smallmatrix}-(11X15+2)\\-(11X15+3)\end{smallmatrix}$	18.92	BOROFLEX	6.34	28.3	28.3	643	0.99773	0.00298
80	2-(11X14), 2-(11X16+5)	18.92	BOROFLEX	6.34	28.3	66.1	670	0.99250	0.00395
81	2-(11X14), 2-(11X16+8)	18.92	BOROFLEX	6.34	28.3	85.0	676	0.99455	0.00289
82	2-(11X14), $1\begin{smallmatrix}-(11X9)\\-(11X8+10)\end{smallmatrix}$	18.92	CADMUM	0.61	28.3	28.3	505	0.99433	0.00312
83	2-(11X14), 2-(11X13)	18.92	CADMUM	0.61	28.3	53.0	594	0.99363	0.00333
84	4-(11X14)	18.92	CADMUM	0.61	28.3	64.3	616	0.99646	0.00321
85	1-(34X17+17)	16.84	—	—	0.0	0.0	595	0.99819	0.00348
86	2-(17X17), 2-(11X1)	16.84	—	—	25.9	25.9	600	1.00270	0.00263
87	1-(17X17+5), 1-(17X17+4)	16.84	—	—	16.8	0.0	587	0.99014	0.00330
88	2-(17X17), $1\begin{smallmatrix}-(17X3+15)\\-(17X3+14)\end{smallmatrix}$	16.84	—	—	42.7	42.7	709	0.99500	0.00325
89	2-(17X17), 2-(17X15+2)	16.84	—	—	59.5	59.5	1,092	0.99790	0.00284
90	2-(17X17), 2-(17X1)	16.84	—	—	25.9	51.1	612	0.99564	0.00288
91	2-(17X17), 2-(17X2)	16.84	—	—	25.9	66.6	646	0.99565	0.00259
92	2-(17X17), 2-(17X4)	16.84	—	—	25.9	75.3	714	0.99117	0.00299
93	2-(17X17), 2-(17X9)	16.84	—	—	25.9	90.0	884	0.99232	0.00312
94	2-(17X17), 2-(17X12)	16.84	—	—	25.9	99.7	986	0.99346	0.00281
95	2-(17X17), 2-(17X15)	16.84	—	—	25.9	114.5	1,088	0.99438	0.00301
96	4-(17X17)	16.84	—	—	25.9	138.7	1,156	0.99544	0.00336
97	2-(17X17), $1\begin{smallmatrix}-(12X1)\\-(13X1)\end{smallmatrix}$	16.84	ALUMINUM	6.25	25.9	25.9	603	0.99568	0.00328
98	2-(17X17), 2-(10X1)	16.84	ZIRCALOY-4	6.52	25.9	25.9	598	0.99742	0.00294
99	2-(17X17), $1\begin{smallmatrix}-(17X3+13)\\-(17X3+14)\end{smallmatrix}$	16.84	304L STEEL	3.02	25.9	25.9	707	0.99585	0.00289
100	2-(17X17), 2-(17X11+11)	16.84	304L STEEL	3.02	25.9	59.5	974	0.99284	0.00299
101	2-(17X17), 2-(17X15+15)	16.84	304L STEEL	3.02	25.9	76.4	1,118	0.99424	0.00299
102	2-(17X17), 2-(17X17+16)	16.84	304L STEEL	3.02	25.9	93.2	1,188	0.99936	0.00362
103	2-(17X17), $1\begin{smallmatrix}-(17X5+15)\\-(17X5+14)\end{smallmatrix}$	16.84	304L STEEL	4.85	25.9	25.9	777	0.99616	0.00371
104	2-(17X17), 2-(17X9+7)	16.84	304L STEEL	4.85	25.9	42.7	898	0.99407	0.00295
105	2-(17X17), $1\begin{smallmatrix}-(17X19)\\-(17X18+16)\end{smallmatrix}$	16.84	304L STEEL	4.85	25.9	93.2	1,223	0.99682	0.00334
106	2-(17X17), $1\begin{smallmatrix}-(17X5)\\-(17X4+16)\end{smallmatrix}$	16.84	COPPER	3.37	25.9	25.9	747	0.99457	0.00315
107	2-(17X17), $1\begin{smallmatrix}-(17X8+5)\\-(17X8+6)\end{smallmatrix}$	16.84	COPPER	3.37	25.9	42.7	861	0.99045	0.00322
108	3-(17X17), 1-(17X16+16)	16.84	COPPER	3.37	25.9	76.4	1,155	0.99111	0.00361
109	2-(17X17), $1\begin{smallmatrix}-(17X8+8)\\-(17X8+9)\end{smallmatrix}$	16.84	COPPER	6.46	25.9	25.9	867	0.99411	0.00299
110	2-(17X17), 2-(17X16+10)	16.84	COPPER	6.46	25.9	59.5	1,142	0.99436	0.00339

**Table 3.C.6.1** Calculated results and critical experimental data for three aligned UO<sub>2</sub> rod clusters in water with side reflectors

Serial No.	Fuel Cluster	Reflector Material	Pitch (mm)	Enrichment (wt %)	Distance Y (mm)	Distance Xc (mm)	500 neutrons/gen.	
							k <sub>eff</sub>	± σ
1	1·23×18, 2·20×18	Uranium	16.84	2.35	0.0	80.6	0.98039	0.00292
2	1·23×28, 2·20×18	Uranium	16.84	2.35	13.21	95.0	0.98954	0.00294
3	1·23×28, 2·20×18	Uranium	16.84	2.35	26.16	98.3	0.99070	0.00278
4	1·23×28, 2·20×18	Uranium	16.84	2.35	39.12	91.9	0.99454	0.00291
5	1·23×18, 2·20×18	Lead	16.84	2.35	0.0	100.6	0.98930	0.00310
6	1·23×18, 2·20×18	Lead	16.84	2.35	6.60	101.1	0.98928	0.00342
7	1·23×18, 2·20×18	Lead	16.84	2.35	32.76	85.0	0.99376	0.00312
8	1·23×18, 2·20×18	Lead	16.84	2.35	∞	65.9	0.98444	0.00295
9	1·23×18, 2·20×18	Lead	16.84	2.35	∞	72.2	0.98274	0.00320
10	3·12×16	Uranium	18.92	4.31	0.0	153.3	0.98676	0.00292
11	3·12×16	Uranium	18.92	4.31	6.60	167.1	0.98918	0.00338
12	3·12×16	Uranium	18.92	4.31	13.21	182.7	0.99466	0.00348
13	3·12×16	Uranium	18.92	4.31	19.56	192.4	0.98730	0.00350
14	3·12×16	Uranium	18.92	4.31	26.16	193.7	0.98936	0.00318
15	3·12×16	Uranium	18.92	4.31	32.76	187.8	0.99898	0.00323
16	3·12×16	Uranium	18.92	4.31	54.05	164.0	0.99046	0.00367
17	3·12×16	Uranium	18.92	4.31	∞	123.5	0.99252	0.00343
18	3·12×16	Lead	18.92	4.31	0.0	177.4	0.98862	0.00339
19	3·12×16	Lead	18.92	4.31	6.60	181.8	0.99083	0.00338
20	3·12×16	Lead	18.92	4.31	19.56	174.3	0.99372	0.00304
21	3·12×16	Lead	18.92	4.31	54.05	143.5	0.99016	0.00322
22	3·12×16	Lead	18.92	4.31	∞	129.7	0.98013	0.00348

**Table 3.C.7.1** Calculated results and critical experimental data for three UO<sub>2</sub> rod clusters in water with absorbers and reflecting walls

Serial No.	Fuel Clusters	Pitch (mm)	Plate Material	t <sub>p</sub> (mm)	G (mm)	X <sub>c</sub> (mm)	Y (mm)	k <sub>eff</sub>	± σ	
1	3-19X16	20.32	—	—	106.5	0.0	0.98658	0.00286		
2	3-19X16	20.32	—	—	112.0	6.60	0.99139	0.00303		
3	3-19X16	20.32	—	—	112.0	13.21	0.98965	0.00275		
4	3-19X16	20.32	—	—	103.6	26.16	0.98721	0.00300		
5	3-19X16	10.32	—	—	95.1	39.12	0.99097	0.00302		
6	3-19X16	20.32	—	—	83.1	∞	0.98323	0.00296		
7	3-19X16	20.32	—	—	82.3	∞	0.97993	0.00290		
8	1-25X18, 2-20X18	16.84	—	—	89.8	0.0	0.99531	0.00287		
9	1-25X18, 2-20X18	16.84	—	—	95.8	6.60	0.99609	0.00307		
10	1-25X18, 2-20X18	16.84	—	—	95.1	13.21	0.99524	0.00353		
11	1-25X18, 2-20X18	16.84	—	—	96.6	16.84	0.99509	0.00306		
12	1-25X18, 2-20X18	16.84	—	—	92.8	23.44	0.98976	0.00310		
13	1-25X18, 2-20X18	16.84	—	—	90.6	30.05	0.99570	0.00280		
14	1-25X18, 2-20X18	16.84	—	—	85.4	39.12	0.99078	0.00326		
15	1-25X18, 2-20X18	16.84	—	—	76.7	67.26	0.98555	0.00288		
16	1-25X18, 2-20X18	16.84	—	—	72.4	∞	0.98809	0.00314		
17	1-25X18, 2-20X18	16.84	—	—	71.9	∞	0.98374	0.00327		
18	1-25X18, 2-20X18	16.84	—	—	95.1	13.21	0.99524	0.00353		
19	1-25X18, 2-20X18	16.84	304L STEEL	3.02	0.0	82.8	13.21	0.99107	0.00303	
20	1-25X18, 2-20X18	16.84	304L STEEL, 1.1 wt% B	2.98	0.0	48.0	13.21	0.98961	0.00318	
21	1-25X18, 2-20X18	16.84	BORAL B	2.92	0.0	26.9	13.21	0.98916	0.00308	
22	1-25X18, 2-20X18	16.84	BOROFLEX	5.46	0.0	29.8	13.21	0.99575	0.00291	
23	1-25X18, 2-20X18	16.84	CADMIUM	0.61	0.0	38.6	13.21	0.99094	0.00322	
24	1-25X18, 2-20X18	16.84	COPPER	3.37	0.0	77.9	13.21	0.98696	0.00297	
25	1-25X18, 2-20X18	16.84	COPPER-CADMIUM	3.57	0.0	54.3	13.21	0.98634	0.00302	
26	3-13X8	25.40	—	—	128.9	0.0	0.99502	0.00370		
27	3-13X8	25.40	—	—	142.5	6.60	0.99130	0.00356		
28	3-13X8	25.40	—	—	141.2	13.21	0.99103	0.00339		
29	3-13X8	25.40	—	—	124.4	26.16	0.98769	0.00342		
30	3-13X8	25.40	—	—	98.0	54.05	0.98502	0.00316		
31	3-13X8	25.40	—	—	82.3	∞	0.98269	0.00303		
32	3-13X8	25.40	—	—	82.4	∞	0.98870	0.00364		
33	3-12X16	18.92	—	—	148.7	0.0	0.99304	0.00359		
34	3-12X16	18.92	—	—	157.4	6.60	0.99105	0.00276		
35	3-12X16	18.92	—	—	158.7	13.21	0.99709	0.00332		
36	3-12X16	18.92	—	—	158.4	19.56	0.99549	0.00345		
37	3-12X16	18.92	—	—	154.5	26.16	1.00108	0.00321		
38	3-12X16	18.92	—	—	138.2	54.05	0.99802	0.00350		
39	3-12X16	18.92	—	—	129.6	∞	0.99537	0.00342		
40	3-12X16	18.92	—	—	129.8	∞	0.99088	0.00349		
41	3-12X16	18.92	—	—	127.5	∞	0.99050	0.00381		
42	3-12X16	18.92	—	—	158.4	19.56	0.99549	0.00345		
43	3-12X16	18.92	304L STEEL	3.02	0.0	137.5	19.56	0.99255	0.00317	
44	3-12X16	18.92	304L STEEL, 1.1 wt% B	2.98	0.0	98.3	19.56	0.98899	0.00353	
45	3-12X16	18.92	BORAL B	2.92	0.0	83.0	19.56	0.99635	0.00302	
46	3-12X16	18.92	BOROFLEX	5.46	0.0	83.7	19.56	0.99275	0.00328	
47	3-12X16	18.92	CADMIUM	0.61	0.0	89.4	19.56	0.99705	0.00332	
48	3-12X16	18.92	COPPER	3.37	0.0	134.7	19.56	0.98733	0.00367	
49	3-12X16	18.92	COPPER-CADMIUM	3.57	0.0	105.7	19.56	0.99261	0.00362	

**Table 3.C.8.1** Summary of characteristics of heterogeneous plutonium benchmark experiments

No.	H/Pu atom ratio	% $^{240}\text{Pu}$	Core radius, cm	Fuel type
16	632	6	23.3	Af-1.8 wt% Pu
17	1003	6	25.0	Af-1.8 wt% Pu
18	1333	6	29.4	Af-1.8 wt% Pu
19	583	16	29.3	Af-2 wt% Pu
20	926	16	32.6	Af-2 wt% Pu
21	1230	16	41.4	Af-2 wt% Pu
22	355	5	16.8	Af-5 wt% Pu
23	666	5	18.1	Af-5 wt% Pu
24	1148	5	25.4	Af-5 wt% Pu
25*	666	5	20.2	Af-5 wt% Pu
26*	666	5	23.8	Af-5 wt% Pu
27	238	8	19.1	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
28	554	8	17.3	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
29	1113	8	24.3	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
30	185	8	27.7	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
31*	185	8	27.7	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
32	391	8	17.4	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
33*	391	8	34.4	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
34	564	8	17.9	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
35*	564	8	37.2	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
36	238	24	24.3	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
37	554	24	21.4	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
38	1113	24	35.3	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>
39	153	18	18.0	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>
40	289	18	16.5	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>
41	922	18	37.4	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>

\* With boron in the moderator

**Table 3.C.8.2** Calculated results for heterogeneous plutonium benchmark experiments.

No.	Fuel type	H / Pu atom ratio (boron PPM)	% $^{240}\text{Pu}$	Core radius, cm	Calculated $k_{\text{eff}}$				
16	Af-1.8 wt% Pu	632	6	23.3	0.9993 ± 0.0040				
17	Af-1.8 wt% Pu	1003	"	25.0	0.9852 ± 0.0041				
18	Af-1.8 wt% Pu	1333	"	29.4	1.0047 ± 0.0039				
19	Af-2 wt% Pu	1333	"	29.3	1.0102 ± 0.0044				
20	Af-2 wt% Pu	583	16	32.6	1.0110 ± 0.0034				
21	Af-2 wt% Pu	926	"	41.4	1.0228 ± 0.0035				
22	Af-5 wt% Pu	926	"	5	16.8	0.9976 ± 0.0052			
23	Af-5 wt% Pu	1230	"	6	18.1	1.0001 ± 0.0043			
24	Af-5 wt% Pu	1230	"	1148	"	25.4	1.0039 ± 0.0048		
25*	Af-5 wt% Pu	355	5	22	Af-5 wt% Pu	666	"	20.2	1.0032 ± 0.0048
26*	Af-5 wt% Pu	355	5	23	Af-5 wt% Pu	1148	"	23.8	0.9964 ± 0.0046
27	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	238	8	24	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	666(100)	"	19.1	0.9954 ± 0.0045
28	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	554	8	25	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	666(285)	"	17.3	0.9913 ± 0.0041
29	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	1113	8	26	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	554	"	24.8	1.0256 ± 0.0038
30	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	185	8	27	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	1113	"	24.3	0.9936 ± 0.0042
31*	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	185	8	28	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	554	"	21.4	1.0035 ± 0.0042
32	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	391	8	29	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	1113	"	35.3	1.0064 ± 0.0036
33*	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	391	8	30	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	238	24	18.0	1.0104 ± 0.0046
34	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	564	8	31	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	554	"	16.5	1.0058 ± 0.0049
35*	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	564	8	32	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	1113	"	37.4	1.0115 ± 0.0036
36	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	238	24	33	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	554	"	Average	1.0039
37	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	554	24	34	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	1113	"		
38	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	1113	24	35	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	238	24		
39	UO <sub>2</sub> -2 wt% PuO <sub>2</sub>	153	18	36	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>	554	"		
40	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>	289	18	37	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>	1113	"		
41	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>	922	18	38	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>	153	18		
				39	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>	289	"		
				40	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>	522	"		
				41	UO <sub>2</sub> -4 wt% PuO <sub>2</sub>				

Table 3.C.9.1 Principal critical parameters and calculated results

No.	Fuel type	Pitch (mm)	Critical number of rods	Boron concentration (wppm)	Calculated $k_{eff}$
1	$\text{UO}_2 - 2.35\% \text{ }^{235}\text{U}$	1.5621	708	0.9	$0.9910 \pm 0.0043$
2	,	1.5621	1201	463.8	$1.0039 \pm 0.0037$
3	,	1.905	383	0.5	$0.9911 \pm 0.0043$
4	,	1.905	1201	568.0	$0.9831 \pm 0.0035$
5	,	2.2098	342	0.9	$0.9921 \pm 0.0041$
6	,	2.2098	885	285.8	$0.9963 \pm 0.0033$
7	$\text{UO}_2 - 2\text{ w/o PuO}_2$ (8% $^{240}\text{Pu}$ )	1.778	469	1.7	$1.0051 \pm 0.0047$
8	,	1.778	761	680.9	$1.0066 \pm 0.0042$
9	,	2.2098	195	0.9	$0.9947 \pm 0.0048$
10	,	2.2098	761	1090.4	$1.0222 \pm 0.0033$
11	,	2.5146	160	1.6	$0.9999 \pm 0.0039$
12	,	2.5146	689	767.2	$1.0147 \pm 0.0034$
Average					1.000

Table 3.C.10.1 Measured and calculated  $k_{eff}$  values

Core No.	Measured values	Calculated values	C/E
1	$1.0002 \pm 0.0005$	$0.9898 \pm 0.0039$	0.9896
2	$1.0001 \pm 0.0005$	$1.0148 \pm 0.0036$	1.0147
3	$1.0000 \pm 0.0006$	$1.0158 \pm 0.0035$	1.0158
4	$0.9999 \pm 0.0006$	$0.9979 \pm 0.0040$	0.9980
5	$1.0000 \pm 0.0007$	$0.9958 \pm 0.0041$	0.9958
6	$1.0097 \pm 0.0012$	$1.0056 \pm 0.0038$	0.9959
7	$0.9998 \pm 0.0009$	$0.9935 \pm 0.0040$	0.9937
8	$1.0083 \pm 0.0012$	$1.0046 \pm 0.0038$	0.9963
9	$1.0030 \pm 0.0009$	$1.0024 \pm 0.0043$	0.9994
10	$1.0001 \pm 0.0009$	$1.0100 \pm 0.0039$	1.0099
11	$1.0000 \pm 0.0006$	$1.0120 \pm 0.0037$	1.0120
12	$1.0000 \pm 0.0007$	$0.9935 \pm 0.0042$	0.9915
13	$1.0000 \pm 0.0010$	$0.9962 \pm 0.0045$	0.9962
14	$1.0001 \pm 0.0010$	$0.9984 \pm 0.0038$	0.9983
15	$0.9998 \pm 0.0016$	$1.0056 \pm 0.0042$	1.0058
16	$1.0001 \pm 0.0019$	$0.9951 \pm 0.0037$	0.9950
17	$1.0000 \pm 0.0010$	$1.0034 \pm 0.0035$	1.0034
18	$1.0002 \pm 0.0011$	$1.0064 \pm 0.0035$	1.0062
19	$1.0002 \pm 0.0010$	$1.0075 \pm 0.0037$	1.0073
20	$1.0003 \pm 0.0011$	$1.0089 \pm 0.0034$	1.0086
21	$0.9997 \pm 0.0015$	$1.0065 \pm 0.0034$	1.0068
Average		1.001	1.003

**Table 3.C.10.2** Comparison of the group averaged  $k_{eff}$ 's for different spacings between assemblies

Spacing between assemblies (cm)	Core No.	Moderator boron Concentration (ppm)	Measured $k_{eff}$	Calculated $k_{eff}$
None	1	0	1.0002 ± 0.0005	0.9898 ± 0.0039
	2	1037	1.0001 ± 0.0005	1.0148 ± 0.0036
	Average $k_{eff}$		1.0002	1.0023
1.636 (1 pitch)	3	764	1.0000 ± 0.0006	1.0158 ± 0.0035
	4	0	0.9999 ± 0.0006	0.9979 ± 0.0040
	11	514	1.0000 ± 0.0006	1.0120 ± 0.0037
	13	15	1.0000 ± 0.0010	0.9962 ± 0.0045
	14	92	1.0001 ± 0.0010	0.9984 ± 0.0038
	15	395	0.9998 ± 0.0016	1.0056 ± 0.0042
	17	487	1.0000 ± 0.0010	1.0034 ± 0.0035
	19	634	1.0002 ± 0.0010	1.0075 ± 0.0037
	Average $k_{eff}$		1.0000	1.0046
3.272 (2 pitches)	5	0	1.0000 ± 0.0007	0.9958 ± 0.0041
	6	0	1.0097 ± 0.0012	1.0056 ± 0.0038
	12	217	1.0000 ± 0.0007	0.9935 ± 0.0042
	16	121	1.0001 ± 0.0019	0.9951 ± 0.0037
	18	197	1.0002 ± 0.0011	1.0064 ± 0.0035
	20	320	1.0003 ± 0.0011	1.0089 ± 0.0034
	Average $k_{eff}$		1.0017	1.0009
4.908 (3 pitches)	7	0	0.9998 ± 0.0009	0.9935 ± 0.0040
	8	0	1.0083 ± 0.0012	1.0046 ± 0.0038
	10	143	1.0001 ± 0.0009	1.0100 ± 0.0039
	21	72	0.9997 ± 0.0015	1.0065 ± 0.0034
	Average $k_{eff}$		1.0020	1.0037
6.544 (4 pitches)	9	0	1.0030 ± 0.0009	1.0024 ± 0.0043

**Table 3.C.11.1** Principal critical parameters and calculated  $k_{eff}$ 

No.	Number of pins	Center separation (cm)	Critical water height above fuel (cm)	Calculated $k_{eff}$
NUSW001	380	1.054	15.2	—
NUSW002	222	1.242	15.2	0.9556 ± 0.0045
NUSW003	138	1.488	30.8	0.9514 ± 0.0047
NUSW004	102	1.742	- 21.3	0.9526 ± 0.0054
NUSW005	85	1.999	15.2	0.9593 ± 0.0053
NUSW006	78	2.276	15.2	0.9637 ± 0.0050
NUSW007	77	2.253	- 3.9	0.9718 ± 0.0045
NUSW008	75	2.507	15.2	0.9682 ± 0.0050
NUSW009	77	2.779	- 43.2	0.9799 ± 0.0044
NUSW010	83	2.995	- 34.1	0.9859 ± 0.0040
NUSW011	96	2.497	- 10.4	0.9615 ± 0.0047
NUSW012	75	1.250	- 12.2	—
			Average	0.9650

**Table 3.C.11.2** Principal critical parameters and calculated  $k_{eff}$ 's

No.	Number of pins	Solution characteristic		Calculated $k_{eff}$
		B (%)	$^{235}U$ (%)	
NUSS001	99	—	—	$0.9760 \pm 0.0046$
NUSS002	114	0.039	—	$0.9674 \pm 0.0050$
NUSS003	113	0.039	—	$0.9807 \pm 0.0049$
NUSS004	133	0.190	—	$0.9774 \pm 0.0048$
NUSS005	83	—	3.68	$0.9776 \pm 0.0039$
NUSS006	133	0.315	3.68	$0.9794 \pm 0.0042$
		Average		0.9764

**Table 3.C.12.1** Critical experimental data for  $UO_2$  rod clusters fully immersed in borated water

No.	Square Pitch (cm)	Boron (a) Concentration g/l	Critical Array Width No. of Rods	Critical (b)(c) Array Length No. of Rods	Total Rods (d) for Criticality
1	1.890	0.0	40	8.92	357
2	1.890	$0.49 \pm 0.06$	40	10.72	429
3	1.890	$1.25 \pm 0.21$	40	14.05	562
4	1.890	$2.15 \pm 0.35$	40	23.07	923
5	1.890	$2.55 \pm 0.07$	40	30.92	1237
6	1.715	0.0	44	11.57	509
7	1.715	$1.03 \pm 0.05$	44	16.75	737
8	1.715	$1.82 \pm 0.30$	44	20.84	917
9	1.715	$2.55 \pm 0.21$	44	27.09	1192

(a) Boron added in the form of  $H_3BO_3$ .

(b) This has the same interpretation as all experiments performed before in this program. A fraction of a row is a row which extends full width but is thinner in length. This is the best interpretation for computer modeling.

(c) The cell boundary of the first fuel row is next to the plexiglas wall of the box.

(d) The error in critical rod number is &lt;0.5%.

**Table 3.C.12.2** Calculated  $k_{eff}$ 's with different Monte Carlo calculational conditions

ケ-ス NO.	500n/b		1000n/b	
	$k_{eff}$	$\pm\sigma$	$k_{eff}$	$\pm\sigma$
1-01	0.99158	0.00323	0.99361	0.00202
1-02	0.95701	0.00331	0.96186	0.00247
1-03	0.94702	0.00301	0.94811	0.00212
1-04	0.97890	0.00280	0.97504	0.00215
1-05	0.98931	0.00287	0.98485	0.00197
1-06	0.99780	0.00371	0.99799	0.00248
1-07	0.98254	0.00324	0.98309	0.00223
1-08	0.98874	0.00298	0.97800	0.00211
1-09	0.98341	0.00275	0.98428	0.00192

**Table 3.C.13.1** Calculated results for UO<sub>2</sub> fuel rods in NaNO<sub>3</sub> solution

No	Lattice			Solution Data				Calculation	
	Square Pitch (mm)	Configuration	Total Number of Rods	Density (g/cm <sup>3</sup> )	Concentration NaNO <sub>3</sub> (g/cm <sup>3</sup> )	Normality NO <sub>3</sub> ± 2%	Critical Height (cm ± 0.1)	k <sub>eff</sub>	σ
1	12.6	24 × 25	600	1.1521	0.247	2.91	80.9	0.99293	0.00352
2	16	18 × 18	324	1.1503	0.243	2.86	76.1	0.99475	0.00327
3		17 × 17	289	1.1523	0.246	2.89	76.2	0.95977	0.00349
4	21.0	16 × 16	256	1.1094	0.178	2.10	89.5	0.98743	0.00348
5		15 × 16	240	1.0565	0.090	1.06	84.8	0.99616	0.00324
6	25.2	20 × 21	420	1.1523	0.246	2.89	85.9	0.98815	0.00308

**Table 3.C.14.1** Calculated results with experimental conditions for each case of triangular lattices in poisonous water

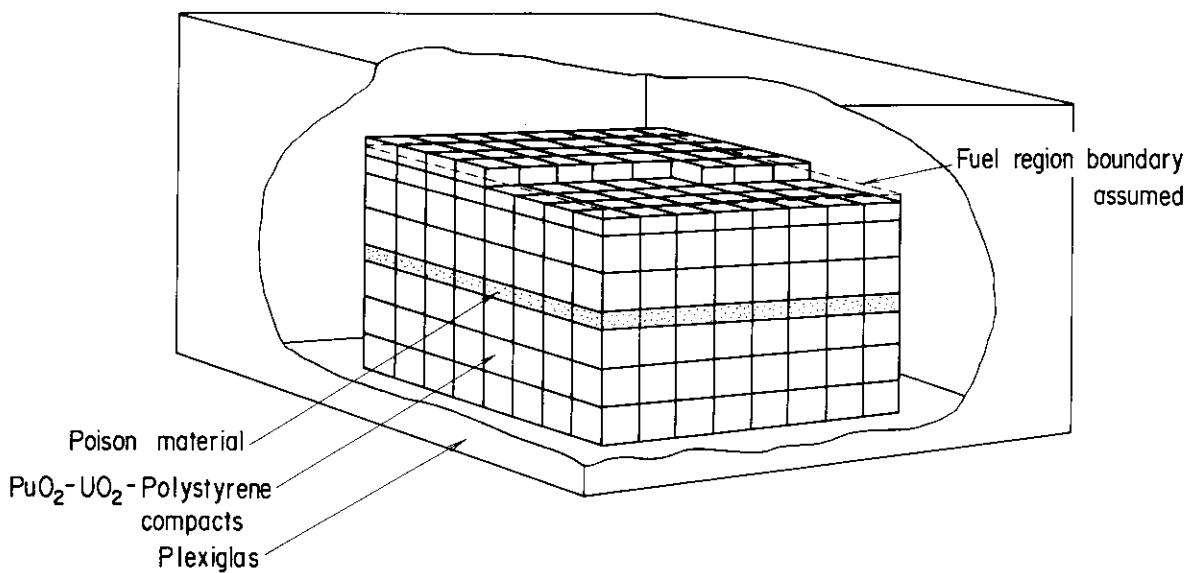
Case No.	Triangular Lattice Pitch (mm)	Number of Fuel Rods Required for Criticality	Neutron Absorber Concentration (g/l)	Total NO <sub>3</sub> (g/l)	Excess Acid Molarity M	Calculated Results	
						k <sub>eff</sub>	σ
1	22.86	223.3	0	---	---	0.95990	0.00339
2	22.86	289.8	0.2307 B	---	---	0.95837	0.00308
3	22.86	363.0	0.4514 B	---	---	0.97193	0.00301
4	22.86	433.2	0.6053 B	---	---	0.97151	0.00309
5	22.86	282.6	0.4293 Cd	0.722	0.004	0.95580	0.00364
6	22.86	391.7	1.060 Cd	1.418	0.004	0.97888	0.00319
7	22.86	283.4	0.0722 Gd	0.230	0.0023	0.95653	0.00348
8	22.86	349.5	0.145 Gd	0.398	0.0036	0.96767	0.00305
9	22.86	422.4	0.213 Gd	0.527	0.0044	0.97949	0.00320
10	27.94	166.6	0	---	---	0.97951	0.00353
11	27.94	225.6	0.158 B	---	---	0.98852	0.00301
12	27.94	349.9	0.380 B	---	---	1.00229	0.00262
13	27.94	232.3	0.0547 Gd	0.1927	0.0021	0.99138	0.00314
14	27.94	341.8	0.1169 Gd	0.2707	0.0021	1.00067	0.00289
15	33.02	187.4	0	---	---	1.01526	0.00308
16	33.02	234.0	0.0643 B	---	---	1.00967	0.00309
17	33.02	434.0	0.2154 B	---	---	1.03261	0.00288
18	33.02	327.5	0.1507 B	---	---	1.01870	0.00318
19	33.02	252.7	0.0257 Gd	0.1571	0.00204	1.01943	0.00290
20	33.02	316.8	0.0440 Gd	0.1805	0.00207	1.02042	0.00303

**Table 3.C.15.1** Calculated results with experimental conditions for  $\text{UO}_2$  rod clusters partially immersed in uranyl-nitrate solution

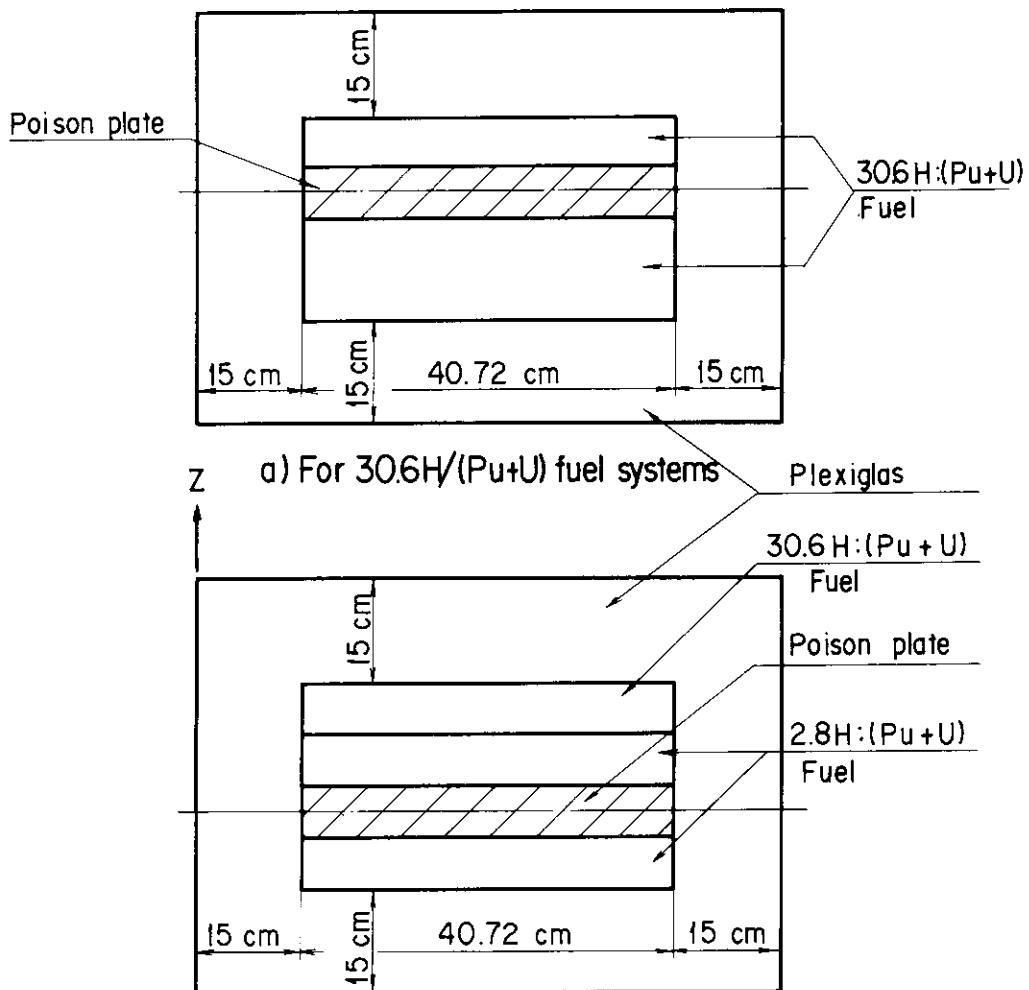
Case No.	Number of Fuel Rods	Triangular Lattice Pitch (mm)	Gadolinium Concentration (g/l)	Critical Solution Height (mm)	Calculated Results	
					$k_{\text{eff}}$	$\sigma$
1	451	22.86	0	373	0.8962	0.0038
			0.124	552	0.9191	0.0040
			0.148	612	0.9197	0.0039
			0.184	744	0.9323	0.0042
5	443	27.94	0	309	0.9084	0.0040
			0.074	413	0.9314	0.0038
			0.099	481	0.9467	0.0040
			0.124	565	0.9622	0.0040
9	421	33.02	0	333	0.9459	0.0041
			0.025	388	0.9542	0.0042
			0.049	460	0.9629	0.0039
			0.074	577	0.9739	0.0038

**Table 3.C.16.1** Calculated results with experimental conditions for mixed-oxide fuel rods in plutonium-uranyl-nitrate solution containing gadolinium and boron

Case No.	Critical Height (cm)	Pu Conc. (g/l)	U Conc. (g/l)	Acid Molarity	Total $\text{NO}_3$ (g/l)	Specific Gravity	$\text{Gd}$ (g/l)	$\text{B}$ (g/l)	Calculated Result	
									$k_{\text{eff}}$	$\sigma$
1	19.205	77.63	180.0	3.4	384.0	1.463	0.02	0	1.0202	0.0050
2	23.066	77.63	180.0	3.4	384.0	1.463	0.258	0	1.0126	0.0041
3	28.227	77.63	180.0	3.4	384.0	1.463	0.515	0	1.0088	0.0040
4	45.753	77.63	180.0	3.4	384.0	1.463	1.040	0	1.0047	0.0039
5	64.506	77.63	180.0	3.4	384.0	1.463	1.280	0	1.0093	0.0033
6	68.862	77.63	180.0	3.4	384.0	1.463	1.338	0	1.0042	0.0036
7	21.20	84.5	182.1	1.75	296	1.438	0.13	0.27	1.0192	0.0044
8	25.37	84.0	183.2	1.88	302	1.442	0.235	0.6	1.0098	0.0039
9	30.49	82.7	180.6	2.01	308	1.444	0.309	0.9	1.0148	0.0041
10	43.86	81.2	180.0	2.18	321	1.447	0.424	1.35	1.0015	0.0034
11	51.97	81.0	180.4	2.09	316	1.451	0.519	1.5	1.0088	0.0030
12	55.18	81.0	180.3	2.21	318	1.452	0.537	1.548	1.0078	0.0030
13	65.42	80.5	180.5	2.24	321	1.454	0.541	1.662	1.0091	0.0031



**Fig. 3.A.1.1** Configuration of PuO<sub>2</sub>-UO<sub>2</sub> polystyrene compacts.



**Fig. 3.A.1.2** Calculational model for PuO<sub>2</sub>-UO<sub>2</sub> polystyrene compacts.

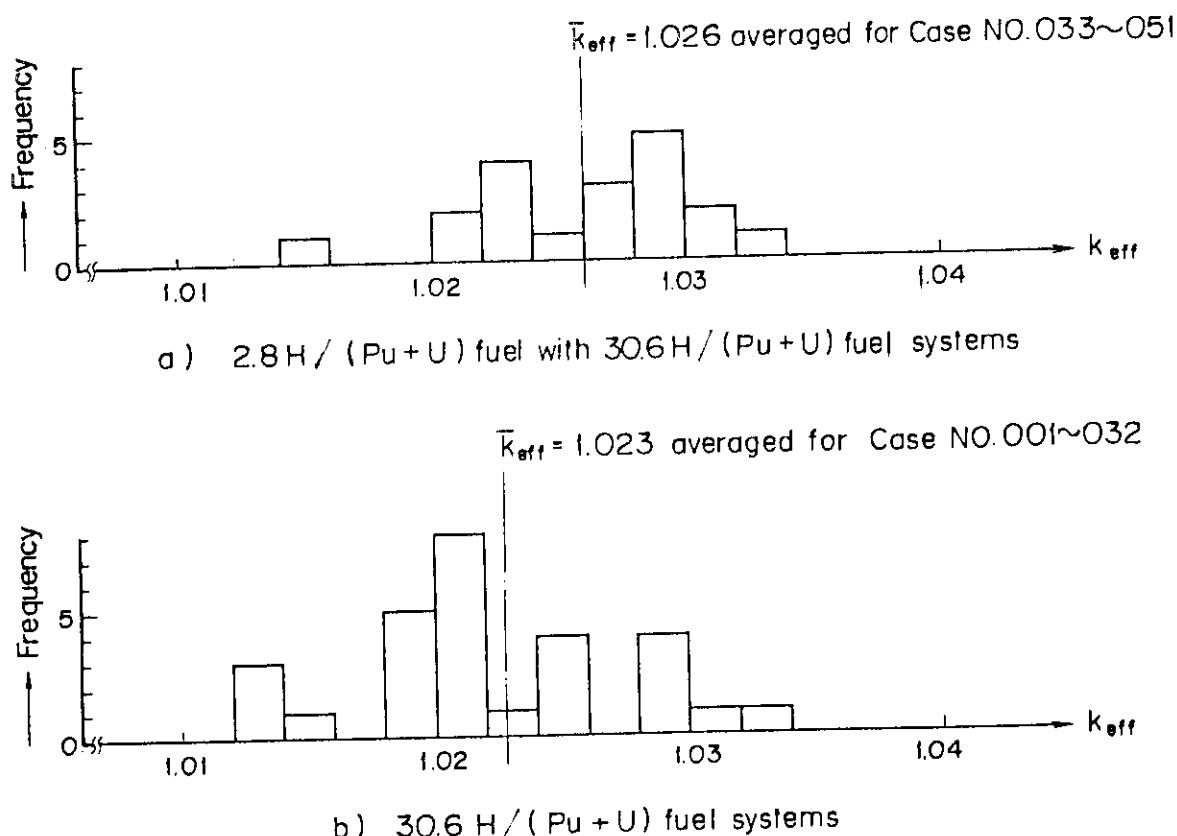


Fig. 3.A.1.3 Histogram of calculated  $k_{eff}$ 's for the two different fuel systems.

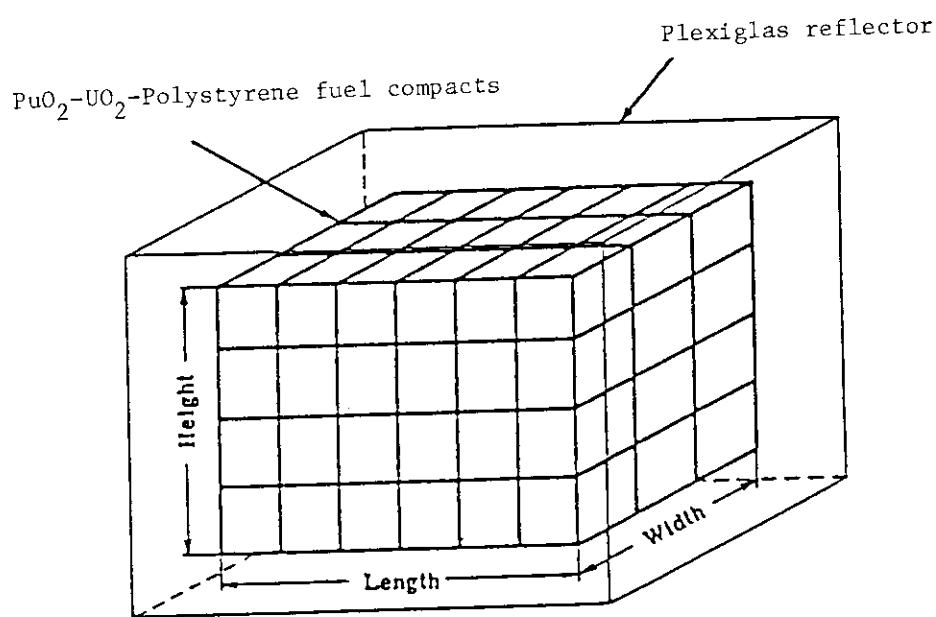


Fig. 3.A.2.1 Calculational model for  $\text{PuO}_2\text{-UO}_2$  polystyrene fuel compacts.

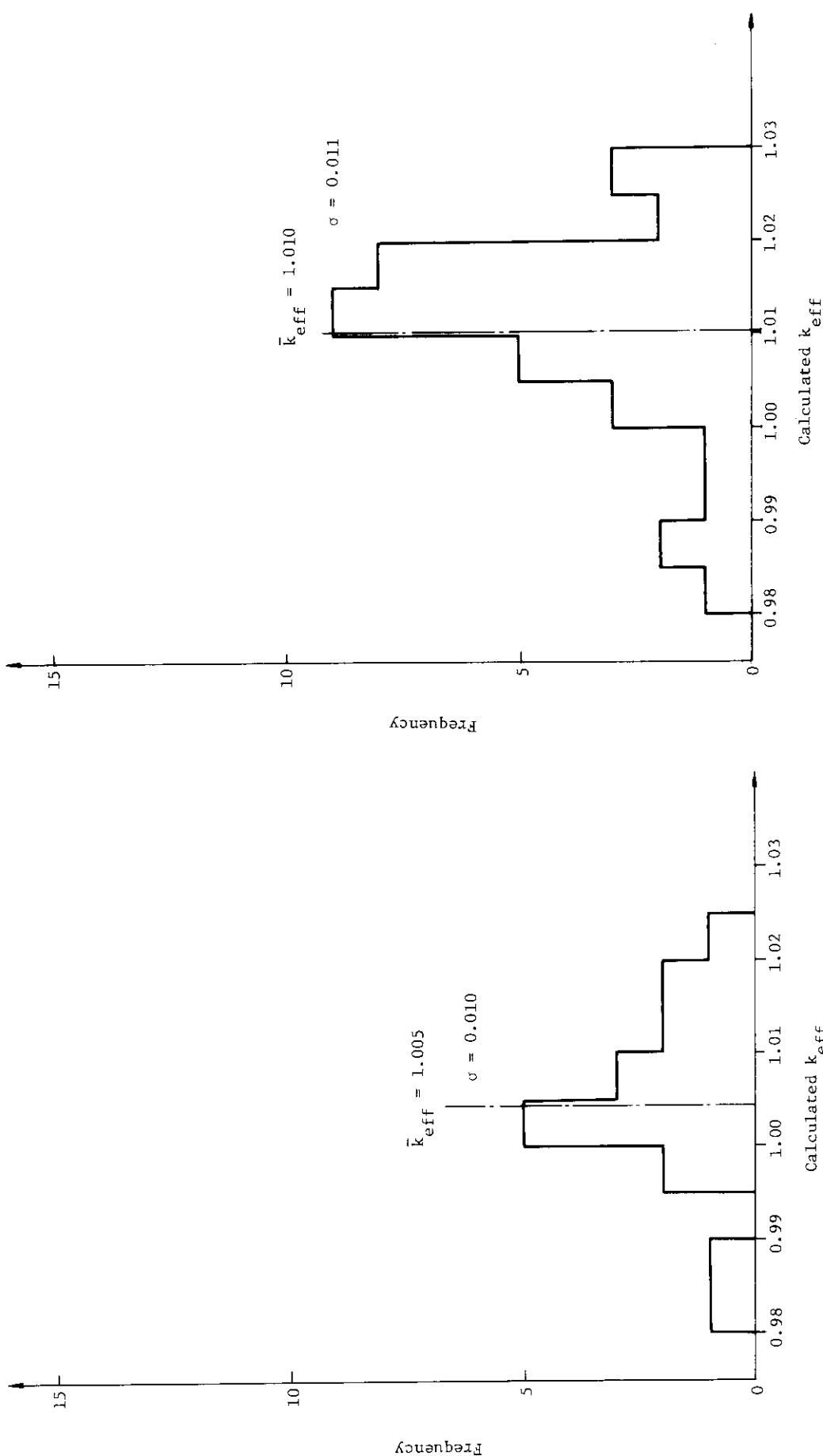


Fig. 3.A.2.2 Histogram of calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub> polystyrene fuel compacts of low (2-8)H/(Pu+U) atomic ratio (17 cases).

Fig. 3.A.2.3 Histogram of calculated  $k_{eff}$ 's for PuO<sub>2</sub>-UO<sub>2</sub> polystyrene fuel compacts of high (30-50)H/(Pu+U) atomic ratio (35 cases).

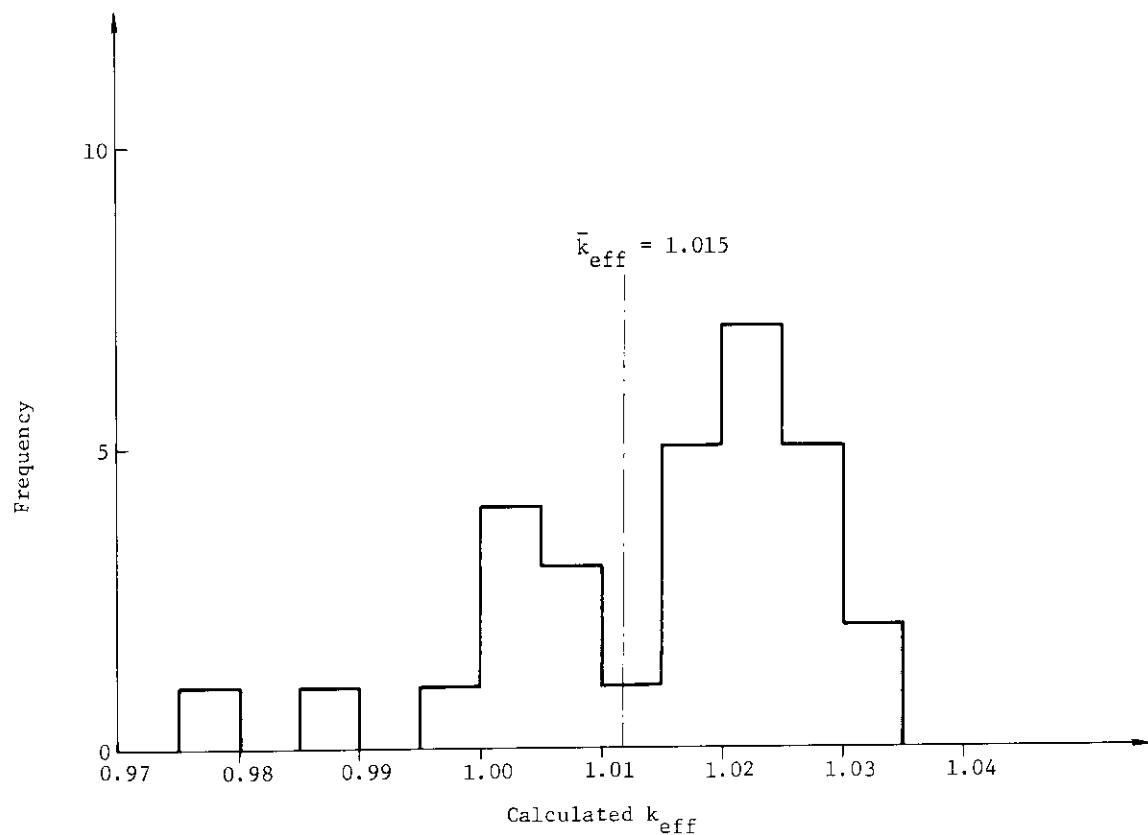


Fig. 3.A.4.1 Histogram of calculated  $k_{\text{eff}}$ 's for  $\text{PuO}_2$  polystyrene fuel compacts.

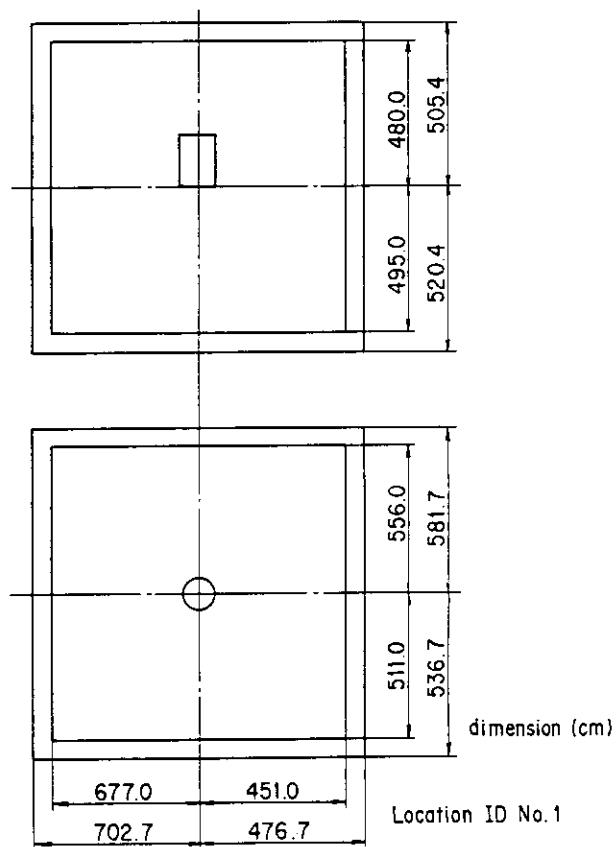


Fig. 3.A.5.1 Calculational model of a reflected single tank.

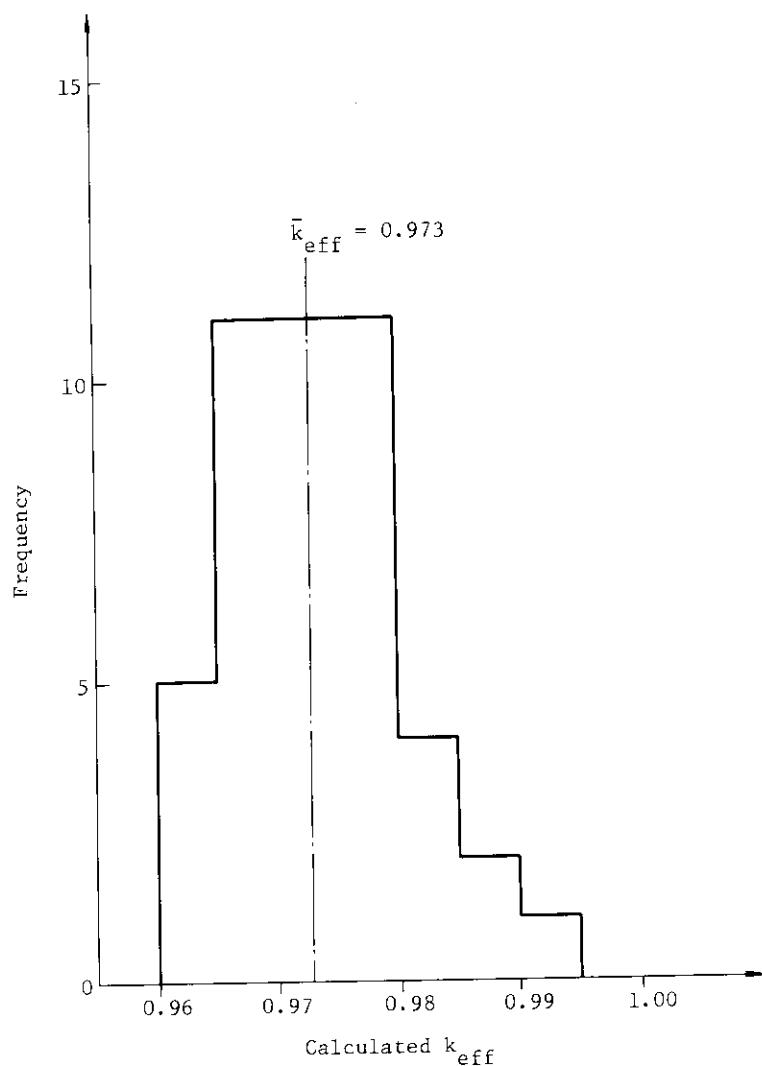


Fig. 3.A.5.2 Histogram of calculated  $k_{eff}$ 's for a single tank (45 cases).

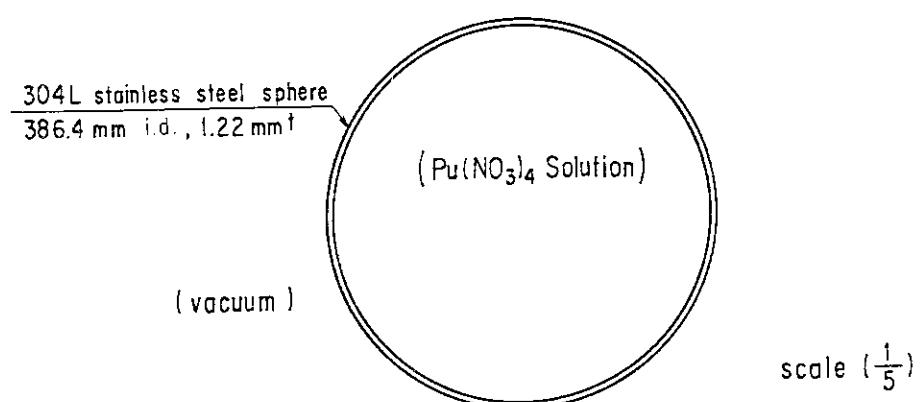
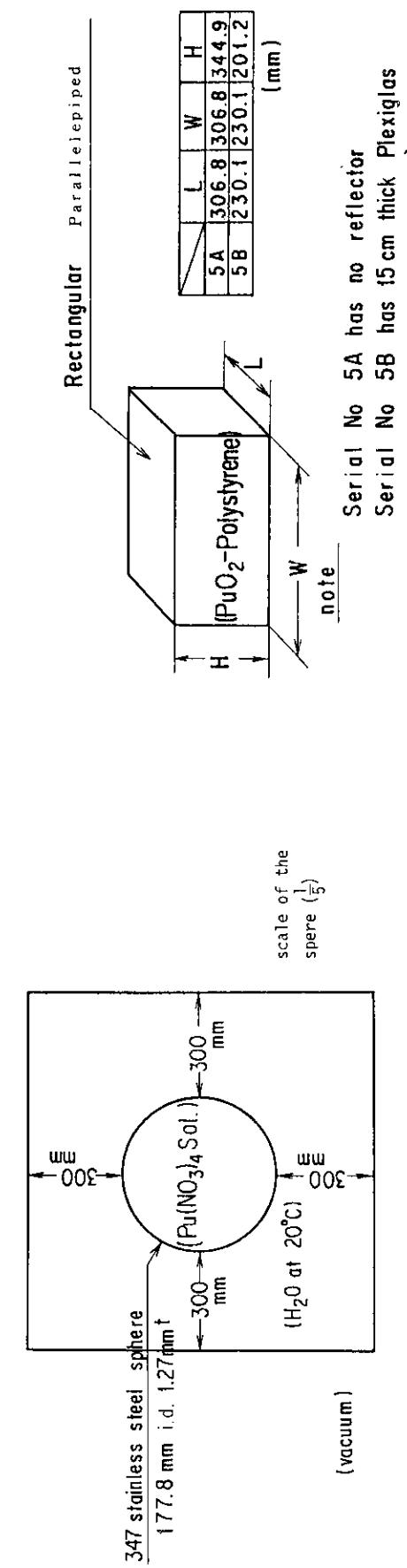
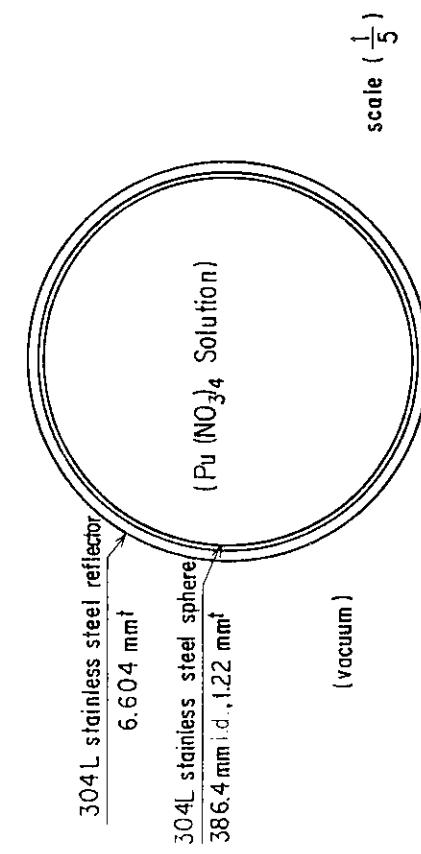


Fig. 3.A.6.1 Calculational model of B.M. No.2.



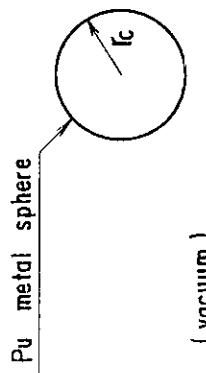
The sphere is at the center of water cubic 300 mm apart from the cubical surface

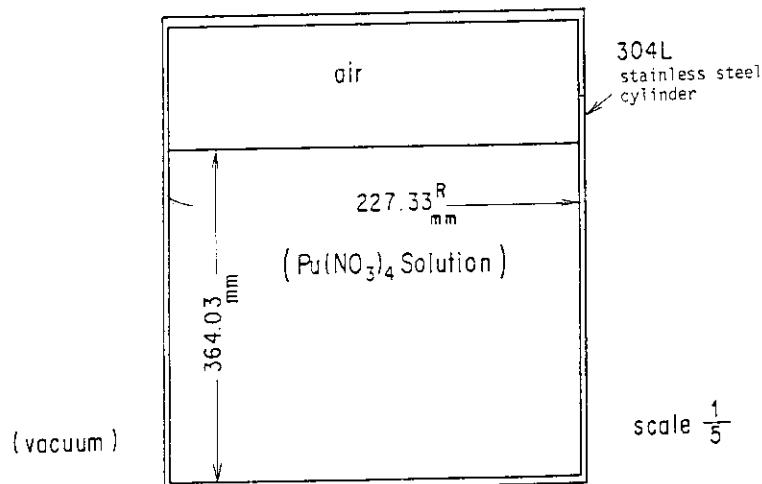
**Fig. 3.A.6.4** Calculational model of B.M. No.5.



**Fig. 3.A.6.5** Calculational model of B.M. No.6 and No.13.

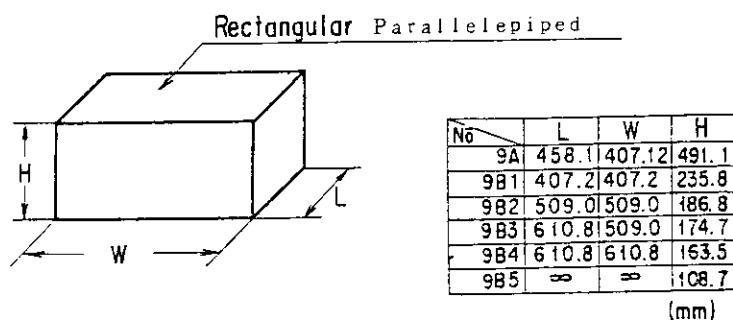
$r_c$  is 63.85 mm for B.M. No 6,  
and 66.6 mm for B.M. No 13,





This calculation was performed without the cylinder vessel due to lack of dimensional data

**Fig. 3.A.6.6** Calculational model of B.M. No.8.



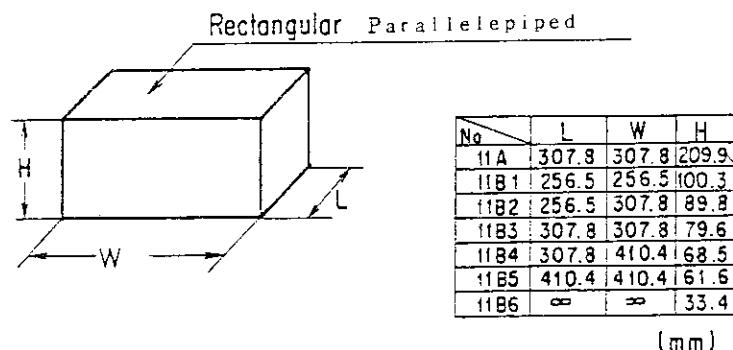
note

Serial No 9A has no reflector

Serial No 9B1 ~ B5 have 15 cm thick

Plexiglas reflector

**Fig. 3.A.6.7** Calculational model of B.M. No.9.



note

Serial No 11A has no reflector.

Serial No 11B1 ~ B6 have 15 cm thick

Plexiglas reflector.

**Fig. 3.A.6.8** Calculational model of B.M. No.11.

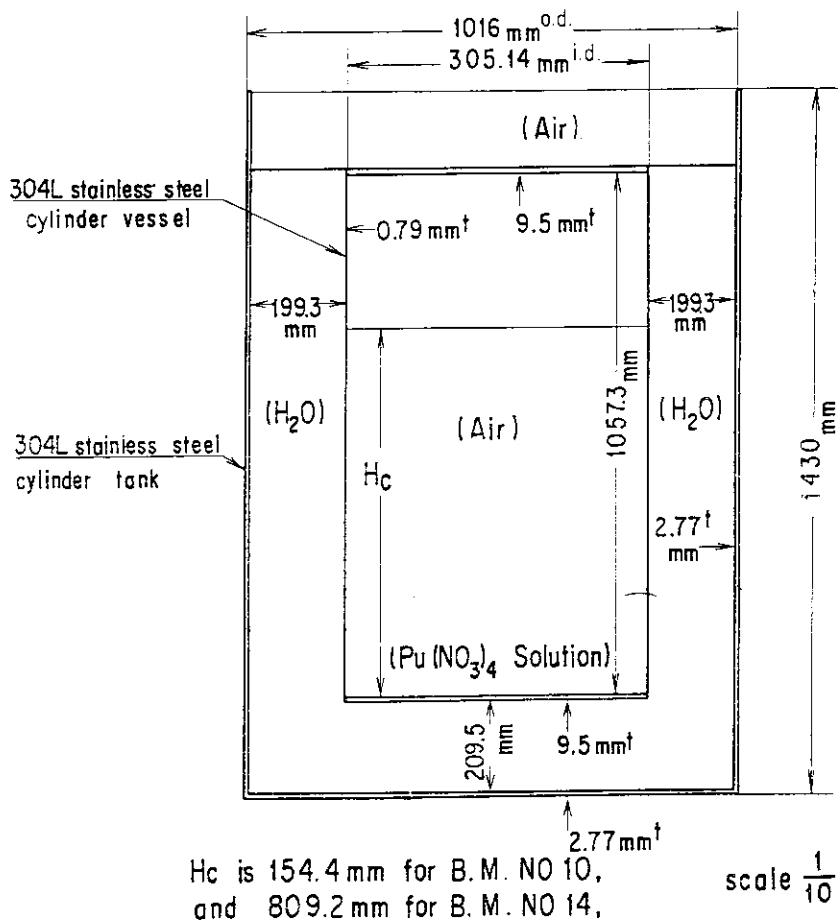
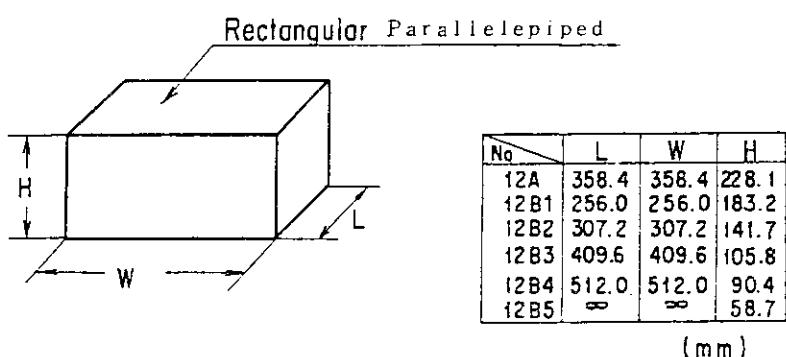


Fig. 3.A.6.9 Calculational model of B.M. No.10 and No.14.



Serial No 12A has no reflector.

Serial No 12B1 ~ B5 have 15 cm thick  
Plexiglas reflector

Fig. 3.A.6.10 Calculational model of B.M. No.12.

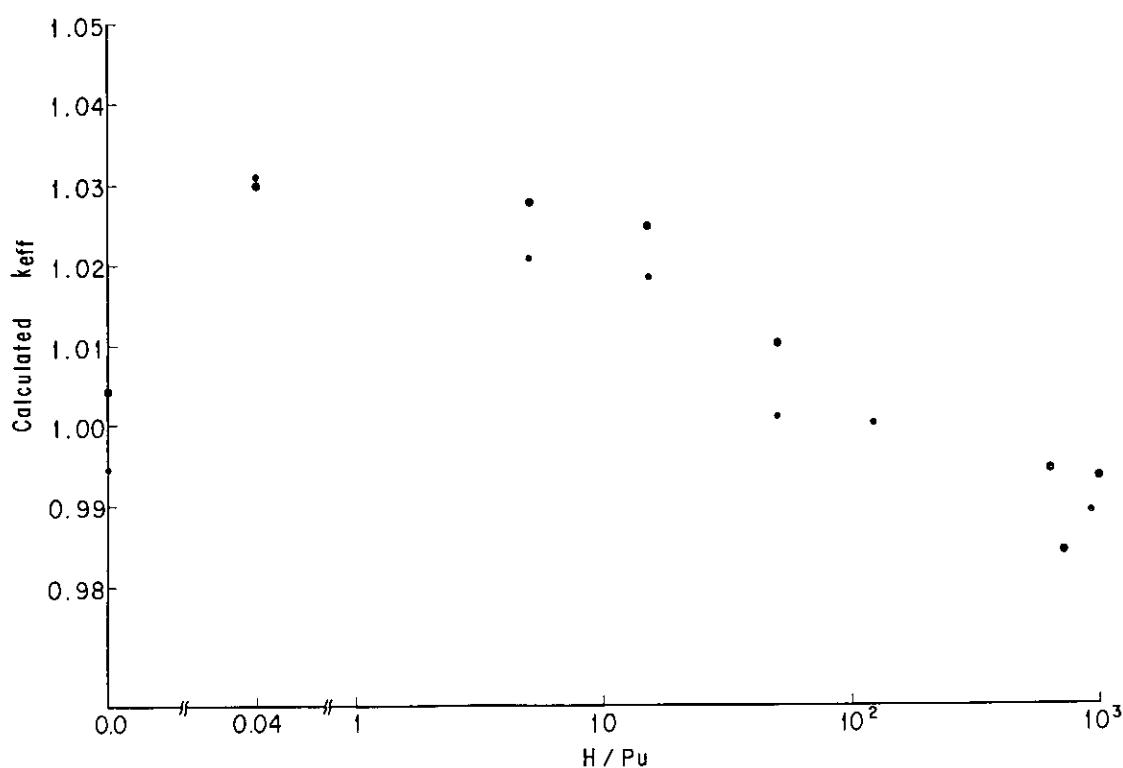


Fig. 3.A.6.11 Decreasing trend of calculated  $k_{eff}$  with increasing  $H/Pu$ .

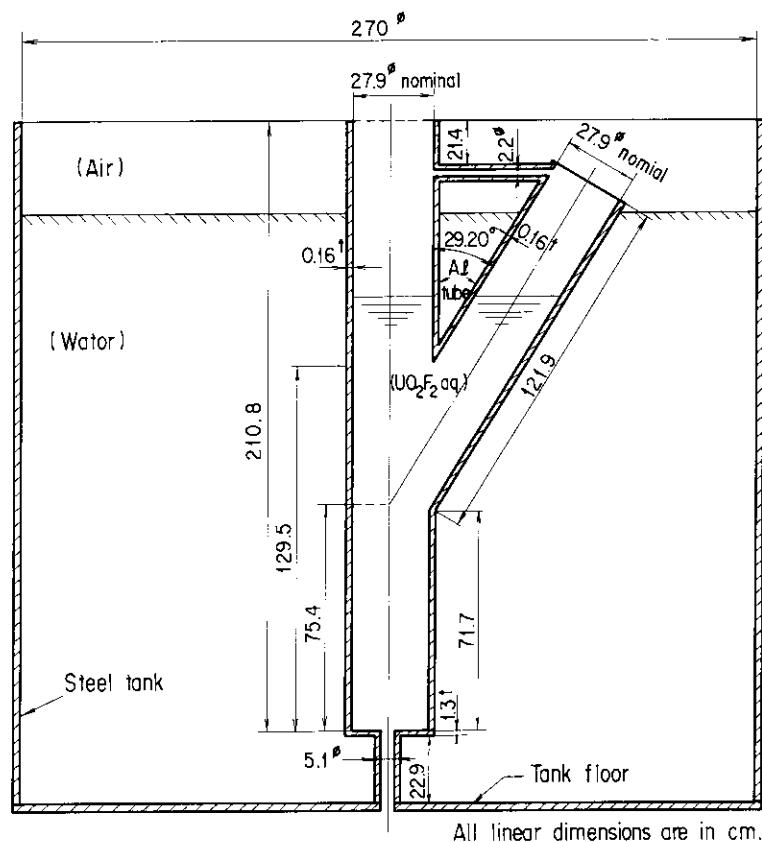
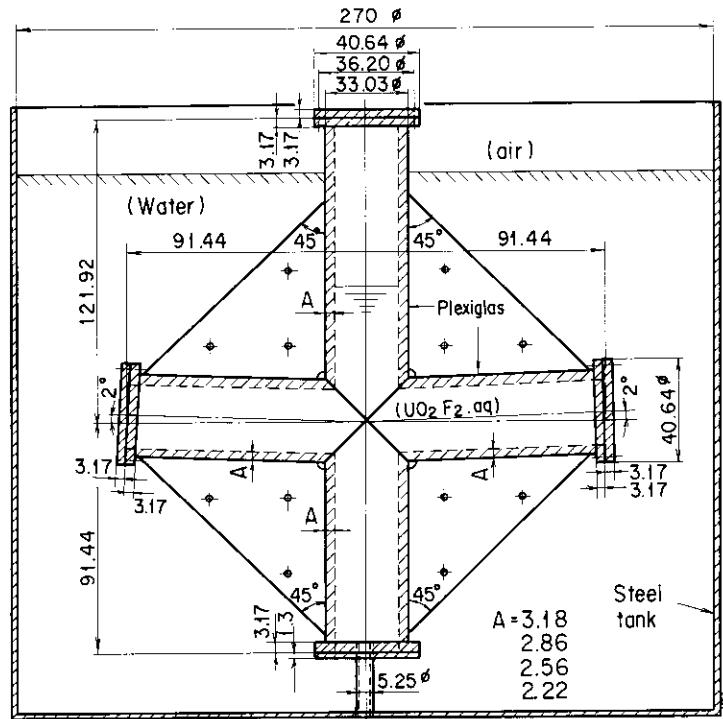
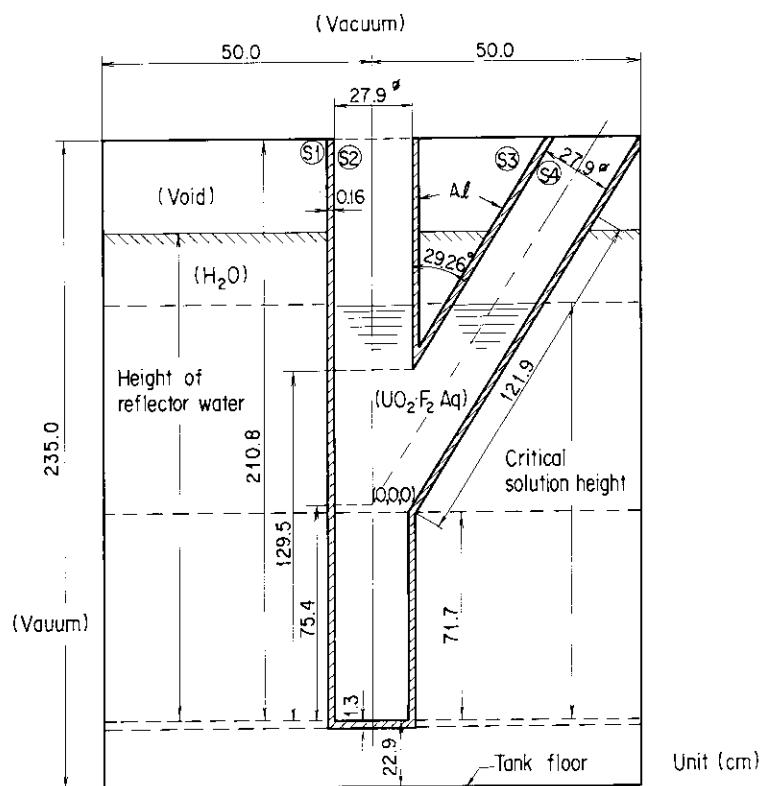


Fig. 3.A.7.1 Cut-out view of the experimental arrangement of "30 degrees lateral".



All linear dimensions are in cm.

**Fig. 3.A.7.2** Cut-out view of the experimental arrangement of "cross".



Surface equations

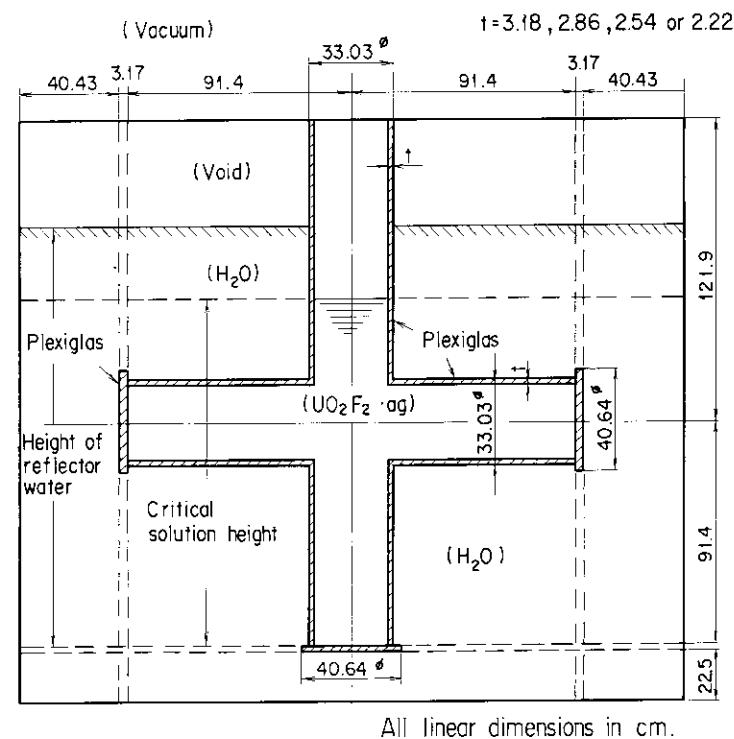
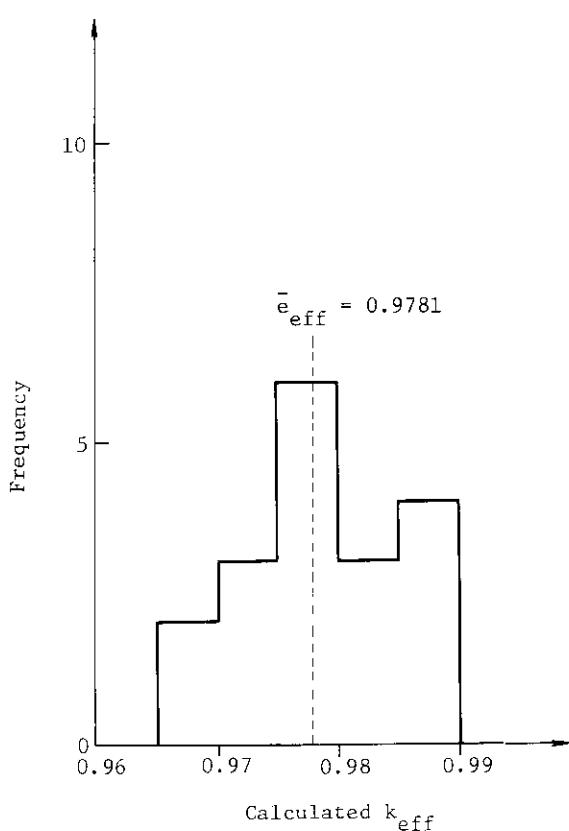
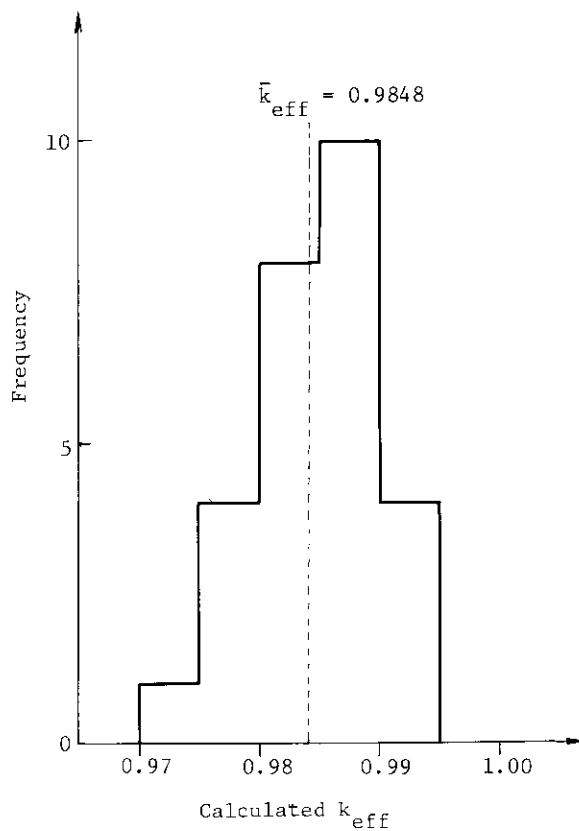
$$\textcircled{S1} \quad X^2 + y^2 - 199.09 = 0$$

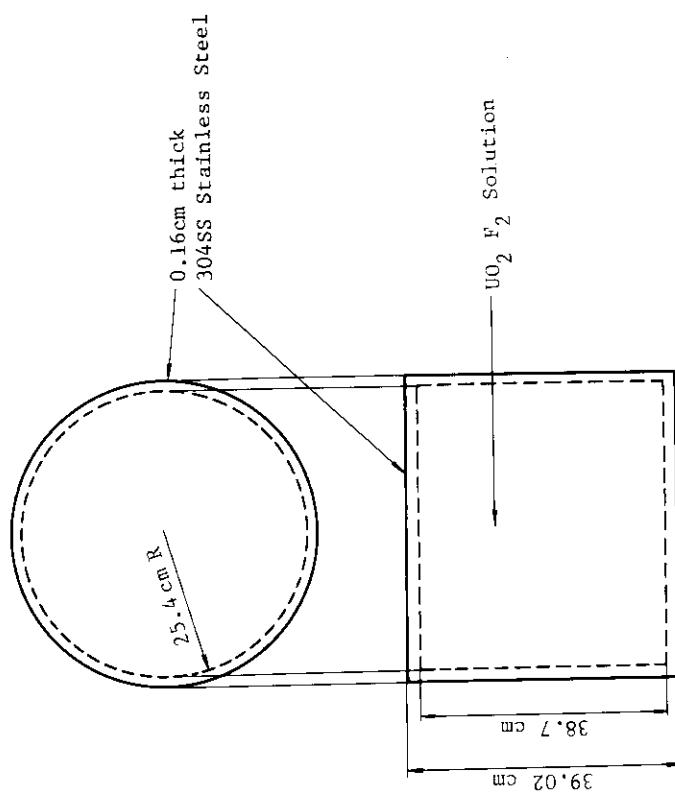
$$\textcircled{S2} \quad X^2 + y^2 - 194.60 = 0$$

$$\textcircled{S3} \quad 1.3139 X^2 + y^2 + 0.3139 Z^2 - 1.1205 yZ - 258.63 = 0$$

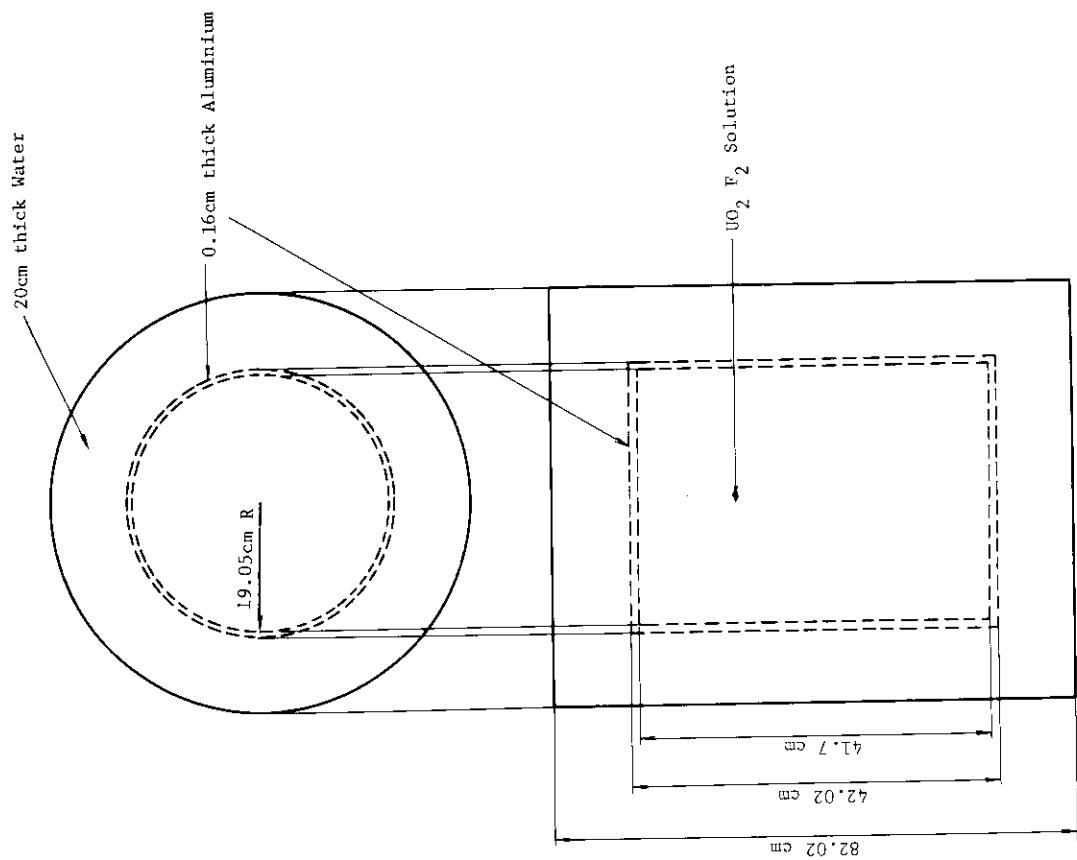
$$\textcircled{S4} \quad 1.3139 X^2 + y^2 + 0.3139 Z^2 - 1.1205 yZ - 255.69 = 0$$

**Fig. 3.A.7.3** Calculational model for "30 degrees lateral".

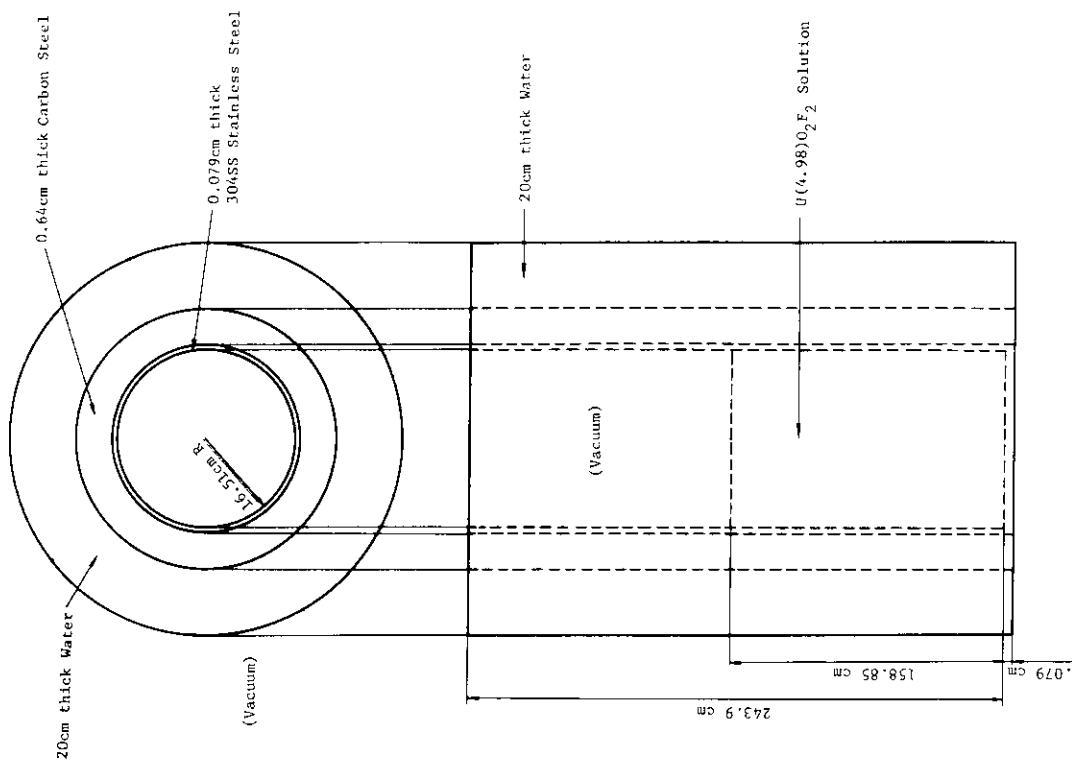
**Fig. 3.A.7.4** Calculational model for "cross".**Fig. 3.A.7.5** Histogram of calculated  $k_{eff}$ 's for "30 degrees lateral" system (18 cases).**Fig. 3.A.7.6** Histogram of calculated  $k_{eff}$ 's for "cross" systems (27 cases).



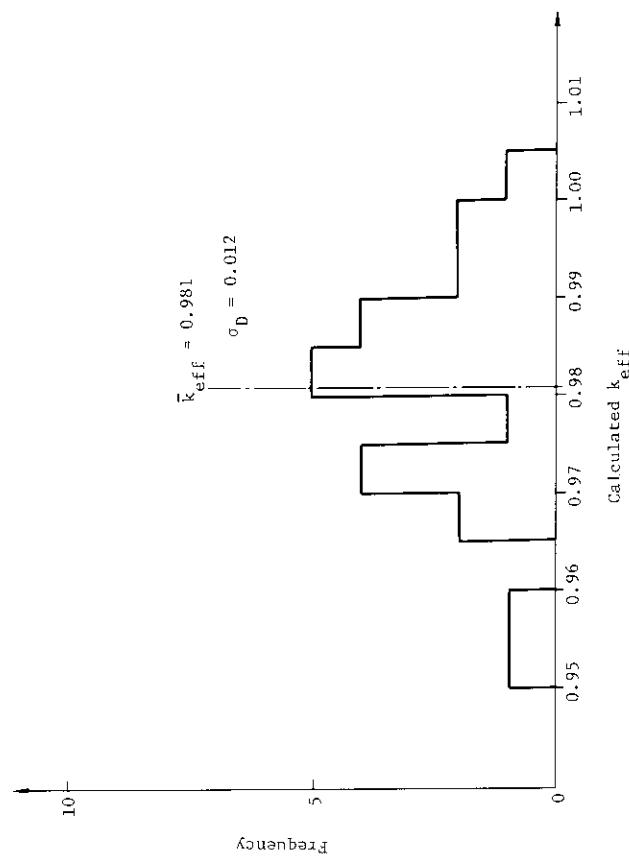
**Fig. 3.A.8.2** Calculational model for a typical unreflected cylinder containing uranyl-fluoride solution.



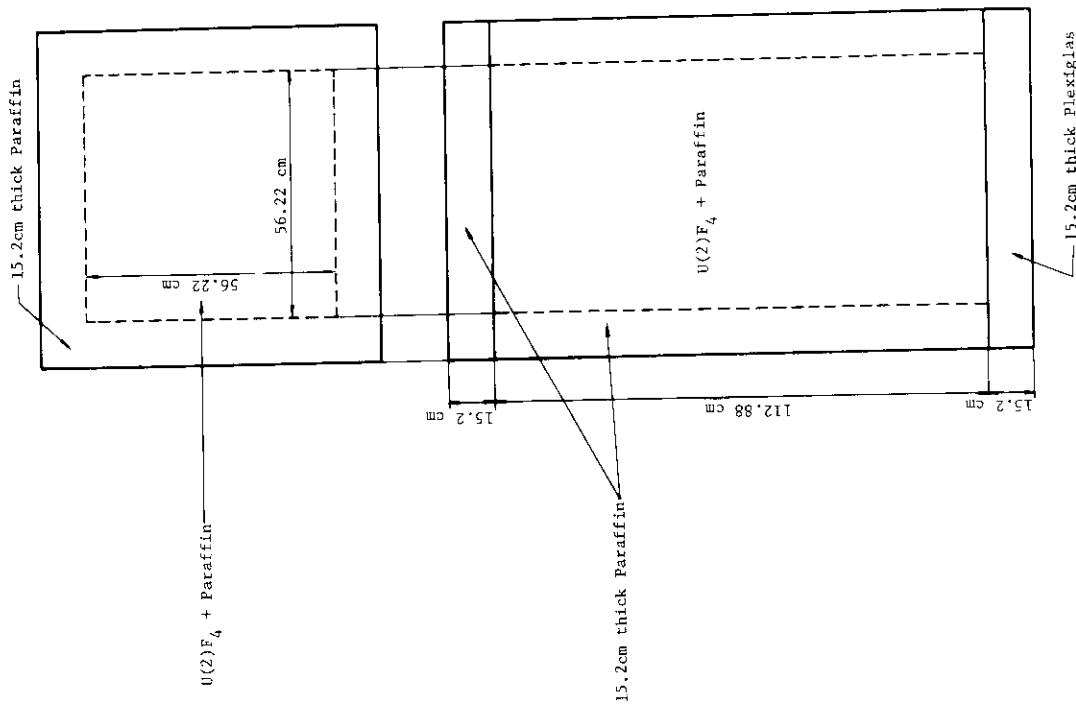
**Fig. 3.A.8.1** Calculational model for a typical water-reflected cylinder containing uranyl-fluoride solution.



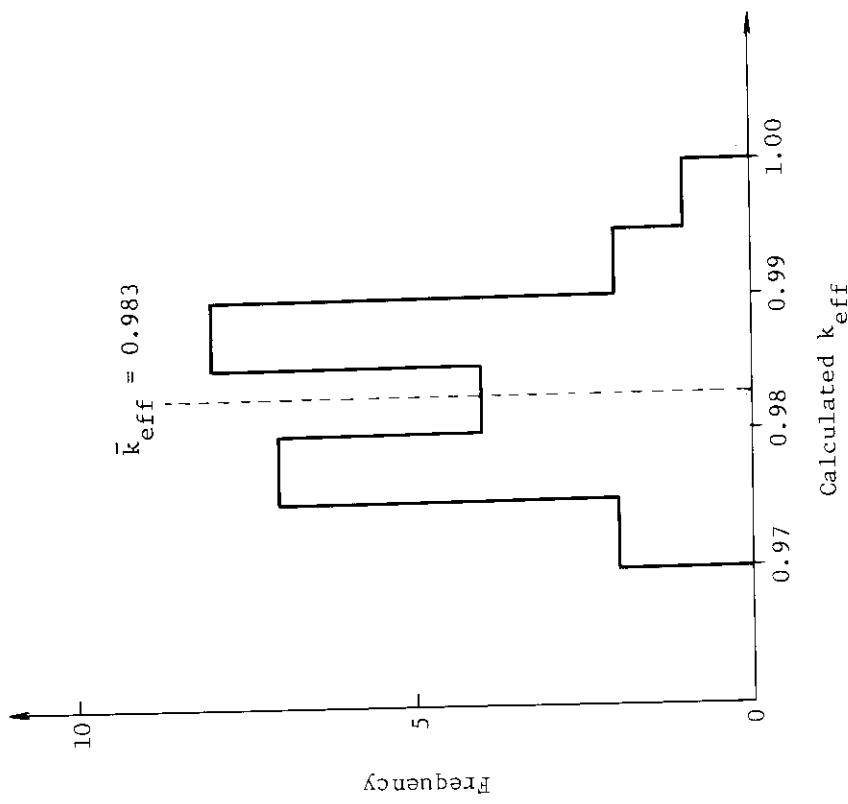
**Fig. 3.A.9.1** Calculational model for a typical experimental case of  $U(4.98)O_2F_2$  solution cylinder surrounded by composite reflector of steel/water.



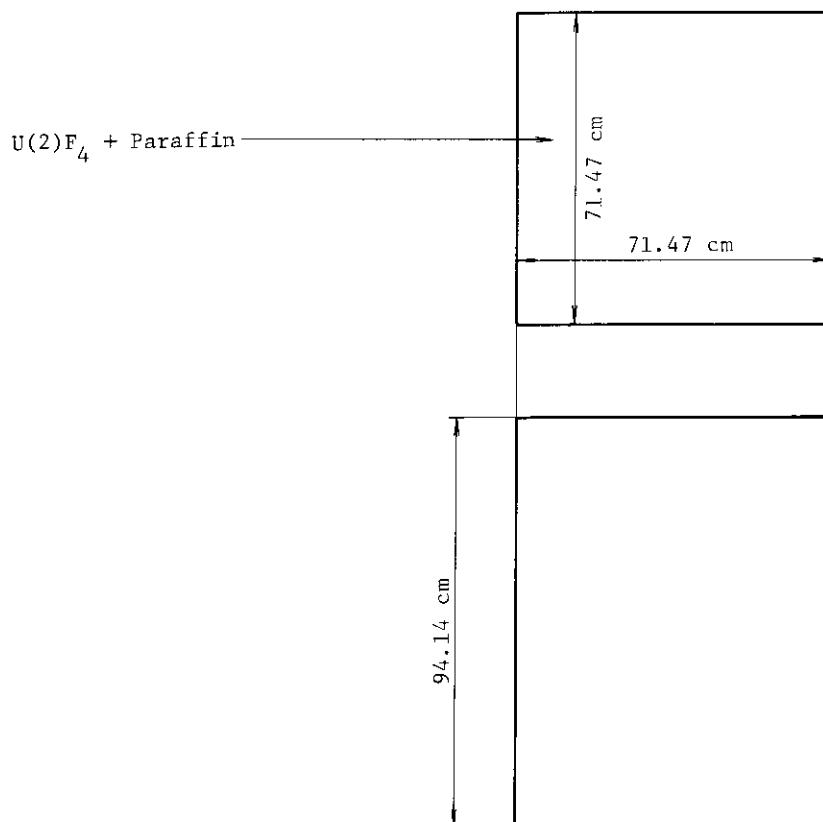
**Fig. 3.A.8.3** Histogram of calculated  $k_{eff}$ 's for cylinders or spheres containing uranyl-fluoride solution.



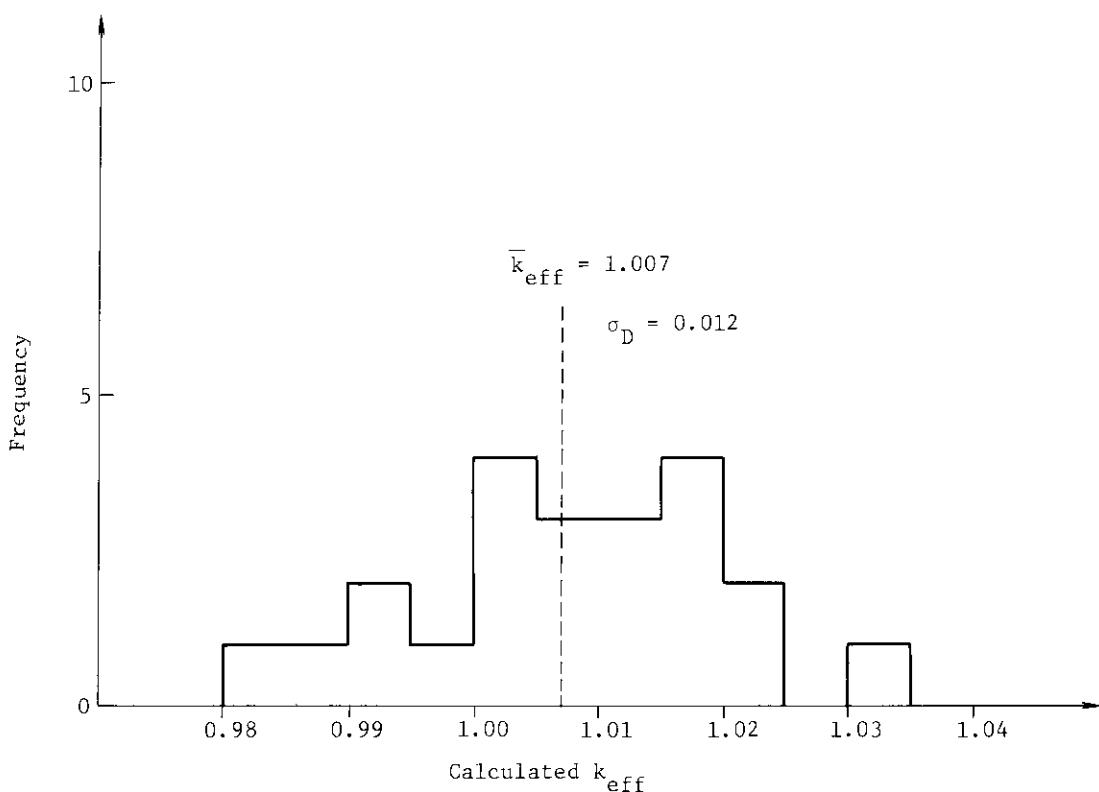
**Fig. 3.A.10.1** Calculational model for a typical reflected pile of UF<sub>4</sub>-paraffin compact fuel.



**Fig. 3.A.9.2** Histogram of calculated  $k_{eff}$ 's for cylinders containing uranyl fluoride solution surrounded by composite reflector.



**Fig. 3.A.10.2** Calculational model for a typical unreflected pile of  $\text{UF}_4$ -paraffin compact fuel.



**Fig. 3.A.10.3** Histogram of calculated  $k_{\text{eff}}$ 's for reflected pile of  $\text{UF}_4$ -paraffin compact fuel.

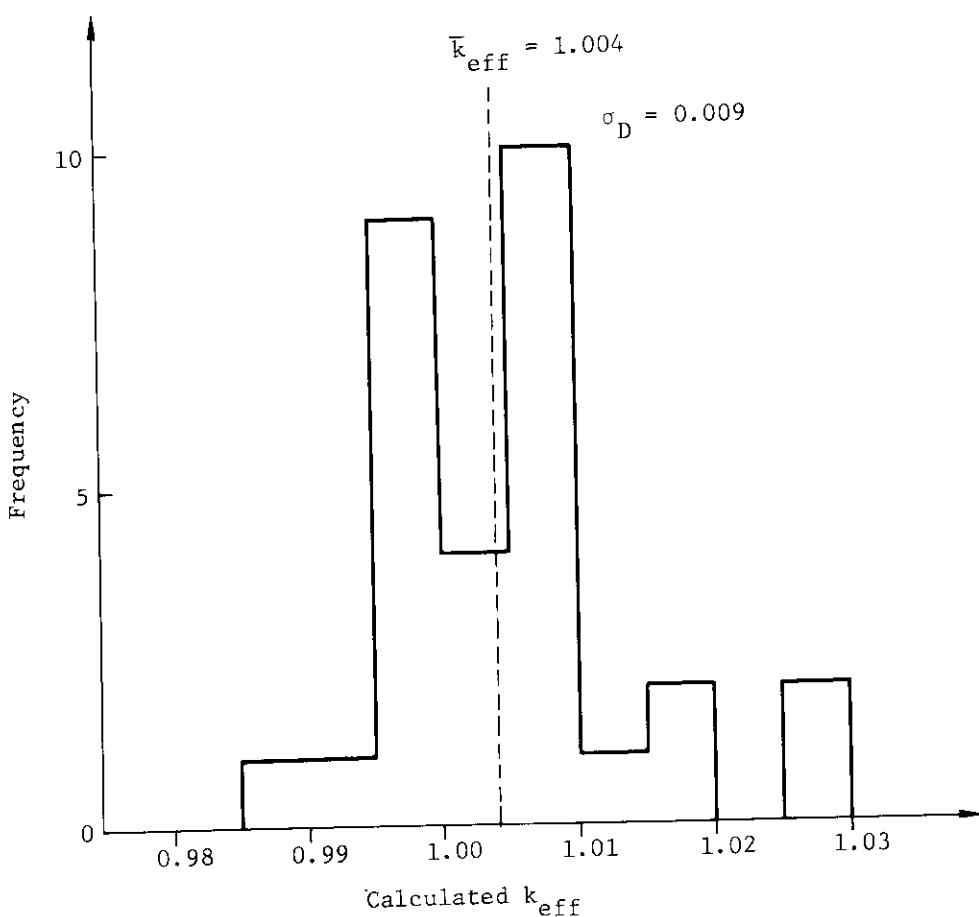


Fig. 3.A.10.4 Histogram of calculated  $k_{eff}$ 's for unreflected pile of  $UF_4$ -paraffin compact fuel.

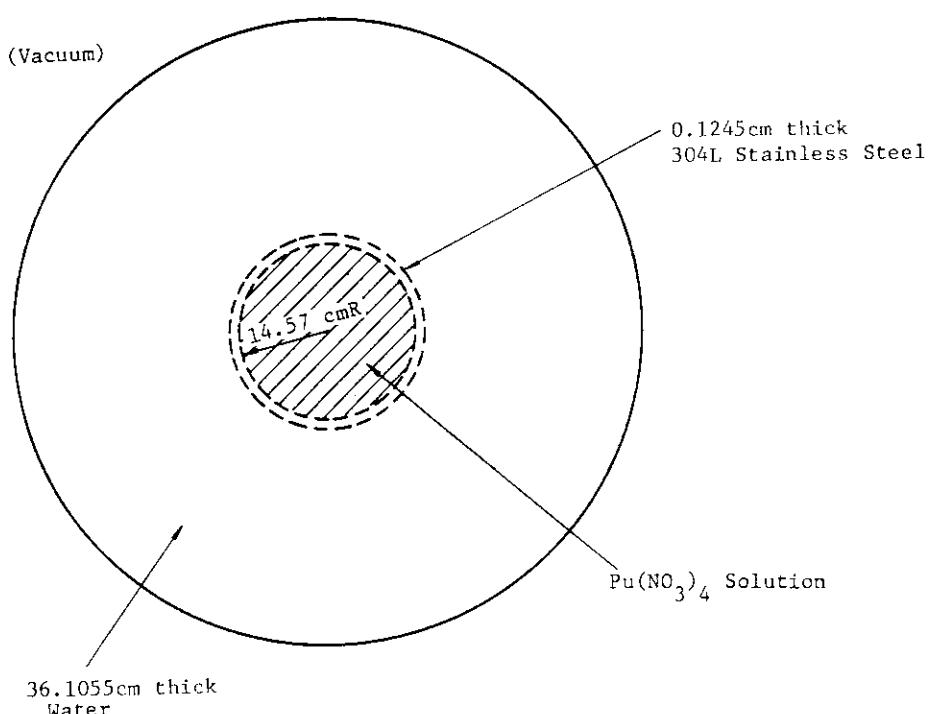
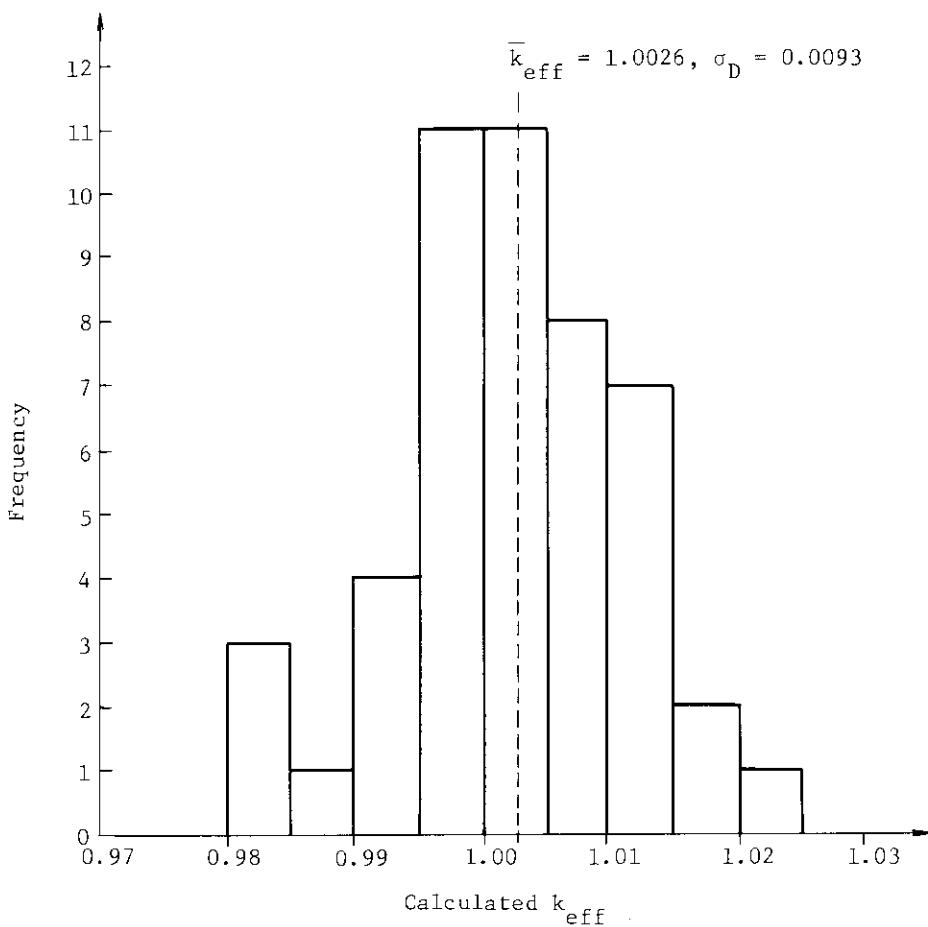
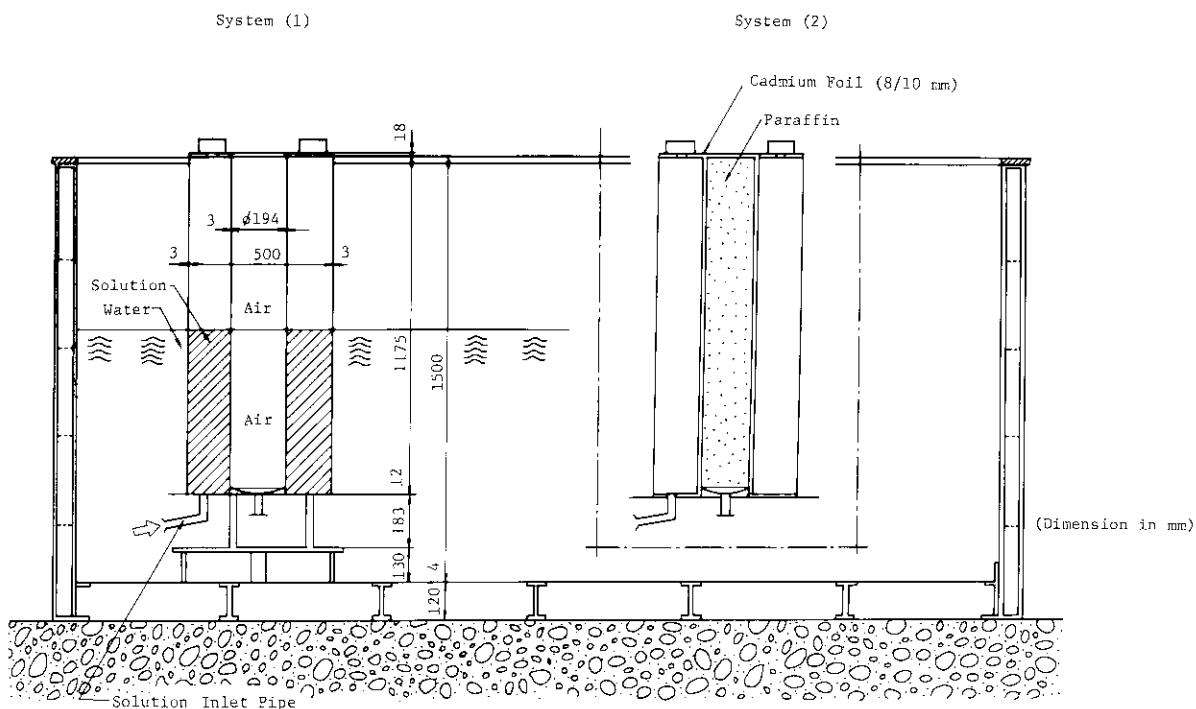


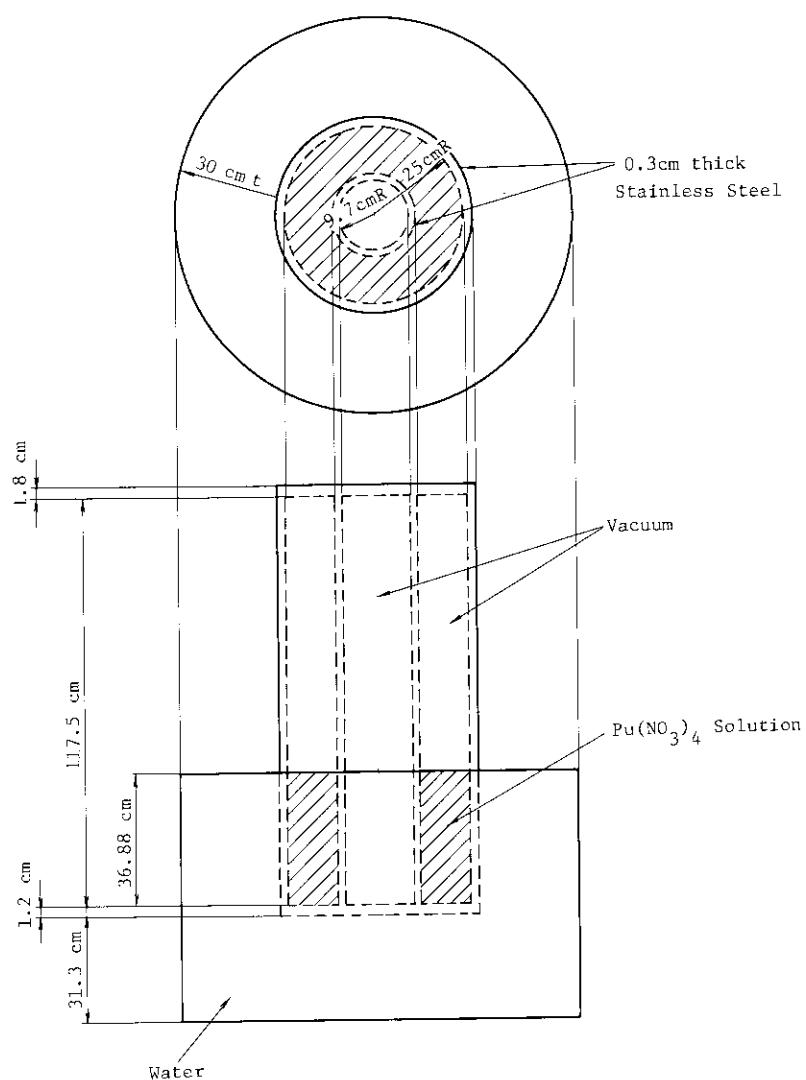
Fig. 3.A.11.1 Calculational model for a typical experimental system, case No.1.



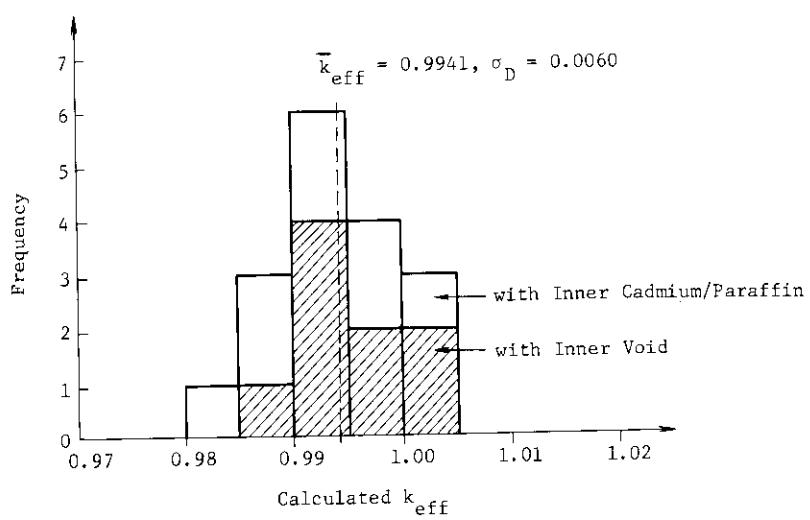
**Fig. 3.A.11.2** Histogram of calculated  $k_{\text{eff}}$ 's for 48 spherical vessels containing uranyl-nitrate solution with various reflectors.



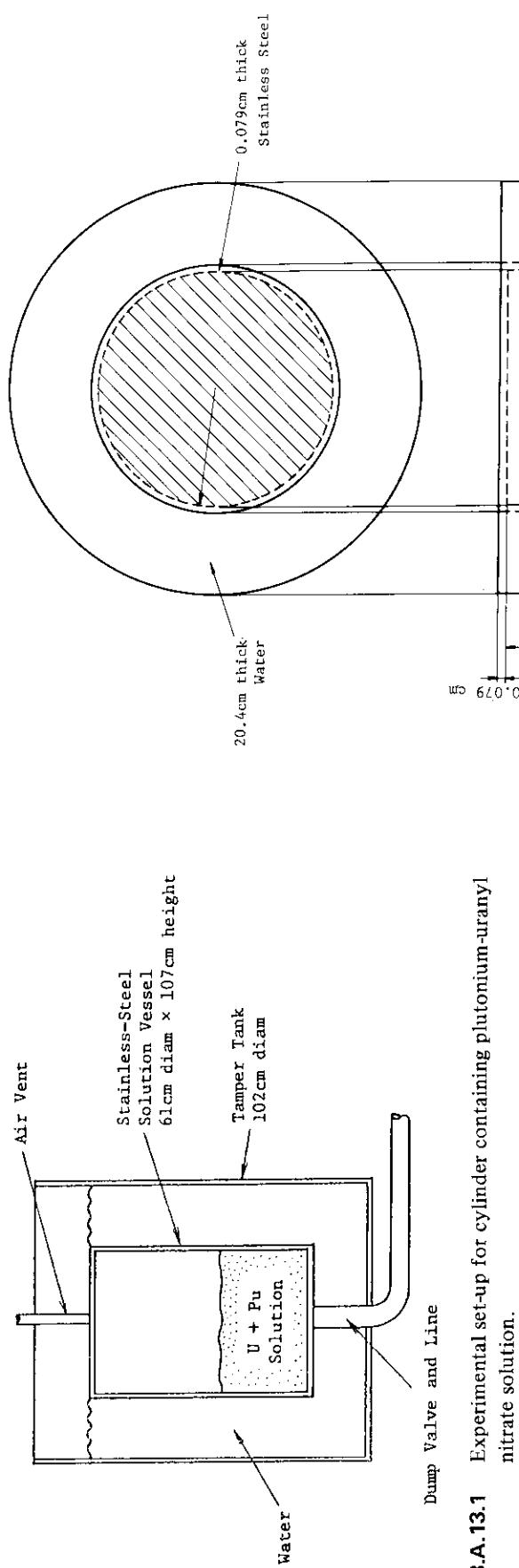
**Fig. 3.A.12.1** Experimental arrangement of annular cylinders containing plutonium-nitrate solution.



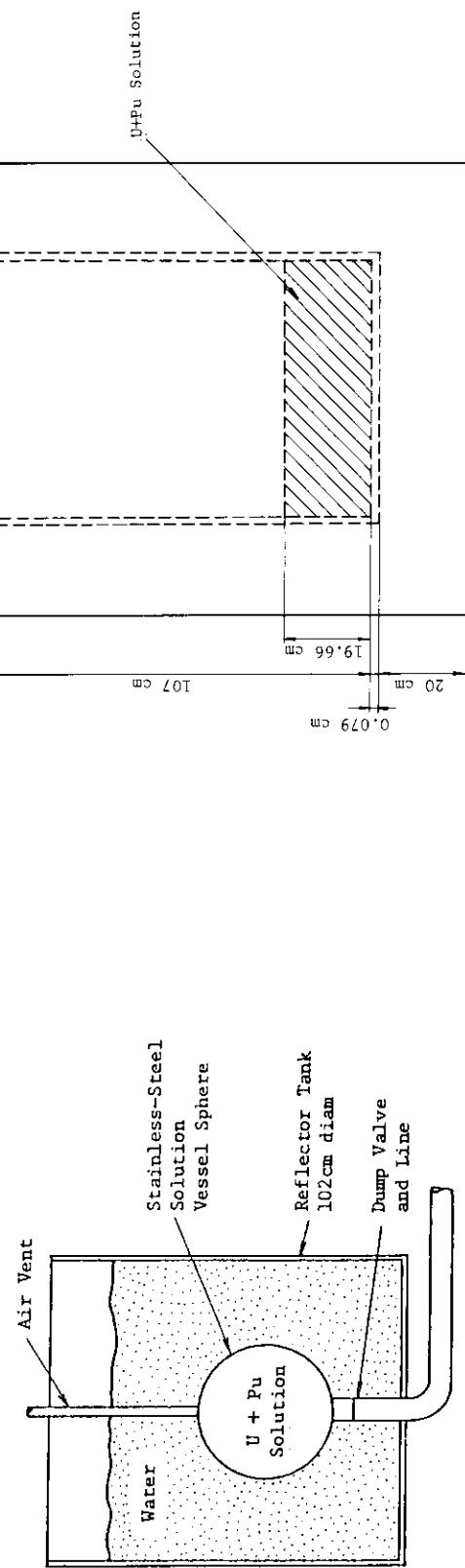
**Fig. 3.A.12.2** Calculational model for a typical case of annular cylinder containing plutonium-nitrate solution.



**Fig. 3.A.12.3** Histogram of calculated  $k_{eff}$ 's for annular cylinders containing plutonium-nitrate solution with or without inner cadmium/paraffin.

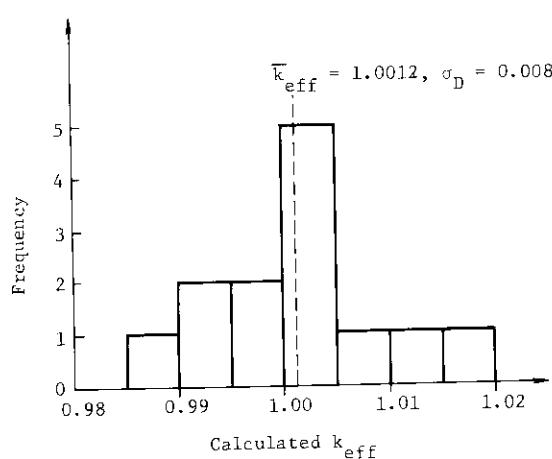


**Fig. 3.A.13.1** Experimental set-up for cylinder containing plutonium-uranyl nitrate solution.

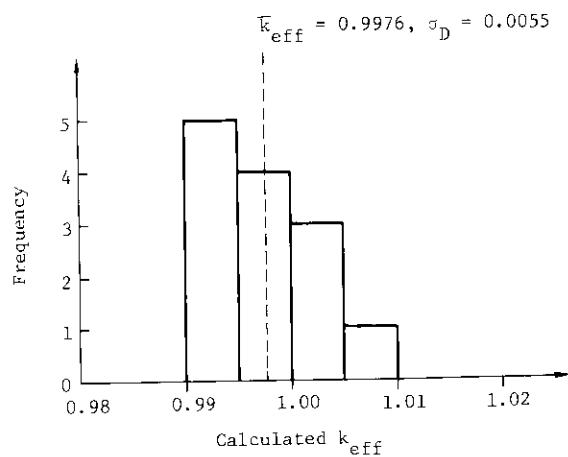


**Fig. 3.A.13.2** Experimental set-up for sphere containing plutonium-uranyl nitrate solution.

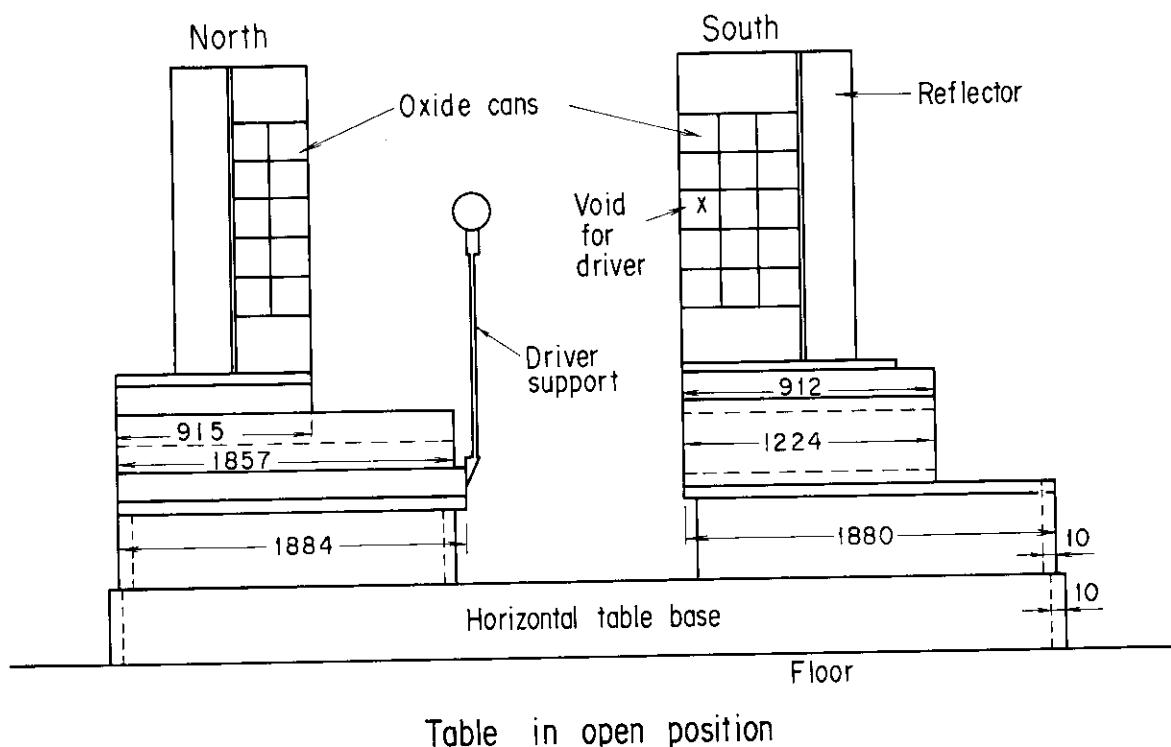
**Fig. 3.A.13.3** Calculational model for a typical cylinder containing plutonium-uranyl nitrate solution.



**Fig. 3.A.13.4** Histogram of calculated  $k_{eff}$ 's for cylinders containing (Pu+U) nitrate solution of 5.6 wt%  $^{240}\text{Pu}$  in Pu.



**Fig. 3.A.13.5** Histogram of calculated  $k_{eff}$ 's for cylinders containing (Pu+U) nitrate solution of 23.0 wt%  $^{240}\text{Pu}$  in Pu.



**Fig. 3.A.14.1** Experimental arrangement of the array of compacted low-enriched uranium-oxide packages.

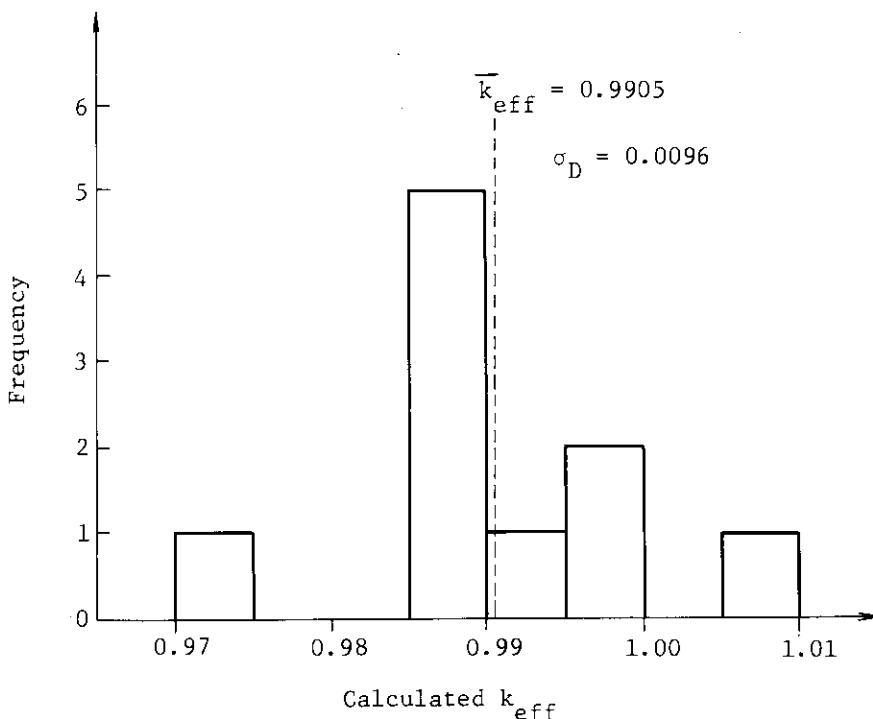


Fig. 3.A.14.2 Histogram of calculated  $k_{eff}$ 's for  $U_3O_8$  package arrays.

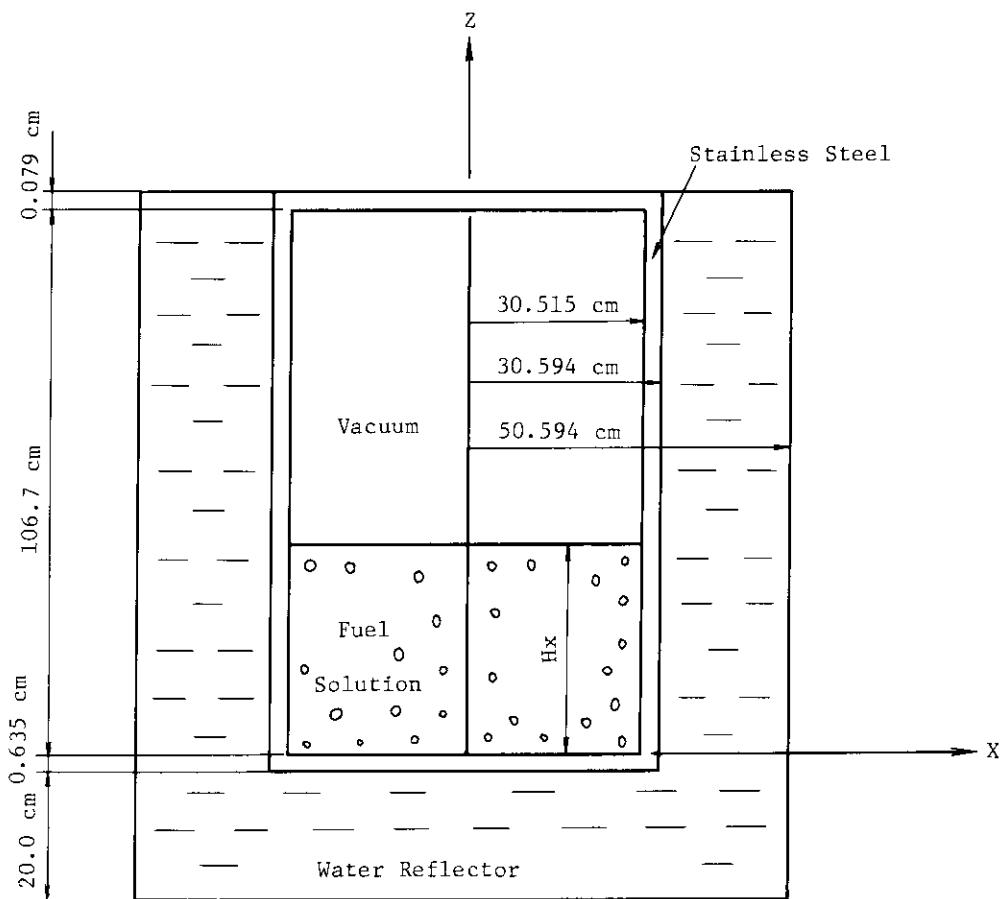
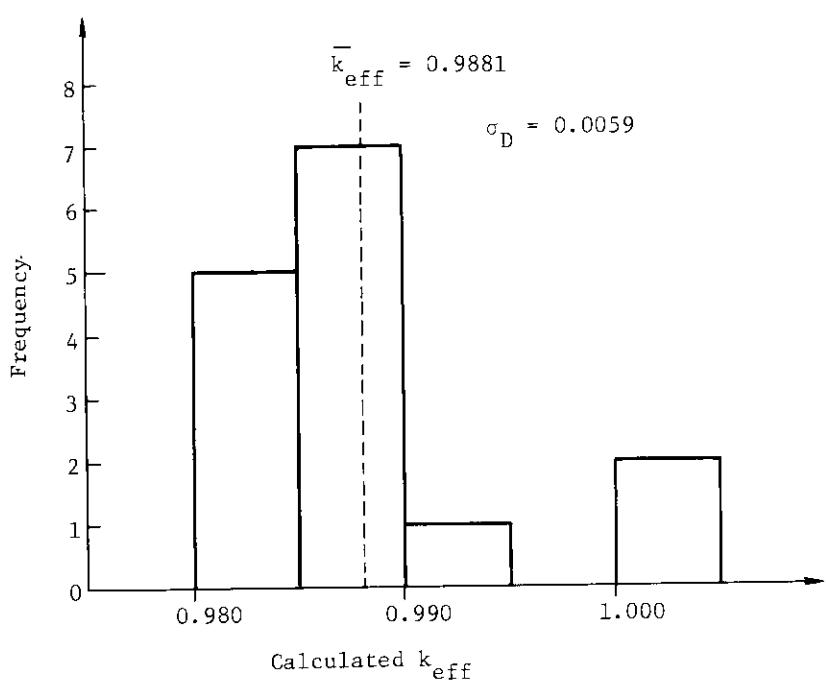
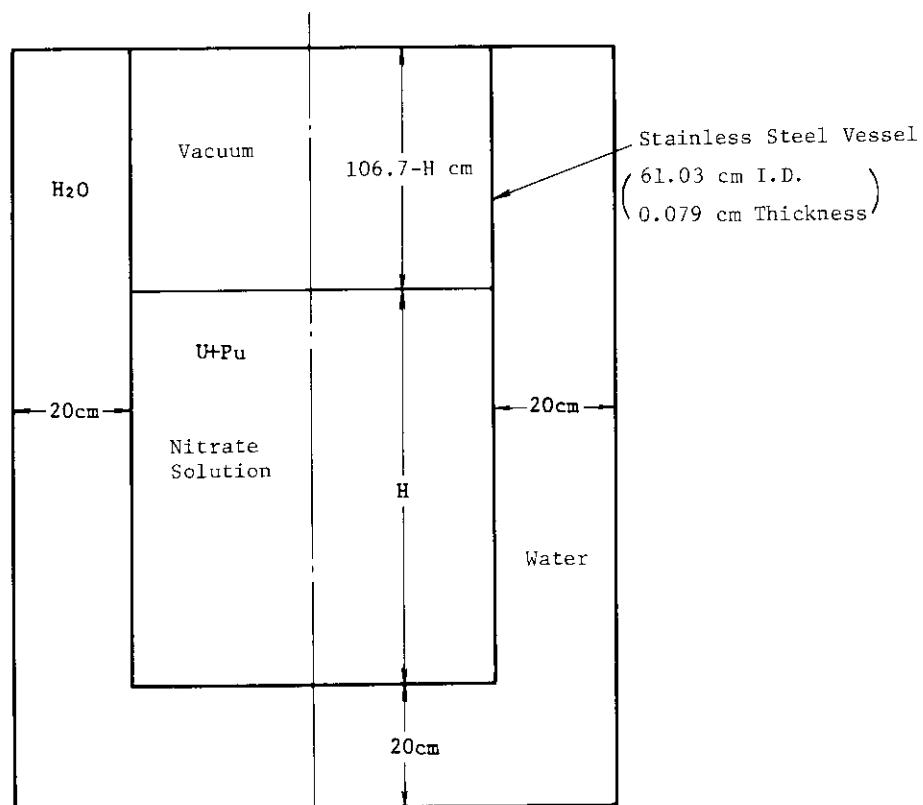


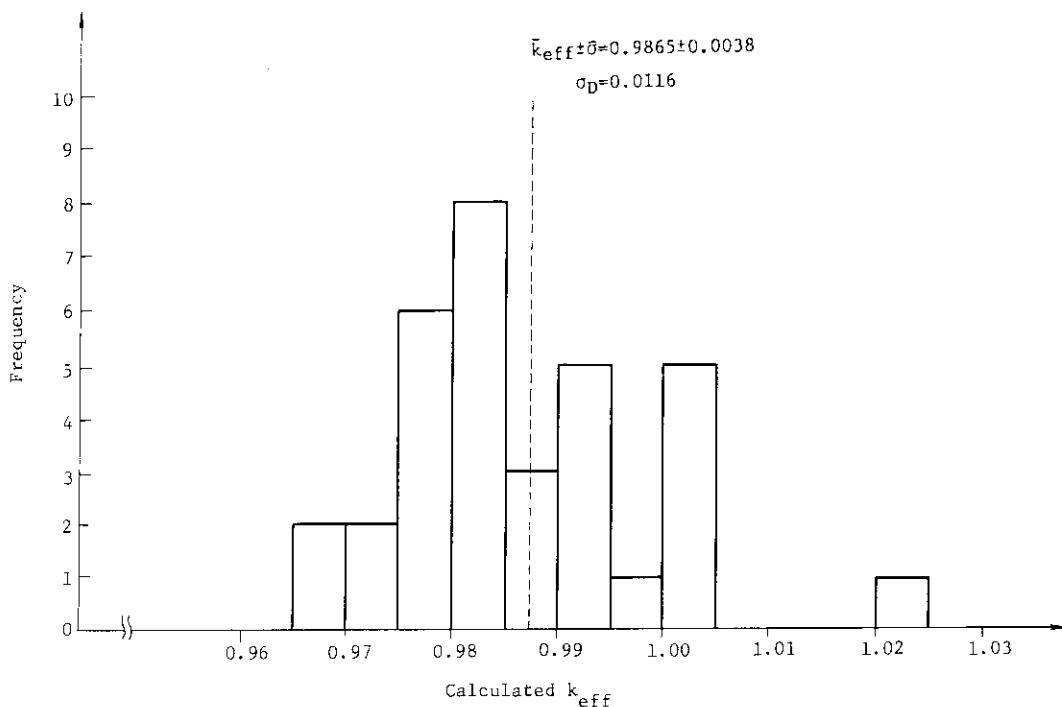
Fig. 3.A.15.1 Calculational model for plutonium-nitrate solution with gadolinium poison in a cylindrical vessel.



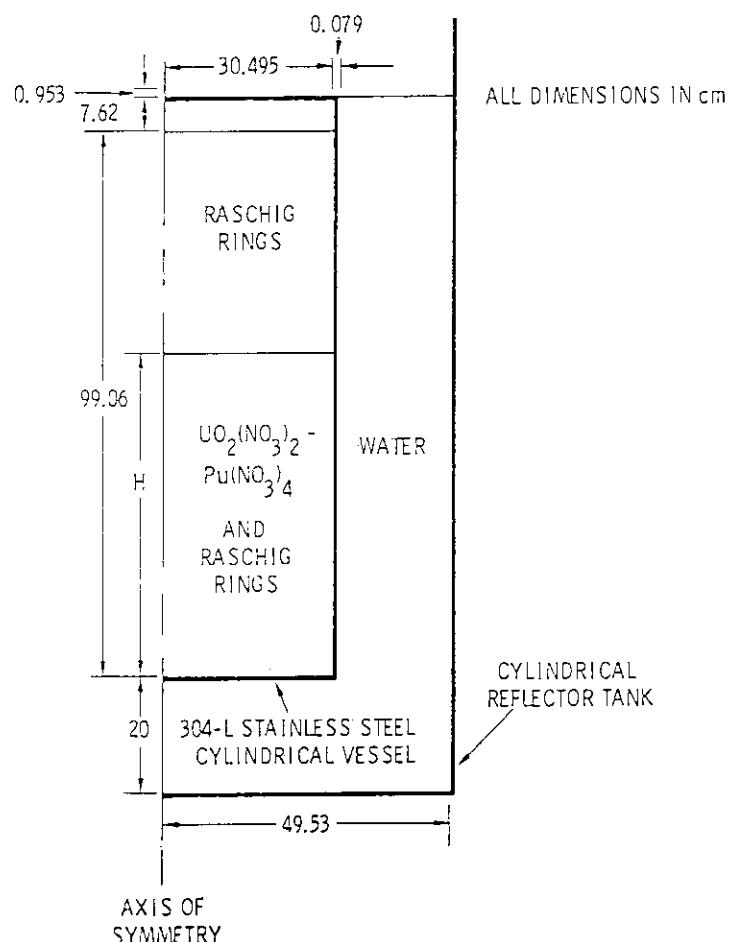
**Fig. 3.A.15.2** Histogram of calculated  $k_{eff}$ 's for plutonium-nitrate solution with gadolinium.



**Fig. 3.A.16.1** Calculational model for a cylinder containing (U+Pu) nitrate solution with soluble neutron absorbers.



**Fig. 3.A.16.2** Histogram of 34 calculated  $k_{\text{eff}}$ 's for cylinders containing Plutonium-uranyl nitrate solution with soluble neutron absorbers.



**Fig. 3.A.17.1** Schema of experimental set up for raschig rings in uranium-plutonium nitrate solution.

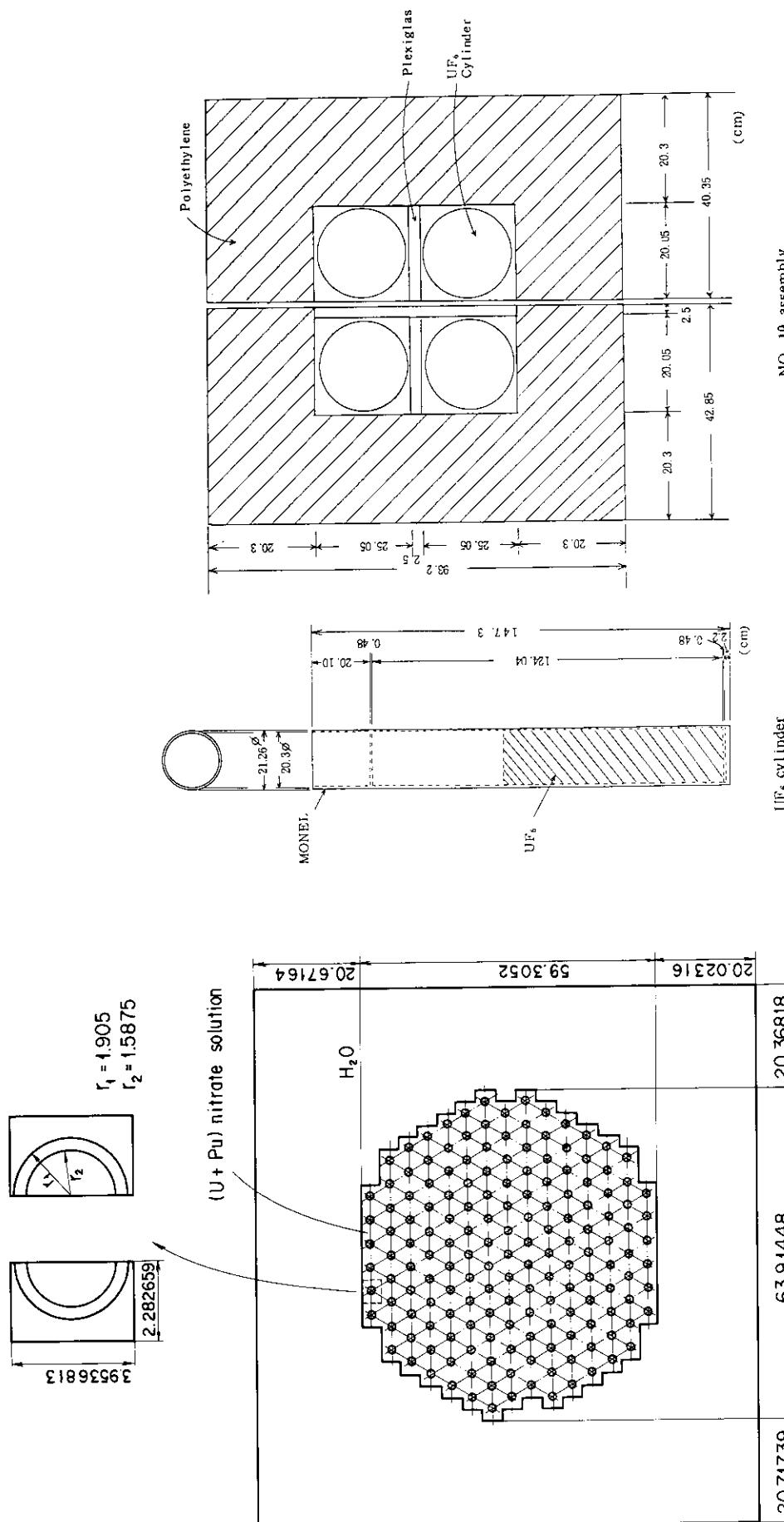
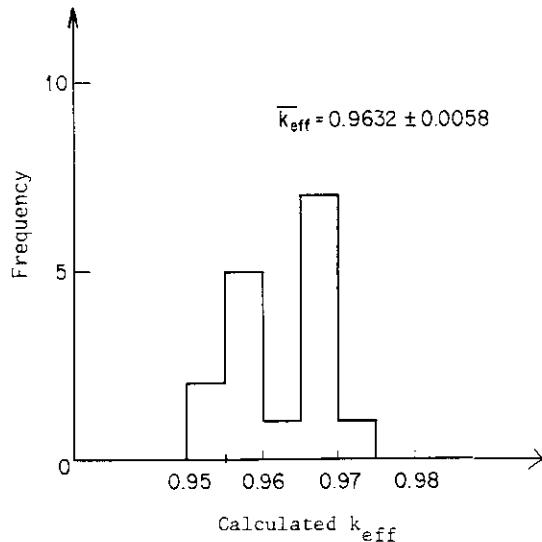
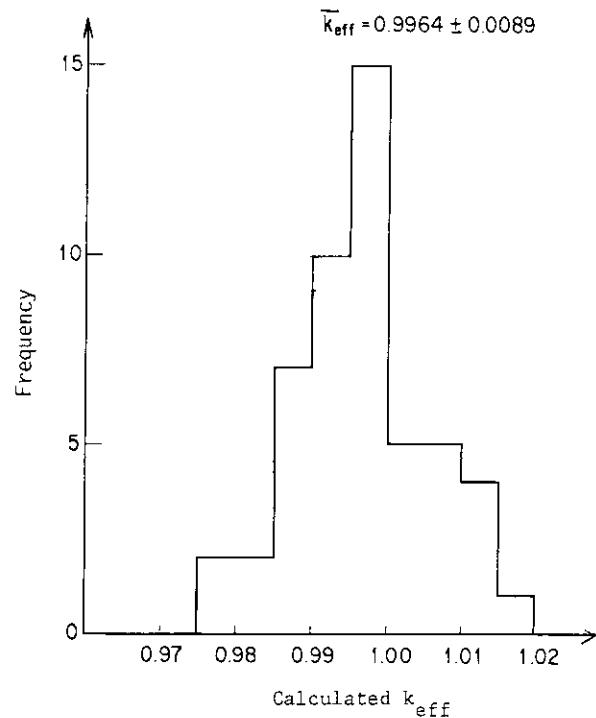


Fig. 3.A.17.2 Calculational model for Raschig rings in uranium-plutonium nitrate solution.  
All dimensions in cm

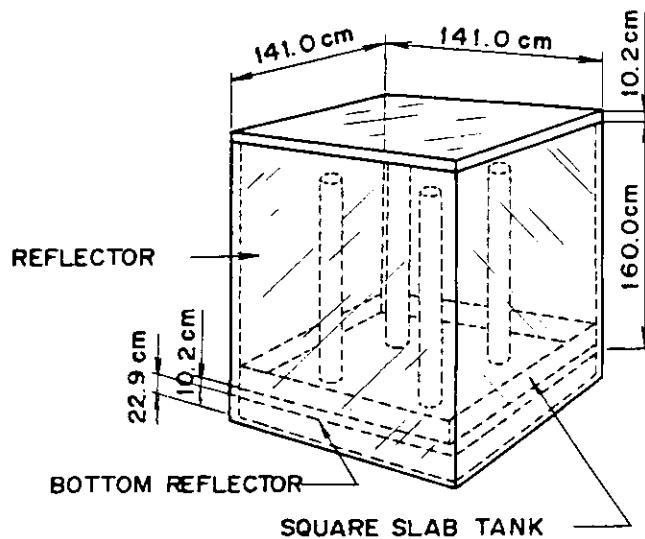
Fig. 3.B.1.1 Calculational models of a  $UF_6$  cylinder and its typical arrangement.



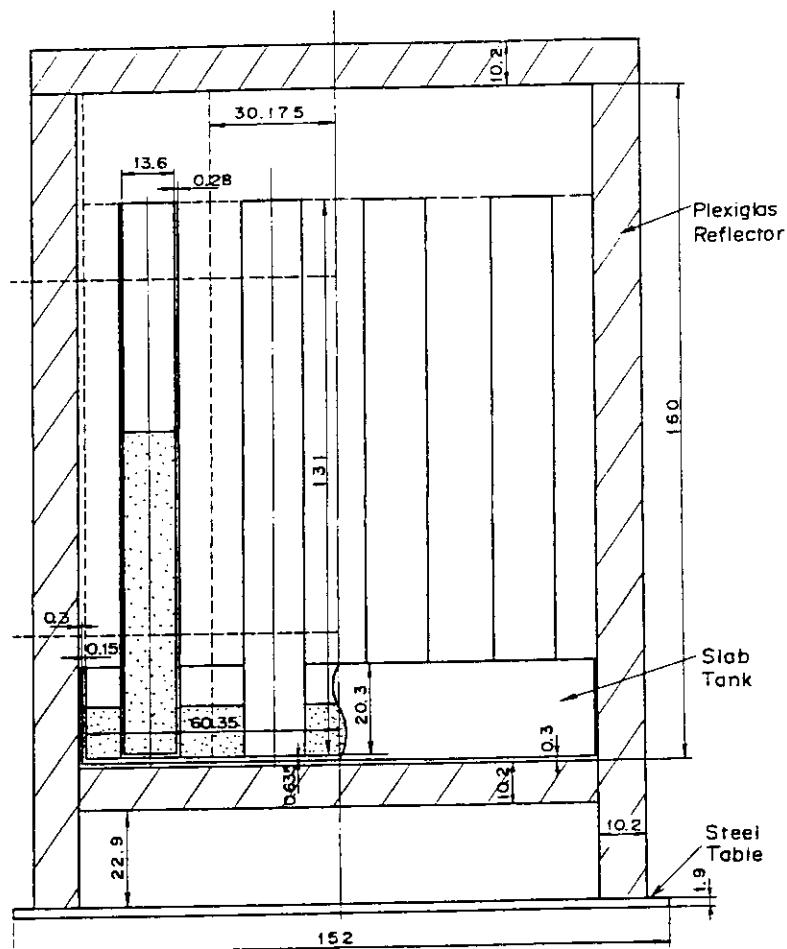
**Fig. 3.B.1.2** Histogram of calculated  $k_{\text{eff}}$ 's for the unreflected  $\text{UF}_6$  cylinder systems (16 cases).



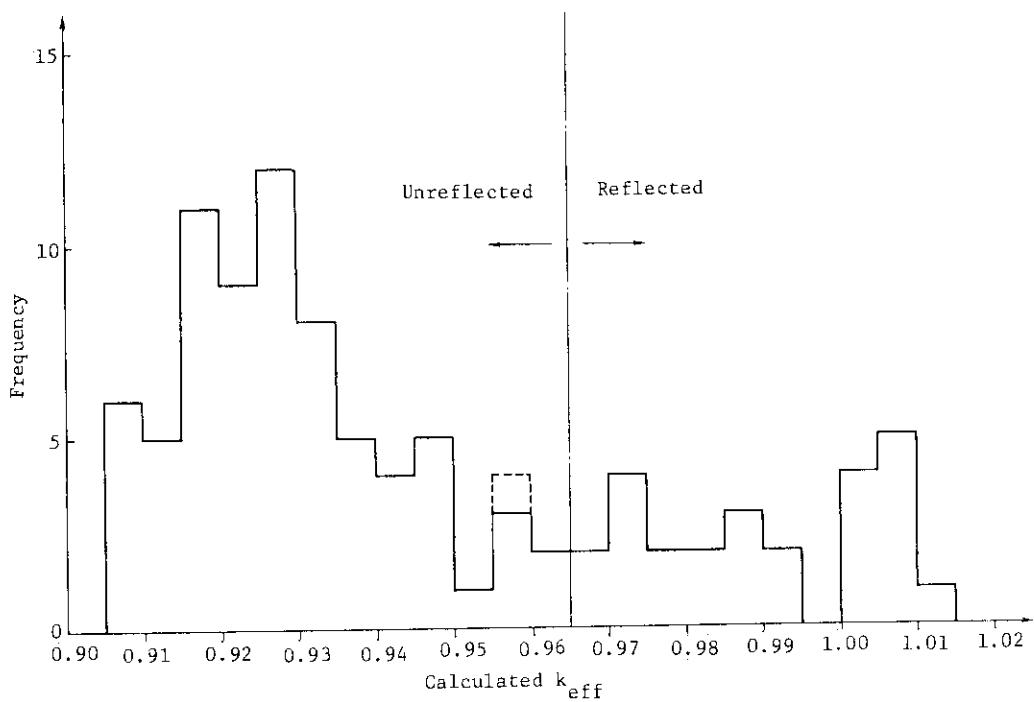
**Fig. 3.B.1.3** Histogram of calculated  $k_{\text{eff}}$ 's for the reflected  $\text{UF}_6$  cylinder systems (51 cases).



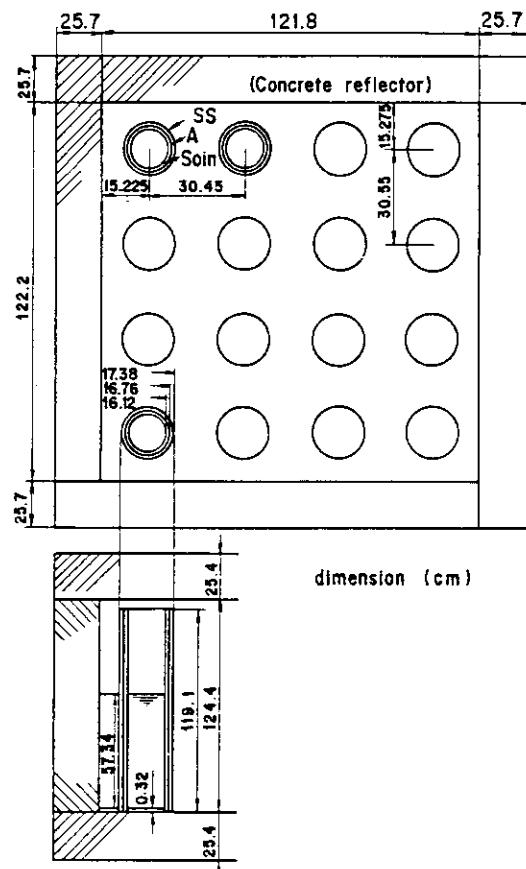
**Fig. 3.B.2.1** Typical reflected experimental configuration.

**Fig. 3.B.2.2**

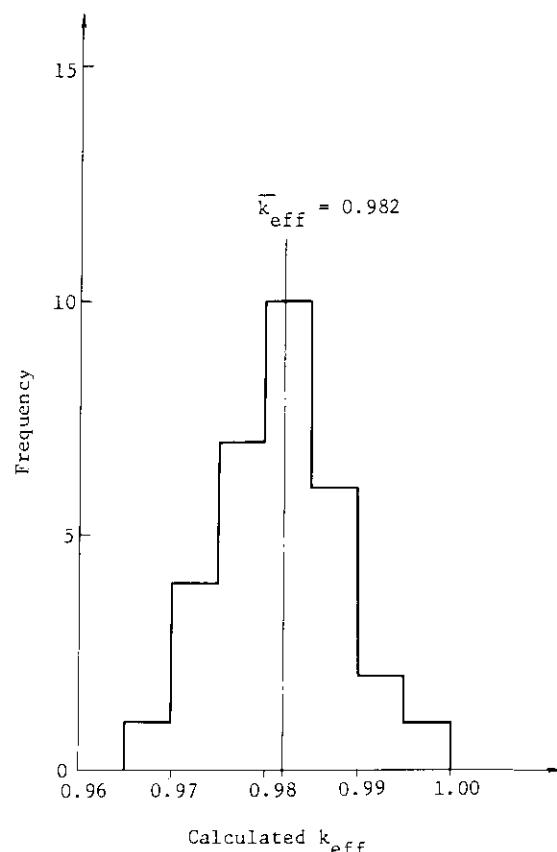
Calculational model for a reflected  $4 \times 4$  cylinder array.



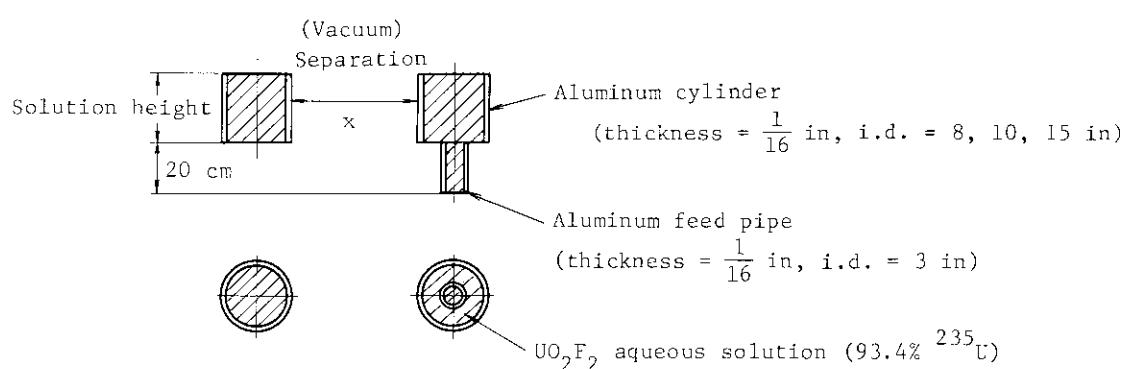
**Fig. 3.B.2.3** Histogram of calculated multiplication factor ( $k_{eff}$ ). Calculated  $k_{eff}$ 's are divided into two groups; one has the values lower than 0.965 corresponding to the unreflected systems (70 cases); the other has the values greater than 0.965 corresponding to the reflected systems (26 cases).



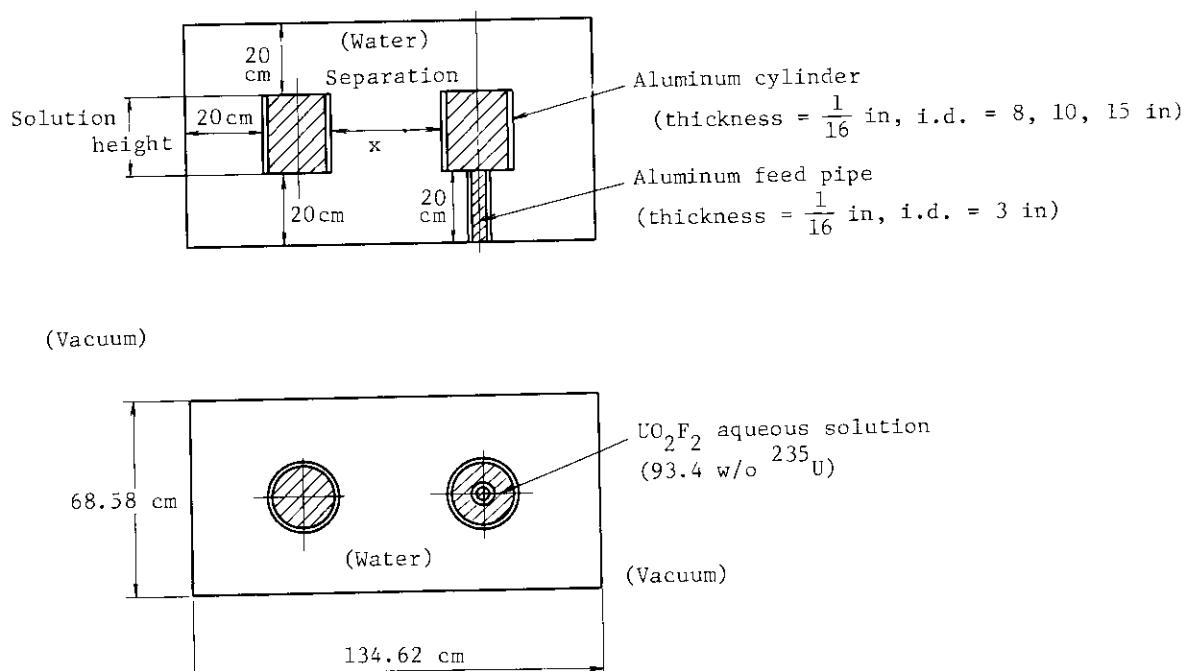
**Fig. 3.B.3.1** Calculational model of reflected sleeved cylinders.



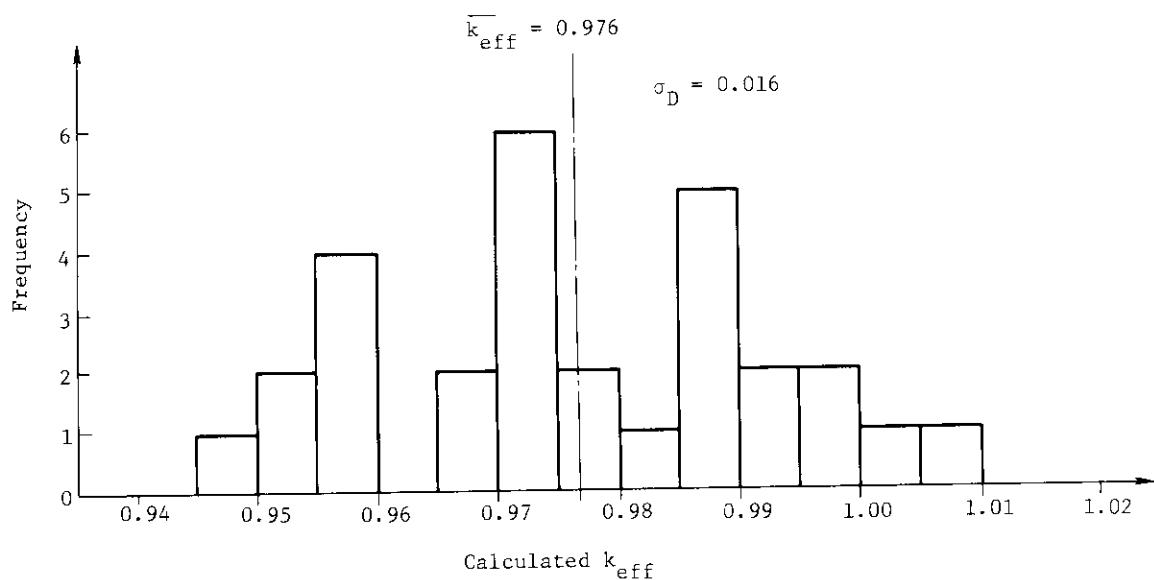
**Fig. 3.B.3.2** Histogram of calculated  $k_{\text{eff}}$ 's for the reflected array of cylinders (31 cases).



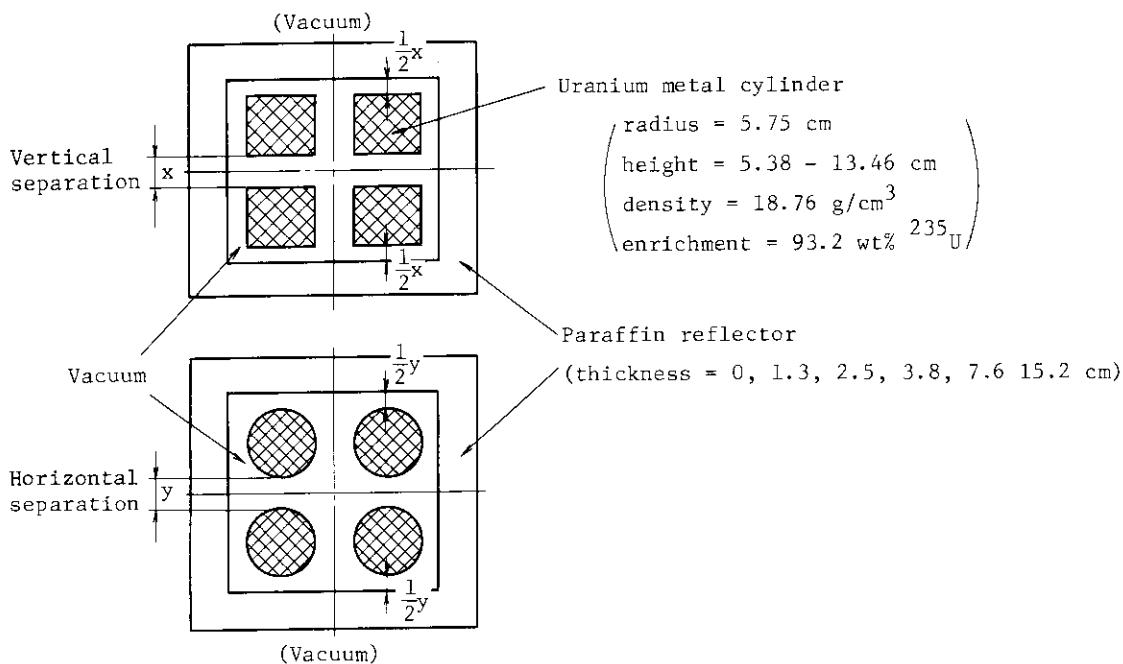
**Fig. 3.B.4.1** Calculational model for two aluminum reactors without water reflector.



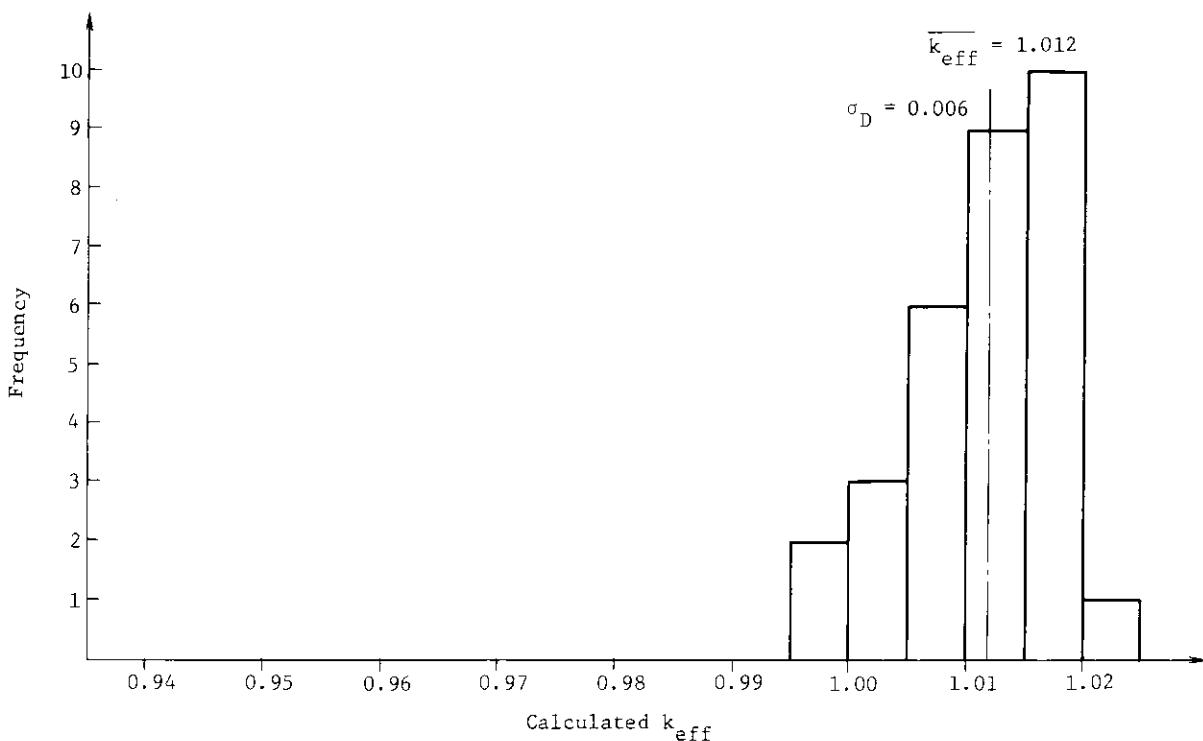
**Fig. 3.B.4.2** Calculational model for two aluminum reactors with water reflector.



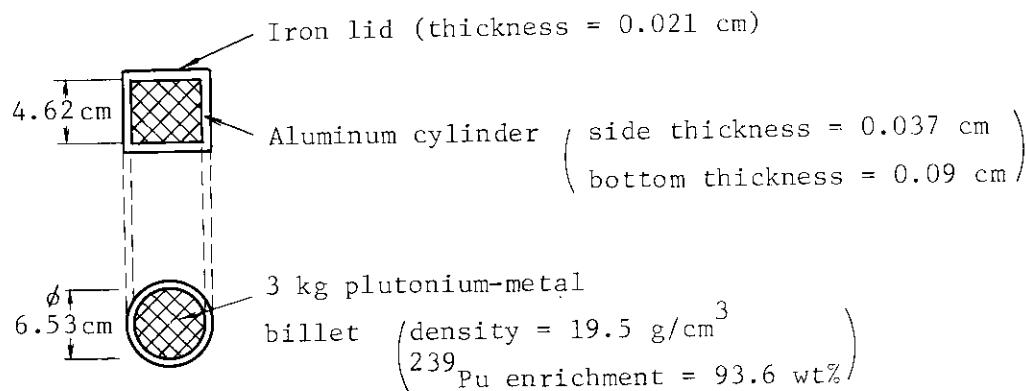
**Fig. 3.B.4.3** Histogram of calculated  $k_{eff}$ 's for two aluminum reactors (29 cases).



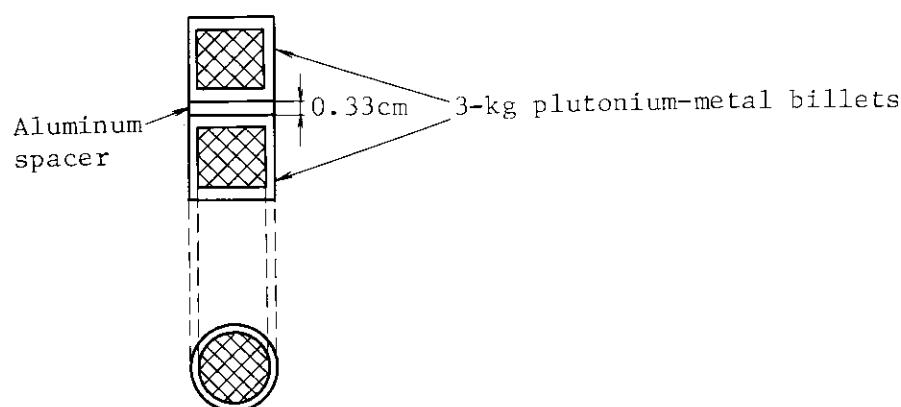
**Fig. 3.B.5.1** Calculational model for a  $2 \times 2 \times 2$  array of uranium-metal cylinders with paraffin reflector.



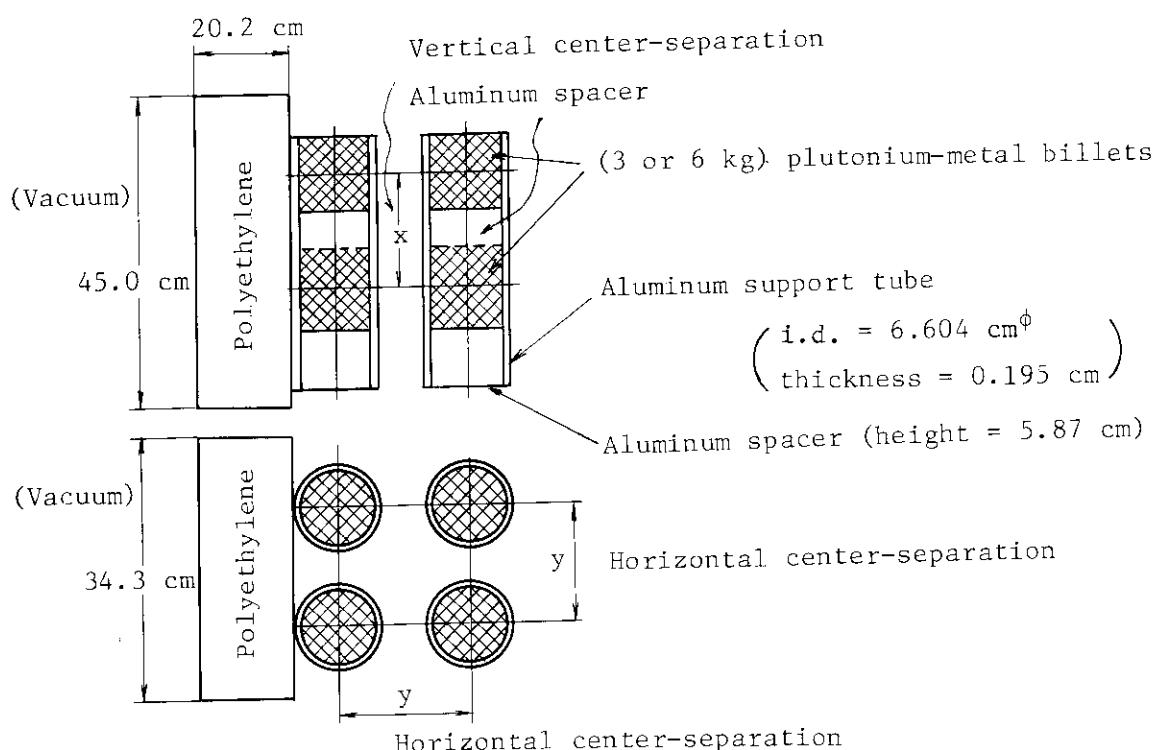
**Fig. 3.B.5.2** Histogram of calculated  $k_{eff}$ 's for cubic arrays of uranium-metal cylinders (31 cases).



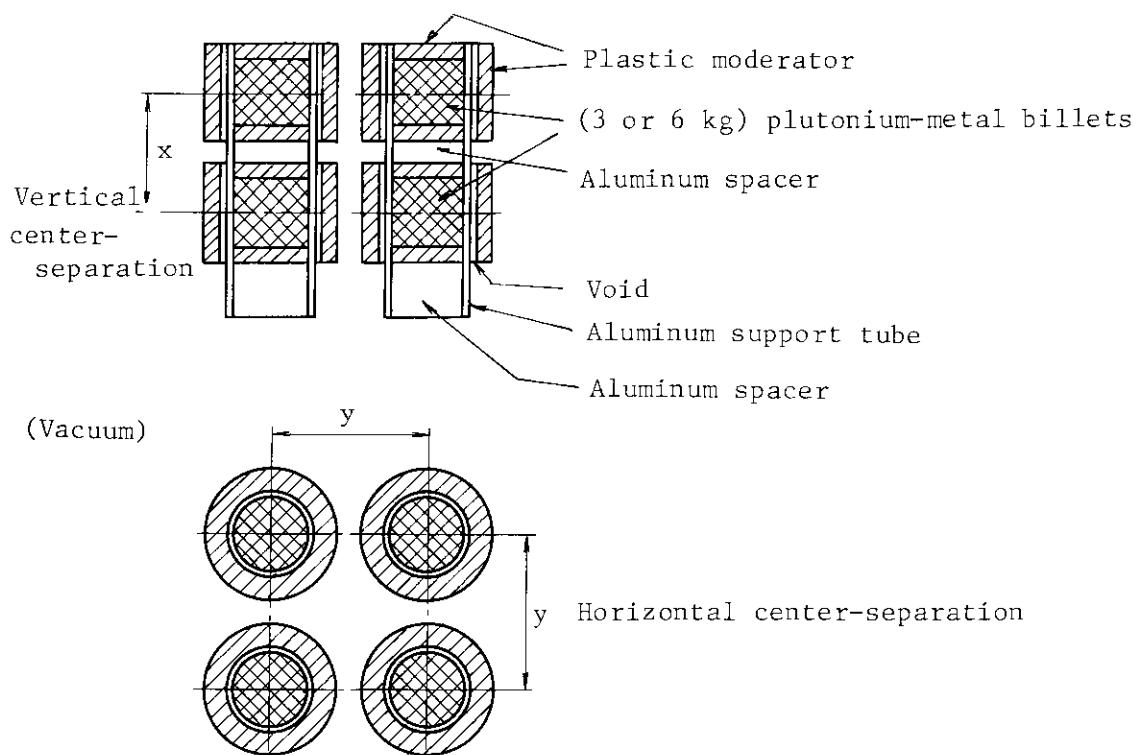
**Fig. 3.B.6.1** Calculational model for a 3 kg plutonium-metal billet.



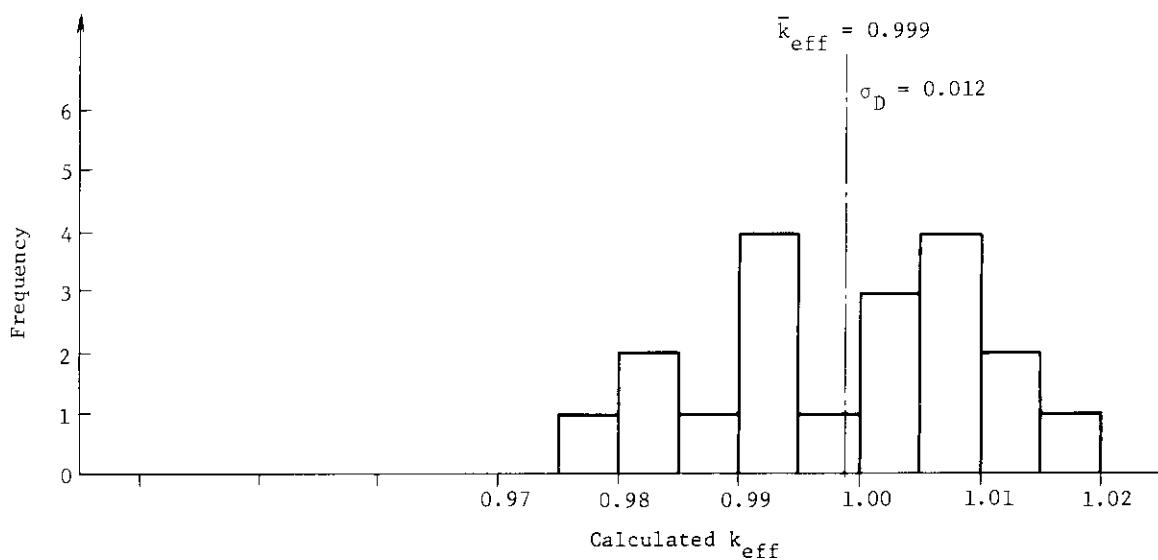
**Fig. 3.B.6.2** Calculational model for a 6 kg plutonium-metal billet.



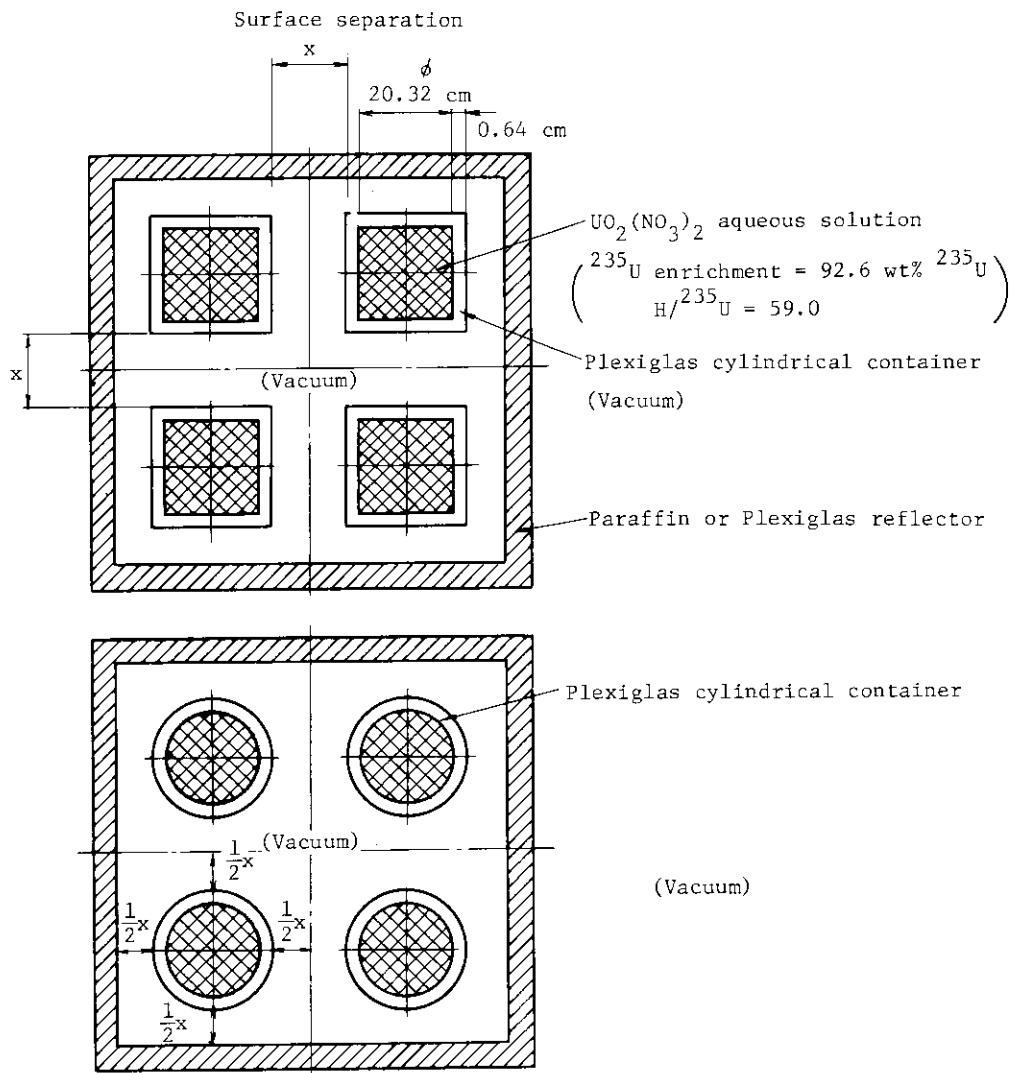
**Fig. 3.B.6.3** Calculational model for a 2X2X2 array of plutonium-metal billets unmoderated with partial reflection.



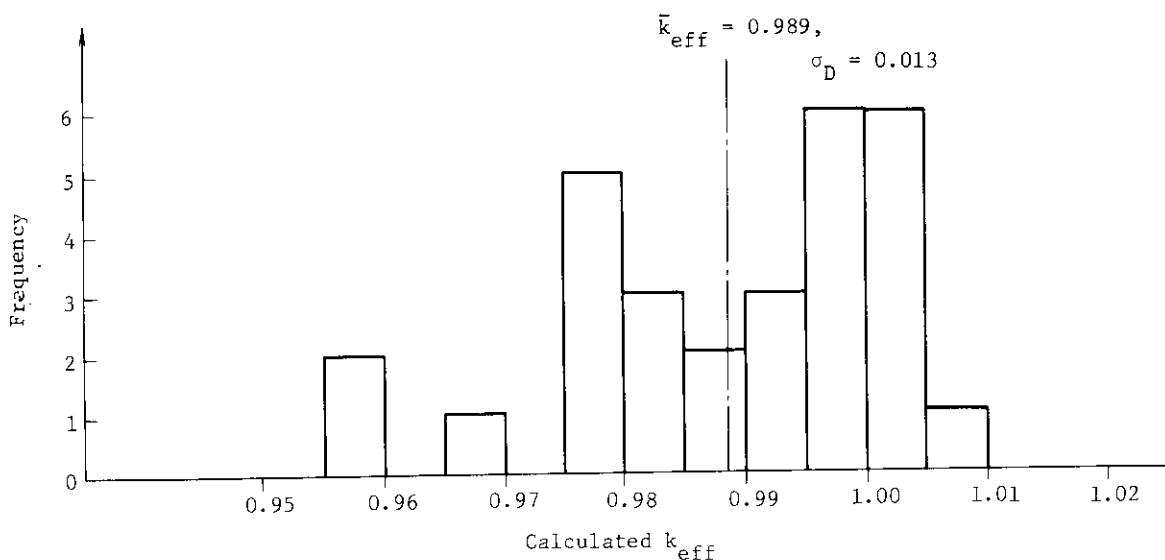
**Fig. 3.B.6.4** Calculational model for a  $2 \times 2 \times 2$  array of plutonium-metal billets moderated without reflectors.



**Fig. 3.B.6.5** Histogram of calculated  $k_{eff}$ 's for arrays of plutonium (19 cases).



**Fig. 3.B.7.1** Calculational model for a  $2 \times 2 \times 2$  array of  $\text{UO}_2(\text{NO}_3)_2$  aqueous solution with reflectors.



**Fig. 3.B.7.2** Histogram of calculated  $k_{\text{eff}}$ 's for cubic arrays of uranyl-nitrate cylinders (29 cases).

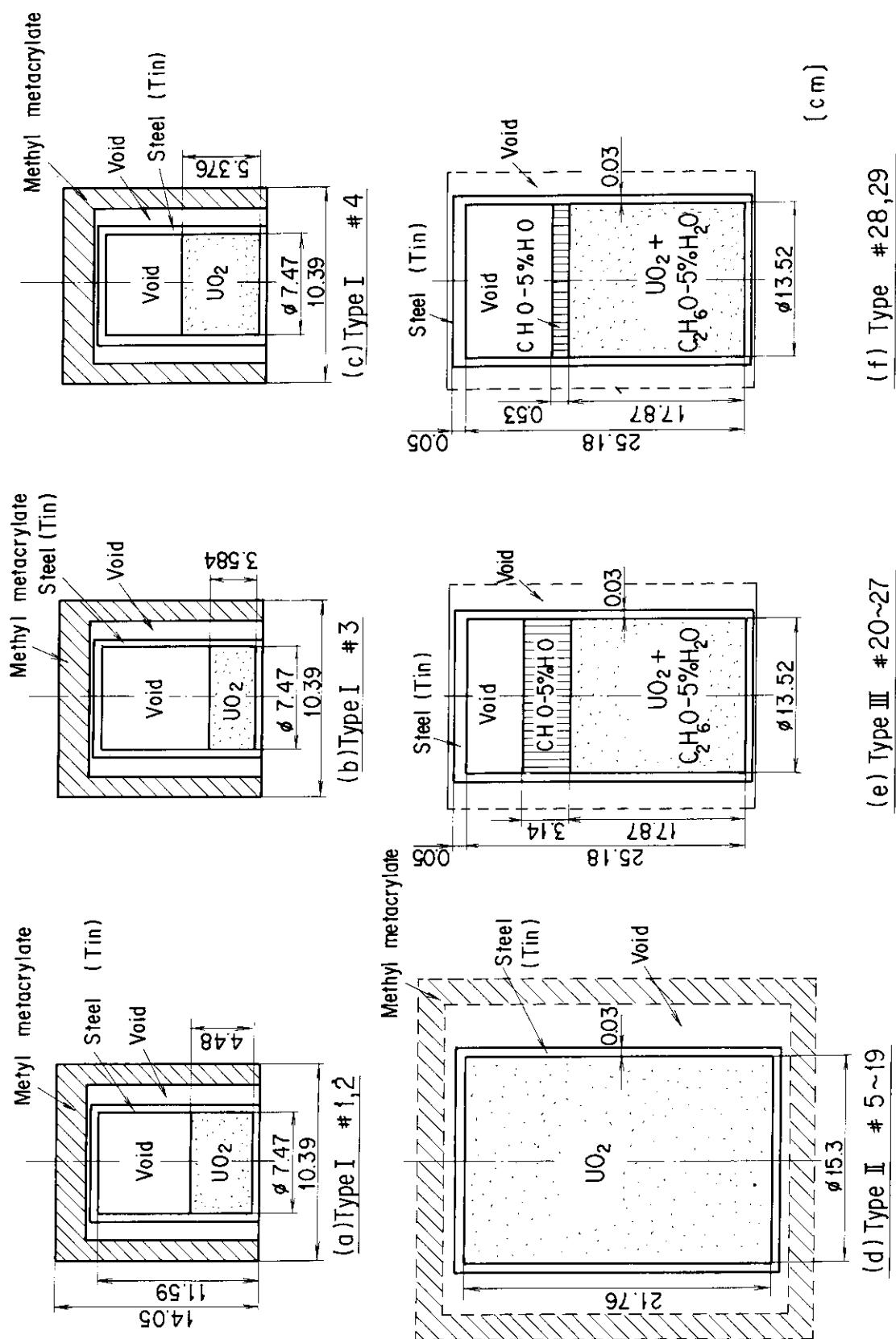
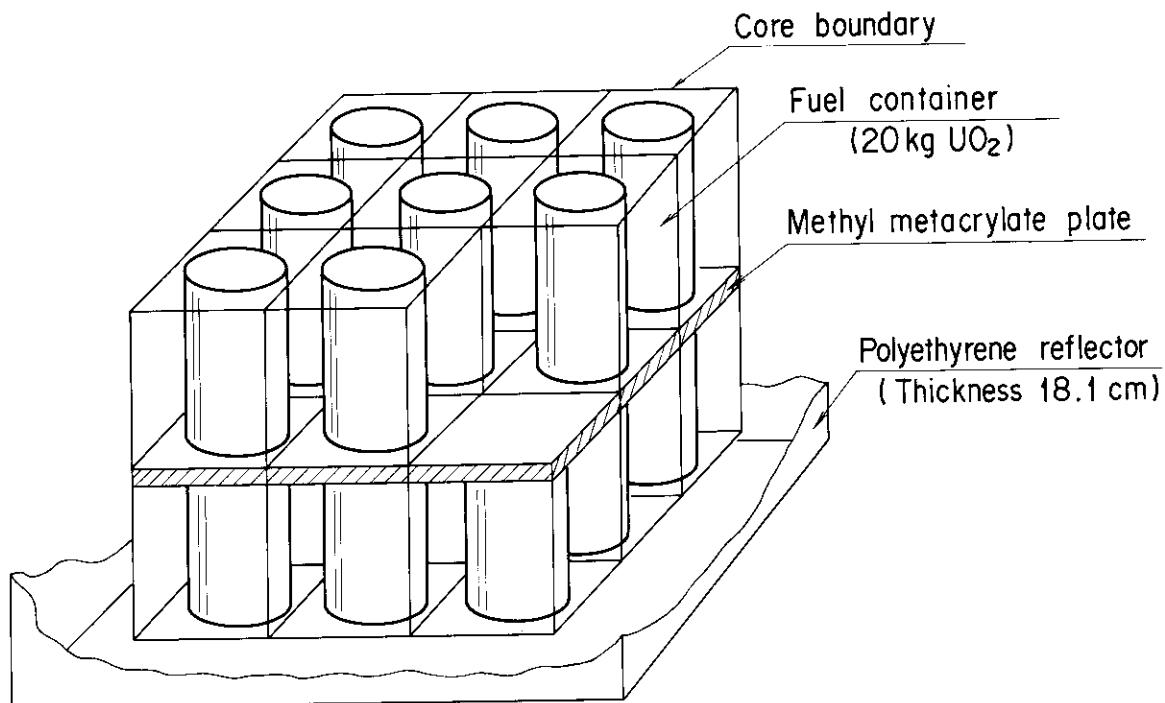
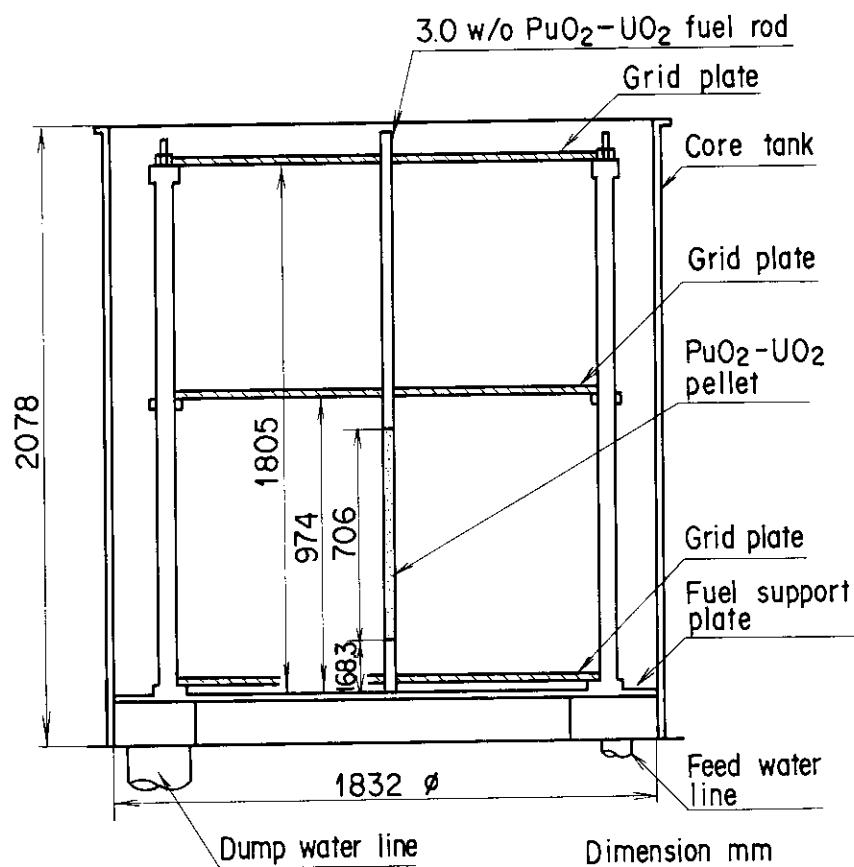


Fig. 3.B.8.1 Modeling of fuel unit.



**Fig. 3.B.8.2** Example of calculational model for array.



**Fig. 3.C.1.1** Vertical cross-sectional view of core tank.

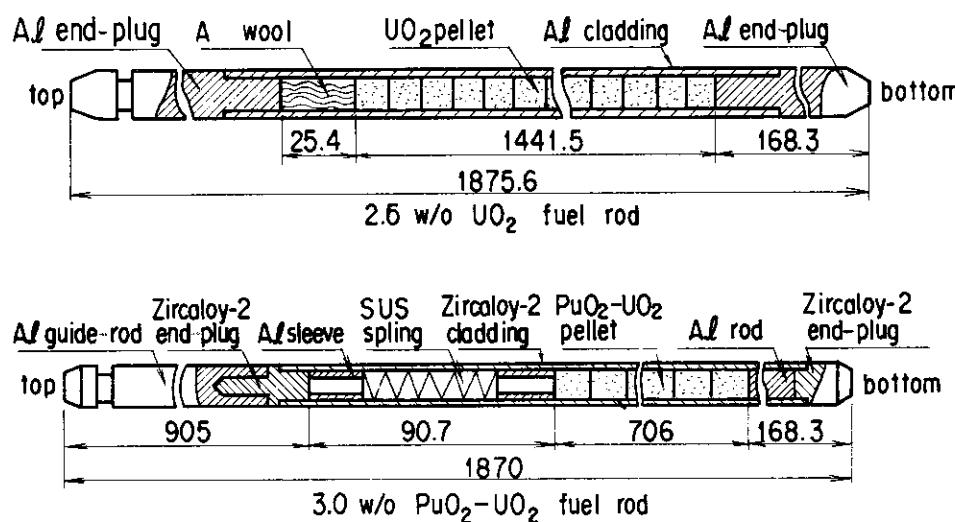


Fig. 3.C.1.2 2.6 wt%  $\text{UO}_2$  and 3.0 wt%  $\text{PuO}_2$ -natural  $\text{UO}_2$  fuel rods.

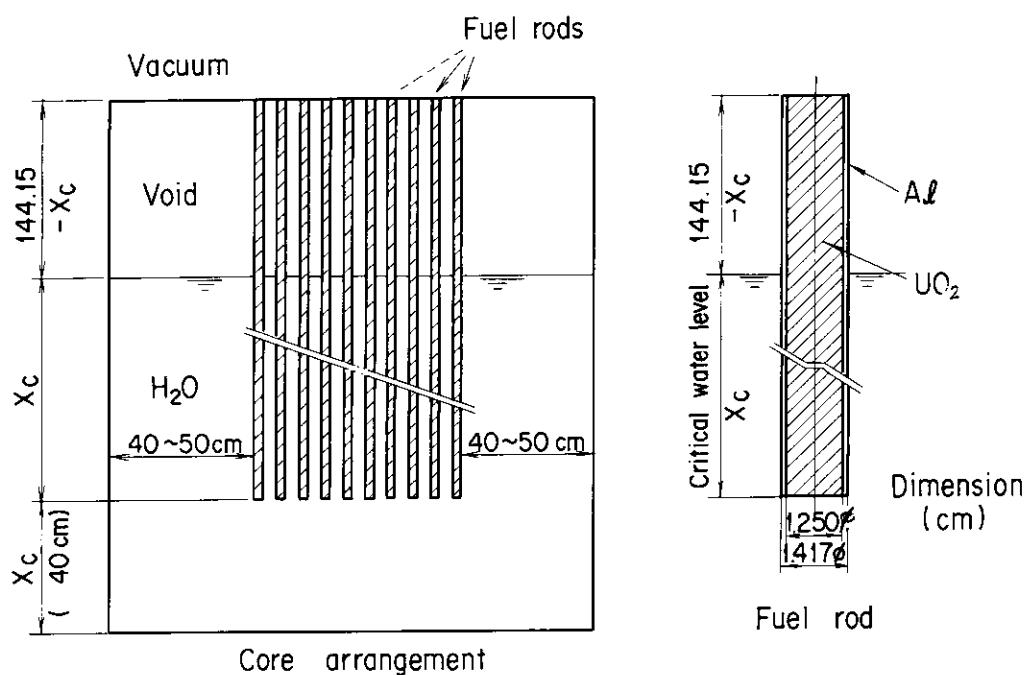
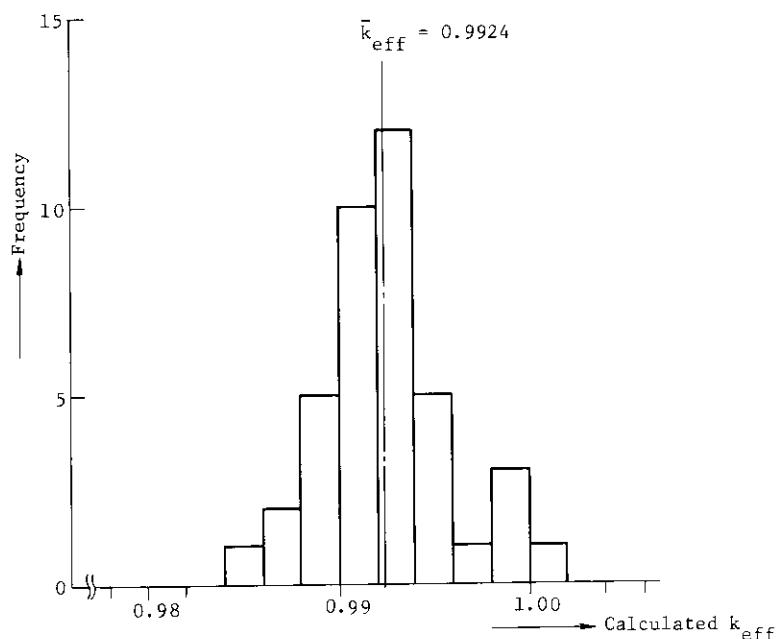
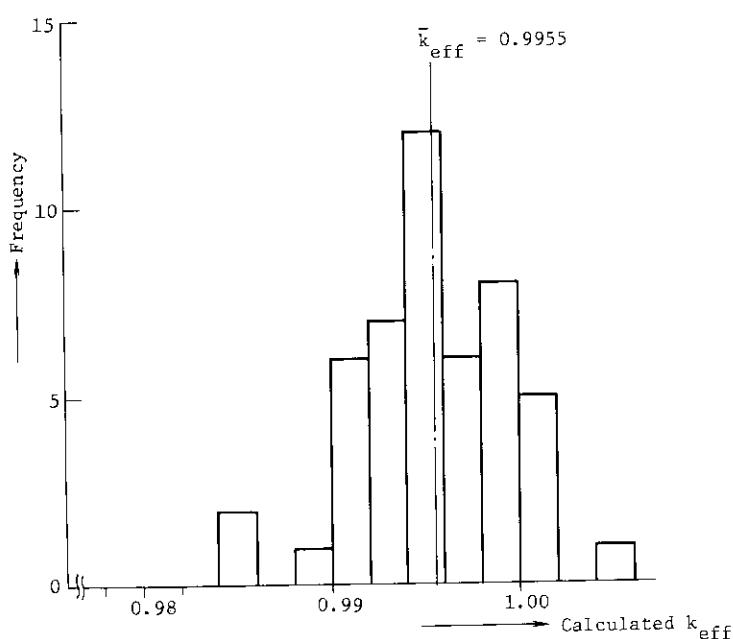


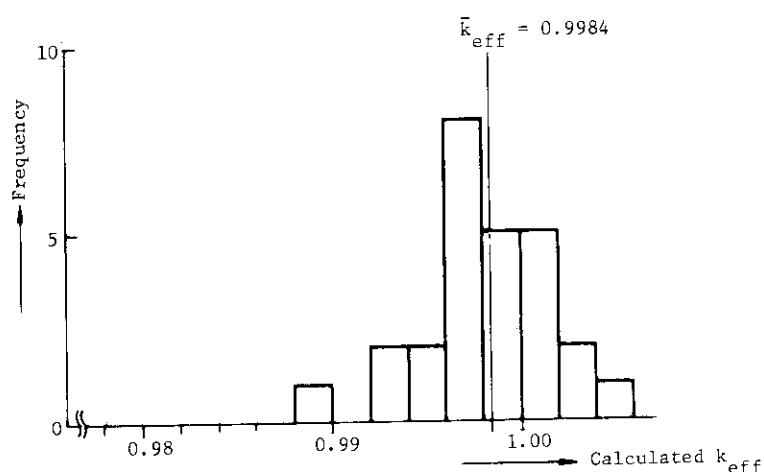
Fig. 3.C.1.3 Calculational model for  $\text{UO}_2$  fuel lattice.



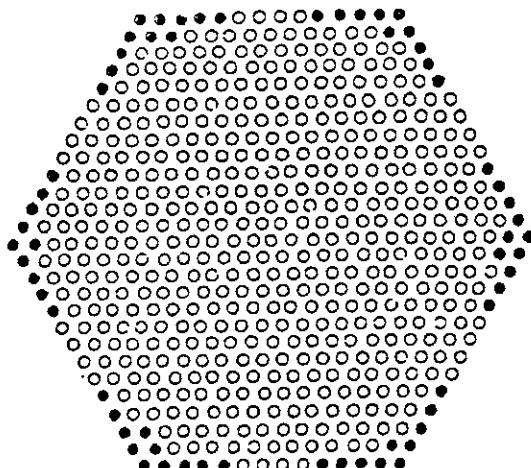
**Fig. 3.C.1.4**  
Histogram of calculated  $k_{eff}$ 's  
for  $\text{U}(2.6)\text{O}_2\text{-H}_2\text{O}$  systems.



**Fig. 3.C.1.5**  
Histogram of  $k_{eff}$ 's for  $\text{PuO}_2\text{-}$   
 $\text{U}(\text{Natural})\text{O}_2\text{-H}_2\text{O}$  systems.

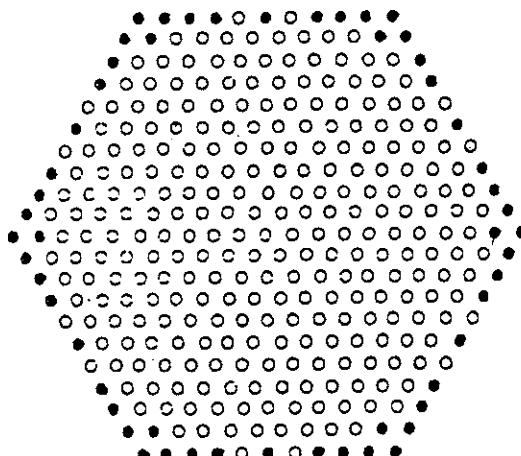


**Fig. 3.C.1.6**  
Histogram of  $k_{eff}$ 's for  $\text{U}$   
 $(2.6)\text{O}_2\text{-poison sheet-H}_2\text{O}$   
systems.



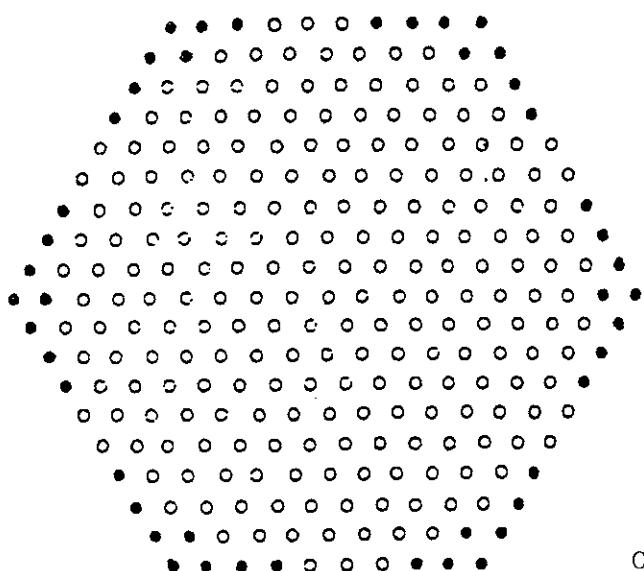
Case No. 11

Pitch = 13.5 mm  
 Rods = 484  
 Holes = 63  
 Critical water height = 85.21 cm



Case No. 12

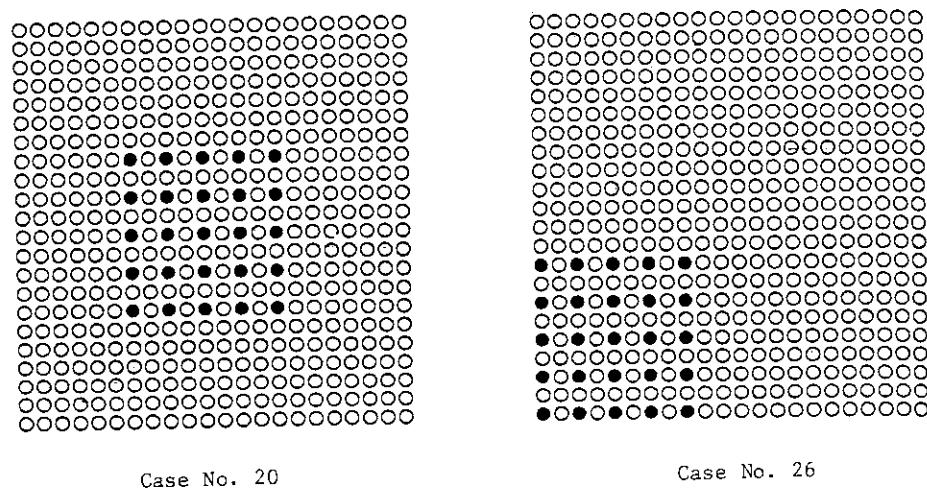
Pitch = 17.2 mm  
 Rods = 277  
 Holes = 54  
 Critical water height = 61.99 cm



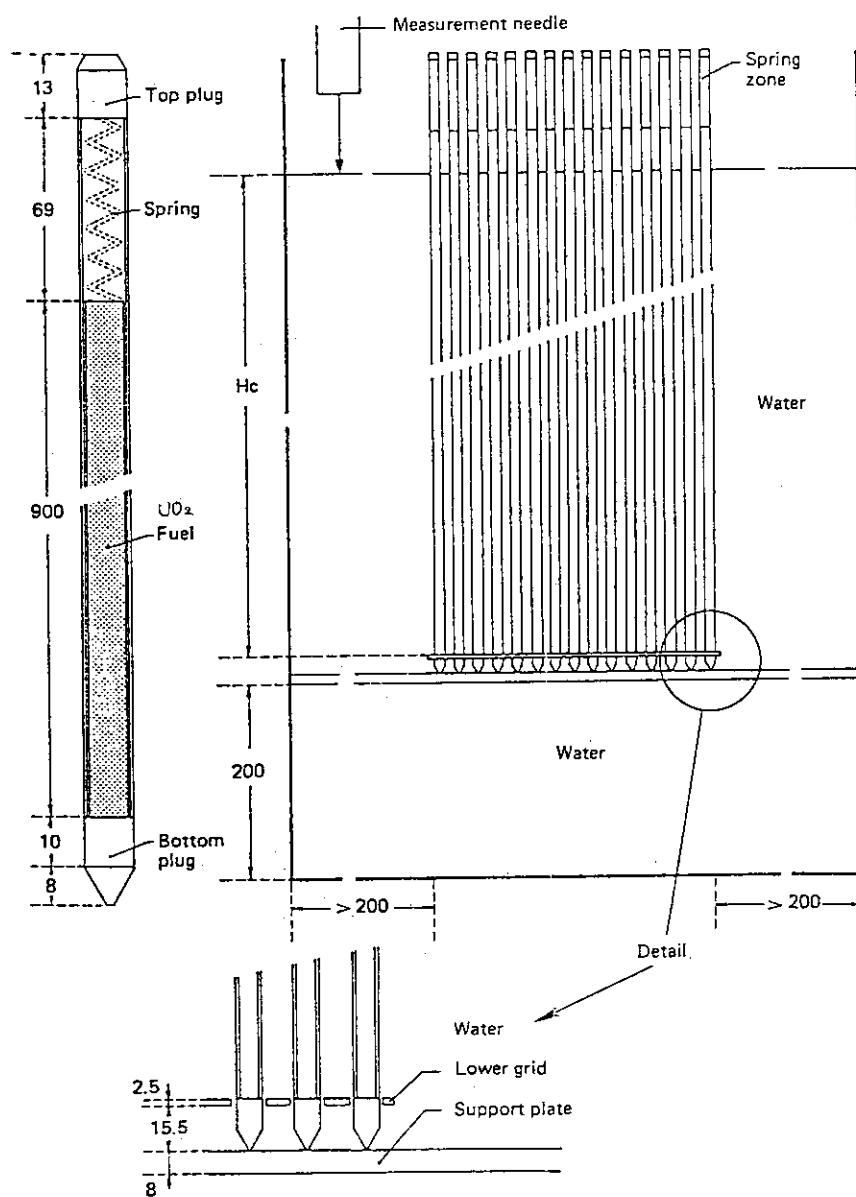
Case No. 13

Pitch = 22.6 mm  
 Rods = 225  
 Holes = 48  
 Critical water height = 70.44 cm

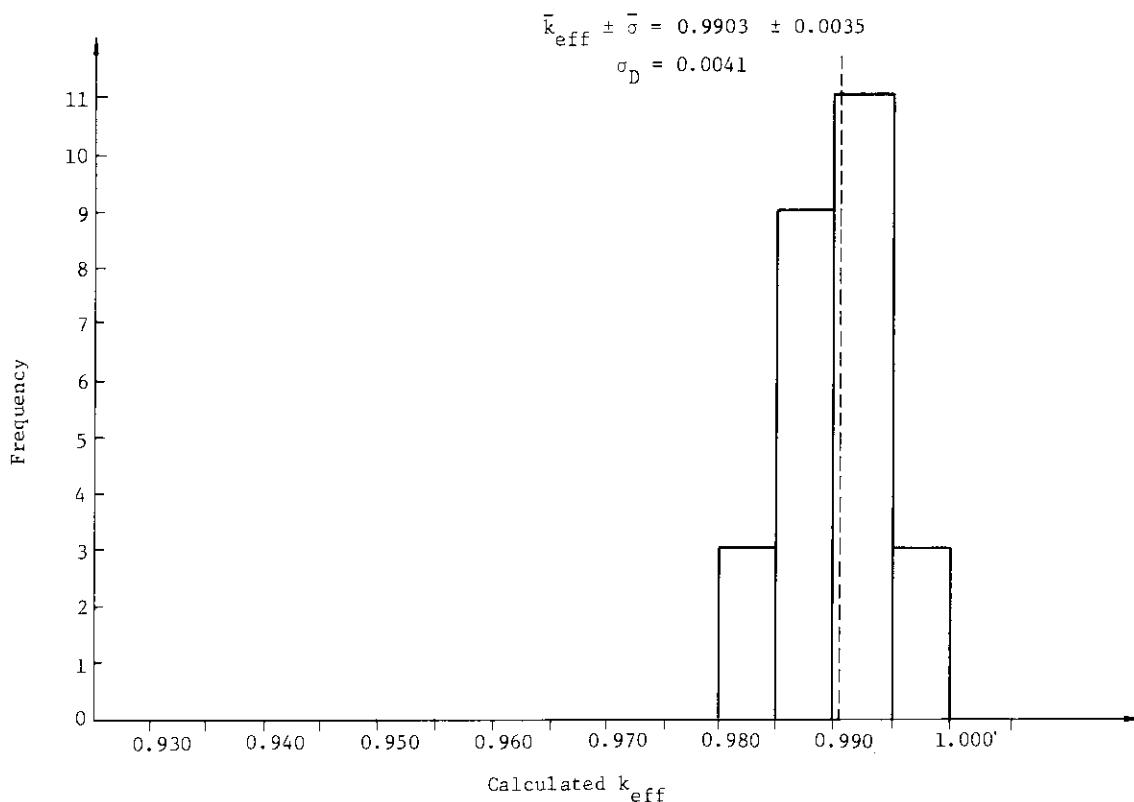
**Fig. 3.C.2.1** Calculational models for representative cases of  $\text{UO}_2$  rod hexagonal lattices.  
Solid circle indicates the removal of rods.



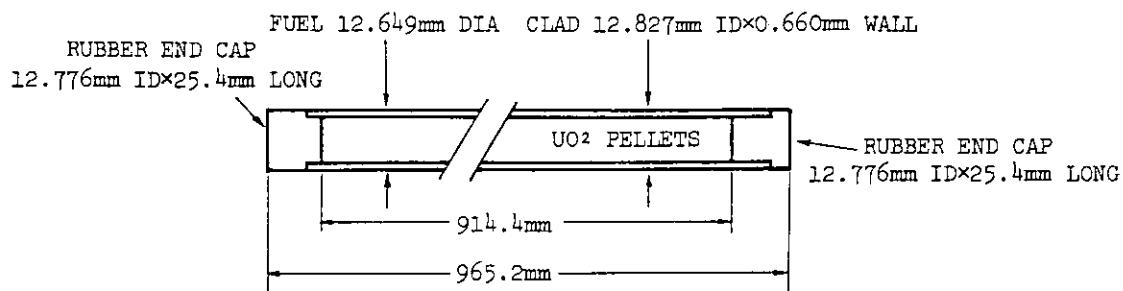
**Fig. 3.C.2.2** Calculational models for representative cases of  $\text{UO}_2$  rod square lattices. Solid circle indicates the removal of rods.



**Fig. 3.C.2.3** Experimental arrangement of  $\text{UO}_2$  rod clusters partially immersed in water.



**Fig. 3.C.2.4** Histogram of calculated  $k_{\text{eff}}$ 's for  $\text{UO}_2$  rod clusters partially immersed in water (26 cases).



CLADDING: 6061 ALUMINUM TUBING

LOADING:

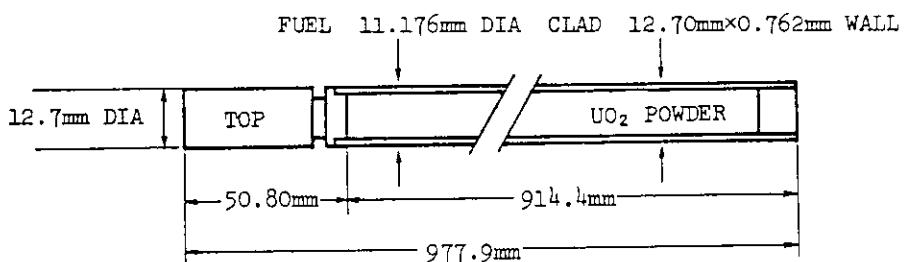
ENRICHMENT -  $4.289 \pm 0.006$  wt%  $^{235}\text{U}$

FUEL DENSITY -  $94.9 \pm 0.55\%$  OF THEORETICAL DENSITY

URANIUM ASSAY -  $88.055 \pm 0.261$  wt% OF TOTAL FUEL COMPOSITION

$\text{UO}_2$  -  $1203.33 \pm 4.12$  g/ROD

**Fig. 3.C.3.1** 4.29 wt%  $^{235}\text{U}$  enriched  $\text{UO}_2$  rods.



CLADDING: 6061 ALUMINUM TUBING SEAL WELDED WITH A LOWER END PLUG OF 5052-H32 ALUMINUM AND A TOP PLUG OF 1100 ALUMINUM

LOADING:

ENRICHMENT -  $2.35 \pm 0.05\text{wt\%}$   $^{235}\text{U}$

FUEL DENSITY - 9.20mg/mm<sup>3</sup> (84% THEORETICAL DENSITY)

URANIUM ASSAY - 88.0wt%

$\text{UO}_2$  - 825g/ROD (AVERAGE)

Fig. 3.C.3.2 2.35 wt%  $^{235}\text{U}$  enriched  $\text{UO}_2$  rods.

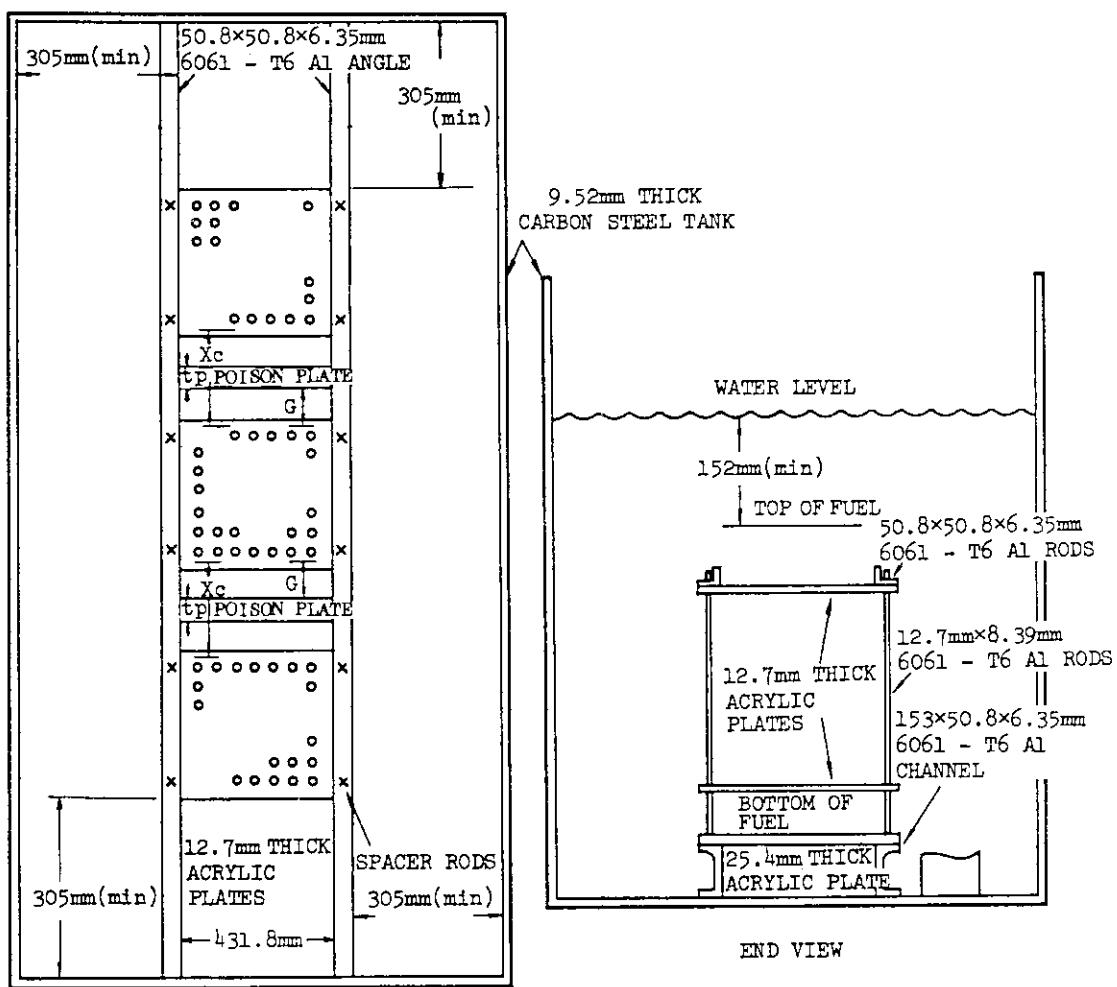
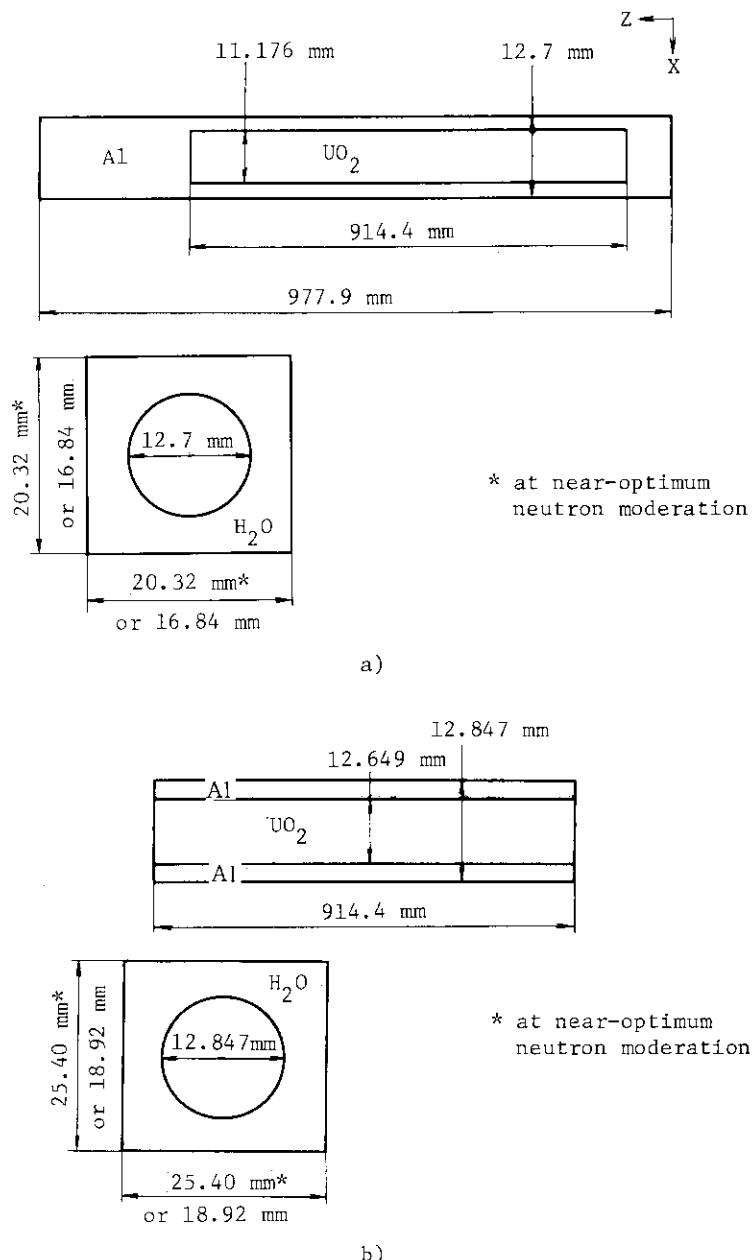
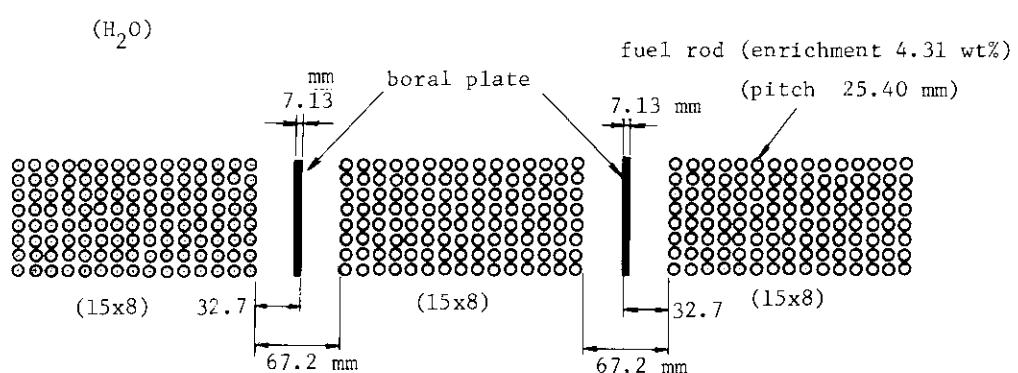
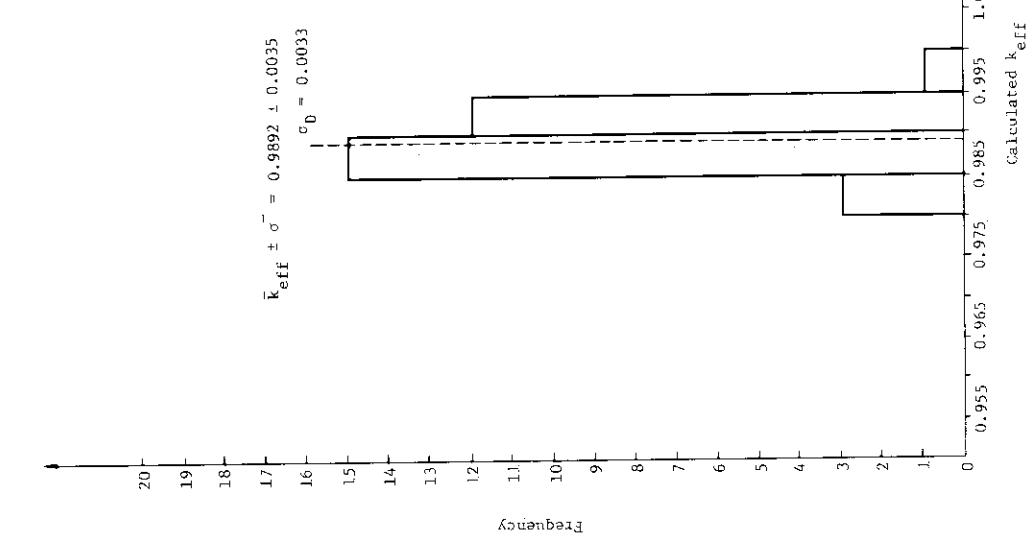


Fig. 3.C.3.3 Experimental arrangement of simulated shipping containers with poison plates.

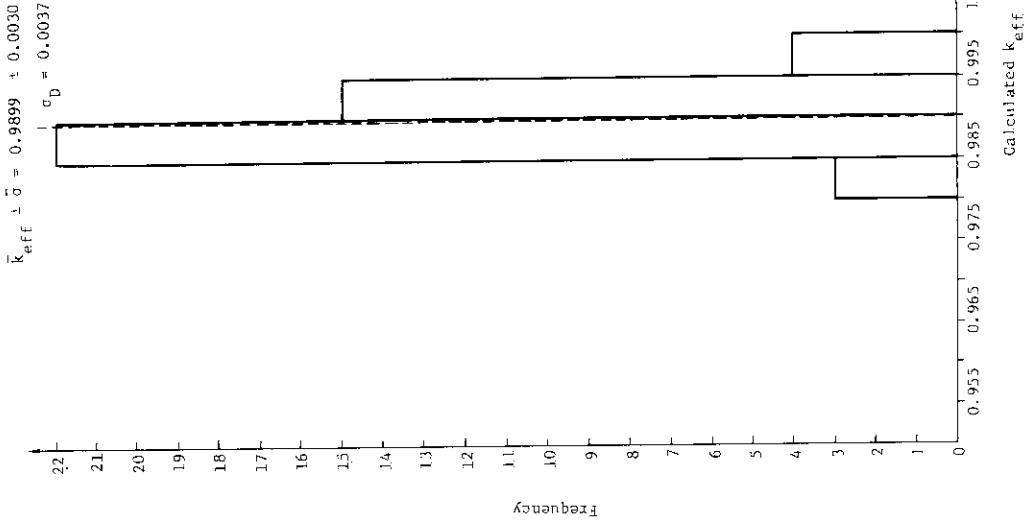
**Fig. 3.C.3.4** Calculational model for UO<sub>2</sub> lattice cell.

- a) 2.35 wt% <sup>235</sup>U fuel rod cell model.  
b) 4.31 wt% <sup>235</sup>U fuel rod cell model.

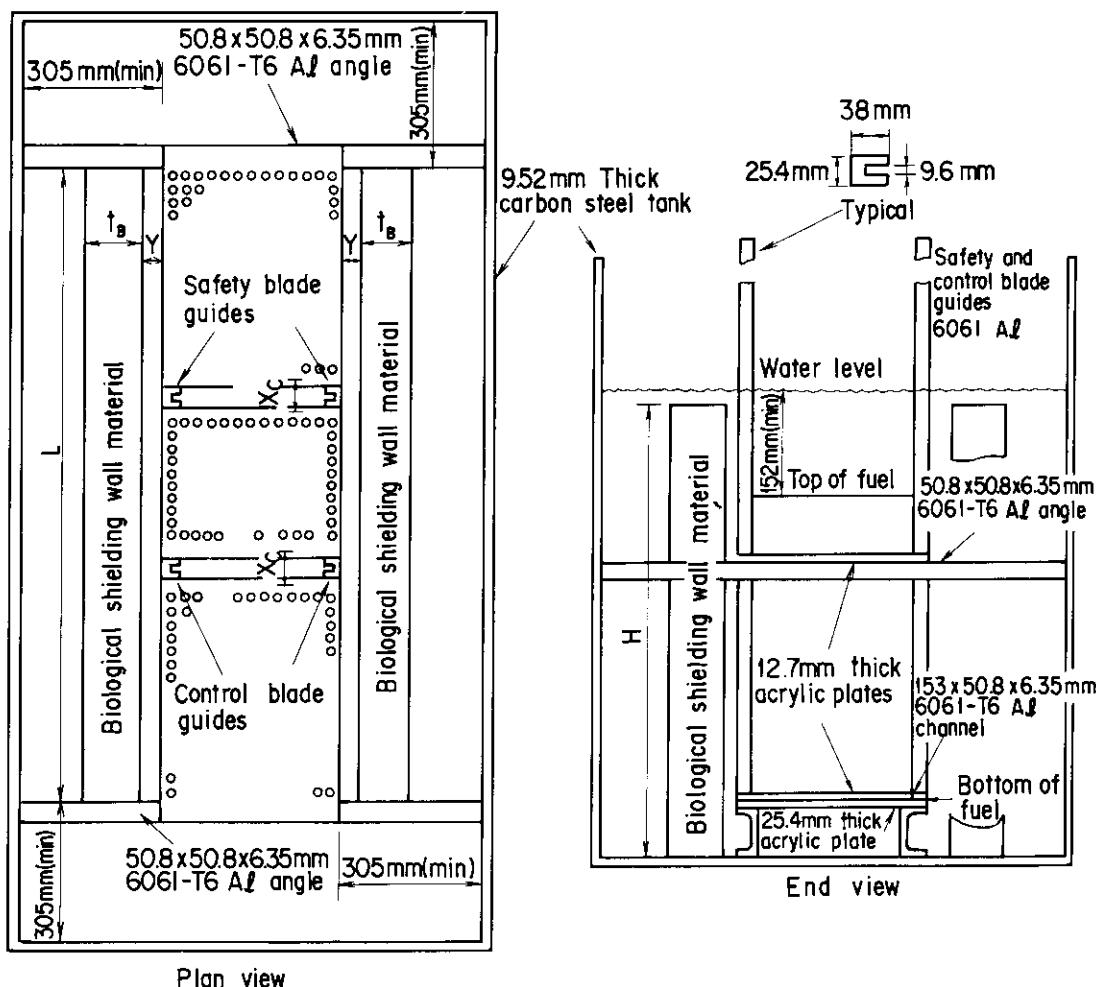
**Fig. 3.C.3.5** A typical KENO-IV calculational model for UO<sub>2</sub> rod clusters submerged in water with boral poison plates.



**Fig. 3.C.3.6** Histogram of calculated  $k_{eff}$ 's for  $UO_2$  rod clusters ( $^{235}U$  enrichment: 4.31 wt%, lattice pitch: 25.40 mm) in water with various poison plates (31 cases).

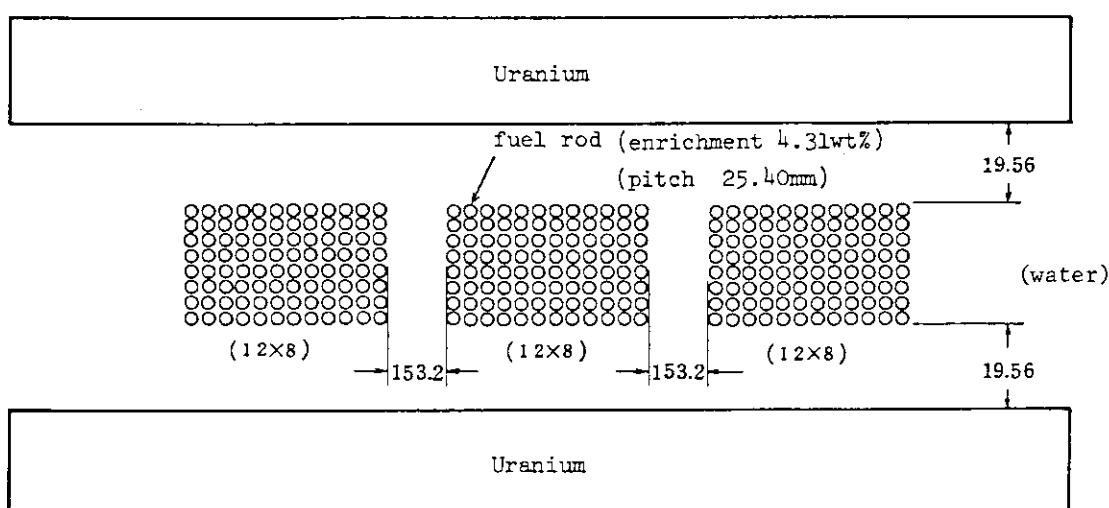


**Fig. 3.C.3.7** Histogram of calculated  $k_{eff}$ 's for  $UO_2$  rod clusters ( $^{235}U$  enrichment: 2.35 wt%, lattice pitch 20.32 mm) in water with various poison plates (44 cases).

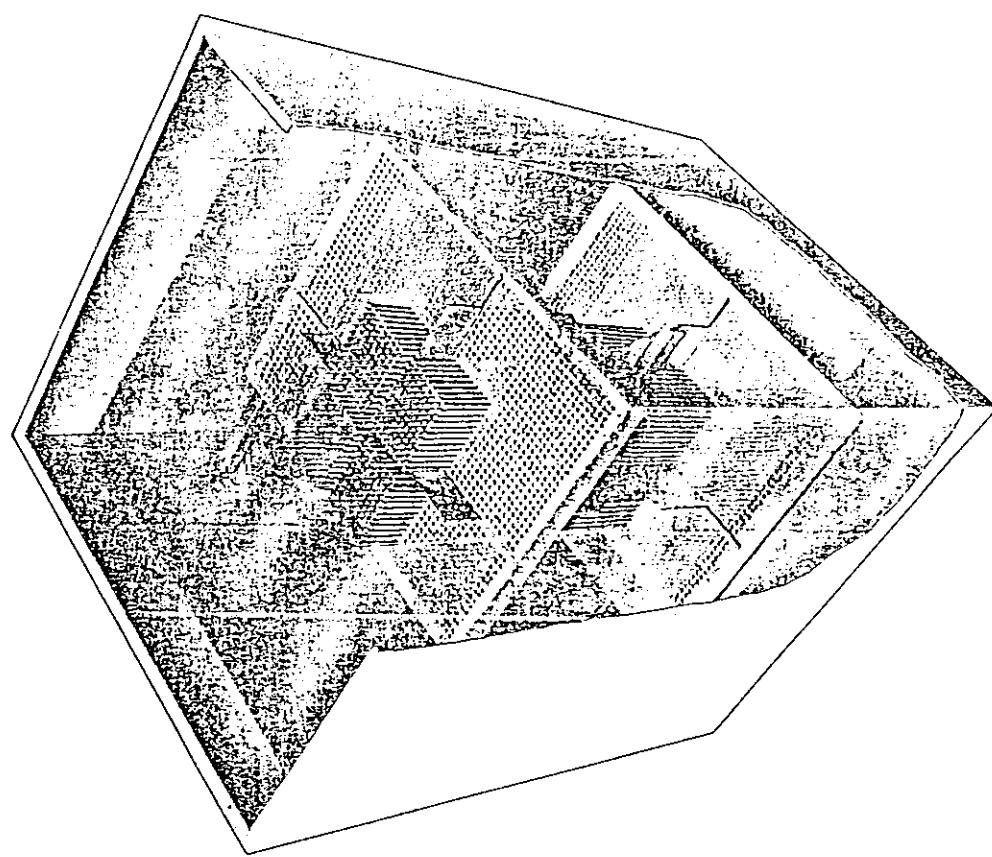


**Fig. 3.C.4.1** Graphical arrangement of simulated shipping container critical experiments.

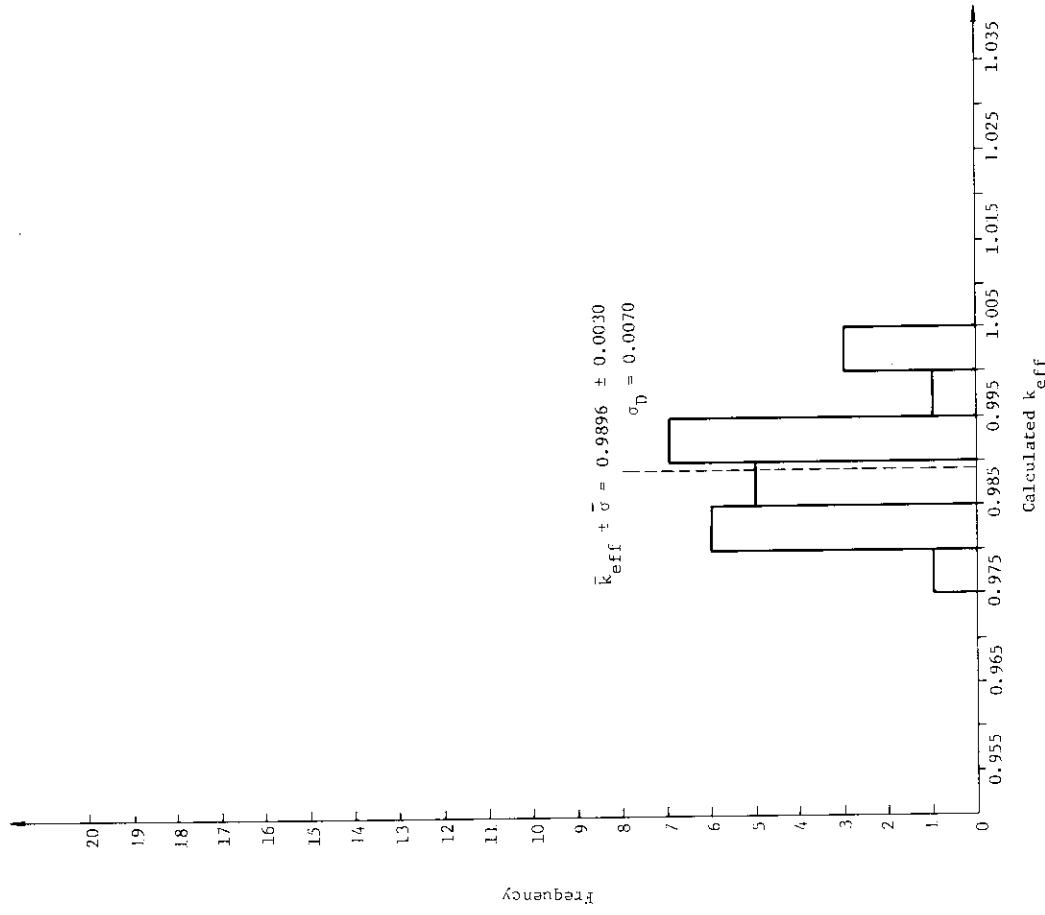
(H<sub>2</sub>O)



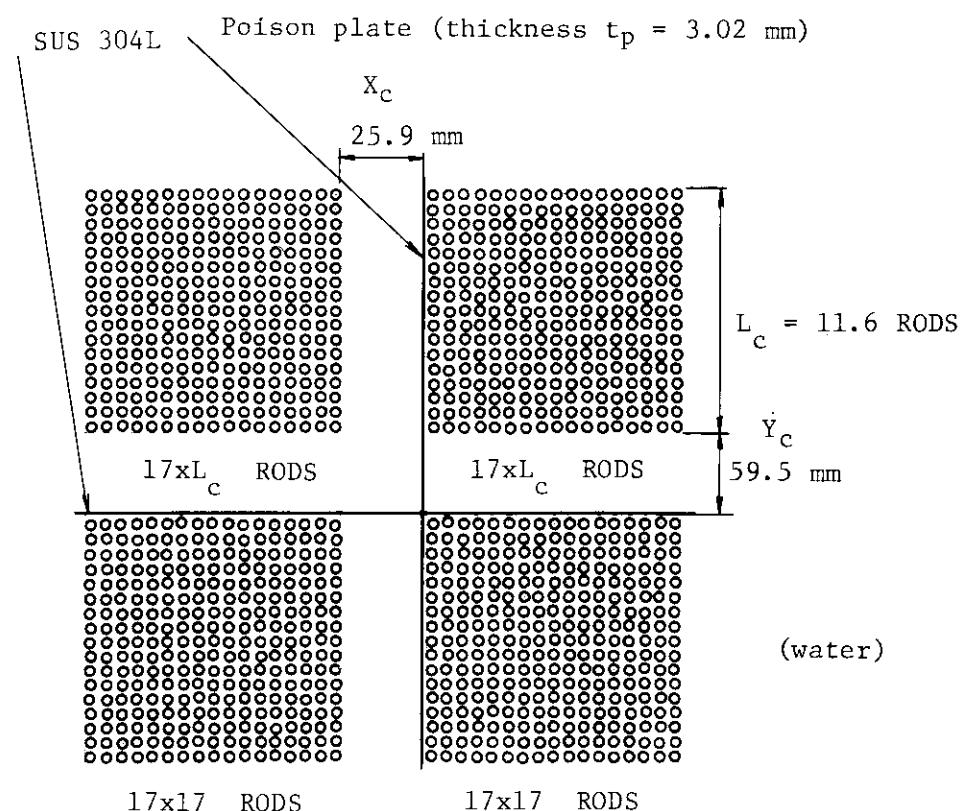
**Fig. 3.C.4.2** A typical KENO-IV calculational model for UO<sub>2</sub> rod clusters submerged in water with reflecting walls.



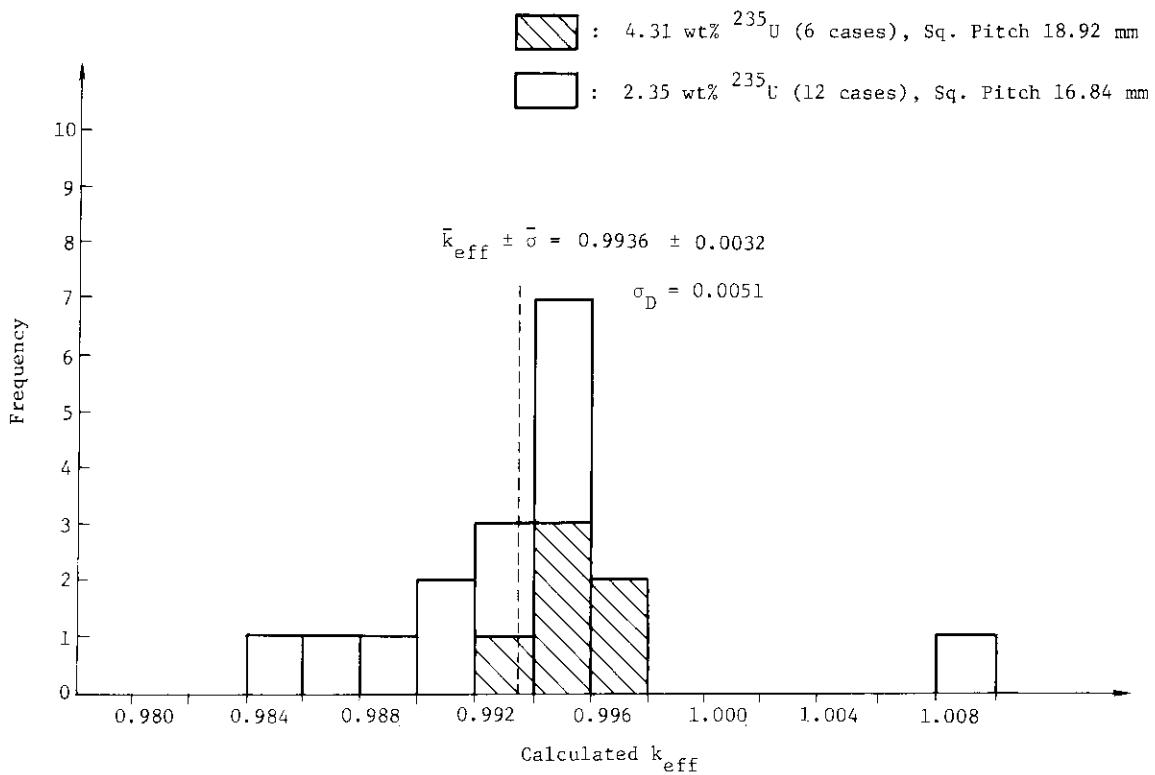
**Fig. 3.C.5.1** Bird's eye view of typical four cross arranged clusters in water pool.



**Fig. 3.C.4.3** Histogram of calculated  $k_{eff}$ 's for  $UO_2$  rod clusters submerged in water with reflecting walls (23 cases).



**Fig. 3.C.5.2** Typical calculational model for four cross arranged clusters (Case No.100).



**Fig. 3.C.5.3** Histogram of calculated  $k_{eff}$ 's for three lined clusters (18 cases).

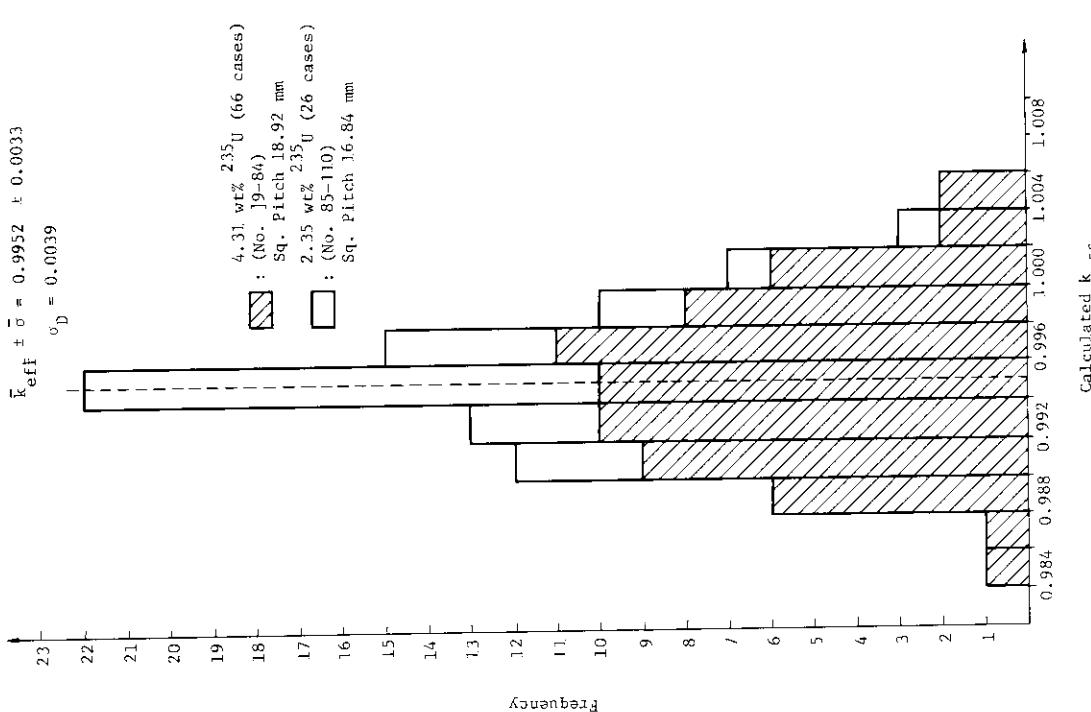


Fig. 3.C.5.4 Histogram of calculated  $k_{eff}$ 's for four cross arranged clusters (92 cases).

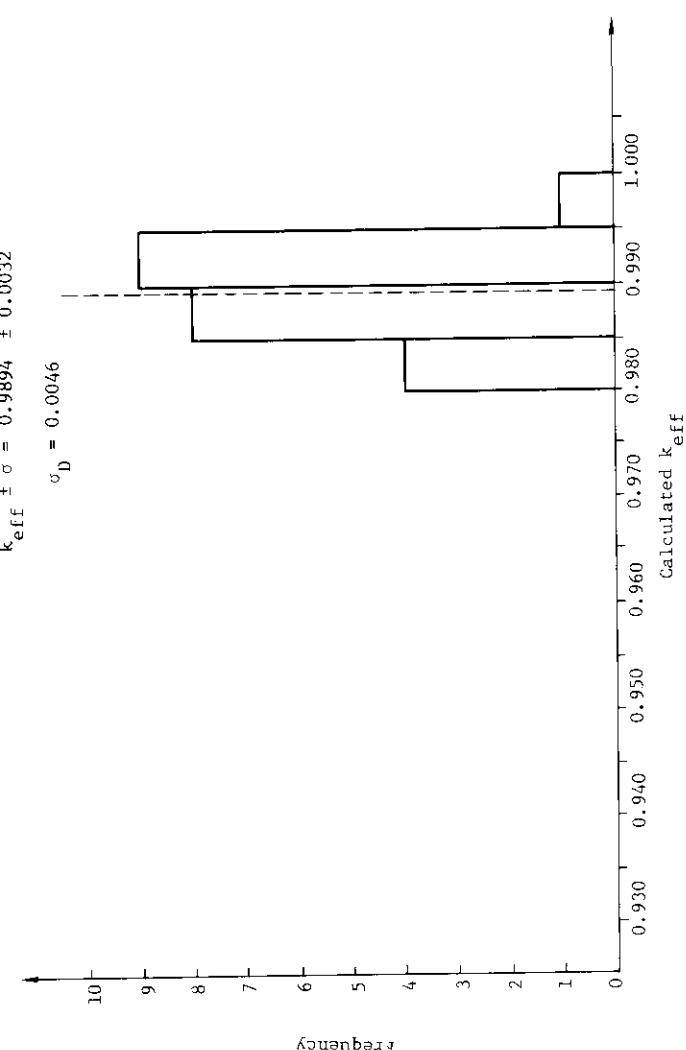
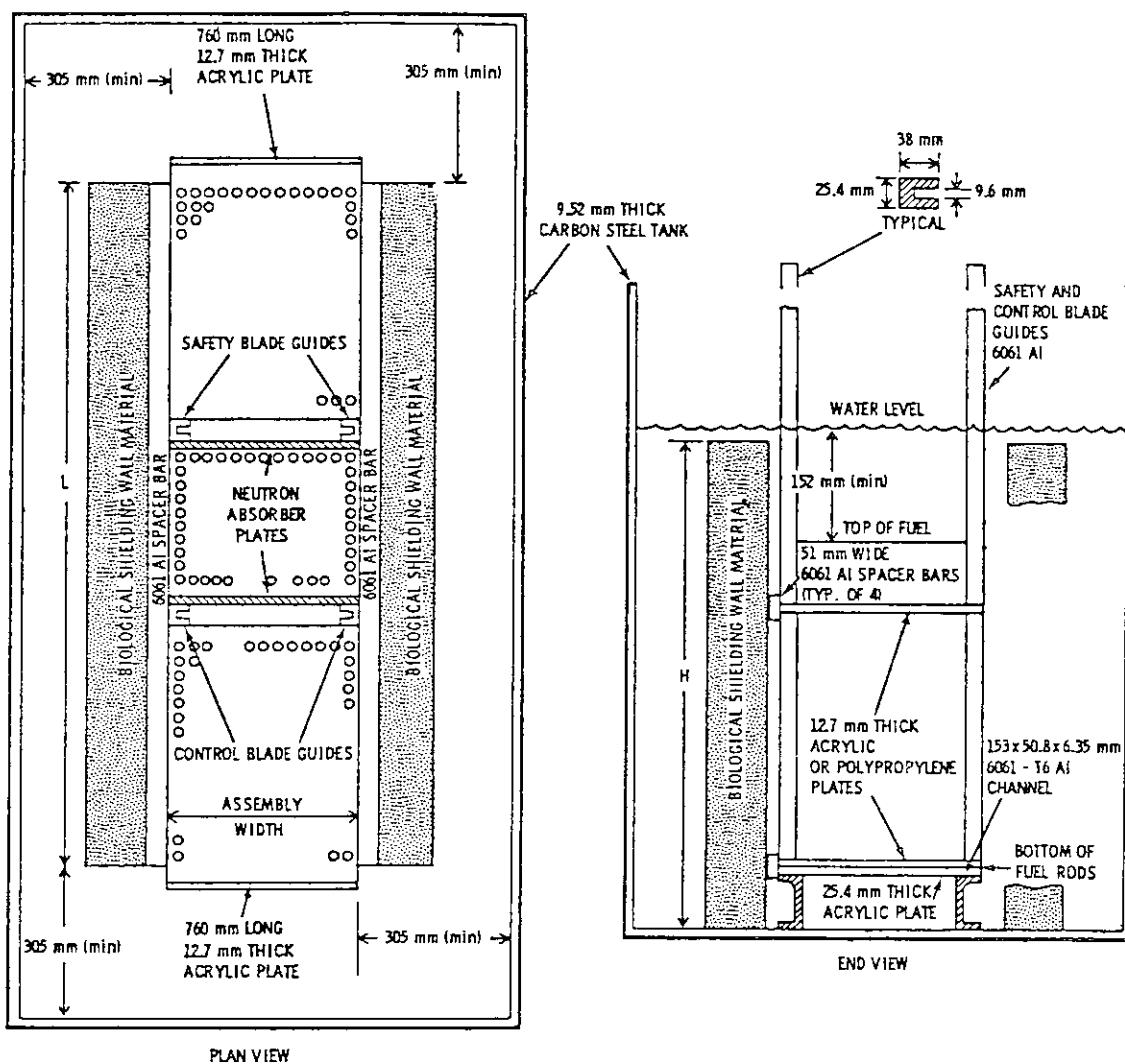
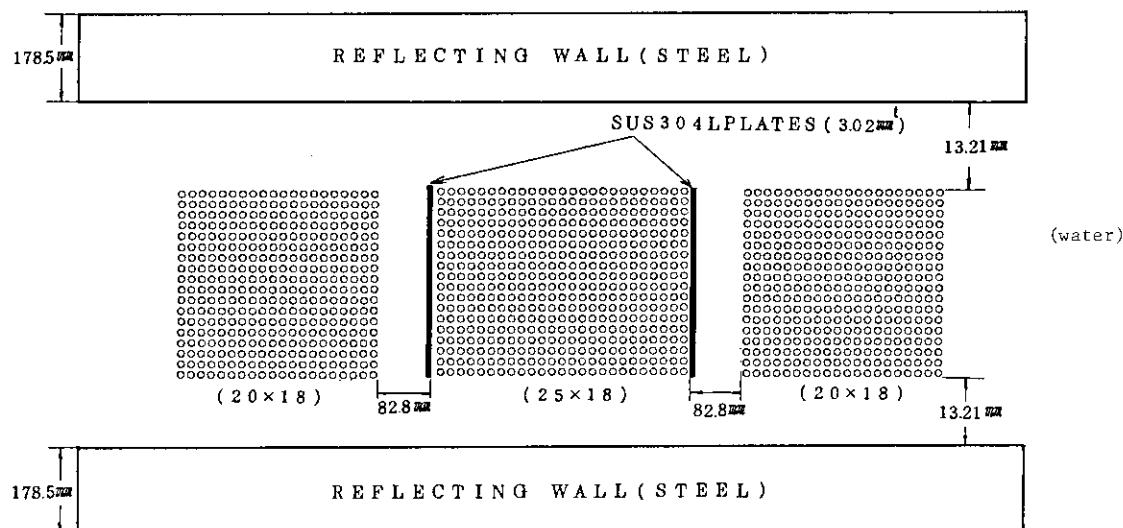


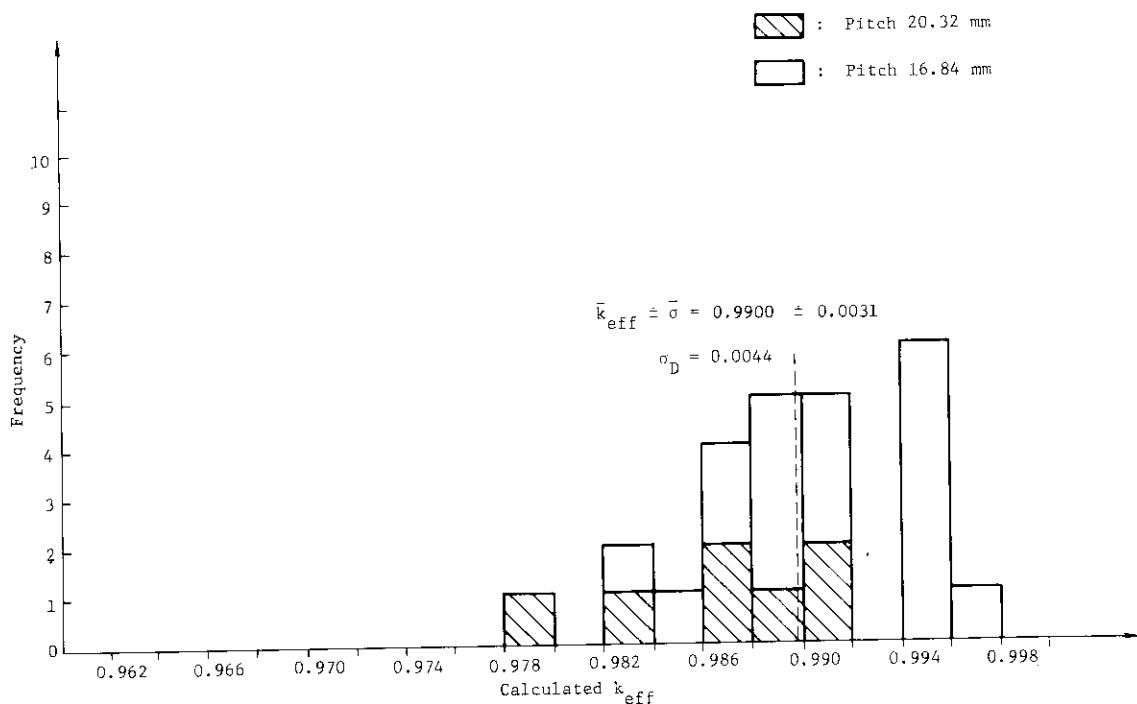
Fig. 3.C.6.1 Histogram of calculated  $k_{eff}$ 's for three aligned  $\text{UO}_2$  rod clusters in water with side reflectors.



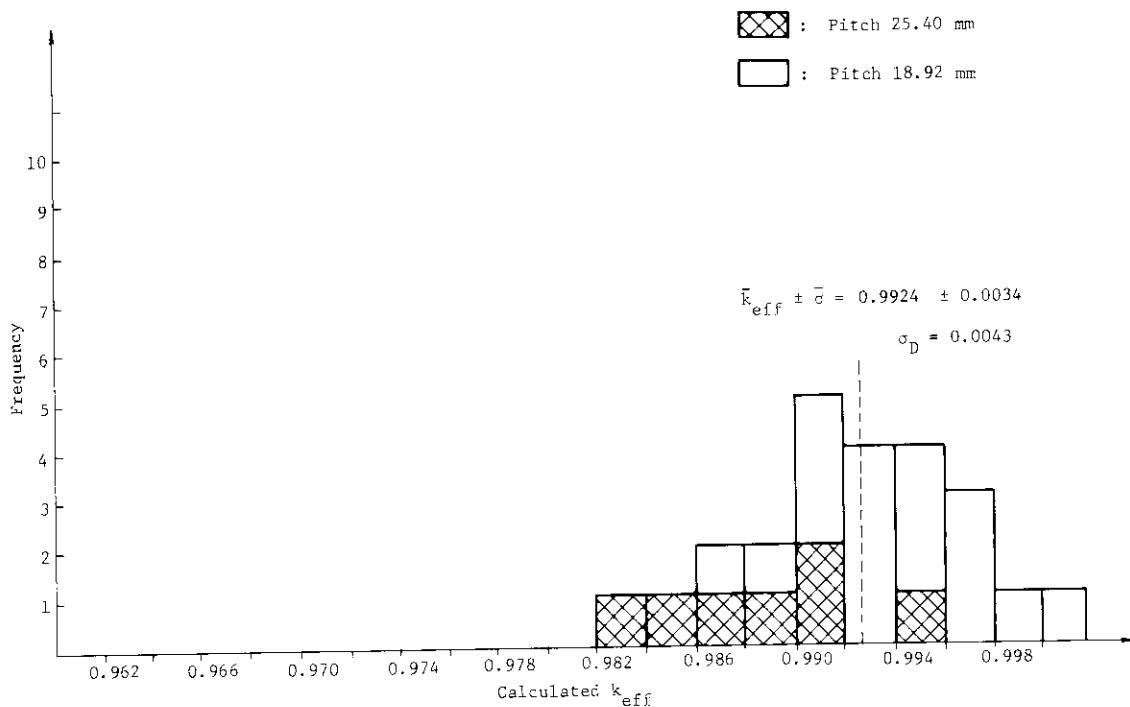
**Fig. 3.C.7.1** Experimental arrangement of three  $\text{UO}_2$  rod clusters in water with absorbers and reflecting walls.



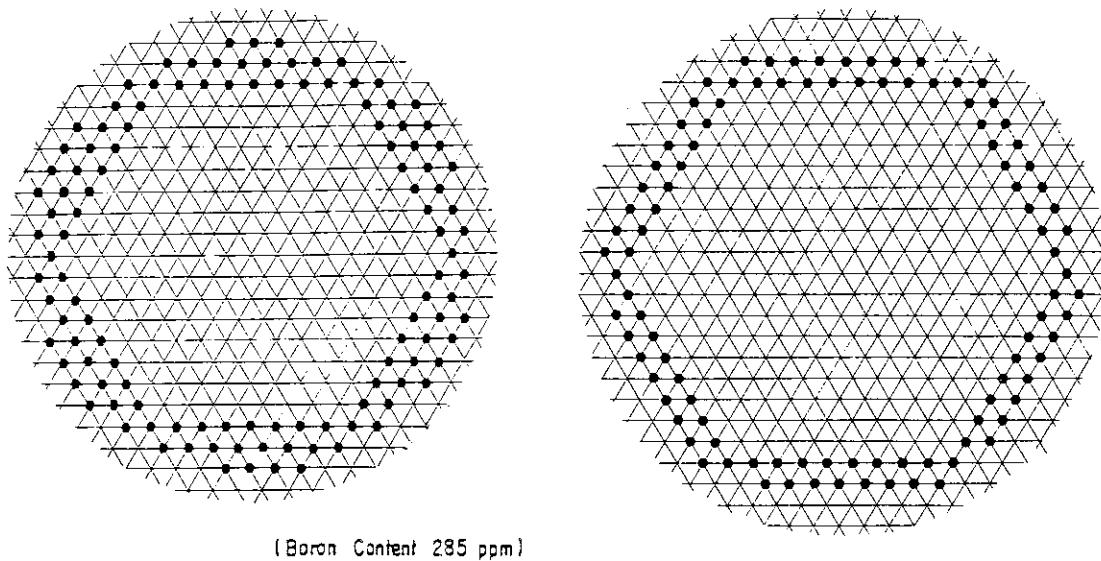
**Fig. 3.C.7.2** Calculational model for three  $\text{UO}_2$  rod clusters in water with absorbers and reflecting walls.



**Fig. 3.C.7.3** Histogram of calculated  $k_{eff}$ 's for three  $UO_2$  rod clusters ( $^{235}U$  enrichment: 2.35 wt%) in water with neutron absorbers and reflecting walls (25 cases).

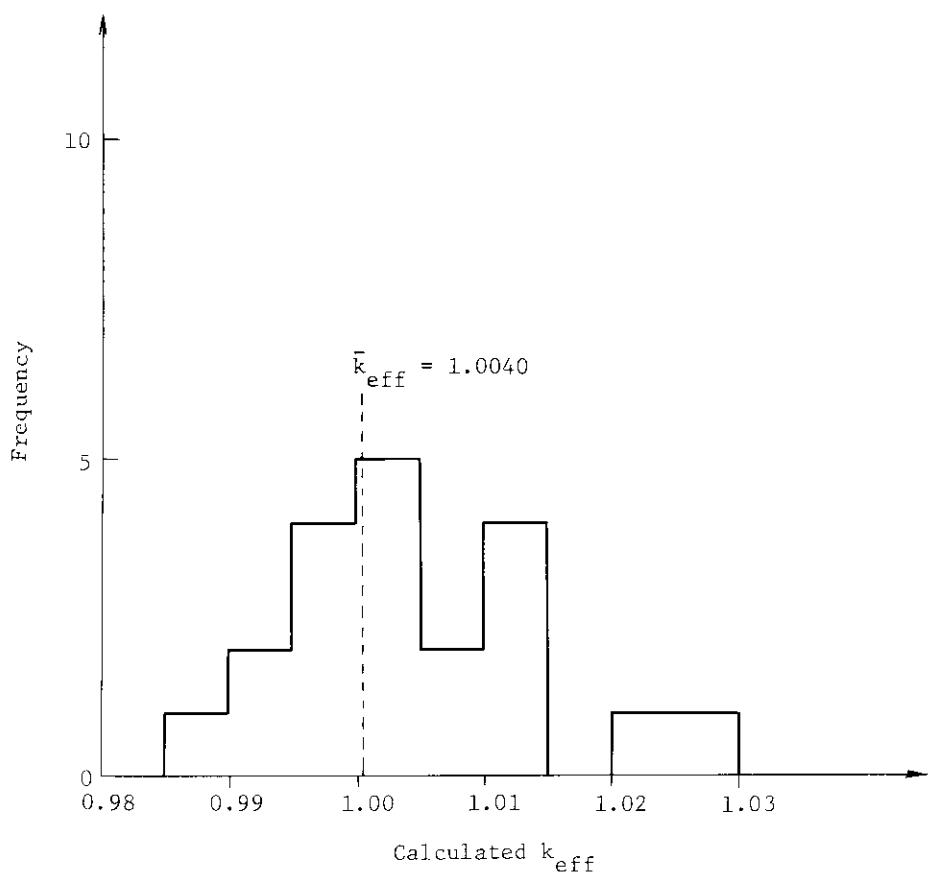


**Fig. 3.C.7.4** Histogram of calculated  $k_{eff}$ 's for three  $UO_2$  rod clusters ( $^{235}U$  enrichment: 4.31 wt%) in water with neutron absorbers and reflecting walls (24 cases).

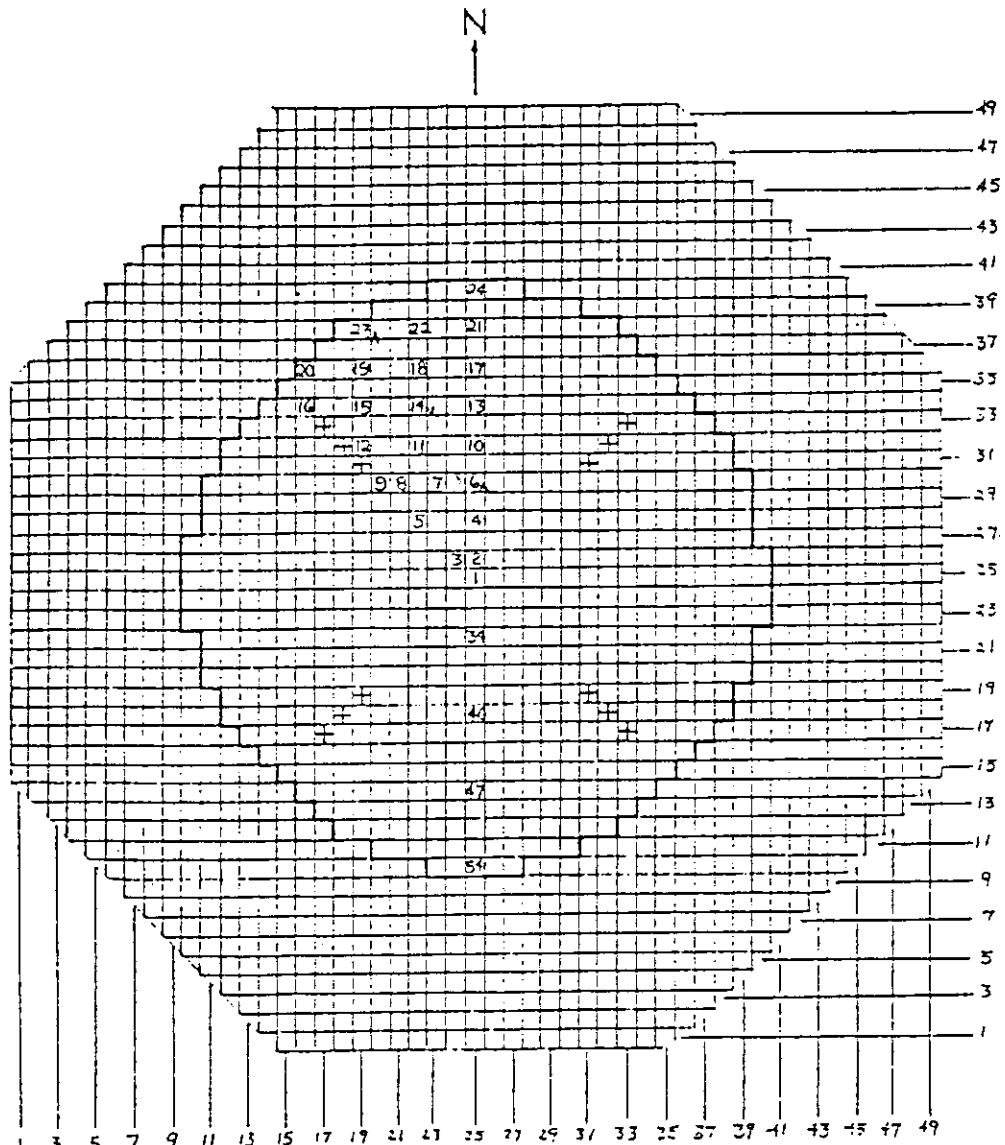


**Fig. 3.C.8.1** Calculational model for B.M. No.26,  
 $N_c=288$ .

**Fig. 3.C.8.2** Calculational model for B.M. No.27,  
 $N_c=320$ .



**Fig. 3.C.8.3** Histogram of calculated  $k_{eff}$ 's for the heterogeneous plutonium benchmark experiments (20 cases).



0.615-INCH  $\text{UO}_2$  CORE, UNBORATED, GAMMA SCAN IRRADIATION

Top Reflector thickness 6.0 inches. Moderator Temp. 21.90 °C

Excess Reactivity 6.88 cents. Boron concentration, < 1 wppm.

Number and Type of Fuel Rods in Core: Top Reflector thickness during  
 $\text{UO}_2$ -2.35%  $^{235}\text{U}$  708 Irradiation 6.0 inches.

A=Axial Scan D=Decay Rod

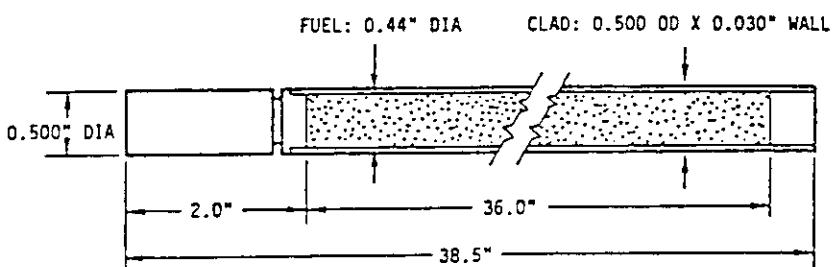
$\beta_{\text{eff.}} = \underline{6.99} \times 10^{-3}$  N=Normalization Rod

Fig. 3.C.9.1 Example of fuel pins loading pattern.

FUEL SPECIFICATIONS: 2.35% ENRICHED UO<sub>2</sub>

FUEL RODS

1. ROD DIMENSIONS



2. CLADDING: 6061 ALUMINUM TUBING SEAL WELDED WITH A LOWER END PLUG OF 5052-H32 ALUMINUM AND A TOP PLUG OF 1100 ALUMINUM.

3. TOTAL WEIGHT OF LOADED FUEL RODS: 917 gm (AVERAGE)

FUEL LOADING

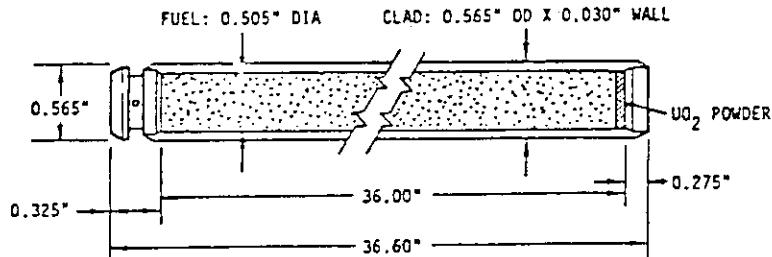
1. FUEL MIXTURE VIBRATIONALLY COMPACTED.
2. 825 gm OF UO<sub>2</sub> POWDER/ROD, 726 gm OF U/ROD, 17.08 gm OF U-235/ROD.
3. ENRICHMENT - 2.35 ± 0.05 w/o U-235.
4. FUEL DENSITY - 9.20 gm/cm<sup>3</sup> (84% THEORETICAL DENSITY).

Fig. 3.C.9.2 UO<sub>2</sub>-2.35 wt% <sup>235</sup>U fuel description.

FUEL SPECIFICATIONS: UO<sub>2</sub> - 2 WT% PuO<sub>2</sub>

FUEL RODS

1. ROD DIMENSIONS



2. CLADDING: ZIRCALOY-2 TUBING WITH PLUGS SEAL WELDED AT BOTH ENDS.

3. TOTAL WEIGHT OF LOADED FUEL RODS: 1340 gms (AVERAGE)

FUEL LOADINGS

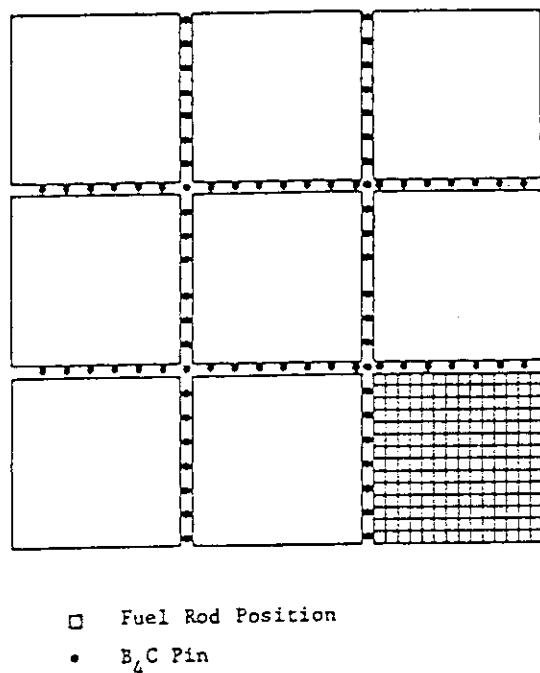
1. PuO<sub>2</sub> MIXED IN NATURAL UO<sub>2</sub> AND VIBRATIONALLY COMPACTED.
2. 1128 gms OF UO<sub>2</sub>-PuO<sub>2</sub> MIX/ROD.
3. CHEMICAL COMPOSITION WT%: Pu/PuO<sub>2</sub> = 88.1 U/UO<sub>2</sub> = 88.0 Pu/MIX = 1.760.
4. PuO<sub>2</sub> IS 2.00 WT% OF TOTAL MIXTURE.
5. FUEL DENSITY - 9.54 gm/cc (~87% THEORETICAL DENSITY).
6. UO<sub>2</sub> POWDER AT THE END OF FUEL COLUMN.
7. THE ISOTOPIC DISTRIBUTION OF PLUTONIUM

8% (NOMINAL) <sup>240</sup>Pu  
ATOM PERCENT

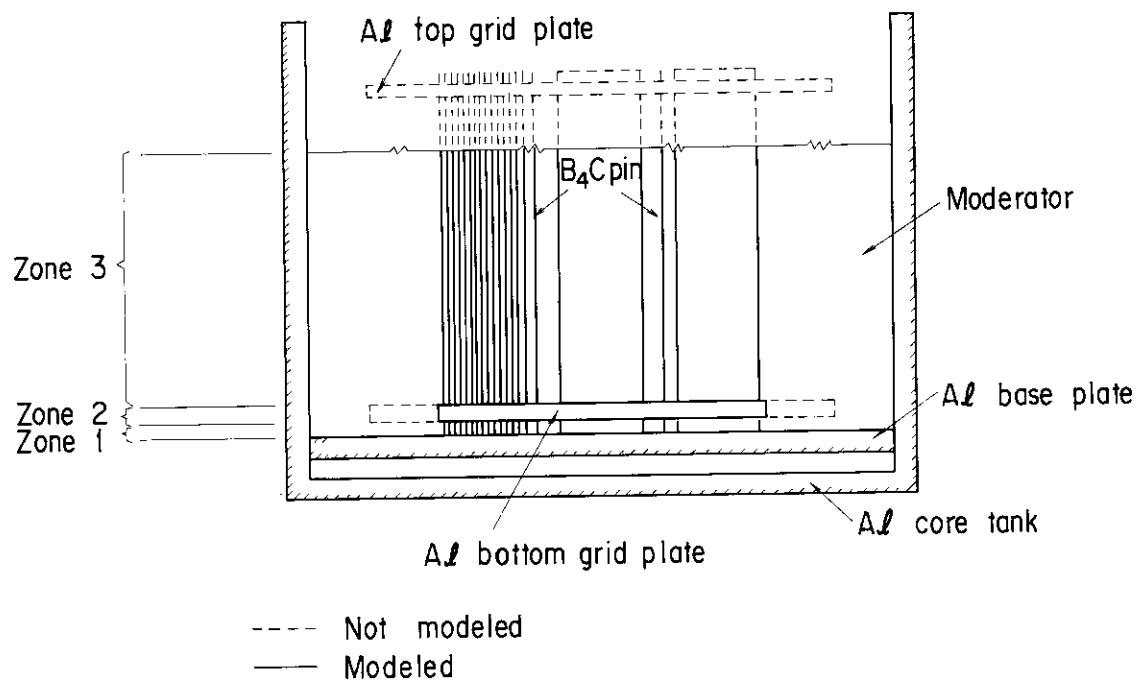
91.615	<sup>239</sup> Pu
7.654	<sup>240</sup> Pu
0.701	<sup>241</sup> Pu
0.031	<sup>242</sup> Pu

8. ANALYSIS DATE: JANUARY 1965
9. SEPARATIONS DATE: APRIL 1952
10. <sup>241</sup>Am CONTENT: NOT KNOWN

Fig. 3.C.9.3 UO<sub>2</sub>-2 wt% PuO<sub>2</sub> (8% <sup>240</sup>Pu) fuel description.



**Fig. 3.C.10.1** An example of core loading patterns, nine arrays separated by one pin pitch with 84 B<sub>4</sub>C pins.



**Fig. 3.C.10.2** KENO calculational model for Core I-IX.

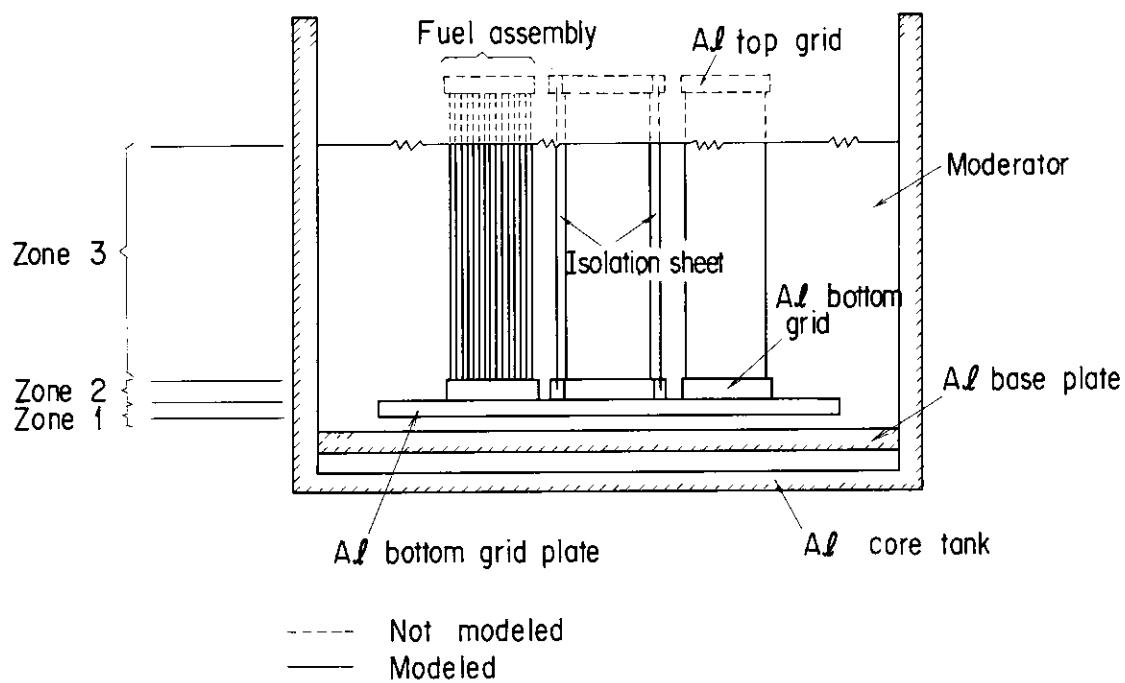


Fig. 3.C.10.3 KENO calculational model for Core X-XXI.

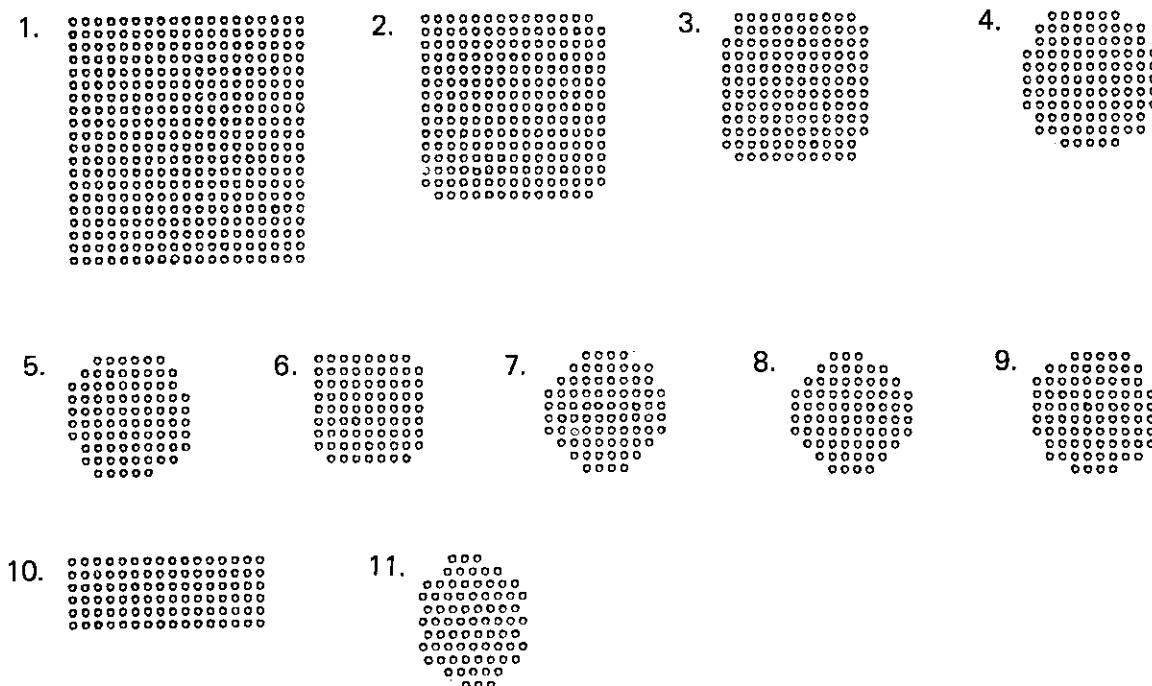


Fig. 3.C.11.1 Lattices of EBOR (Experimental Beryllium Oxide Reactor) fuel pins in water.

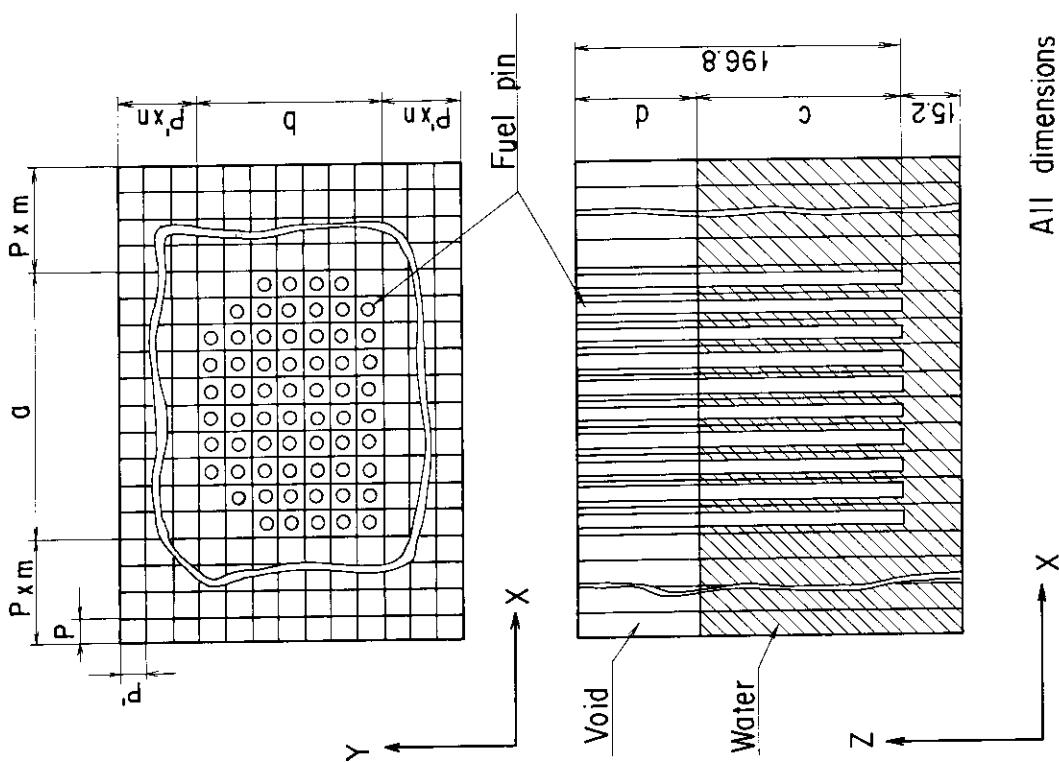


Fig. 3.C.11.3 Geometrical model of fuel pins in water.

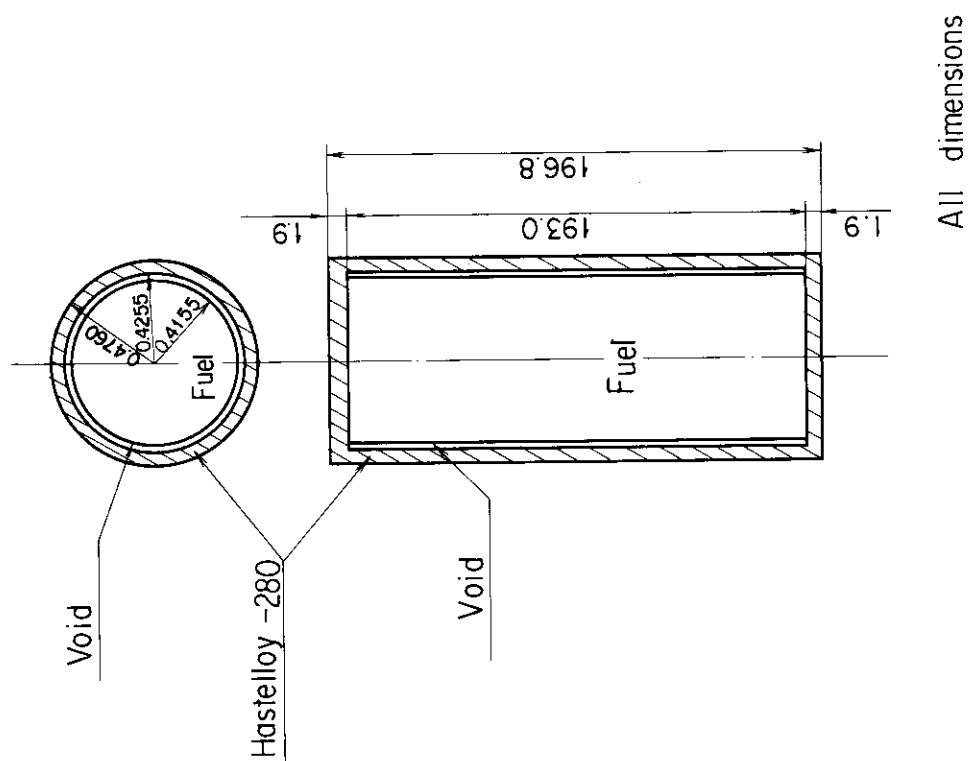
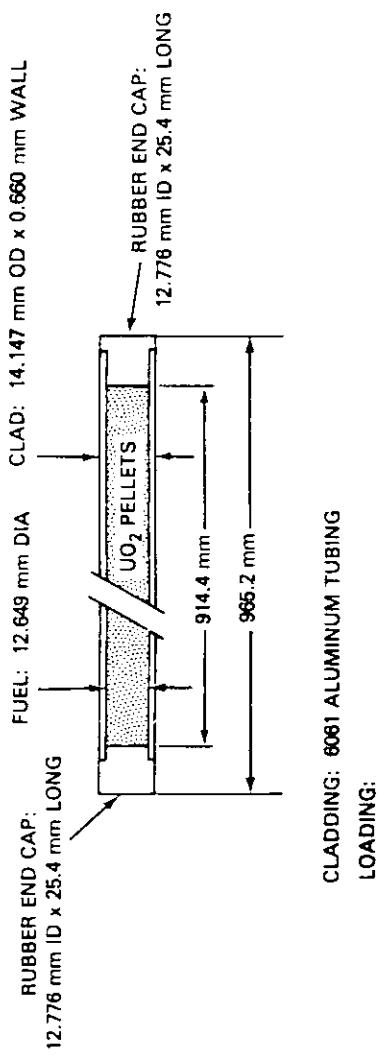


Fig. 3.C.11.2 Schema of the fuel pin model.



ENRICHMENT - 4.31 ± 0.01 WT% <sup>236</sup>U  
 FUEL DENSITY - 94.9 ± 0.66% OF THEORETICAL DENSITY  
 URANIUM ASSAY - 88.065 ± 0.281 WT% OF TOTAL FUEL COMPOSITION  
 $UO_2 \cdot 1203.38 \pm 4.12 \text{ g/ROD}$

END CAP:  
 DENSITY - 1.321 g/cm<sup>3</sup>  
 COMPOSITION - C-58 ± 1 WT%      Si-0.3 ± 0.1 WT%  
 H-6.5 ± 0.3 WT%  
 Ca-11.4 ± 1.8 WT%

Fig. 3.C.12.1 Schema of the fuel rod used in the experiment.

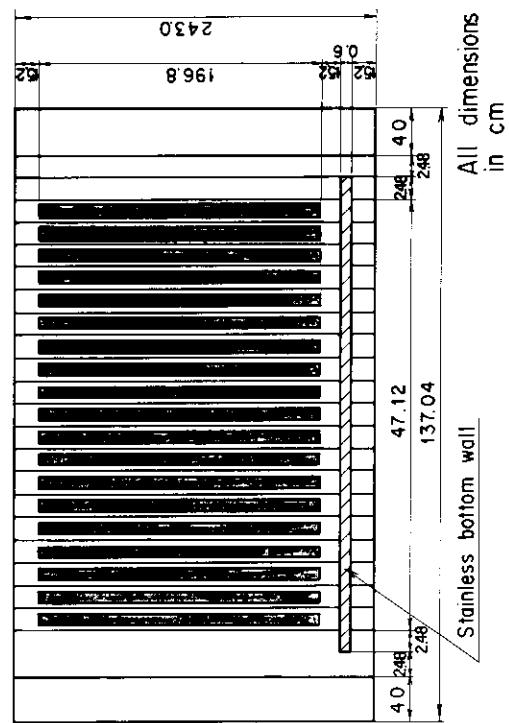
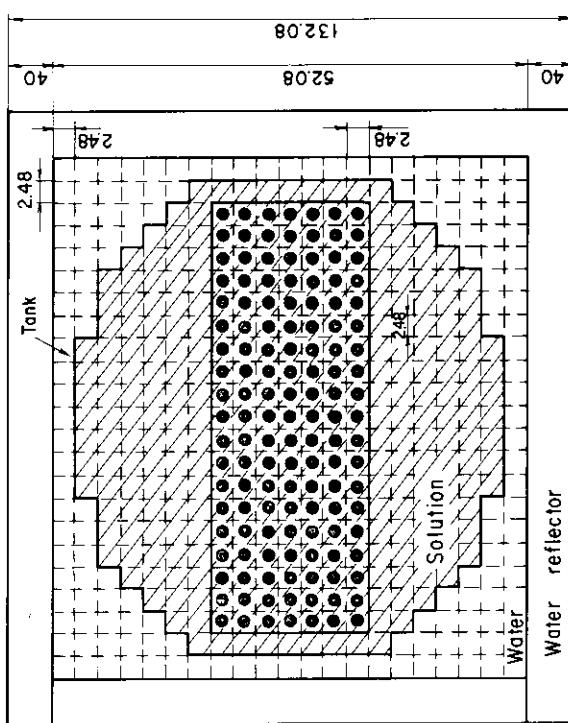


Fig. 3.C.11.4 Geometrical model of fuel pins in solution.

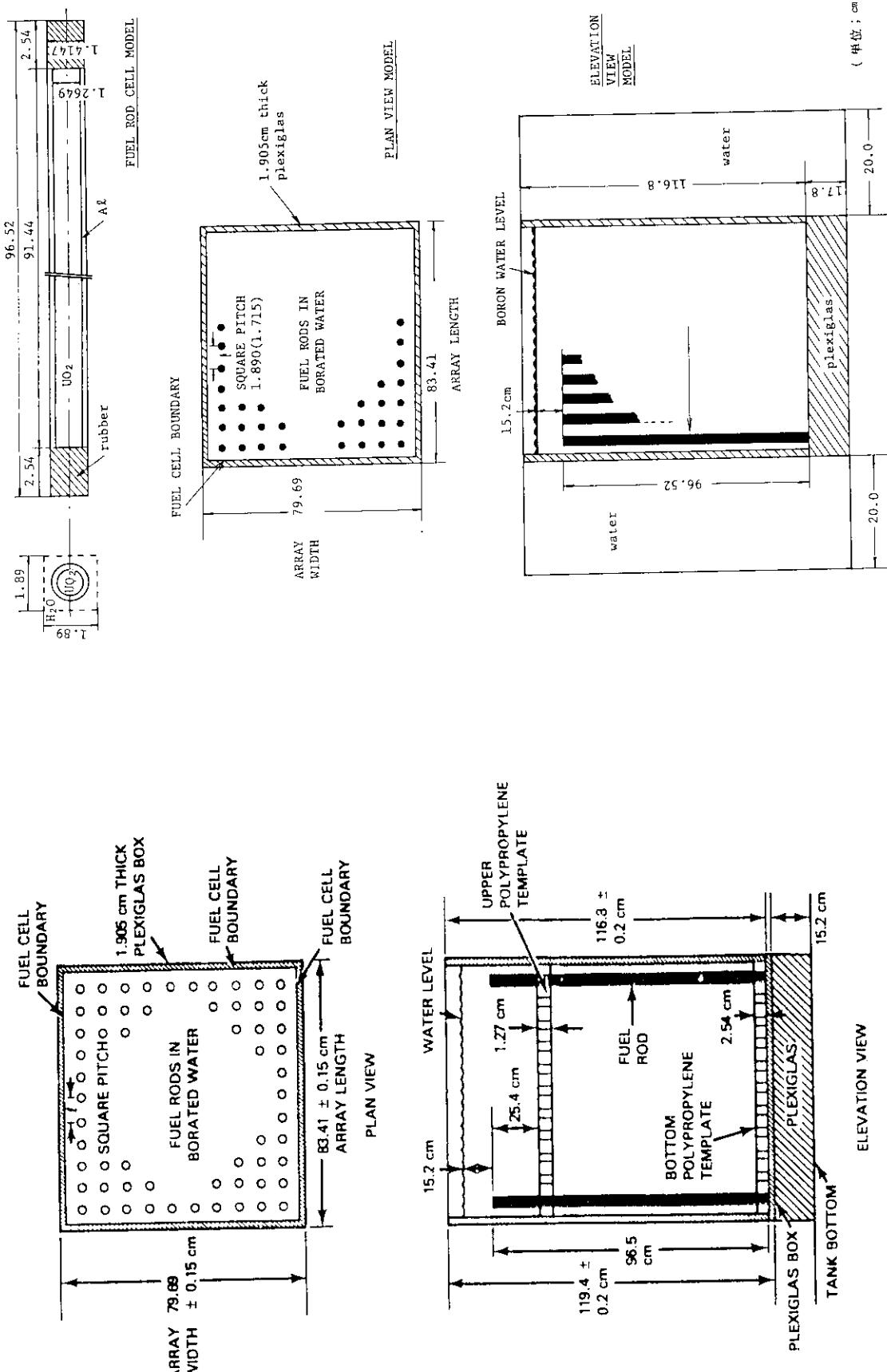
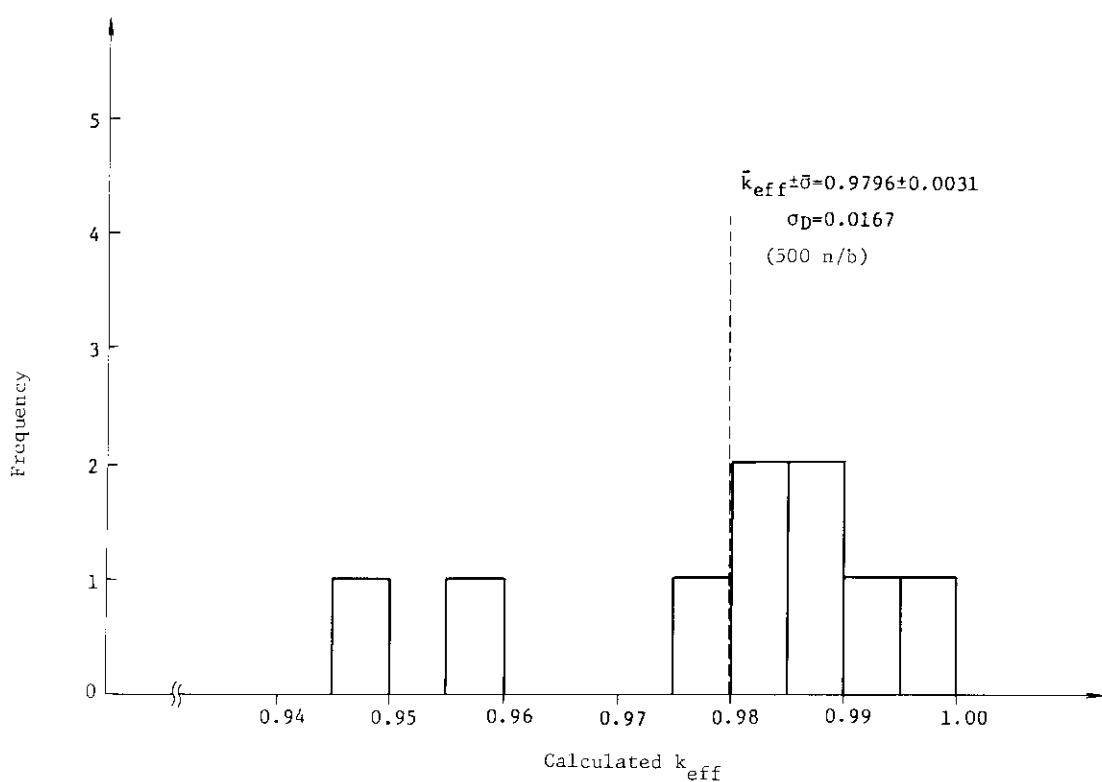
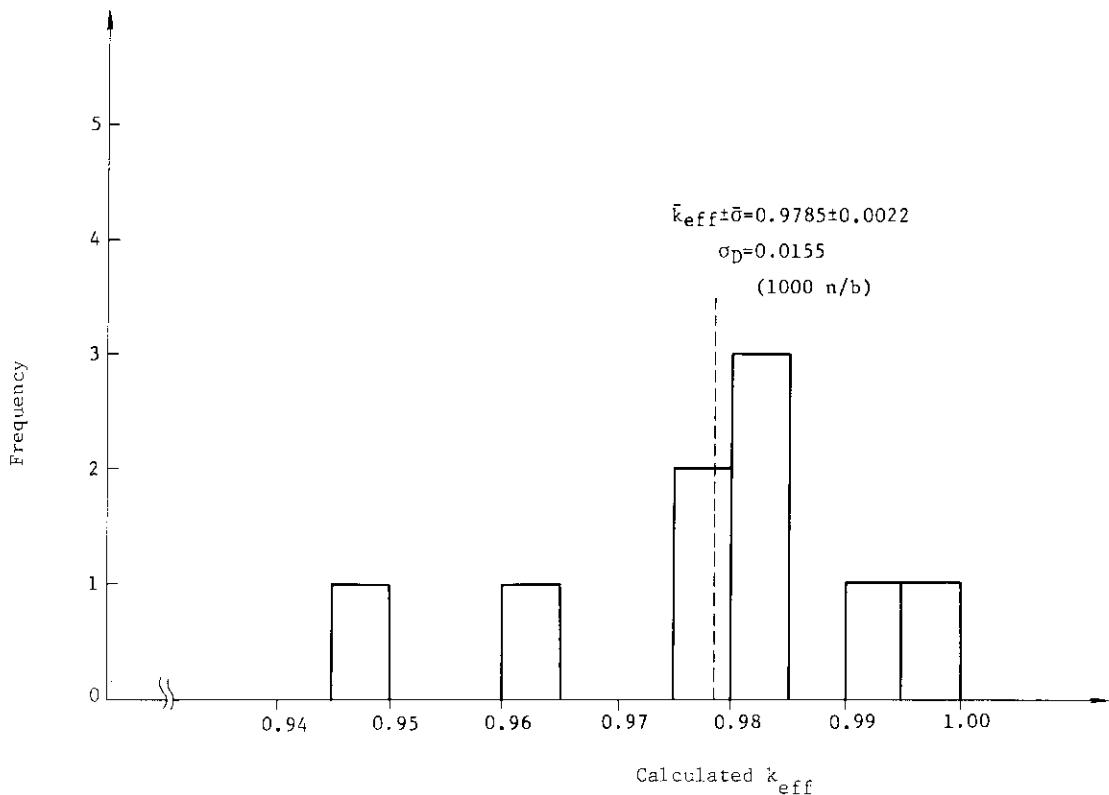


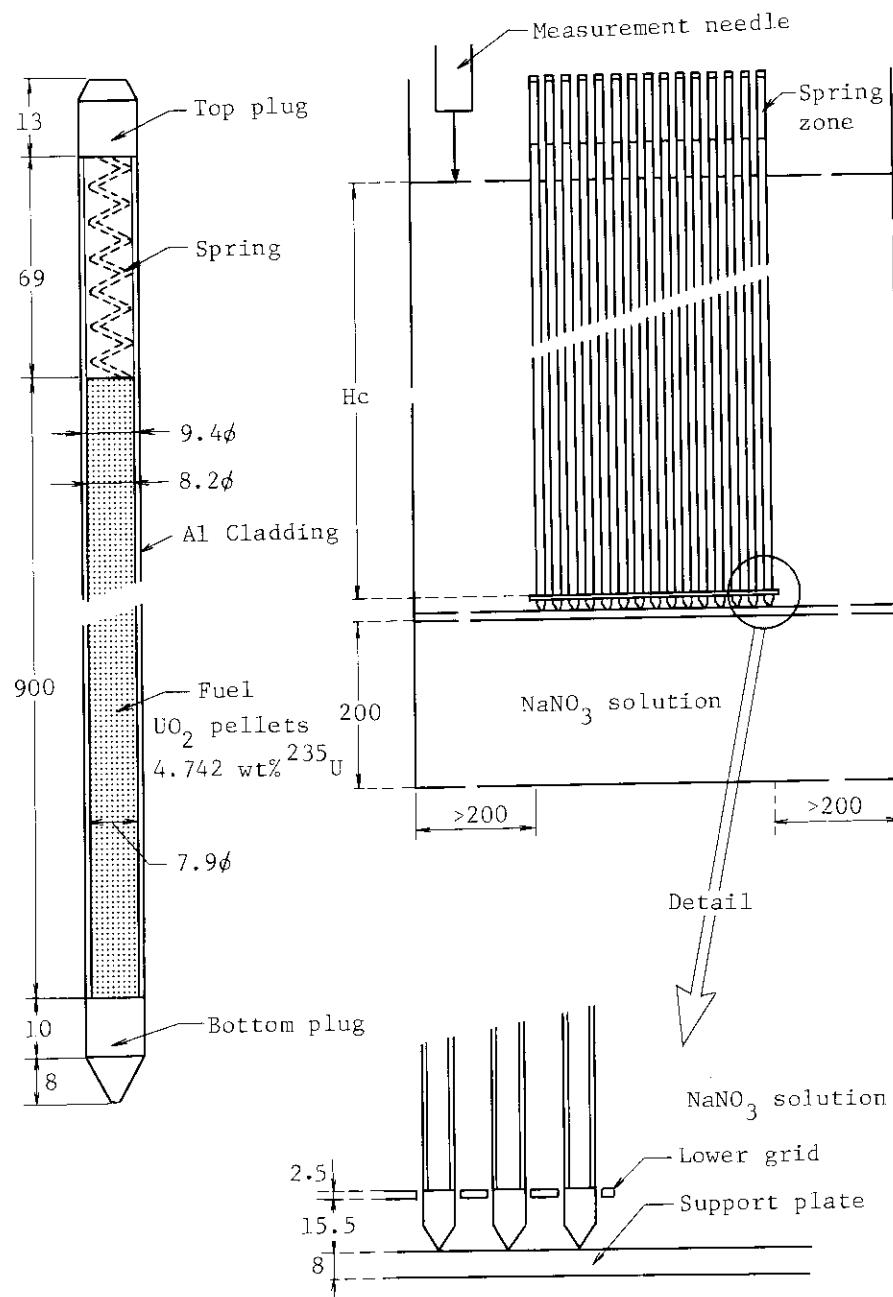
Fig. 3.C.12.2 Experimental arrangement of the  $\text{UO}_2$  rod clusters immersed in borated water.  
Fig. 3.C.12.3 Calculational model for  $\text{UO}_2$  rod cluster fully immersed in borated water.



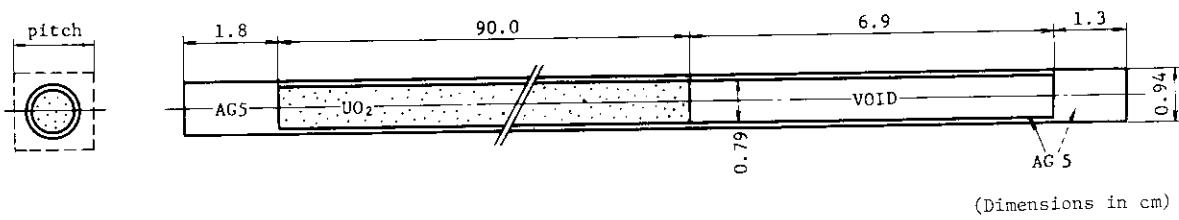
**Fig. 3.C.12.4** Histogram of calculated  $k_{eff}$ 's for the  $UO_2$  rod clusters fully immersed in borated water (1).



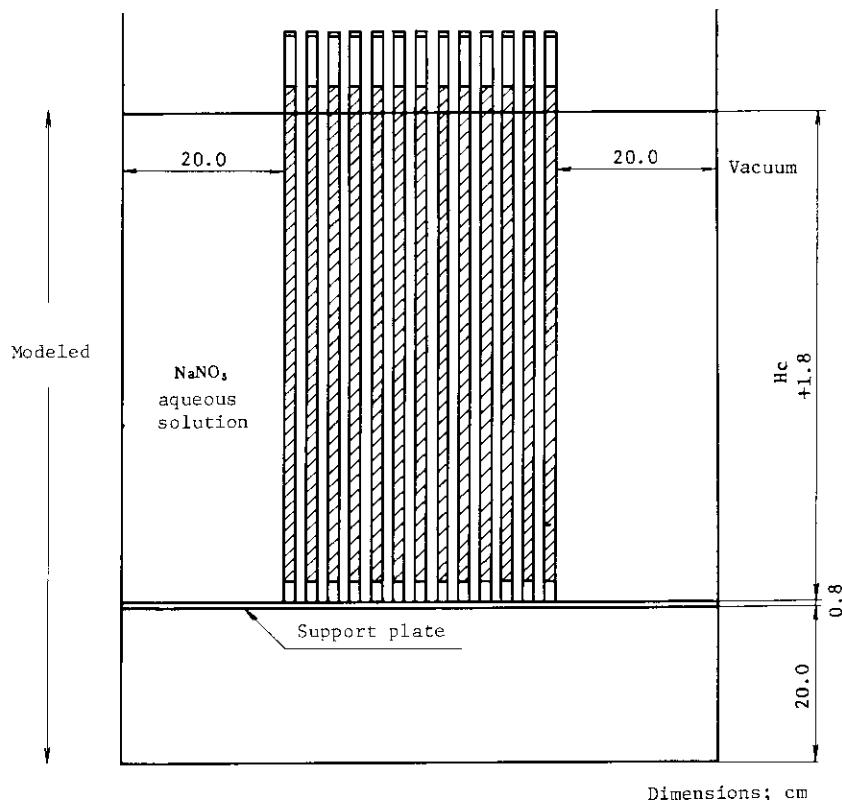
**Fig. 3.C.12.5** Histogram of calculated  $k_{eff}$ 's for the  $UO_2$  rod clusters fully immersed in borated water (2).



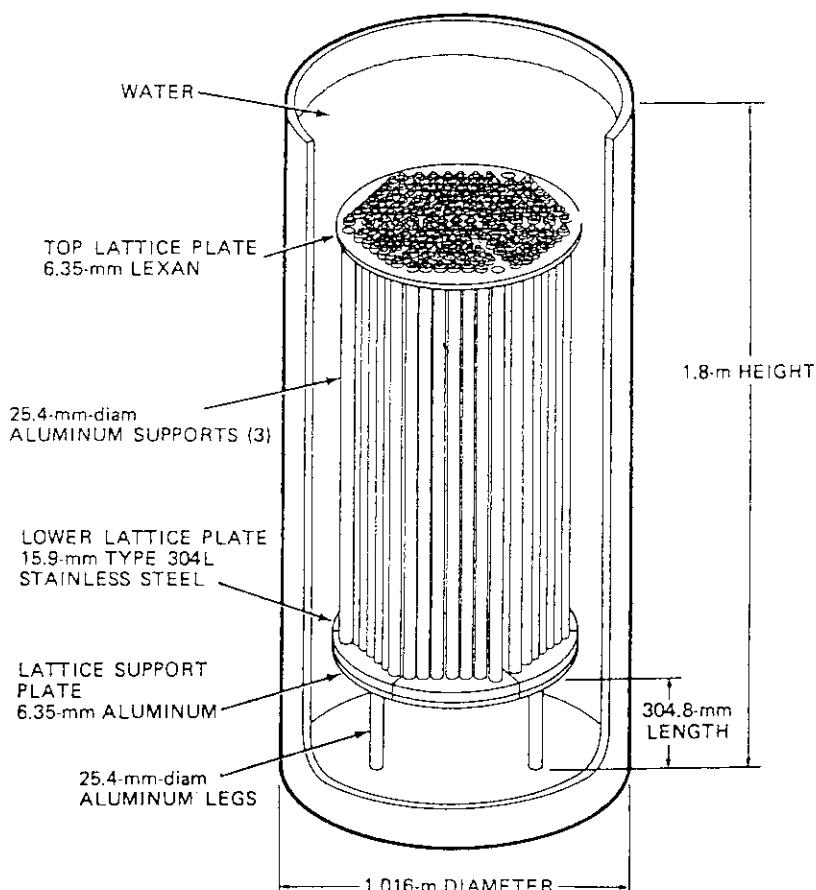
**Fig. 3.C.13.1** Main dimensional figures of the fuel rods and the experimental basket (in millimeters).



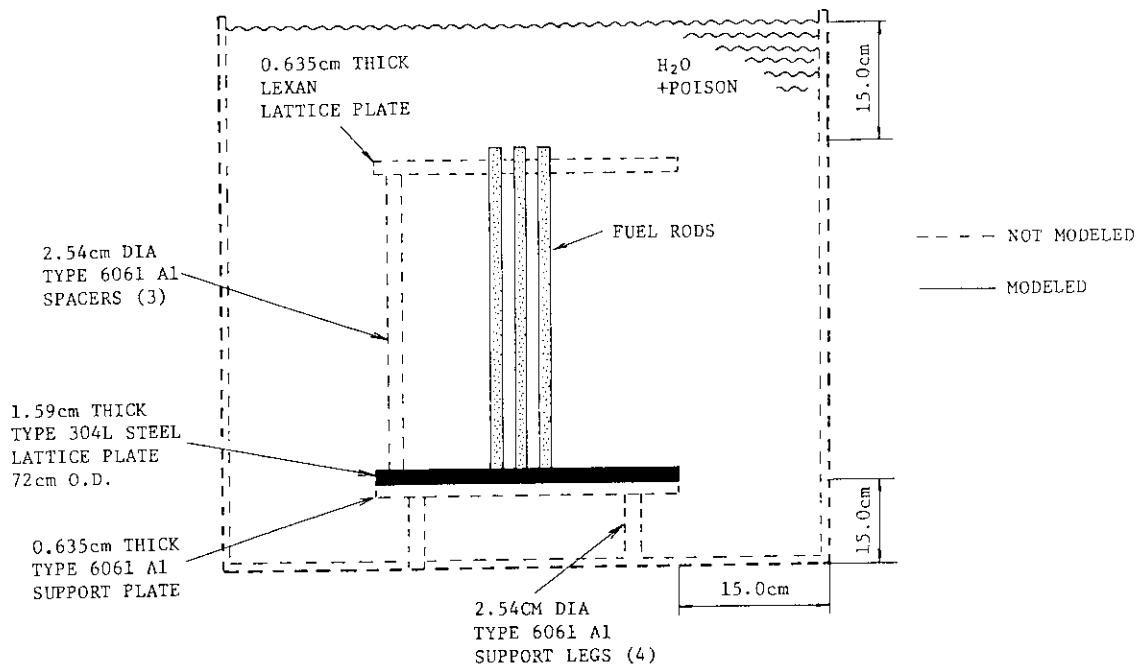
**Fig. 3.C.13.2** Calculational model for a fuel rod.



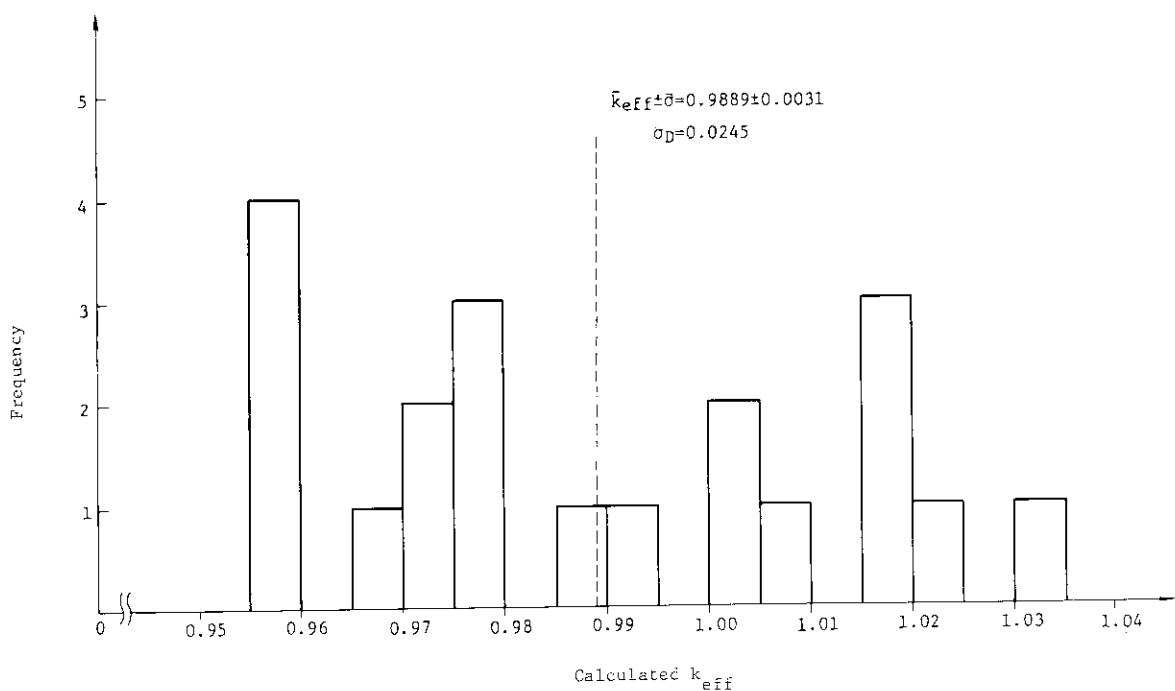
**Fig. 3.C.13.3** Calculational model for a typical experimental system.



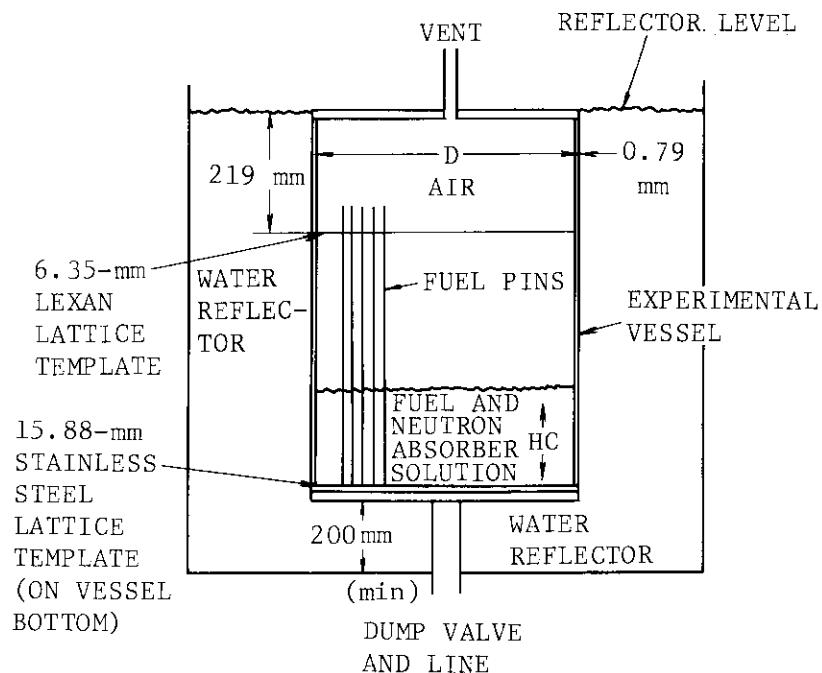
**Fig. 3.C.14.1** Experimental arrangement of the triangular lattice in poisonous water tank.



**Fig. 3.C.14.2** Calculational model for  $UO_2$  fuel rods, triangularly arranged in poisonous water tank.



**Fig. 3.C.14.3** Histogram of calculated  $k_{eff}$ 's for triangular lattice in poisonous water.



#### URANYL NITRATE COMPOSITION

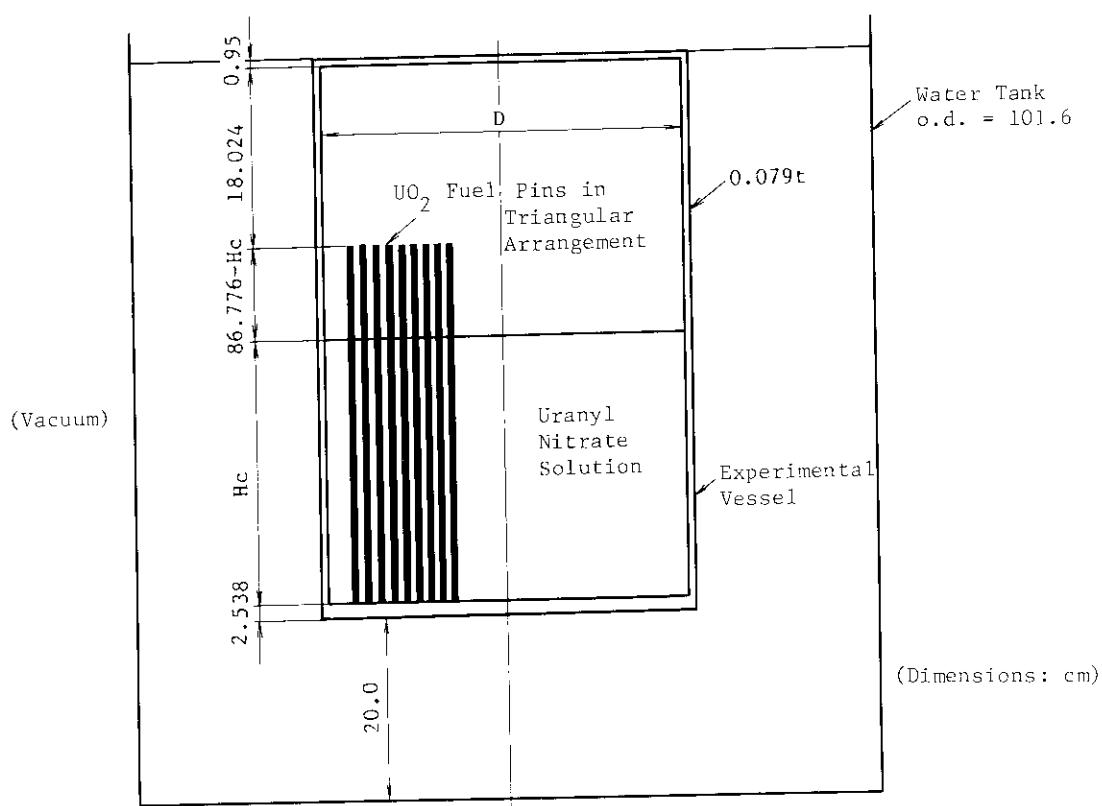
URANIUM CONCENTRATION (g/l)	195.8
SPECIFIC GRAVITY	1.254
FREE ACID MOLARITY	1.52

#### DESCRIPTION OF EXPERIMENTAL VESSELS

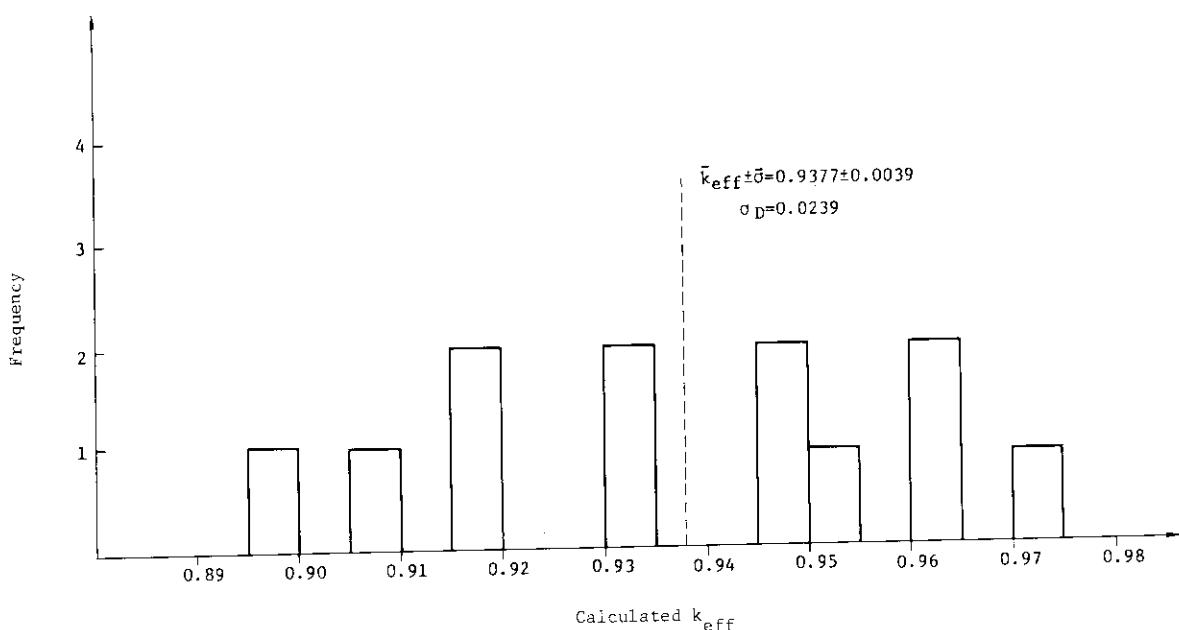
END THICKNESS (TOP AND BOTTOM) (mm)	9.5
EXPERIMENTAL VESSEL WALL THICKNESS (mm)	0.79
EXPERIMENTAL VESSEL HEIGHT (mm) (outside)	1067
WATER REFLECTOR o.d. (mm)	1016

INSIDE VESSEL TRIANGULAR DIAMETER, D (mm)	LATICE PITCH (mm)	NUMBER OF FUEL RODS
557.5	22.9	451
658.0	27.9	433
761.6	33.0	421

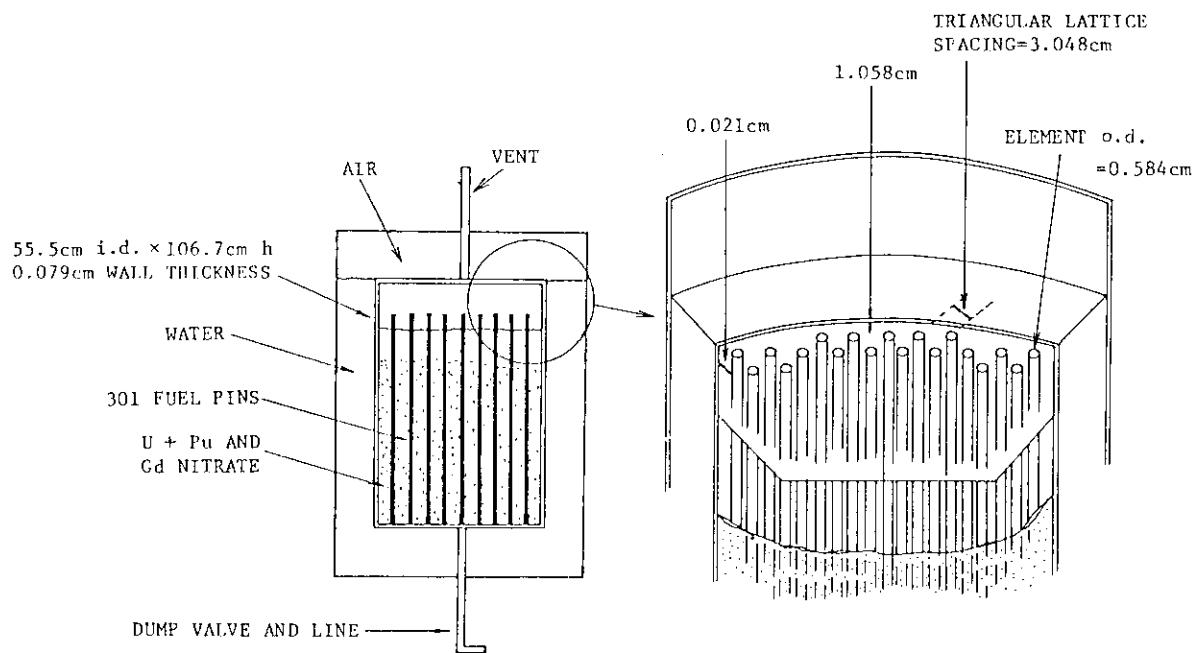
**Fig. 3.C.15.1** Experimental arrangement for  $\text{UO}_2$  rod clusters partially immersed in uranyl-nitrate solution.



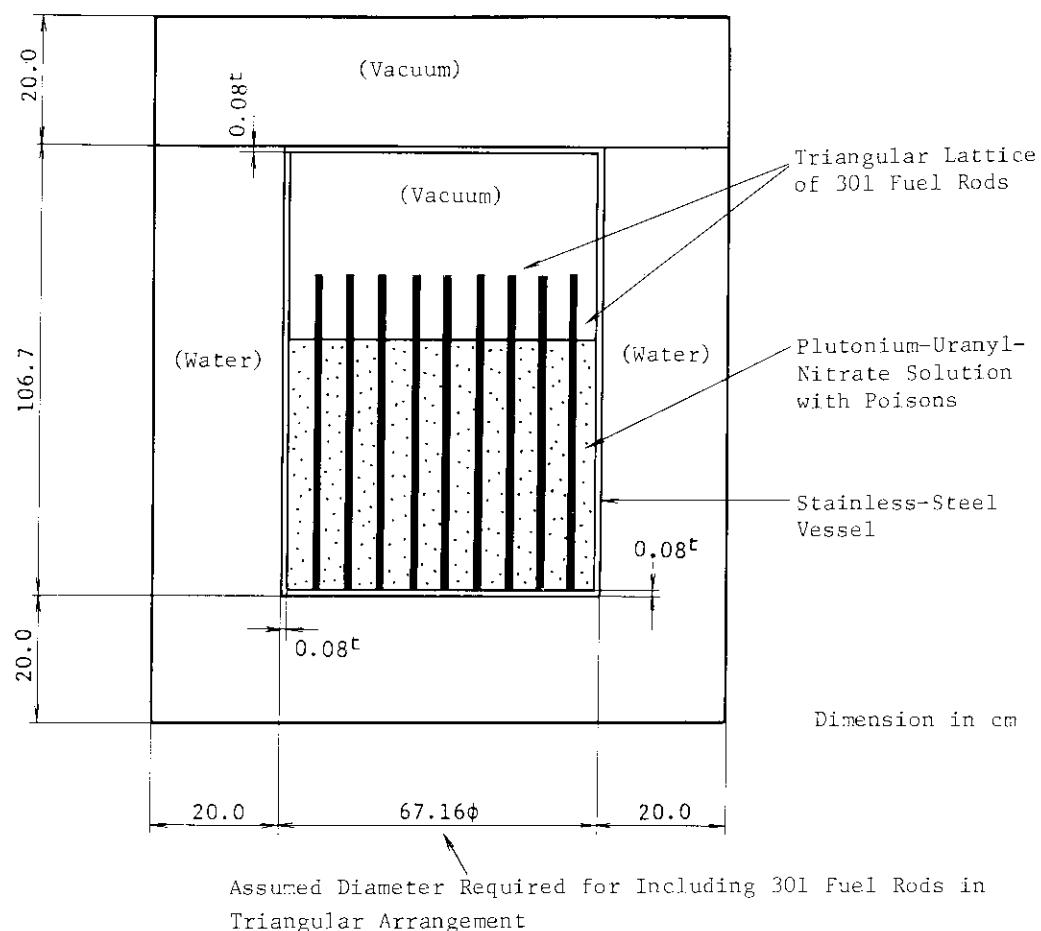
**Fig. 3.C.15.2** Calculational model for  $UO_2$  rod clusters partially immersed in uranyl-nitrate solution.



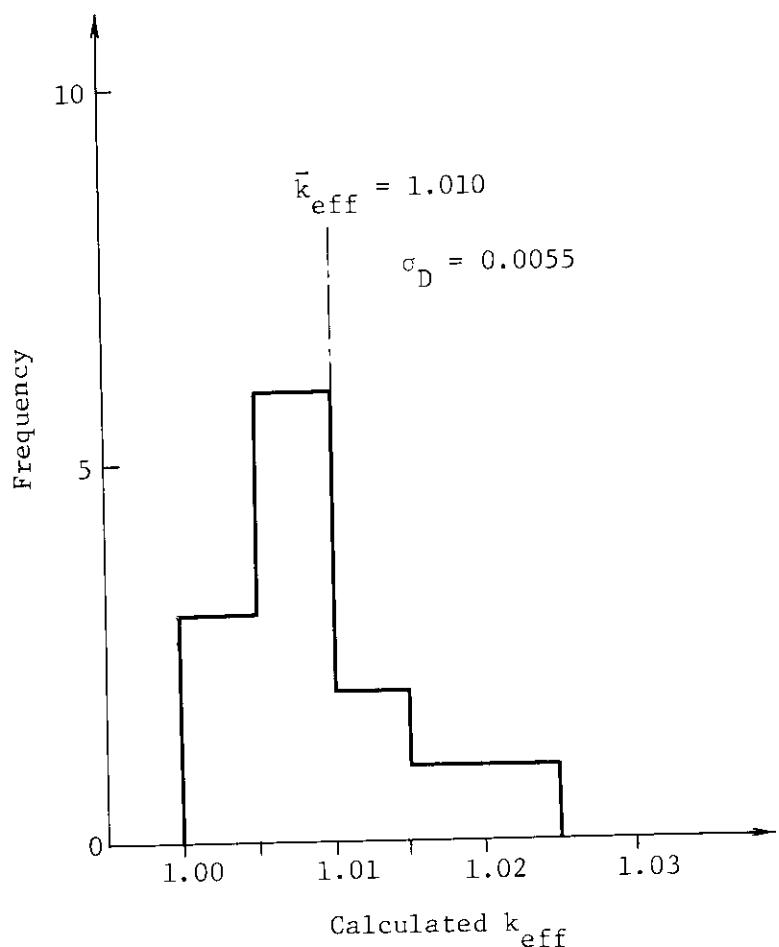
**Fig. 3.C.15.3** Histogram of 12 calculated  $k_{eff}$ 's for rod clusters partially immersed in uranyl-nitrate solution.



**Fig. 3.C.16.1** Experimental arrangement for  $\text{UO}_2\text{-PuO}_2$  fuel rods in fissile solution containing gadolinium and boron.



**Fig. 3.C.16.2** Calculational model for  $\text{PuO}_2\text{-UO}_2$  fuel rods in plutonium-uranyl-nitrate solution containing gadolinium and boron.



**Fig. 3.C.16.3** Histogram of calculated  $k_{eff}$ 's for  $PuO_2$ - $UO_2$  fuel rods in plutonium-uranyl-nitrate solution containing gadolinium and boron.

## 4. Discussions and Interpretations

As explained in Chapter 3, calculated  $k_{eff}$ 's have a wide range of bias errors in comparison with statistical errors predicted by the KENO-IV Monte Carlo calculations. There might be some factors determining the trends of the bias errors. Let us try to make clear the effects of these factors, referring to the experimental data and the calculational results described in Chapter 3. In the following sections, they are quoted as the corresponding section number, for example, (3.A.1) if the "Pile of PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene Compacts with Various Poison Plate Materials" is cited in the discussion.

### 4.1 Effect of Reflecting Condition

In most of homogeneous multi-unit (interacting) experimental systems such as (3.B.1), (3.B.2), (3.B.4) and (3.B.7), where both reflected and unreflected conditions are studied in the same experiment, the calculated  $k_{eff}$ 's for the reflected arrays are discerned to be higher than those for the unreflected arrays. One can see this trend in Figs. 3.B.1.2, 3.B.1.3 and 3.B.2.3, or in Tables 3.B.4.1 and 3.B.7.1. In the experiments (3.B.2) and (3.B.7), where the thickness of reflector is changed from case to case, the calculation predicts the higher  $k_{eff}$  value in the case of the thicker reflecting wall, as shown in Figs. 4.1 and 4.2. In these experimental systems most neutrons are considered to leak from fuel units. Consequently, interactions and reflections of the leaked neutrons are considered to play a dominant role in determining the reactivity in these experiments, since the calculated  $k_{eff}$ 's for other non-interacting systems such as homogeneous single-unit systems or heterogeneous systems don't have such a remarkable trend of bias errors. The KENO-IV Monte Carlo calculational method might have some problems in predicting the leaked neutron behavior, together with neutron interactions between units.

### 4.2 Effect of Moderating Condition

In under-moderated homogeneous single- or multi-unit systems, it is discerned that the lower the moderating ratio (H/(Pu or U)), the higher the calculated  $k_{eff}$ 's. For example, Fig. 3.A.6.11 shows the trend of the calculated  $k_{eff}$ 's with the variation of the H/Pu ratio of plutonium fuel of various shapes. To endorse this effect further, we can mention that the calculated  $k_{eff}$ 's for uranium metal arrays (3.B.5) and plutonium metal arrays (3.B.6) have averaged values of 1.012 and 0.999, respectively, as shown in Figs. 3.B.5.2 and 3.B.6.5, higher than those for plutonium or uranium solution unit arrays such as (3.B.2), (3.B.3), (3.B.4) and (3.B.7).

This trend in the calculated  $k_{eff}$ 's is also seen for heterogeneous systems. The calculated  $k_{eff}$ 's for UO<sub>2</sub> rod clusters in water at near-optimum moderation, such as in the system (3.C.3), have the average value of 0.990, as shown in Figs. 3.C.3.6 and 3.C.3.7, which is lower than that of 0.995 for UO<sub>2</sub> rod clusters in water at under-moderation, such as in the system (3.C.5).

On the other hand, in the case of over-moderating condition, the calculated  $k_{eff}$  bias seems to increase with increasing moderating ratio, as shown in Fig. 4.3. This trend is also discerned in homogeneous systems, for example, if Fig. 3.A.6.11 is observed carefully.

Variation of the moderating condition affects the neutron energy dominating nuclear reactions of which cross section data do not have uniform errors in a wide range of neutron energy. Accordingly, the bias errors are considered to be introduced into the calculated  $k_{eff}$ 's with their particular trends and uncertainties from case to case, depending on the variation of moderating condition.

### 4.3 Effect of Soluble Neutron Absorber

Neutron absorber, such as boron or gadolinium, absorbs mainly low energy neutrons and makes neutron energy spectrum harden. Accordingly, its existence is considered to have the reverse effect on the calculated  $k_{eff}$ 's as compared with the effect of moderating material. There are some experiments, where neutron absorbers are used to simulate a spent fuel storage or dissolver. **Figures 4.4 and 4.5** show that the calculated  $k_{eff}$  bias changes with boron and gadolinium concentrations in the plutonium-uranyl nitrate solution used in the experimental system (**3.A.16**). It is seen from these figures that the KENO-IV calculation predicts higher  $k_{eff}$  value for lower concentration of the neutron absorber. The similar bias trend in the calculated  $k_{eff}$ 's is also seen in the experiment (**3.C.14**), as shown in **Fig. 4.6**, where boron, cadmium, gadolinium or their combination is resolved in the water moderator around the  $UO_2$  rod clusters. In this figure, the Ft/At ratio (thermal neutron fissions versus thermal neutron absorptions, obtained by the KENO-IV calculation) is used instead of the neutron absorber concentration as an illustrative variable. The lower the neutron absorber concentration becomes, the softer the neutron spectrum and the smaller the Ft/At ratio becomes. Consequently, the reduction of the neutron absorber concentration in solution systems, considered to be at over-moderation, has the same effect as the increase of moderating ratio, introducing a positive trend in the calculated  $k_{eff}$ 's as discussed in the previous section.

On the other hand, a reverse trend is discerned in the calculated  $k_{eff}$  for the experiments (**3.C.8**) and (**3.C.10**), as shown in **Figs. 4.7 and 4.8**. In this case, neutrons are presumed to be at under-moderation. The reduction of the neutron absorber concentration results in the decrease of the calculated  $k_{eff}$  bias, as is also observed in the previous section.

### 4.4 Effect of Fuel Composition

Since numerical data of neutron cross sections of the fuel material play an important role in computing neutron multiplication factors, the error existing in the cross section data can be a direct cause to produce bias errors in the calculated  $k_{eff}$ 's.

Take the experiment (**3.A.2**) for an example, where the plutonium content in (Pu+U) mixed oxide fuel is varied with the H/Pu moderating ratio held approximately constant. The average value of the calculated  $k_{eff}$ 's for the reflected 30.0 wt% Pu content fuel listed in **Table 3.A.2.4**, is 1.0098, lower than 1.0128 for the reflected 7.89 wt% Pu content fuel in **Table 3.A.2.6**.

Furthermore, take the experiment (**3.A.13**) for another example, where the  $^{240}Pu$  content in Pu is varied in plutonium-uranyl-nitrate solution with a water reflector. As shown in **Figs. 3.A.13.4 and 3.A.13.5**, the average value of calculated  $k_{eff}$ 's for the 5.6 wt%  $^{240}Pu$  content fuel is 1.0012, higher than 0.9976 for the 23.0 wt%  $^{240}Pu$  content fuel.

These trends of the calculated  $k_{eff}$  bias are successfully predicted by the following multiple regression equation.<sup>67)</sup>

$$\begin{aligned}
 k_{\text{eff,exp}} = & +0.000167 - 0.09471 * (At/A - 0.8446) \\
 & + 0.3795 * (Pu/(Pu+U) - 0.3613) \\
 & - 1.1453 * ((Pu/(Pu+U))^2 - 0.3117) \\
 & + 0.7926 * ((Pu/(Pu+U))^3 - 0.3004) \\
 & - 0.09596 * ((\text{fissile})/(Pu+U) - 0.4606) \\
 & + 0.08482 * ((\text{fissile})/(Pu+U))^2 - 0.3739),
 \end{aligned}$$

$At/A$  : thermal neutron absorptions versus total neutron absorptions obtained by the KENO-IV calculation.

The above equation was induced from 243 cases of benchmark calculations on the homogeneous reflected single-unit systems selected from those in Subchapter 3.A.

Fuel compositions in nuclear fuel facilities cover so many varieties that a particular code system cannot predict neutron multiplication factors with sufficient accuracy for all these widely ranged fuel compositions. Therefore, it is considered to be advantageous for criticality safety evaluation to classify the experimental systems by indexing the representative fuel compositions and to estimate the critical values and their deviations calculated by the KENO-IV and MGCL code system according to the indexing. **Table 4.1** shows the results of the classification. In the case of homogeneous low enriched uranium system, for example, the critical  $k_{\text{eff}}$  value to be calculated by the KENO-IV and MGCL code system is 0.986, and the lower limit of the critical value is 0.965. Therefore, if  $k_{\text{eff}}$  value of a system to be evaluated is calculated by the KENO-IV and MGCL code system to be lower than 0.965, the system is concluded to be subcritical. These values listed in **Table 4.1** are obtained by processing statistically some 400 cases of benchmark calculations selected from those described in Chapter 3.

#### 4.5 Effect of Distance Between Units

The calculated  $k_{\text{eff}}$ 's for the homogeneous multi-unit (interacting) systems (3.B.6), (3.B.7) and (3.B.4) are shown in **Figs. 4.9, 4.10** and **4.11**, respectively, as a function of separation distance between units. The KENO-IV code tracks each neutron from birth to death. Since computing time restricts number of neutrons to be tracked, calculational difficulty increases rapidly with increasing the distance between units due to the lack of neutrons sufficient for the simulation of interaction between units.

**Figures 4.9** and **4.10** show that the further the units of array separate each other, the lower the calculated  $k_{\text{eff}}$ 's become. In the related experimental systems (3.B.6) and (3.B.7), the neutron leakage from each unit is significant, and leaked neutrons contribute much to the system reactivity through reflection and interaction. Let us remove the reflectors surrounding the units from the calculational model. Calculated  $k_{\text{eff}}$ 's would be low enough, 0.75 on an average for the system (3.B.6) and 0.63 for the system (3.B.7). For such a system, with increasing the separation distance between units, the difficulty in computing  $k_{\text{eff}}$  by the KENO-IV code becomes remarkable.

On the other hand, no specific trend of the calculated  $k_{\text{eff}}$  is discerned in **Fig. 4.11** for the experiment (3.B.4), where two aluminum cylinders containing uranyl fluoride solution are positioned parallel in a water tank. The cylinders are some 100 cm high and 20-40 cm in the inner diameter, big enough to have most of neutrons kept in each unit. The contribution of leaked neutrons to the system reactivity is minor. The calculation for the assumed unreflected two parallel cylinders would give an averaged  $k_{\text{eff}}$  value of 0.93, rather close to the averaged calculated  $k_{\text{eff}}$  of 0.98 for the real system. Therefore, the calculational difficulties

of simulation of interaction don't produce serious error in  $k_{eff}$  calculation for this experiment.

It is possible that this calculational difficulty might be a cause for the calculated  $k_{eff}$  bias trend pointed out in the section 4.1.

**Table 4.1** Validation results for reflected simple geometry systems with the (MGCL, KENO-IV) code system

Fuel	Index of Composition	Calculated $k_{eff}$ at Critical	Estimated Lower Limit	Number of Samples	Statistics St. Deviation Distribution
Homogeneous	Low-Enriched Uranium	0.986	0.965	40	0.008
	High-Enriched Uranium	0.985	0.954	68	0.013
	Plutonium	1.008	0.980	71	0.011
	Mixed Oxide ( $UO_2 + PuO_2$ )	1.013	0.980	45	0.008
	Mix. Solution $UO_2(NO_3)_2 + Pu(NO_3)_4$	1.010	0.980	10	0.008
Heterogeneous	Low-Enriched Uranium	0.995	0.978	88	0.007
	Plutonium	1.004	0.964	9	0.010
	Mixed Oxide ( $UO_2 + PuO_2$ )	0.997	0.980	50	0.007

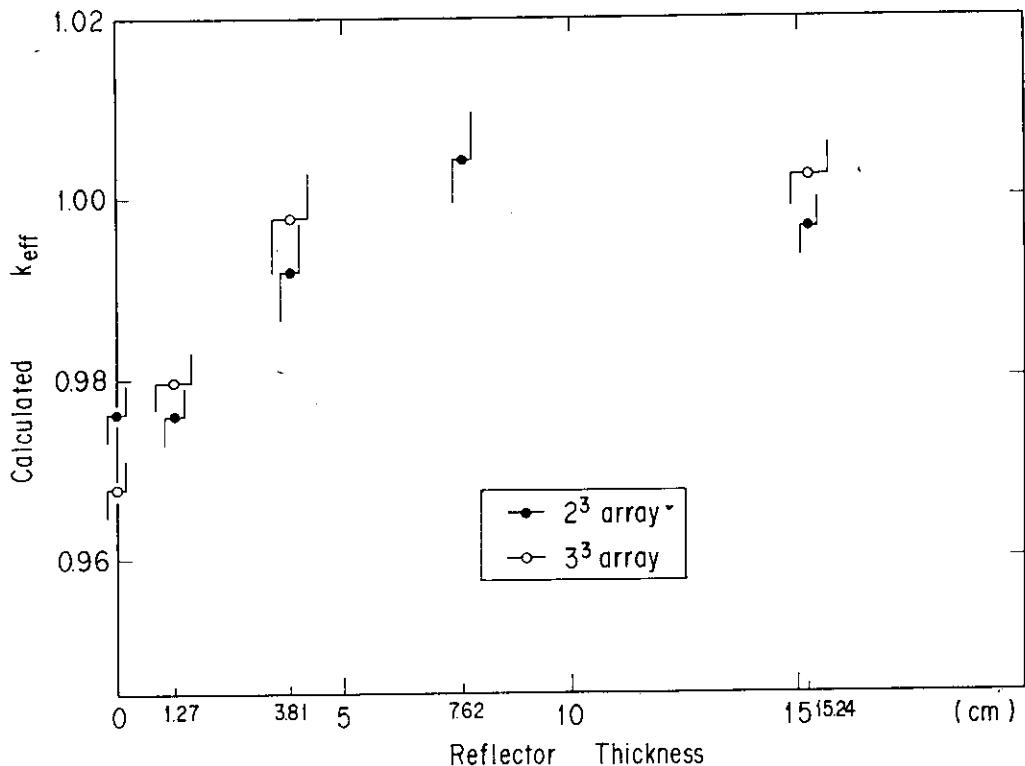


Fig. 4.2 Calculated  $k_{eff}$  as a function of reflector thickness surrounding the array of uranyl-nitrate solution cylinders.

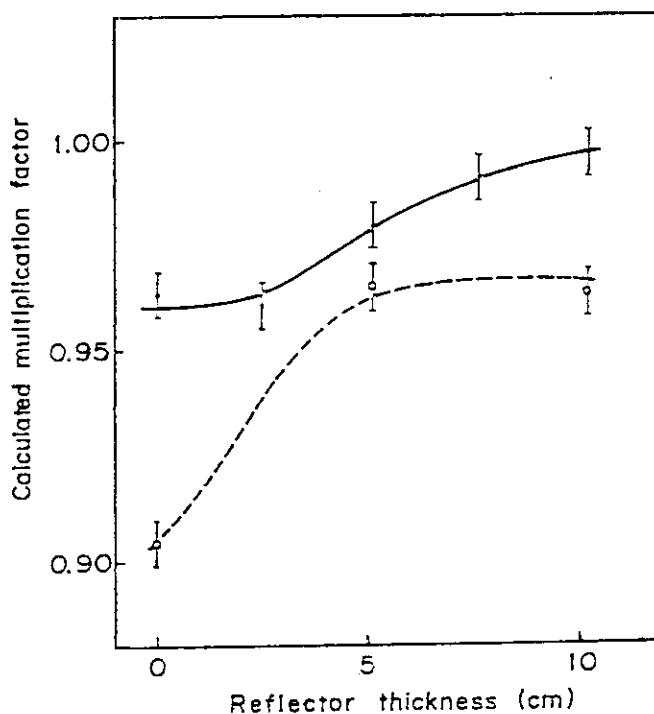
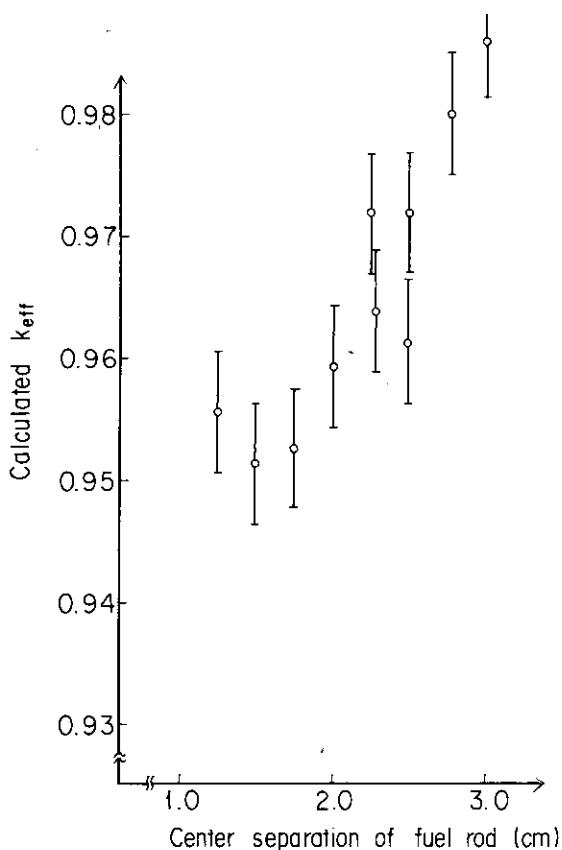
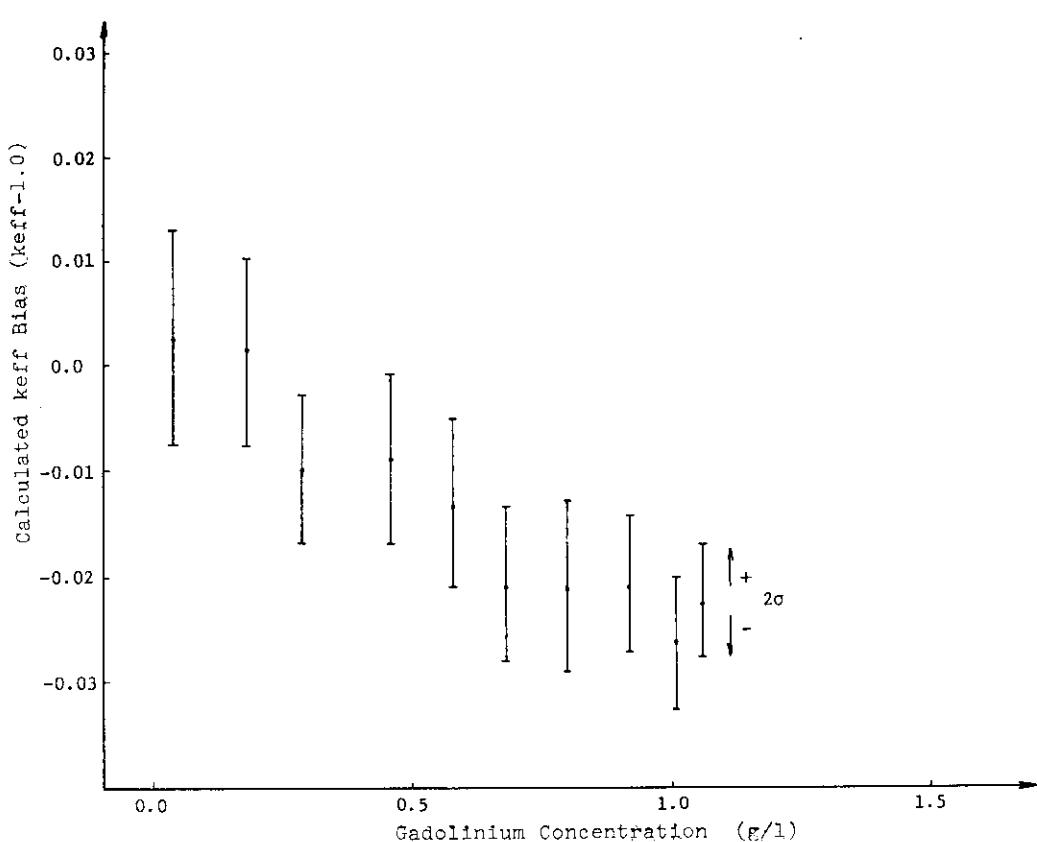


Fig. 4.1 Variation of calculated  $k_{eff}$  with reflector thickness. Closed circles indicate the change of  $k_{eff}$  with the top and side reflector thickness. Bottom reflector thickness is fixed to be 10.2 cm. Open circles indicate the change of  $k_{eff}$  with the bottom reflector thickness. There are no reflectors on the top and the side of the experimental apparatus.

**Fig. 4.3**

Variation of  $k_{eff}$  with center separation of fuel rods in water in the experiment (3.C.11).

**Fig. 4.4** Variation of calculated  $k_{eff}$  bias with gadolinium concentration in the experiment (3.A.16).

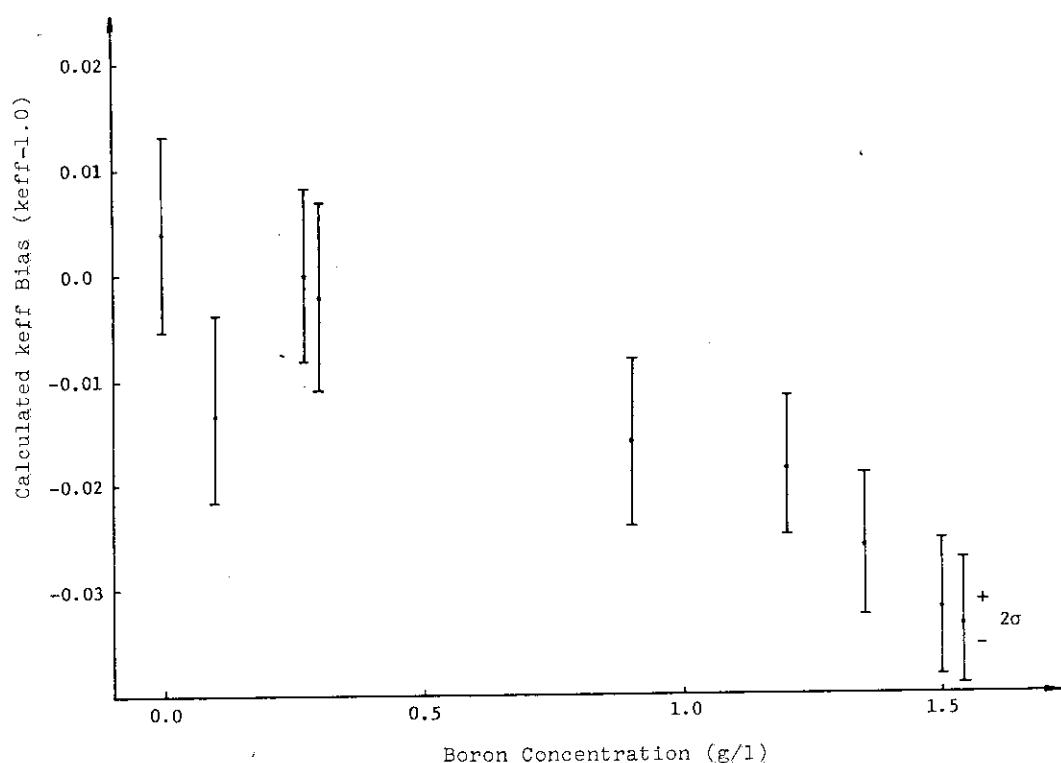


Fig. 4.5 Variation of calculated  $k_{eff}$  bias with boron concentration in the experiment (3.A.16).

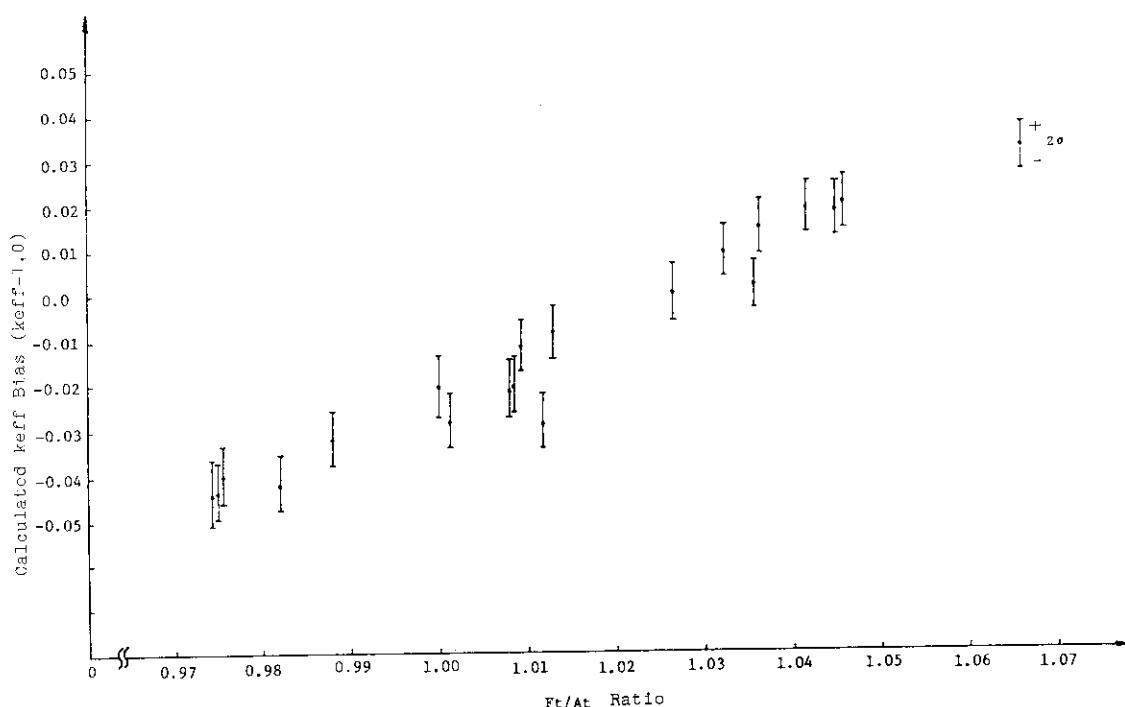
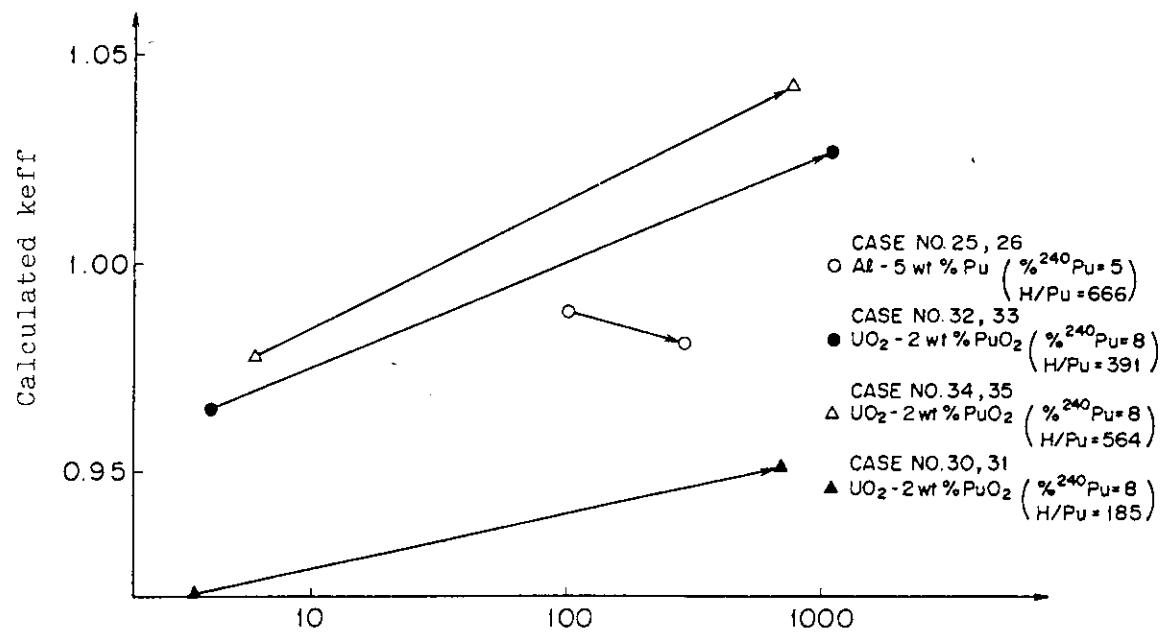
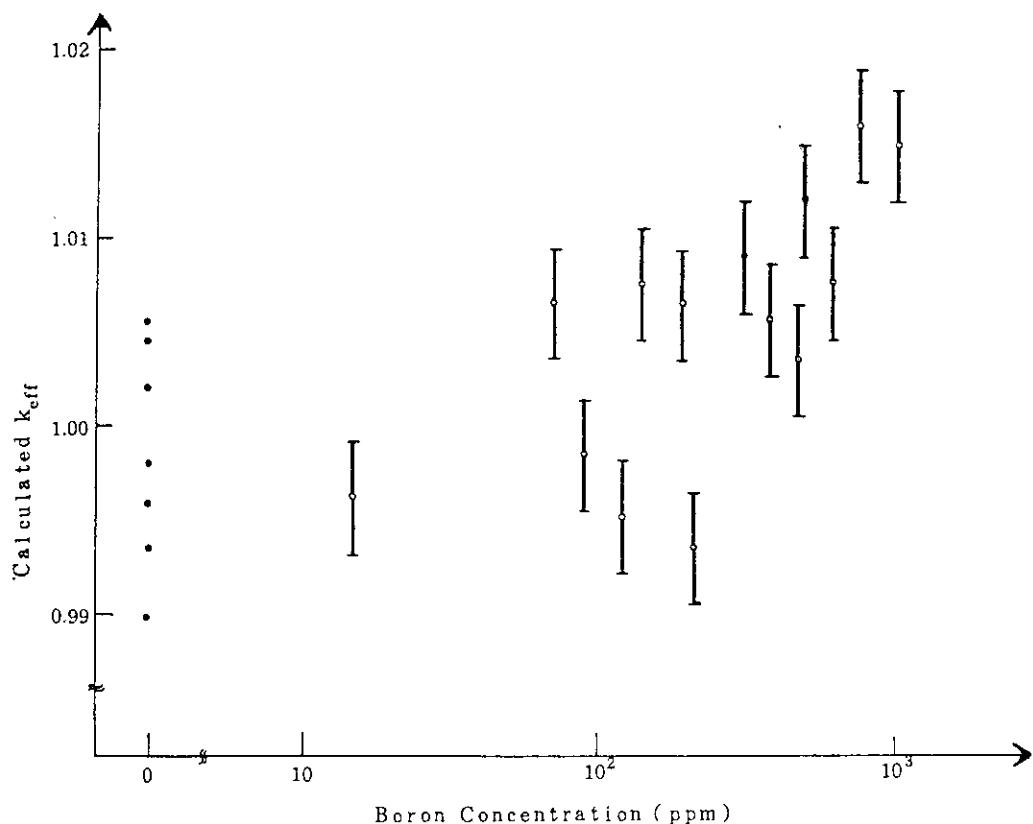


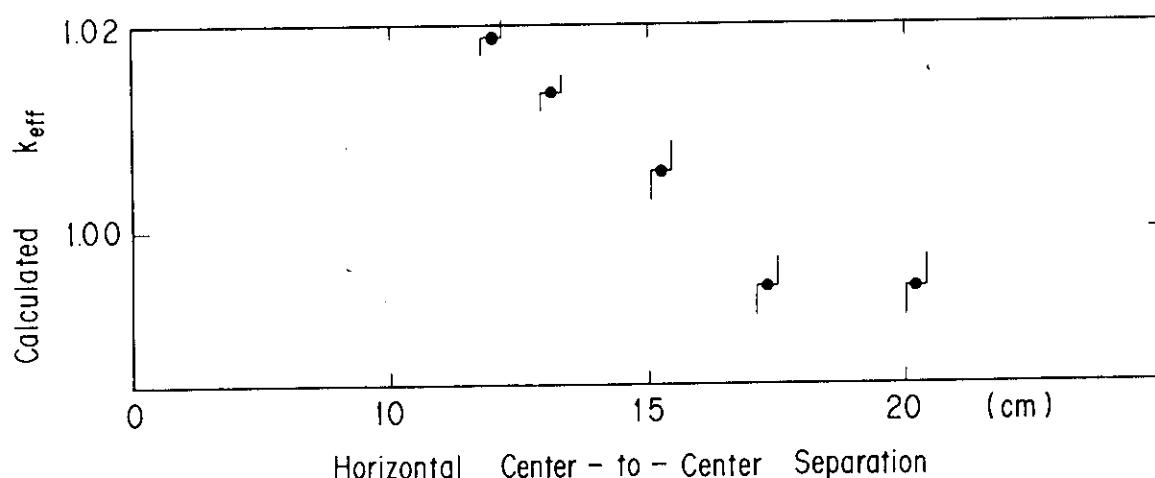
Fig. 4.6 Variation of calculated  $k_{eff}$  bias with Ft/At ratio in the experiment (3.C.14).



**Fig. 4.7** Variation of calculated  $k_{eff}$  with boron concentration in the experiment (3.C.8).



**Fig. 4.8** Variation of calculated  $k_{eff}$  with boron concentration in the experiment (3.C.10).

**Fig. 4.9** Calculated  $k_{eff}$  as a function of center-to-center separation

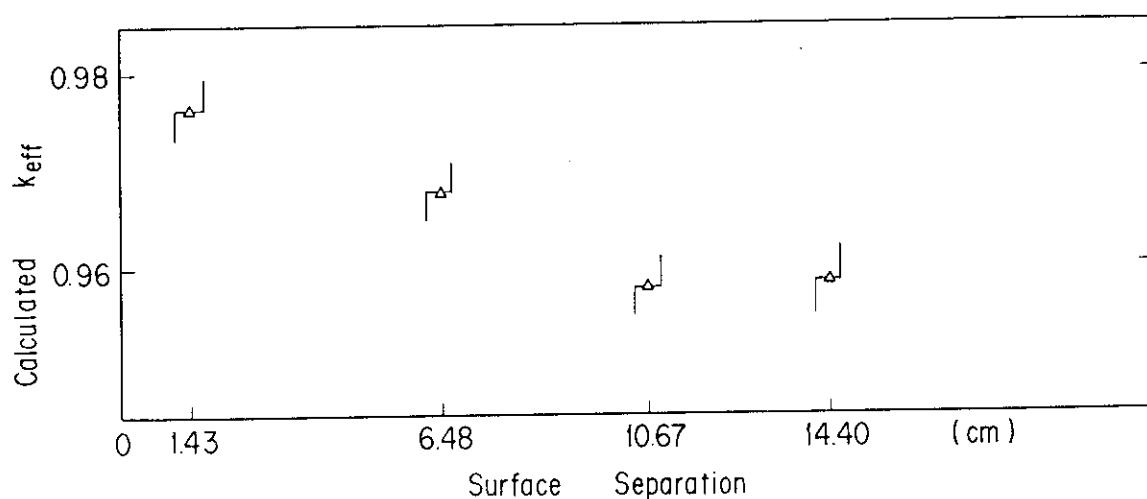
of plutonium metal cylinders in array disposition.

Array size :  $4^3$ 

Unit : 6 kg Pu billet

Moderator : None

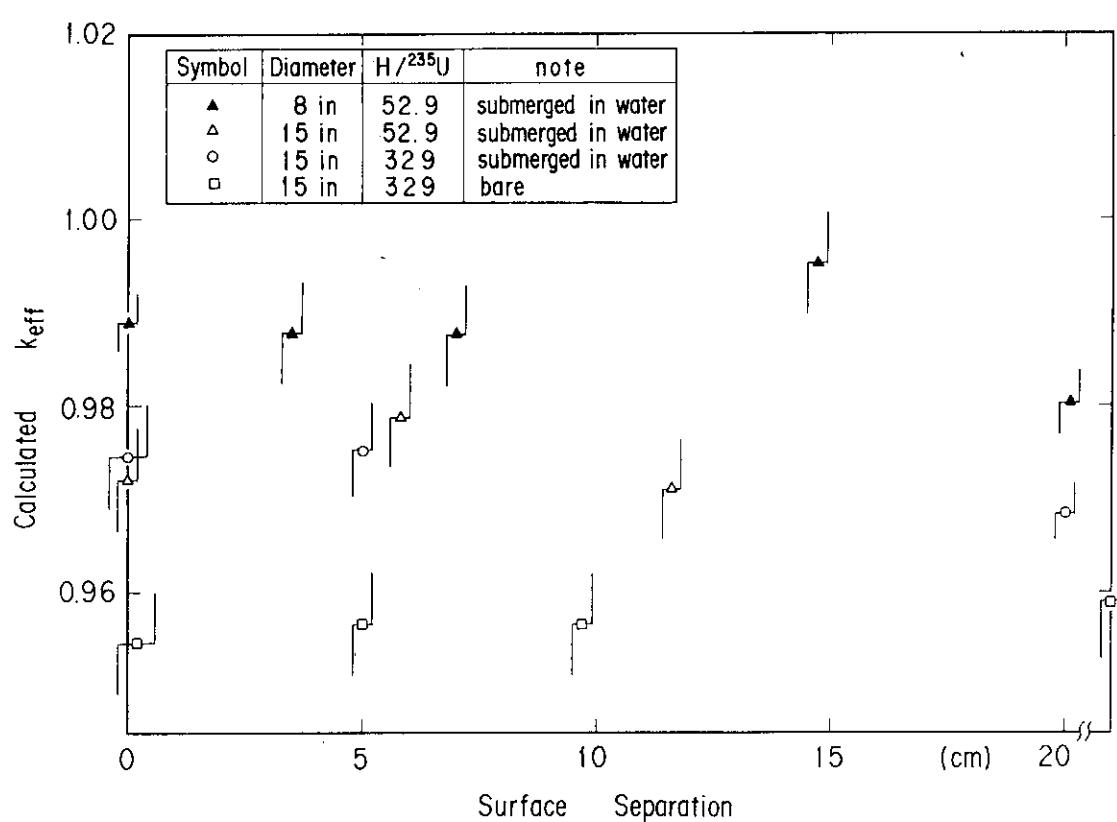
Reflector : None

**Fig. 4.10** Calculated  $k_{eff}$  as a function of surface separation of uranyl-nitrate solution cylinders in array disposition.Array size :  $2^3 - 5^3$ 

Unit : Uranyl-nitrate solution

Moderator : None

Reflector : None



**Fig. 4.11** Calculated  $k_{eff}$  v.s. surface separation of two aluminum cylinders containing uranyl-fluoride solution.

## 5. Conclusions

Brief descriptions have been presented on critical experiments and benchmark calculations by the aid of illustrative figures and tables. They render the presently available information on validation calculations concerning the JACS code system. Reported are 1394 cases in all, divided into 502 cases in 17 kinds of homogeneous single-unit systems, 331 cases in 8 kinds of homogeneous multi-unit systems and 561 cases in 16 kinds of heterogeneous systems.

Factors determining bias errors in the calculated  $k_{eff}$ 's are confirmed as follows:

- 1) reflecting conditions,
- 2) moderating conditions,
- 3) soluble neutron absorber,
- 4) fuel composition,
- 5) distance between units,

with their particular causes and effects.

As a result, in case of criticality analyses of nuclear fuel cycle facilities having these factors, it should be recommended that  $k_{eff}$ 's calculated by the JACS code system should be corrected for these bias errors.

## Acknowledgment

The authers would like to thank Dr. T. Shimooke at the Department of Reactor Safety Research, JAERI, for useful suggestions. They also thank members of working group on evaluation of the KENO-IV code in Nuclear Code Committee of Japan who cooperated in the calculation and evaluation work.

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## References

- 1) Katakura J., Naito Y. and Komuro Y. : Trans. Am. Nucl. Soc., **41**, 329 (1982).
- 2) Naito Y. et al. : "MGCL-Processor: A Computer Code System for Processing Multigroup Constants Library MGCL," JAERI-M 9396 (1981).
- 3) ENDF/B Summary Document BNL-NCS-17541 2nd ed. (1975).
- 4) Nakagawa T. ed. : "Summary of JENDL-2 General Purpose File," JAERI-M 84-103 (1984).
- 5) Bondarenko I.I. ed. : "Group Constants for Nuclear Reactor Calculations," Consultants Bureau, New York (1964).
- 6) Katsuragi S. et al. : "JAERI Fast Reactor Group Constants Systems Part I," JAERI 1195 (1970).
- 7) Wright R.Q. et al. : ORNL-TM-2679 (1969).
- 8) Macdougall J.D. : AEEW-M318 (1963).
- 9) Ozer O. : BNL-17134 (1974).
- 10) Petrie L.M. and Cross N.F. : ORNL-4938 (1975).
- 11) Naito Y. et al. : "MULTI-KENO: A Monte Carlo Code for Criticality Safety Analysis," JAERI-M 83-049 (1983).
- 12) Koyama K. et al. : "ANISN-JR: A One-Dimensional Discrete Ordinates Code for Neutron and Gamma-Ray Transport Calculations," JAERI-M 6954 (1977).
- 13) Phoades W.A. : CCC-276 (1977).
- 14) Fowler T.B. et al. : ORNL-TM 2496 (1970).
- 15) Bierman S.R., Durst B.M. and Clayton E.D. : BNWL-2129 (1976).
- 16) Komuro Y. et al. : "KENO-IV Code Benchmark Calculation (2) (A Pile of PuO<sub>2</sub>-UO<sub>2</sub>-Polystyrene Compact Fuel)," JAERI-M 9105 (1980) [in Japanese].
- 17) Bierman S.R. and Clayton E.D. : Nucl. Sci. Eng., **61**, 370 (1976).
- 18) Bierman S.R., Clayton E.D. and Hansen L.E. : Nucl. Sci. Eng., **50**, 115 (1973).
- 19) Lane R.C. and Perkins O.J.E. : AWRE O 32/68 (1986).
- 20) Richey C.R. et al. : Nucl. Sci. Eng., **23**, 150 (1965).
- 21) Bierman S.R. et al. : Nucl. Appl., **6**, 23 (1969).
- 22) Bierman S.R. and Clayton E.D. : Nucl. Technol., **11**, 185 (1971).
- 23) Rothe R.E. and Oh I. : Nucl. Technol., **41**, 207 (1978).
- 24) Nomura Y. et al. : "KENO-IV Code Benchmark Calculation (5) (Cylinders or a Tank Containing Uranyl Nitrate Solution)," JAERI-M 9108 (1980) [in Japanese].
- 25) Jenquin U.P. and Bierman S.R. : NUREG/CR-0210 (1978).
- 26) Nomura Y. et al. : "KENO-IV Code Benchmark Calculation (6) (Plutonium Fuel in Various Shape)," JAERI-M 9201 (1980) [in Japanese].
- 27) Johnson E.B. : Y-DR-129 (1970).
- 28) Nomura Y. et al. : "KENO-IV Code Benchmark Calculation (8) (Intersecting Cylinders of Aqueous Uranyl Fluoride Solutions)," JAERI-M 9085 (1980) [in Japanese].
- 29) Johnson E.B. and Cronin D.F. : Neutron Physics Division Annual Progress Report, ORNL-3714, p.31 (1964).
- 30) Johnson E.B. and Newlon C.E. : Neutron Physics Division Annual Progress Report, ORNL-4280, p.47 (1968).
- 31) Raffety S.J. and Mihalczo J.R. : Y-DR-14 (1969).
- 32) Lloyd R.C. et al. : Nucl. Sci. Eng., **25**, 165 (1966).
- 33) Colomb G., Mangin D. and Maubert L. : CEA-N-1898 (1973).
- 34) Lloyd R.C. and Clayton E.D. : Nucl. Sci. Eng., **60**, 143 (1976).
- 35) Tuck G. and Oh I. : NUREG/CR-0674 (1979).
- 36) Katakura J. et al. : "Benchmark Calculation with the Nuclear Criticality Safety Evaluation Code System JACS," JAERI-M 9859 (1981) [in Japanese].
- 37) Schmid L.C. : BNWL-1522-3 (1971).
- 38) Lloyd R.C. and Clayton E.D. : BNWL-B-482 (1976).
- 39) Thomas J.T. : Y-DR-128 (1974).
- 40) Katakura J. et al. : "KENO-IV Code Benchmark Calculation (3) (UF<sub>6</sub> Container Cylinder Critical Array)," JAERI-M 9025 (1980) [in Japanese].
- 41) Tuck G. and Clark H.E. : Nucl. Sci. Eng., **40**, 407 (1970).
- 42) Katakura J. et al. : "KENO-IV Code Benchmark Calculation (4) (Slab-Cylinder Critical Configuration of Uranium Nitrate Solution)," JAERI-M 9026 (1980) [in Japanese].

- 43) Callihan D. et al. : K-406 (1949).
- 44) Nomura Y. et al. : "Benchmark Calculations for Neutron-Interacting Systems by the Criticality Code-System JACS," JAERI-M 86-082 (1986) [in Japanese].
- 45) Thomas J.T. : ORNL-TM 868 (1964).
- 46) Morton III J.R. et al. : UCRL-50175 (1966).
- 47) Schuske C.L. and Paxton H.C. : Nucl. Technol., **30**, 101 (1975).
- 48) Magnuson D.W. : Y-DR-120 (1973).
- 49) Tsuruta H. et al. : "Critical Sizes of Light Water Moderated UO<sub>2</sub> and PuO<sub>2</sub>-UO<sub>2</sub> Lattices," JAERI 1254 (1977).
- 50) Komuro Y. et al. : "KENO-IV Code Benchmark Calculation (10) (A Light Water Reactor Critical Assembly)," JAERI-M 9147 (1980) [in Japanese].
- 51) Manaranche J.C. et al. : Nucl. Sci. Eng., **71**, 154 (1979).
- 52) Bierman S.R., Clayton E.D. and Durst B.M. : PNL-2438 (1977).
- 53) Bierman S.R. et al. : NUREG/CR-0073 (1978).
- 54) Nomura Y. et al. : "KENO-IV Code Benchmark Calculation (9) (Three Clusters of UO<sub>2</sub> Rods Lattice)," JAERI-M 9168 (1980) [in Japanese].
- 55) Bierman S.R. et al. : NUREG/CR-0796 vol.1 (1979).
- 56) Bierman S.R. and Clayton E.D. : NUREG/CR-1547 (1980).
- 57) Bierman S.R., Durst B.M. and Clayton E.D. : NUREG/CR-0796 vol.2 (1981).
- 58) Bierman S.R. and Clayton E.D. : NUREG/CR-1784 (1981).
- 59) Nomura Y. et al. : "KENO-IV Code Benchmark Calculation (7) (Water Moderated Triangle Lattices of Plutonium Enriched Rods)," JAERI-M 9079 (1980) [in Japanese].
- 60) Smith R.I. and Konzek G.J. : EPRI NP-196 (1976).
- 61) Baldwin M.N. et al. : BAW-1484-7 (1979).
- 62) Johnson E.B. : ORNL-ENG 2 (1976).
- 63) Durst B.M., Bierman S.R. and Clayton E.D. : NUREG/CR-2709 (1982).
- 64) Manaranche J.C. et al. : Nucl. Technol., **50**, 148 (1980).
- 65) Lloyd R.C., Durst B.M. and Clayton E.D. : Nucl. Sci. Eng., **71**, 164 (1979).
- 66) Lloyd R.C. et al. : Nucl. Sci. Eng., **78**, 121 (1981).
- 67) Nomura Y. and Shimooke T. : Nucl. Technol., **65**, 340 (1984).